<sup>1</sup>H-NMR SPECTRAL EVIDENCES FOR THE DOUBLE-CHAIR CONFORMATION OF THE 9-AZABICYCLO[3.3.1]NONANE-3,7-DIONE SYSTEM

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N-Benzenesulfonyl-9-azabicyclo[3.3.1]nonane-3,7-dione was synthesized, and its conformation was shown to be of a double-chair form by <sup>1</sup>H-nmr spectrum evidences.

Of the conformation of bicyclo[3.3.1]nonanes, a chair-boat as well as twin-chair form are known today. In general, the 9-azabicyclo[3.3.1]nonan-3-one system is known<sup>1</sup> to exist in the doublechair conformation, but the carbocyclic bicyclo[3.3.1]nonan-3-one has been shown to undergo 'chair'-'boat' isomerization in the cyclohexanone ring as evidenced by the nuclear magnetic resonance  $(nmr)^2$  and electron spin resonance study.<sup>3</sup> The bicyclo[3.3.1]nonane-3,7-dione system has been reported to be in the doublechair conformation from the nmr spectrum evidence by Dreiding et al.<sup>4</sup> and from the ir spectrum evidence by Eglinton et al.,<sup>5</sup> but 9,10-benzobicyclo[3.3.2]dec-9-ene-3,7-dione was reported<sup>6</sup> to be in equilibrium between a double-chair and double-boat form.<sup>7</sup> As for 9-azabicyclo[3.3.1]nonane-3,7-dione, there have been no reports describing its conformation. In the present paper, we describe the <sup>1</sup>H-nmr evidence for the double-chair conformation of 9-azabicyclo[3.3.1]nonane-3,7-dione.

N-Benzenesulfonyl-9-azabicyclo[3.3.1]nonane-3,7-dione (VI), which was needed for some of our works concerning the structurereactivity relationships of the bicyclo[3.3.1]nonane-3,7-dione system,<sup>8</sup> was synthesized as follows.



III, R≈SO<sub>2</sub>Ph



The Robinson-Schöpf condensation of  $\beta$ -ethoxyglutaraldehyde (I) with ammonium chloride and acetonedicarboxylic acid gave an amino ketone (II),<sup>9</sup> which was converted to a sulfonamide (III), m.p. 131-133°; IR (KCl): 1715, 1350, 1165 cm<sup>-1</sup>. The ketol (IV), m.p. 181-183°; IR (KCl): 3400 cm<sup>-1</sup>, was obtained in 78% yield upon treatment of III with boron tribromide. Its acetate (V), m.p. 136-137°; IR (KCl): 1750, 1710 cm<sup>-1</sup>, displays an <sup>1</sup>H-nmr signal<sup>10</sup> due to the C<sub>7</sub>-H as a nine-line multiplet (W<sub>1</sub>=22Hz) at  $\delta$  4.91ppm,

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the feature of which suggests an equatorial orientation for the  $C_7$ -substituent in II-V. Oxidation of IV with chromium trioxidepyridine gave the diketone (VI), m.p. 161-162°; IR (CHCl<sub>3</sub>): 1730 cm<sup>-1</sup>; MS: m/e 293(M<sup>+</sup>, 6%), in 80% yield. In the <sup>1</sup>H-nmr spectrum of VI (Fig. 1), the signal for the axial hydrogens at  $C_2$ ,  $C_4$ ,  $C_6$ ,  $C_8$  appears as a double-doublet (J=16Hz, 6Hz) at  $\delta$  2.71ppm, and that for the equatorial counterparts as a doublet (J=16Hz) at  $\delta$  2.38ppm. In addition, the bridgehead hydrogens at  $C_1$  and  $C_5$  positions appear as a pseudo-triplet centered at  $\delta$  4.93ppm with half-width of 10Hz and base-width of 22Hz.<sup>11</sup> These results indicate that VI exists in the double-chair conformation despite the dipolar repulsion resulting from the two facing ketone carbonyls.

Peracid oxidation of VI with m-chloroperbenzoic acid in boiling methylene chloride gave a keto lactone (VII) [m.p. 165-167°; IR (KC1): 1750, 1720 cm<sup>-1</sup>; 68% yield] possibly via the through-space participation<sup>8</sup> of the facing carbonyls.



Fig. 1 The nmr spectrum of compound VI

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## NOTES AND REFERENCES

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- See, for example, T. Momose, S. Atarashi, and O. Muraoka, Tetrahedron Letters, 1974, 3697.
- 9. The N-methyl derivative of II was prepared by Stetter et al.;H. Stetter and R. Mehren, Liebigs Ann. Chem., 1967, 709, 170.
- 10. <sup>1</sup>H-nmr spectra were measured for the CDCl<sub>3</sub> solution with TMS as an internal standard on a Hitachi R-22 spectrometer operating at 90MHz.
- 11. In the tropane system in which the piperidine ring is in a chair form, the half-width and the base-width of the two bridgehead protons are reported to be 9.5-10.2Hz and 17.5-21Hz, respectively: see C.-Y. Chen and R. J. W. Le Fèvre, J. Chem. Soc., 1965, 3473.

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