

## HISPANOLONE, A NEW FURANODITERPENE

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Hispanolone, a new furanoditerpene occurring in Ballota hispanica (Labiatae), is attributed the structure [I] with normal labdane skeleton.

Continuing our examination of Mediterranean Labiatae and specially of the genus Ballota, we have started with the investigation of Ballota hispanica Neck. ex Nym. (= Ballota hirsuta Benth.) growing in Spain. We refer here on the structure of a new furanoditerpene, hispanolone, occurring in this species.

Hispanolone is a crystalline product, m.p. 142-144° (from pet. ether-AcOEt),  $[\alpha]_D^{22} - 17.6^\circ$  (CHCl<sub>3</sub>; c, 0.33). Elemental analysis and MS agree with the C<sub>20</sub>H<sub>30</sub>O<sub>3</sub> formula. The IR spectrum shows bands at 3450 (OH), 1690 (C=O) and 875 cm<sup>-1</sup> (furan). The occurrence of a β-substituted furan ring is supported by the typical NMR signals (100 MHz, CDCl<sub>3</sub>) at 6.22, 7.18 and 7.30 δ; other signals include three singlets of tertiary methyl groups at 0.87, 0.90 and 1.17 δ, a doublet for a secondary methyl at 1.11 δ (J 6.5 Hz) coupled with a quartet of a methine proton at 2.72 δ; this signal collapses into a singlet on irradiation at 1.11 δ; a complex pattern for two protons (AB part of an ABX system) occurs in the 2.27-2.59 δ region.

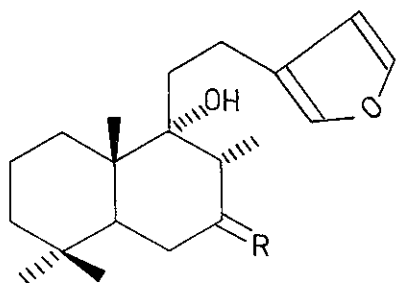
The mass spectrum of hispanolone shows prominent peaks at 318 ( $M^+$ ), 194, 123, 109, 95 and 81 m/e.

As other species of the same genus (B.nigra subsp. foetida and B.rupestris) contain<sup>1-3</sup> furanoditerpenes similar to marrubiin and marrubenol but carrying an additional keto group on position C-7, the above reported data suggest the structure and the stereochemistry depicted in [I] for hispanolone.

The configuration of the secondary methyl on C-8 must be equatorial, as supported by the coupling constant of the doublet (J 6.5 Hz): an axial methyl group should have a larger value (J 8 Hz)<sup>4</sup>. The determination<sup>5</sup> of CD on hispanolone shows a negative Cotton effect at 287 nm,  $\Delta\epsilon_{287} = -0.98^\circ$ ,  $[\theta]_{287} = -3210^\circ$  (EtOH; c, 0.259); this result is better consistent with an equatorial  $8\alpha\text{-CH}_3$  than with an axial  $8\beta\text{-CH}_3$  configuration: compare the data reported<sup>4</sup> for product [II],  $[\theta]_{290} = -1820^\circ$  (MeOH) and for product [III],  $[\theta]_{289} = -6570^\circ$  (MeOH).

The reduction of hispanolone with  $\text{NaBH}_4$  in EtOH-dioxane (1:1) solution gives a mixture of the two epimeric alcohols, separable by preparative TLC. The first product (60% yield) has m.p. 128-130° (from acetone-hexane),  $[\alpha]_D^{27} = -6.6^\circ$  ( $\text{CHCl}_3$ ; c, 0.21); MS 320 m/e ( $M^+$ ); IR  $3300\text{ cm}^{-1}$  (broad, OH), no C=O absorption; NMR (100 MHz,  $\text{CDCl}_3$ ) 3.84  $\delta$  (four lines,  $J_{ea} = J_{ea''} = J_{ee} = 3.5\text{ Hz}$ , equatorial  $7\beta\text{-H}$ ): hence the product has the structure [IV]. The second product (40% yield) has m.p. 102-105° (from hexane),  $[\alpha]_D^{27} = +13.8^\circ$  ( $\text{CHCl}_3$ ; c, 0.18); MS 320 m/e ( $M^+$ ); IR  $3450\text{ cm}^{-1}$  (OH), no C=O absorption; NMR (100 MHz,  $\text{CDCl}_3$ ) 3.43  $\delta$  (six lines,  $J_{aa'} = J_{aa''} = 10.5\text{ Hz}$ ,  $J_{ae} = 5\text{ Hz}$ , axial  $7\alpha\text{-H}$ ): hence the product has structure [V].

