In the present paper, we will deal with our results concerning that step of the reaction in which 2-nitrohydraze compounds (B) are converted into 2-phenylbenzotriazole-1-oxide (C). As given in Scheme II, a series of 2-nitrohydrazobenzene derivatives was prepared.

Scheme II

Cyclization thereof gave the corresponding 2-(4-X-phenyl)-6-Y-benzotriazole-1-axides, respectively.

Hydrazocompounds were prepared by reacting 2-nitrofluoro-benzene with the corresponding 4-X-phenylhydrazine, which, in turn, were prepared by SnCl₂ or Na₂SO₃ reduction of 4-X-ben-zene diazonium solts.

zene diazonium solts.

The cyclisation reaction was followed spectrophotometricaly in 40% aqueous propanol by measuring the intensity of an absorption band around 300 nm corresponding to benzotriazole oxide. The effect of pH and substitution was investigated. The kinetic measurements showed the reaction to be 1 st order in hydrazocompound (at constant pH in the region of 5,5—10,5) and the reaction rote to be pH dependent. Rate constants for cyclisation of hydrazocompounds are linearly pH dependence (the pH dependence of log k has a slope equal to 1). Obviously, the reactions rate depends on concentration of the hydrazocompound, on concentration of hydrazocompound on the substituent. A mechanism of the cyclisation suggested on the basis of the above data is also in agreement with some quantum mechanical calculations.

quantum mechanical calculations.

Scheme III

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SYNTHESES AND REACTIONS OF 2-AMINO-3-CYANO-4,5-8IS (HETARYL) FURANS AND 4-R-5,6-BIS(HETARYL)

FUROPYRIMIDINES

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2-Amino-3-cyanofuran derivatives are not only interesting in aspect of their preparation but as the possibility of their use in another syntheses as well $^{1-3}.\,$

In this report, 2-amino-3-cyano-4,5-bis(2-furyl)furan la and 2-amino-3-cyano-4,5-bis(2-thienyl)furan lb have been obtained by reaction from the corresponding acyloines and moleonaitile. These derivatives have been utilized in another synthesis for the preparation of the furopyrimidines II—IV, furo-1,2,3-triazinones VI and Schiff's bases VII as shown the following charts:

The structures of the synthesized compounds were determined by means of their IR, UV, ¹H-NMR and mass spectra. Spectral data of 2-furyl- and 2-thienyl derivatives have been compared with each other. The biological activity of some compounds mentioned above has been studied also.

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PO 10

MANNICH BASES OF 2-MERCAPTOBENZOTHIAZOLE AND THEIRS ANTIMYCOBACTERIAL AND ANTIVIRAL ACTIVITY

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The present data on the Mannich reaction of 2-mercaptobenzothiazole (2-MBI) with primary amines do not offer a satisfactory explanation, why with some amines monoderivatives and with the others bisderivatives are obtained $^{\rm L}$.

We have found that the amines with ρK_B around 3—5 afford only bisderivatives, while those with ρK_B around 9—14 gave exclusively monoderivatives $z^{2/3}$. In further series experiments we employed as aminocomponents hydrozides of aliphatic and aromatic acids. At both types of hydrozides we obtain only bisderivatives of Mannich bases.

Several of the prepared compounds showed remarkable anti-mycobacterial activity mainly against Mycobacterium tubercu-losis resistant on the isonicotinylhydrazide^{4,5}.

A number of these derivatives was studied also for antiviral activity. The antiviral tests were carried out in cell cultures using vaccinia, Newcastle disease and western equine encepholomyelits viruses. Several of derivatives tested have shown middle or slight degree of inhibition of virus multiplication.

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PO 11

A NEW METHOD OF PREPARATION OF DERIVATIVES OF PARTIALLY HYDROGENATED DIBENZO[c,h][1,2,6,7]TETRA-ZECINE AND ITS ANALOGUE

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5,6,7,12,13,14-Hexohydrodibenzo [c,h] [1,2,6,7] tetrazecine-7,14-dione and 5,6,7,12,13,14-hexohydrodipyrido[3,4-c,h][1,2,6,7] tetrazecine-5,12-dione were prepared by heoting the hydrazide of sallcylic and 3-hydroxyisonicotinic acid, respectively. Their structure was verified by elementary analysis, infra-red and mass spectra.

PO 12

CONCERTED AND STEPWISE CYCLOADDITIONS OF 1-BENZYL -4-N-ACYLIMINO-1,2,4-TRIAZOL WITH ISOTHIOCYANATES

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In contrast to numerous 1,3-dipolar cycloadditions, following a concerted reaction mechanism, there are few 1,5-intermolecular cycloadditions known. Since the extension of 1,3-dipols with one conjugated double bond transforms them to 1,5-dipols, theoretically there can be as much as 98 1,5-dipols.

In our case the azomethinimine system has been extended on the nitrogen end with a carbonyl group. The charge separation and its dissipation of such a dipol predisposes the molecula of 1-benzyl-4-N-acylimino-1,2,4-triazol 1 to both 1,3-concerted and 1,5-stepwise reaction manners.

We have investigated the possible role of nonpolar and aprotic polar solvents on the stabilisation of mesomeric forms lb and lc and on alternative transition states. Using benzene as solvent, I adds to aromatic isothioeyanates in 1,3-concerted manner to give stable aduct on C S band and unstable aduct on C N bond, together with other compounds, arising either from still another mechanism, or as splitting products of unstable C N aduct of I. Polar apratic solvents, like DMF, DMSO or HMPT favour 1,5-dipolar structure, as they bring more solvation stabilization needed for enhanced charge separation. Concerted [6₅ 2₃] reaction being disallowed, the cyclic aduct IV arises by stepwise addition. In the first step anionic terminus of 1,5-dipol atacks the corbon atom of isothiocyanate, giving noncyclic intermediate. This could be isolated and consequently cyclises to IV.

PO 13

5-NITRO-2-FURYLVINYLATION OF PHOSPHOROUS COMPOUNDS

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The synthesis of 5-nitro-2-furylvinylenbromide (I) which contains a sufficiently reactive bromine has opened the path for a new synthesis of biologically high active 5-nitro-2-furylethylene derivatives.

We studied a 5-nitro-2-furylvinylation of the tertiary phosphines (e.g. triphenylphosphine), tertiary phosphites (e.g. triethylphosphite) and alkali metal diphenylphosphides in the nonpolar

We have found that the presence of a phosphorous group on vinylene group of the nitrofurane derivative effects significantly its biological properties.