$$(1) \qquad \qquad (2) \qquad (3) \qquad (3)$$

$$\underbrace{ \begin{bmatrix} \operatorname{iCH}_2 \operatorname{i-OH} \\ \operatorname{Br} \end{bmatrix}^{\operatorname{iCH}_2 \operatorname{i-OH}}_{\operatorname{CO}} + \underbrace{ \begin{bmatrix} \operatorname{iCH}_2 \operatorname{i-OH} \\ \operatorname{iCH}_2 \end{array}^{\operatorname{i-OH}}_{\operatorname{O}} \\ \underbrace{ \begin{bmatrix} \operatorname{iCH}_2 \operatorname{i-OH} \\ \operatorname{iCH}_2 \end{array}^{\operatorname{i-OH}}_{\operatorname{OH}} \\ \underbrace{ \begin{bmatrix} \operatorname{iCH}_2 \operatorname{i-OH} \\ \underbrace{ \begin{bmatrix} \operatorname{iCH}_2 \operatorname{i-OH}_2 \end{array}^{\operatorname{i-OH}}_{\operatorname{OH}} \\ \underbrace{ \begin{bmatrix} \operatorname{iCH}_2 \operatorname{i-OH}_2 \end{array}^{\operatorname{i-OH}}_{\operatorname{OH}} \\ \underbrace{ \begin{bmatrix} \operatorname{iCH}_2 \operatorname{i-OH}_2 \end{array}^{\operatorname{i-OH}}_{\operatorname{OH}} \\ \underbrace{ \begin{bmatrix} \operatorname{iCH}_2 \operatorname{i-OH}_2 \end{array}^{\operatorname{i-OH}_2 }} \\ \underbrace{$$

2. The anodic oxidation of some lactams

The anodic oxidation of N-methyllactams (6, n = 1.2) in acetonitrile with EtaNOTs as an electrolyte gave N-methylmides (7, n = 1.2) and N-methylhydroxylactams (6, n = 1.2) in moderate yields. In a similar manner, the oxidation of N-methylcaptoriactam (6, n = 3) afforded caprolactam (9) and N-hydroxymethylcaprolactam (10) in 22 and 19% yields, respectively. The last reaction is now being applied to a new synthesis of some attlebydges. aldehydes.

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NOVEL 1,3,5-BENZOTRIAZEPINE SYNTHESES

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Seven-membered heterocycles such os 1,4-diozepines and triazepines are of great pharmacological interest. During the past several years, we have investigated the possibility of devising novel syntheses of the 1,3,5-benzotriazepine skeleton. This heterocyclic system has yet scarcely been explored and only a very few outhenticated derivatives of this system have been described. This is probably, due to the fact that, the most obvious syntheses of benzotriazepines may alternatively lead to substituted benzimidozoles:

In order to minimize the danger of the formation of isomeric benzimidozole derivatives, an auxiliary ring was fused to the

Two simple, rational syntheses of condensed 1,3,5-benzotriaze-pines will be discussed. Some of the obtained products dis-played CNS activity.

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LE I 16

SOME SUBSTITUTION DERIVATIVES OF ERGOLINE-I WITH PROLACTIN-INHIBITING ACTIVITIES

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By introduction of alkyl groups from alkyl halides in the dimethylformamide medium into the molecules of D-8-cyano-methylergaline-1 and 6-norfestuclavine, respectively, prepared by a series of reactions from D-dihydroergotamine, the corresponding D-6-alkyl-8-cyanomethylergalines-1 and 6-alkyl and logues of festuclavine were prepared. Some D-6-alkyl-8-cyanomethylergalines-1 were also prepared by reacting sodium cyanide with D-6-alkyl-8-chlormethylergolines-1, obtained from D-6-alkyl-8-hydroxymethylergalines-1.

Some D-6-alkyl-8-cyanomethylergolines-1 were converted to D-6-Joines D-o-drivin-c-ganometriyergoines-i were converted to D-o-dikyl-8-ergolin-l-ylacetamides, either via the corresponding 8-carboxymethyl acids, their methyl esters, and their hydrazides, or by direct addition of water to the nitrile groups in the parent compounds in a heterogeneous biphasic pyridinic-aqueous medium in the presence of tetraalkylammonium bases.

alum in the presence of tetradikylommonium bases.

1-Nitroso derivatives of D-6-alkyl-8-ergolin-1-ylacetamides were prepared by the action of excess nitrous acid upon the corresponding hydrazides of D-6-alkyl-8-ergolin-1-ylacetic acids and decomposition by aqueous ammonia of the intermediate 1-nitroso derivatives of azides of the acids.

