SYNTHETIC STUDIES ON β -LACTAM ANTIBIOTICS. PART 5. A SYNTHESIS OF 7β -ACYLAMINO-3-METHYL-1-OXADETHIA-3-CEPHEM-4-CARBOXYLIC ACIDS⁺

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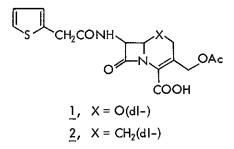
Chloroazetidinone <u>10</u>, prepared from 6-APA, was etherified with propargyl alcohol and zinc chloride to a 2:1 mixture of <u>cis</u>- and <u>trans</u>-ethers <u>18</u>. The separated <u>cis</u>-ether <u>18</u> was converted into a keto ylide <u>21d</u>, which was cyclized by intramolecular Wittig reaction to 3-methyl 1-oxacephalosporin <u>22</u>. From this compound, four optically active 3-methyl oxacephalosporins <u>25a-d</u> were prepared. Interestingly all of them except <u>25d</u>, 1-oxacephalexin, exhibited antibacterial activity as four to eight times high as that of the corresponding cephalosporins.

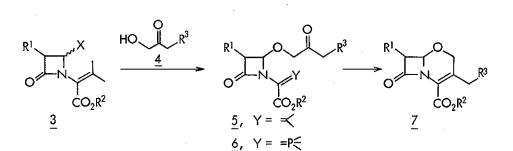
⁺ Dedicated to Professor Robert B. Woodward on his sixtieth birthday. The reports that (\pm) -l-oxacephalothin $\underline{l}^{la,c}$ and (\pm) carbacephalothin $\underline{2}^{lb,c}$ possessed half a potency in antibacterial activity as that of cephalothin have evoked our interest in connection with the question of the extent of the activity of optically active l-oxacephalosporins with different substituents and the structure-activity relationship among them. In this communication, we describe a synthesis of optically active 3-methyl l-oxacephalosporin nucleus and its 7-acylamino derivatives.

In order to construct the 1-oxacephalosporin nucleus, we first adopted the route shown in Scheme 1 consisting of etherification of an azetidinone $\underline{3}$ with a hydroxyacetone derivative $\underline{4}$, and conversion of the resulting $\underline{5}$ into a ylide $\underline{6}$ and following Wittig cyclization of $\underline{6}$ to the desired nucleus $\underline{7}$.

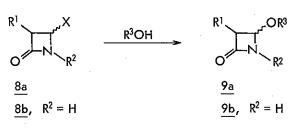
Etherification of N-substituted azetidinones <u>8a</u> (X = Cl) with an alcohol using Lewis acid,²⁾ silver tetrafluoroborate,^{la,c)} or stannic chloride³⁾ is known (Scheme 2). While most phthalimido derivatives <u>8a</u> (R¹ = phthalimido) gave <u>trans</u>ethers <u>9a</u>,^{2,3)} as major products, an azidoazetidinone <u>8a</u> (R¹ = N₃) yielded a 1:1 mixture of <u>cis</u>- and <u>trans</u>-isomers <u>9a</u>.^{la,c)} Acid alcoholysis of <u>8a</u>, where R¹ and X forms an oxazoline ring, gave a <u>trans</u>-ether <u>9a</u>⁴⁾ exclusively. Reaction of N-unsubstituted azetidinones <u>8b</u> proceeded more easily. Methanolysis of a chloro derivative <u>8b</u> (X = Cl) to <u>trans 9b</u> (R¹ = phthalimido)⁵⁾ and a reaction of a compound <u>8b</u> (X = N₃ or SO₂Et, R¹ = H) giving <u>9b</u> (R¹ = H)⁶⁾ were reported.

