

FLASH VACUUM PYROLYSIS OF 3,5-DISUBSTITUTED β -PHENETHYLSULFONYL AZIDES

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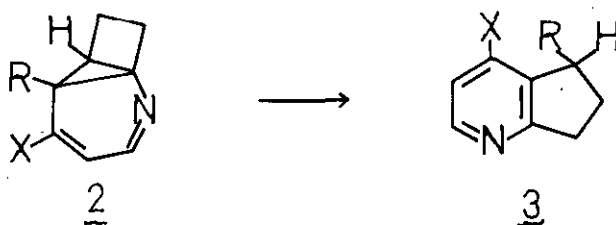
(Submitted in honor of Prof. T. Nozoe's 77th birthday)

Flash vacuum pyrolysis of 3,5-dimethyl- β -phenethylsulfonyl azide at 650° gives appreciable quantities of the normal nitrene cyclization products as well as some 2-cyclopropyl-3,5-dimethylpyridine. 3-Chloro-(5H)- and (7H)-pyridine are obtained from 3,5-dichloro- β -phenethylsulfonyl azide. Possible mechanisms are proposed to account for the results.

The flash vacuum pyrolysis (FVP) of a number of β -phenethylsulfonyl azides has been reported and a tentative mechanism proposed for the formation of dihydropyridines.¹ We now describe the results obtained from the FVP of some 3,5-disubstituted β -phenethylsulfonyl azides (1) which necessitate some refinement in the proposed mechanism. In the final step of the latter a hydrogen migration [$2(R = H) \longrightarrow 3(R = H)$] occurred from the position originally meta to the ethyl side

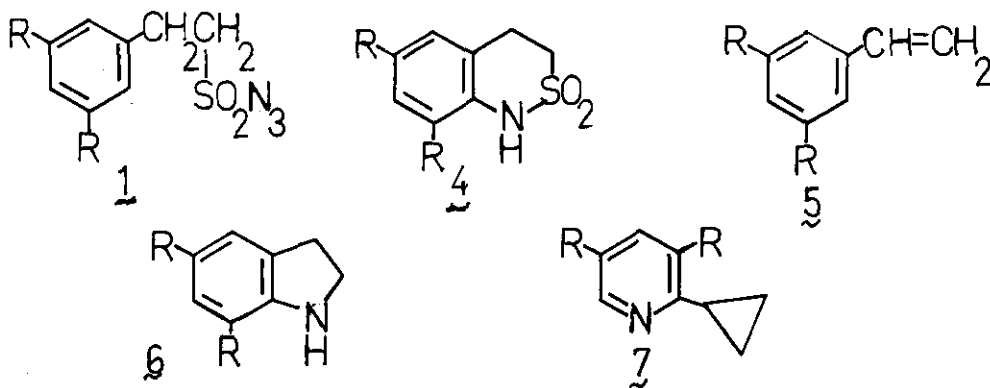
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chain to the carbon that now became C₅ in the dihydropyridine (3).



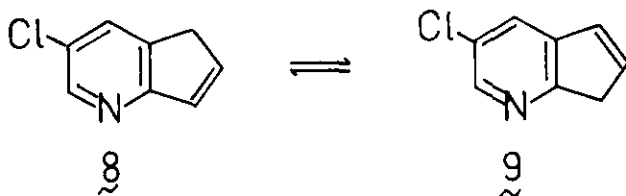
It was of interest, therefore, to determine whether or not such a migration would occur if the meta positions were blocked and, if not, what pathway would be followed.

3,5-Dimethyl- β -phenethylsulfonyl azide (1; R = Me)² underwent FVP at 400°C in a nitrogen flow (10mm Hg pressure) to give 6,8-dimethyl-3,4-dihydro-2,1-benzothiazine-2,2-dioxide (4; R = Me)³ (46%), mp 187-188°, 3,5-dimethylstyrene (5; R = Me) (3.8%)³, and 5,7-dimethylindoline (6; R = Me) (trace). As expected,¹ FVP of 1 at 650° gave a lower yield of 4 (20.5%) and a corresponding larger amount of 6 (14.8%). In addition, however, a new type of product was obtained, namely 2-cyclopropyl-3,5-lutidine (7; R = CH₃) (2.9%), bp 45-48°/0.05 mm; picrate, mp 154-156°(EtOH); identical with a sample prepared (49% yield) from 3,5-lutidine and cyclopropyllithium in boiling toluene. No pyridine derivative 3 was detected.