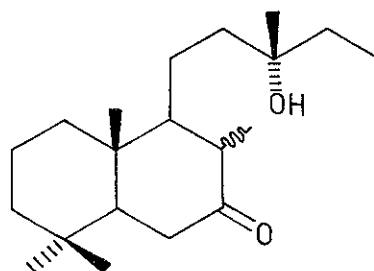
It is interesting to remark that the  $[\alpha]_D$  values and the 7-H pattern of hispanolone and its reduction products closely parallel the values and the pattern found for villenolone [VI], a diterpene



[I] R = O

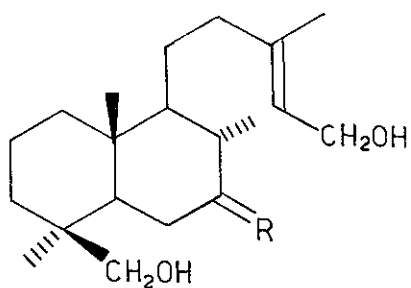
[IV] R = 7 $\alpha$ -OH, 7 $\beta$ -H

[V] R = 7 $\alpha$ -H, 7 $\beta$ -OH



[II] 8 $\alpha$ -CH<sub>3</sub>

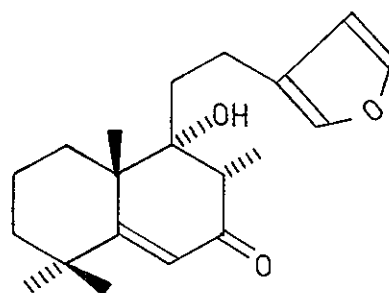
[III] 8 $\beta$ -CH<sub>3</sub>



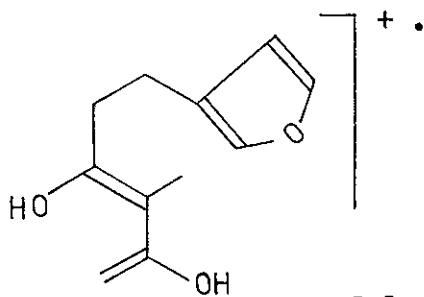
[VI] R = O

[VII] R = 7 $\alpha$ -OH, 7 $\beta$ -H

[VIII] R = 7 $\alpha$ -H, 7 $\beta$ -OH



[IX]



[X]

occurring in Sideritis chamaedryfolia (Labiatae), and for its reduction products, whose structures have been fully elucidated<sup>6</sup> as [VII] and [VIII].

As the absolute configuration of villenolone was proved<sup>6</sup> to be normal, these data bring further support to assign hispanolone the absolute configuration [I].

Table 1

[ $\alpha$ ] values for hispanolone, villenolone and their derivatives

hispanolone [I]	[ $\alpha$ ]	- 17.6°
7 $\alpha$ -OH alcohol [IV]		- 6.6°
7 $\beta$ -OH alcohol [V]		+ 13.8°
villenolone [VI]		- 4.1°
7 $\alpha$ -OH alcohol [VII]		- 10.4°
7 $\beta$ -OH alcohol [VIII]		+ 22.9°

A product with structure [I] had been obtained by Li/NH<sub>3</sub> reduction<sup>7</sup> of "Leonotis compound Y" [IX], a diterpenoid occurring in Leonotis leonurus (Labiatae). The data of this reduction product, m.p. 145-146° (from benzene-hexane) and [ $\alpha$ ]<sub>D</sub><sup>20°</sup> - 19° (CHCl<sub>3</sub>; c, 0.96), are in good agreement with those of hispanolone. As this product had been directly transformed<sup>7</sup> into isoambreinolide, the absolute stereochemistry of hispanolone [I] is therefore proved.

We wish to draw the attention on the very strong peak occurring in the MS of hispanolone at 194 m/e: it has to be attributed to the C<sub>11</sub>H<sub>14</sub>O<sub>3</sub> fragment [X] arising from the retro Diels-Alder cleavage of the  $\Delta_{7,8}$  enolic form of [I].

All the products reported gave satisfactory elemental analyses.

The present work was supported in part by the Italian National Research Council (C.N.R.), Roma.

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Received, 5th November, 1977