(840)





Scheme 1



Scheme 2

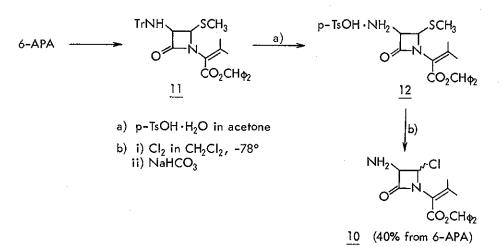
Concerning preparation of the ylide <u>6</u>, we planned to follow the method developed in this laboratory.⁷⁾ Different from previously reported preparations of ylides used in Wittig cyclization,^{8,9)} this method effectively converts the α -isopropylideneacetate moiety into the α -ylidoacetate moiety.

Compound <u>11</u>, prepared from 6-APA (6-aminopenicillanic acid), 10,11 was deprotected to the amine tosylate <u>12</u>, which was chlorinated to give a 4 to 1 mixture of <u>cis</u>- and <u>trans-10</u>. The free base 10 was unstable and stored as its tosylate.

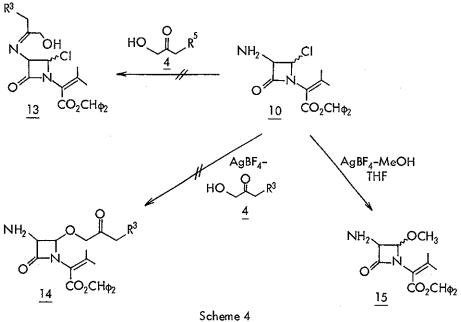
First, an intramolecular etherification of an azomethine 13 giving a <u>cis</u>-ether <u>14</u> solely was planned. However, several attempts to prepare <u>13</u> by condensation of the amino chloride <u>10</u> with hydroxyacetones <u>4</u> failed (Scheme 4). Then, methoxylation of chloride <u>10</u>, a model experiment, was carried out. Treatment of <u>10</u> with silver tetrafluoroborate and methanol in tetrahydrofuran afforded a 1:1 mixture of <u>cis</u>- and <u>trans</u>ethers <u>15</u>. Thus, etherification of the amine <u>10</u> with hydroxyacetones <u>4</u> (\mathbb{R}^3 = OH or OAc) in the presence of silver tetrafluoroborate was attempted. Unfortunately, decomposition of the β -lactam ring resulted.

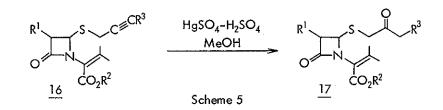
Next, etherification of <u>10</u> with propargyl alcohol was examined. Nayler et al. utilized a propargyl group as a synthon for an acetonyl group exemplified by a successful conversion of a propargyl ether <u>16</u> into a ketone <u>17</u> (Scheme 5).^{9,12)} The amine <u>10</u> was treated with silver tetrafluoroborate in propargyl alcohol to give a 1:1 mixture of <u>cis-</u> and <u>trans-</u> ethers <u>18</u> (Scheme 6). By using less expensive anhydrous zinc

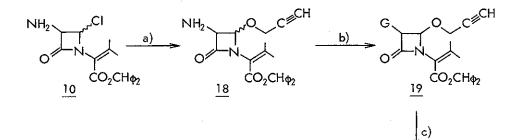
(842)

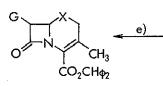




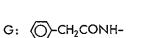








<u>22</u>, X = O <u>23</u>, X = S



	d)	- C
$\frac{21a}{21b}, Y = O$ $\frac{21b}{OH}, Y = \frac{A^{H}H}{OH}$ $\frac{21c}{CI}, Y = \frac{A^{H}H}{CI}$	a) a-1 a-2 a-3	reagent AgBF ₄ ZnCl ₂ SnCl ₂

$$\underline{21d}$$
, Y = =P ϕ_3

b) $C_6H_5CH_2COCI$, pyridine c) H₂SO₄-HgSO₄-MeOH-H₂O

20

yield 38%

39%

14%

ĊO₂CH_{φ2}

cis-trans

1:1

2:1 3:2

d)	reagent		product
۱°	i) O ₃ ,	ii) Me ₂ S	<u>21a</u>
2°	Zn-ĤO/	Ac -	<u>21b</u>
3°	SOCI2-p	у	<u>21c</u>
4°	Ф ₃ Р -		<u>21d</u>