Cyclopropylpyridine ζ ($R = CH_3$) was stable under the FVP conditions. Reinvestigation of the FVP of β -phenethylsulfonyl azide (λ ; $R = H$) at 650° gave rise to a trace of a compound with the same retention time as authentic 2-cyclopropylpyridine (ζ ; $R = H$) but the quantity formed was too small to permit collection from the g.l.c. and thus positive identification could not be made. A similar result was obtained by carrying out the FVP at 750° . ζ ($R = H$) too was stable at 650° and did not rearrange to dihydropyridine under FVP conditions.

In contrast, FVP of 3,5-dichloro- β -phenethylsulfonyl azide (λ ; $R = Cl$), mp $79-80^\circ$ (aq. EtOH), at 650° gave ξ ($R = Cl$)⁴ (2.4%), η ($R = Cl$) (4.8%), mp $156-157^\circ$ (benzene), ζ ($R = Cl$) (2.7%), and a mixture of 3-chloro-(5H)- θ and 3-chloro-(7H)-1-pyridine (ρ), mp $58-59^\circ$ (43.9% yield).⁵ No cyclopropylpyridine derivative (ζ)

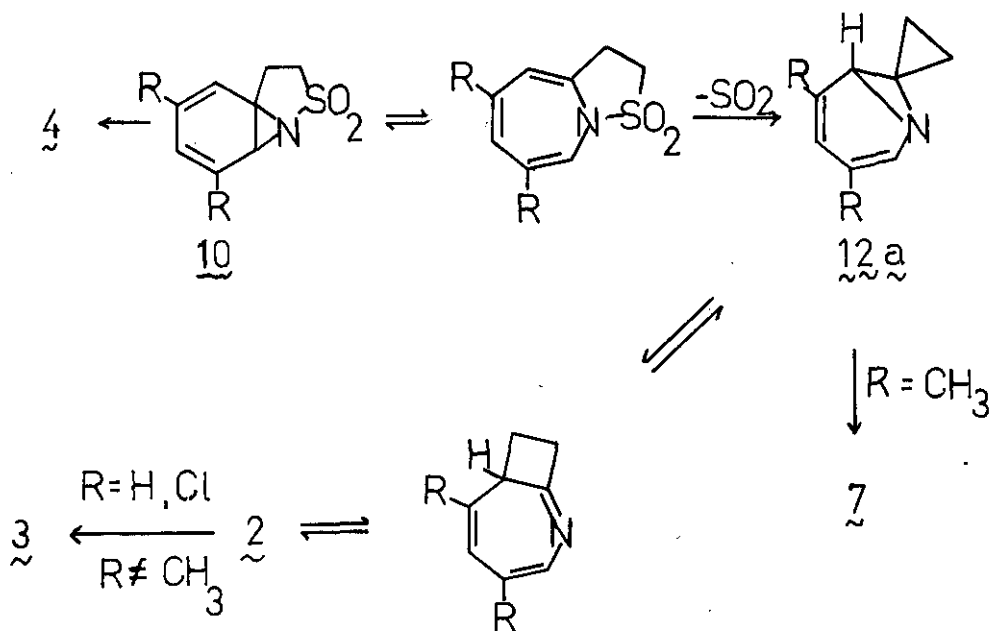


could be detected by g.l.c..

In the tentative mechanism proposed earlier¹ the proton migration step [$\rho \rightarrow \xi$ ($R = H$)] occurs at the end of the reaction sequence, long after the SO_2 is eliminated. The formation of appreciable quantities of η ($R = Me$) [and the product ξ of its thermal decomposition¹] at 650° ⁷ when migration [$\rho \rightarrow \xi$ ($R = Me$)] is apparently not feasible suggests that either the 3,5-dimethyl groups stabilize the intermediate sultam considerably, thus diverting the reaction towards formation

