

PL 1

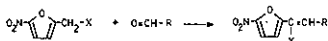
A NEW SYNTHESIS OF 5-NITROFURYLETHYLENE COMPOUNDS


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The methods of synthesis known up to this date of 5-nitrofuryl-ethylene derivatives give possibilities of preparing only some types of these compounds. The new found ways in this area enabled to synthesise new type of 5-nitrofuryl-ethylene compounds.

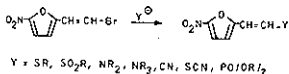
The first way is based on the use of the acidic hydrogen of some nitrofuryl derivatives in the condensation reactions with carbonyl compounds.



where X = SO₂R / SR /  / NR₂ / NO₂ /

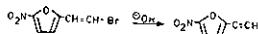
This method is suitable for preparing three and foursubstituted 5-nitrofuryl-ethylene derivatives. The modified Horner-Wittig reaction on suitable ketone was used in the case, when X was an electron-donating group.

The second way is more common and give possibilities to prepare unknown compound of 5-nitrofuryl-ethylene series. This method is based on the use of 5-nitrofurylvinylbromide (I) in the nucleophilic vinylic substitution reaction, where the reagents are compounds containing a free electron pair

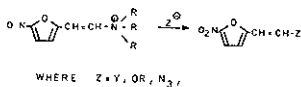


nitrogen heterocycles and so on

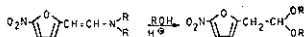
The elimination of HBr goes by the action of strong bases on I and the 5-nitrofurylacetylene arises.



The use of 5-nitrofurylvinylammonium salts is more advantageous while the reaction proceeds in polar solvent and this compound reacts with more reagents than compound I.



The enamines of 5-nitrofuryl-ethylene type are the potential source for preparation of derivatives of unknown 5-nitrofuryl-acetaldehyde



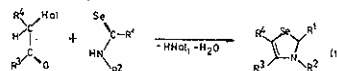
PL 2

RINGSCHLUSSREAKTIONEN AN SELENOCARBONAMID-DERIVATIVEN ZU 1,3-SELENAZOLEN, 1,3,4-SELENAJAZINEN UND 1,3,4-SELENAJAZOLEN

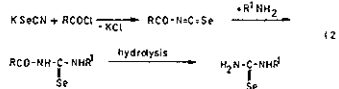
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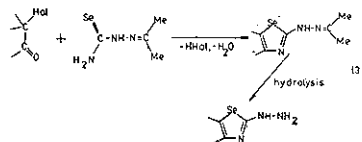
Selenazoles are the selenium analogues of thiazoles, and although the first derivatives of 1,3-selenazole were prepared by Hofmann as early as 1889, only a few publications on this subject have appeared in the literature until now. Hofmann was a student of Hantzsch's and transferred his thiazole synthesis to the selenazoles. This most useful and versatile of all the thiazole syntheses is the reaction of an α-halo ketone or aldehyde with a thioamide. For his purpose Hofmann used selenourea instead of thioamide. Even nowadays this method is still the most usual one. It permits a wide variety by choosing suitable reaction partners as to be seen in formula 1.



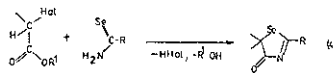
The selenoureas, needed for condensation, can usually be prepared by addition of hydrogen selenide to a solution of the corresponding cyanamides or carbodilimides. They can be less dangerously and quite smoothly synthesized according to Douglas via the isoselenocyanates by the alkaline hydrolysis of the corresponding acyl-substituted selenoureas (formula 2).



2-Hydrozinoselenazoles have been prepared by the use of selenosemicarbazide as selenocarbonamide. Starting from acetone selenosemicarbazone one can prepare the 2-isopropylidenehydrozinoselenazoles in a smooth reaction by condensation with α-halocarbonyl compounds. Their careful acidic hydrolysis leads to the free 2-hydrozino-selenazoles (formula 3).



On the other hand, the α-halocarbonyl component offers the possibility of variation too. For instance, α-halocarboxylic acids, or rather their esters, form derivatives of 4-oxoselenazoline (formula 4).



First investigations on the reactivity of the 1,3-selenazoles were undertaken by Haginiwa. He came to the conclusion that the 5-position of selenazoles is slightly reactive towards electrophilic substitution. This reactivity is still further increased by substituents in the 2-position, which can exert the +M-effect. We found during our investigations multiple nitration depending on the conditions of the reaction and the substrate, whereas Haginiwa only described the formation of the 5-nitro derivatives. Thus, the 2-benzamino-4-ary(alkyl)selenazoles form the corresponding 5-nitro derivatives under mild conditions using the nitrate/sulfuric acid method. The use of cold mixed nitric and sulfuric acids also effects the phenyl groups which may be present, leading to dinitro or trinitro compounds (formula 5).