e) dioxane, reflux

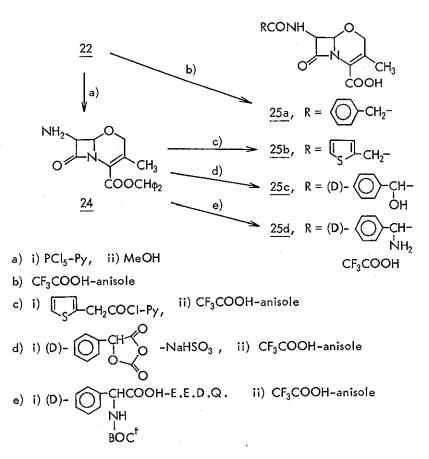
Scheme 6

chloride as the condensing agent, the cis-trans ratio was improved to 2:1. The reaction in the presence of stannic chloride was sluggish and the yield was less satisfactory. The resulting cis- and trans-ethers were separated by silica gel chromatography. The pure cis-18 was acylated to give amide 19. Mercuric sulfate catalyzed hydration of the propargyl group of 19 proceeded well as expected to yield a ketone 20. Smooth conversion of the isopropylideneacetate 20 into ylide 21d via 21a, 21b, and 21c, was performed by following the method developed in this laboratory.⁷⁾ Wittig cyclization of 21d proceeded slowly but cleanly in refluxing dioxane to give the cyclized product 22 in good yield. The structure of 22 was confirmed by similarity of its spectra to those of the corresponding cephalosporin 23 (Table 1). A higher reactivity of the β -lactam ring in 22 was suggested from its slightly higher carbonyl stretching frequency and lower electron densities on the carbon atoms at 6 and 7 positions were indicated from a smaller coupling constant $J_{6,7}$. Side chain cleavage of <u>22</u> with phosphorous pentachloride proceeded satisfactorily to produce amine 24 (Scheme 7).

The usual methods used in modification of the C_7 -side chain and deprotection of the ester group of cephalosporins were applied to the oxa-series without any difficulties and four 1-oxa-cephalosporins <u>25a-d</u> were obtained. The compound <u>25a-c</u> showed antibacterial activity¹³⁾ against Gram-positive and -negative bacteria as four to eight times high as that of

(845)

,		<u>22</u>	<u>23</u>
ir (CHCl ₃) cm ⁻¹		1 792	1785
nmr (CDCl₃) δ, ppm	C ₁₀ C ₂ C ₆ C ₇	1.97s 4.15s 4.97d (3.5 Hz) 5.65dd (3.5, 9 Hz)	2.06s 3.03, 3.39 ABq 4.92d (4.5 Hz) 5.73dd (4.5, 9 Hz)
$uv(CH_2Cl_2)$ nm (e)		267.5 (7,760)	264 (6,500)



cephalosporin analogs, whereas the cephalexin analog 25d was almost inactive. Although 25d as its trifluoroacetate was stable, the free base was proved to be unstable. The ultraviolet absorption maximum of 25d (255 nm, ε 7,080) in a slightly alkaline solution disappeared with a half life time of about 3 hr. This contrasted with the stability of cephalexin under the same conditions where the absorption maximum remained unchanged. Accordingly, we concluded that 25d was decomposed by intramolecular aminolysis of the reactive β -lactam ring during the in vitro assay.

Christensen et al.¹⁴⁾ further reported a synthesis of (\pm) -l-oxacefamandol, which showed doubled activity of that of cefamandol. Recently, Nayler et al.¹⁵⁾ reported a synthesis of several optically active amides <u>25</u> by a method similar to that described in this report. In this case, the etherification was carried out by refluxing <u>8b</u> (R¹ = tritylamino, X = SO_2CH_3 or R¹ = phenoxyacetamido, X = OAc) in propargyl alcohol and toluene in the presence of zinc acetate.

Acknowledgement

We are grateful to Dr. T. Yoshida of this laboratory for his helpful discussion on the present work.