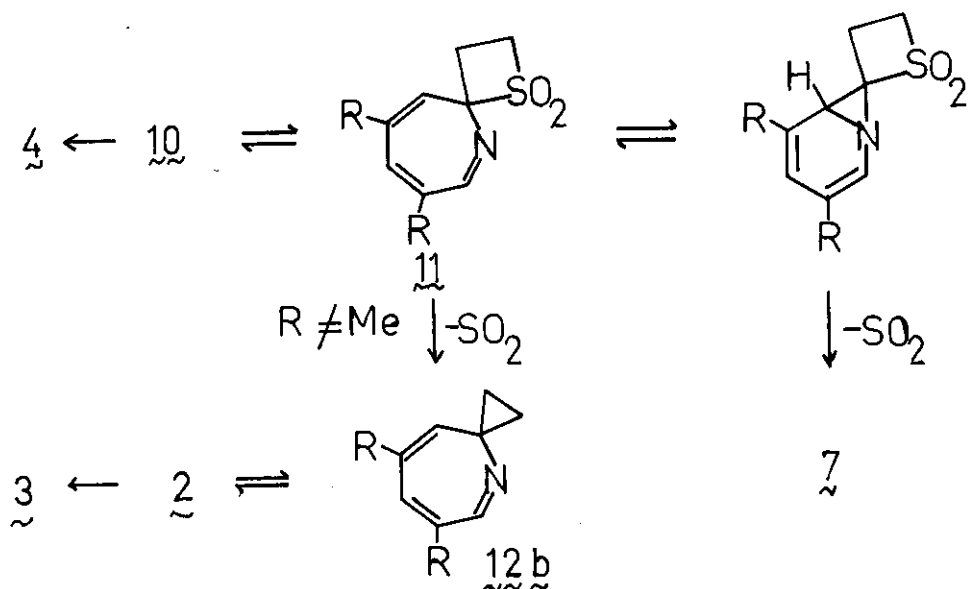
of $\frac{4}{3}$ ($R = \text{Me}$), or loss of SO_2 occurs much later in the sequence with all the preceding steps being reversible. In the former case, the proposed¹ scheme would have to be refined somewhat as in Scheme I below. If the latter case obtains then a possible explanation might

SCHEME I

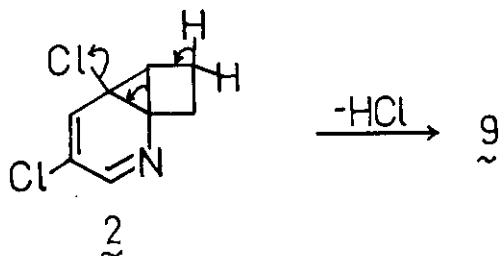


be that outlined in Scheme 2, though again it would be necessary to assume that loss of SO_2 from $\frac{11}{11}$ is rendered more difficult when $R = \text{CH}_3$. An intermediate such as $\frac{12}{12}$ (a or b) is required by the observation that both α - (present work)⁸ and β -methyl- β -phenethylsulfonyl azides¹ give the same ratio of 6- to 7-methyldihydropyridines.

SCHEME 2



3-Chloro-(7H)-1-pyridine (9) may arise directly from 2 (R = Cl).



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2. All new compounds were completely characterized by microanalysis and infrared, nmr, and mass spectrometry.

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4. C. S. Marvel, C. G. Overberger, R. E. Allen, H. W. Johnson, J. H. Saunders, and J. D. Young, J. Am. Chem. Soc., 68, 861 (1946).
5. Nmr (CDCl₃) δ 8.39 and 8.28 (2 d, $J_{2,4} = 2.5$ Hz, 1H, H₂), 7.65 and 7.53 (2 d, $J_{2,4} = 2.5$ Hz, 1H, H₄), 7.07-6.67 (m, 2H, vinylic H), 3.60 - 3.30 (m, 2H, CH₂). The ratio of the peaks at δ 8.39 and 8.28, and at δ 7.65 and 7.53 changes from 3.9:1 to 1.5:1 on keeping the CDCl₃ solution with the δ 8.39 and 7.65 peaks decreasing in intensity and those at δ 8.28 and 7.53 increasing. This shows that ζ initially formed tautomerizes to ξ in solution, as reported for other (5H)-1-pyridines.⁶
6. M. M. Robinson, J. Am. Chem. Soc., 80, 6254 (1958). C.B. Reese, ibid., 84, 3979 (1962). G. Bergson and A.-M. Weidler, Acta Chem. Scand., 16, 2464 (1962). A. G. Anderson, Jr., and H. L. Ammon, Tetrahedron Lett., 2579 (1966).
7. Under these conditions, no ζ (R = H) is isolated from ζ (R = H).¹
8. Sodium 1-phenyl-2-propanesulfonate, mp >310°; 1-phenyl-2-propanesulfonyl chloride, bp 81-83°/0.1 mm; 1-phenyl-2-propanesulfonyl azide, bp 90-93°/0.1 mm.

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