The 6-alkyl analogues of readcare. D-8-cyanomethylergoline-I, and D-8-ergolin-I-ylacetomide produced marked antinidation and antilactation effects in biological tests. On introduction of the ethyl, propyl or allyl groups into the position 6 in the ergoline cycle the prolactin-inhibiting activities of the compounds were markedly enhanced in comparison with the activities of the corresponding 6-methyl derivatives.

THE SYNTHESIS OF GEMINALLY DISUBSTITUTED 4,5-DIHYDRO-6H-CANTHIN-6-ONES

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In the post decade some alkalaids of the eburnane group, e.g. vincamine 1, have attracted much interest, especially in view of their known biological activity. In continuation of our bro-

$$\begin{array}{c} & & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & &$$

oder programme in this field we focused our attention on the synthesis of their degradation products — 4,5-dihydrocanthin-6-ones — particularly on (—)-4-ethyl-4-propyl-4,5-dihydro-641-canthin-6-one 2, which was readily obtained by degradation of (+)-vincamine 1.

Starting from racemic and optically active 2,2° and 3,3-disubstituted 3-methoxycarbonylpropionic acids 6—9 the appropriate amid-esters 10—13 were prepared, which on Bischler-Napierol-skii cyclization followed by selenium dehydrogenation afforded the desired 4,5-dihydrocanthin-6-ones 2—5.

The synthesis of succinic acid-esters 6—9 will be discussed briefly together with the physico-chemical properties of tryptomides 10—13 and especially of conthin-6-ones 2—5.

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SYNTHESIS OF SUBSTITUTED 1(2H)-ISOQUINOLINONES AND 8-OXOBERBINES FROM HOMOPHTALIC ANHYDRIDES AND AZOMETHINES

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Recently we reported that the reaction of 1,3-isochromandiones (homophtalic anhydrides) with acyclic and cyclic azomethines can be used as a method for synthesis of 4-carboxy-1(2H)-isoquinolinones, 13-carboxy-8-oxoberbines and 14-carboxy-hexadehydroyohimbanes^{1,2}.

-isoquinolinones, 13-carboxy-8-oxoberbines and 14-carboxy-hexadehydroychimbanes 1-2. Homophtolic anhydrides 1 react analogously with azomethines of aromatic, heteroaromatic and aliphatic aldehydes and ketones, and aliphatic amines 2, as well as with 1-alky[ary])-6.7-dimethoxy-3.4-dihydroisoquinolines 4 (refluxing of 1 and 2, resp. 4 in benzene or dichloroethone, extraction of 3, resp. 5 with NaOH, aq), where trans-2,3-disubstituted or 2,3,3-trisubstituted-4-carboxy-3.4-dihydro-1(2H)-isoquinolinones 3 (R = H) and 14-alky[ary])-13-carboxy-8-oxoberbines 5 (R = H), resp. are obtained. The relative configuration of 3 is proved chemically and by NMR-investigation of their corresponding methyl esters 3 (R = Me)1-2. The spatial structure of C-13 and C-14 can not be established by NMR-analysis of 5 (R = Me). Only in the case of the reaction between 1 (R1 = H or MeO) and ethoxymethylene-aniline (2, R2 = H, R3 = Eto, R4 = Ph). A-anilinomethylene-1,3-isochromandiones 6 (R1 = H or MeO) are obtained. In conditions of alkaline hydrolysis they are converted into 4-carboxy-1(2H)-isoquinolinones 7 (R = H, R1 = H) or MeO, R2 = Ph). The NMR-spectra of their methyl esters show a great similarity with the spectrum of the ester 7 (R = R2 = Me, R1 = H), which was obtained by us from the known

acid 7 (R \rightleftharpoons H)3. In these two cases an aldal condensation between the CH-acidic anhydrides and the azomothine takes place

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R = H or Me $R^1 = H$ or MeO $R^2 = Me$ or Ph

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ON THE MECHANISM OF THE AMINATION OF 3-SUBSTITUTED DERIVATIVES OF 1,2,4-TRIAZINE WITH POTASSIUM AMIDE IN LIQUID AMMONIA

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Amination of 3-methylthio-1,2,4-triazine with potassium amide in liquid ammonia at -75 °C yields the corresponding 3-amino compound and in addition 3,3'-bis(methylthio)-5,5'-bi-1,2,4-triazine. It has been proved using the corresponding 3-methylthio-[4-15N]-1,2,4-triazine that amination goes for about 93% via a ring opening - ring closure sequence (SNANRORC-mechonism)¹ (see Scheme)