References

- (1) a) L. D. Cama and B. G. Christensen, J. Am. Chem. Soc., 1974, 96, 7582; b) R. N. Guthikonda, L. D. Cama, and B. G. Christensen, <u>ibid</u>, 1974, 96, 7584; c) B. G. Christensen, S. P. Ratcliffe, R. William, and N. Plainfield, Germ. Offen., 2,355,209 (May 16, 1974).
- (2) S. Wolfe and M. P. Goeldner, <u>Tetrahedron Letters</u>, 1973, 5131.
- (3) a) S. Wolfe, J.-B. Ducep, K.-C. Tin, and S.-L. Lee,
 <u>Can. J. Chem.</u>, 1974, <u>52</u>, 3996; b) S. Wolfe, Jap. Offen.,
 51-4,138 (Aug. 6, 1975).
- (4) D. F. Corbett and R. J. Stoodley, <u>J. Chem. Soc.</u>, <u>Perkin I</u>, 1974, 185.
- (5) J. C. Sheehan, D. Ben-Ishai, and J. U. Piper, <u>J. Am</u>. <u>Chem</u>. Soc., 1973, 95, 3064.
- (6) K. Clauss, D. Grimm, and G. Prossel, <u>Liebigs Ann. Chem.</u>, 1974, 539.
- (7) S. Yamamoto, N. Haga, T. Aoki, S. Hayashi, H. Tanida, andW. Nagata, Series Part 3, to be published.
- (8) a) R. Scartazzini, H. Peter, H. Bickel, K. Heusler, and R. B. Woodward, <u>Helv. Chim. Acta</u>, 1972, <u>55</u>, 408; b) R. Scartazzini and H. Bickel, <u>ibid</u>., 1972, <u>55</u>, 423; c) R. Scartazzini, J. Gosteli, H. Bickel, and R. B. Woodward, ibid., 1972, 55, 2567.
- (9) a) J. H. C. Nayler, M. J. Pearson, and R. Southgate,
 <u>Chem. Comm.</u>, 1973, 58; b) J. H. C. Nayler, N. F.
 Osborne, M. J. Pearson, and R. Southgate, <u>J. Chem. Soc.</u>,

Perkin I, 1976, 1616.

- (10) J. C. Sheehan and K. R. Henery-Logan, J. <u>Am. Chem Soc.</u>, 1962, <u>84</u>, 2983.
- (11) a) E. G. Brain, I. McMillan, J. H. C. Nayler, R.
 Southgate, and P. Tolliday, <u>J. Chem. Soc.</u>, <u>Perkin I</u>,
 1975, 562; b) J. P. Clayton, J. H. C. Nayler, R.
 Southgate, and P. Toliday, <u>Chem. Comm.</u>, 1971, 590.
- (12) a) J. H. C. Nayler, M. J. Pearson, and R. Southgate,
 <u>Chem. Comm.</u>, 1973, 57; b) M. A. Harris, I. McMillan, and
 J. H. C. Nayler, J. Chem. Soc., <u>Perkin I</u>, 1976, 1612.
- (13) The <u>in vitro</u> assay was kindly carried out by Dr. T. Yoshida and his co-workers in this laboratory.
- (14) a) R. A. Firestone, J. L. Fahey, N. C. Maciejewicz, G. S. Petel, and B. G. Christensen, <u>J. Med. Chem.</u>, 1977, <u>20</u>, 551; b) B. G. Christensen and R. W. Ratcliffe in "Annual Reports in Medicinal Chemistry" ed. by F. H. Clarke, Academic Press, New York, 1976, p. 271.
- (15) a) E. G. Brain, C. L. Branch, A. J. Eglington, J. H. C. Nayler, N. F. Osborne, M. J. Pearson, J. C. Smale, R. Southgate, and P. Tolliday, "Recent Advances in the Chemistry of β-lactam Antibiotics" ed. by J. Elks, The Chemical Society, Burlington House, London, 1977, p. 204;
 b) M. J. Pearson and C. R. Branch, Jap. Offen., 51-149,295 (Dec. 22, 1976).

Received, 30th July, 1977