

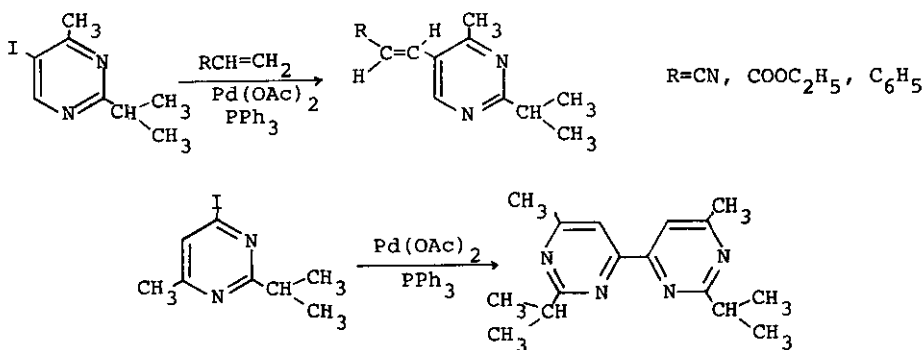
CROSS-COUPLING REACTION OF 5-SUBSTITUTED 4-IODOPYRIMIDINES WITH OLEFINS IN THE PRESENCE OF PALLADIUM COMPLEX

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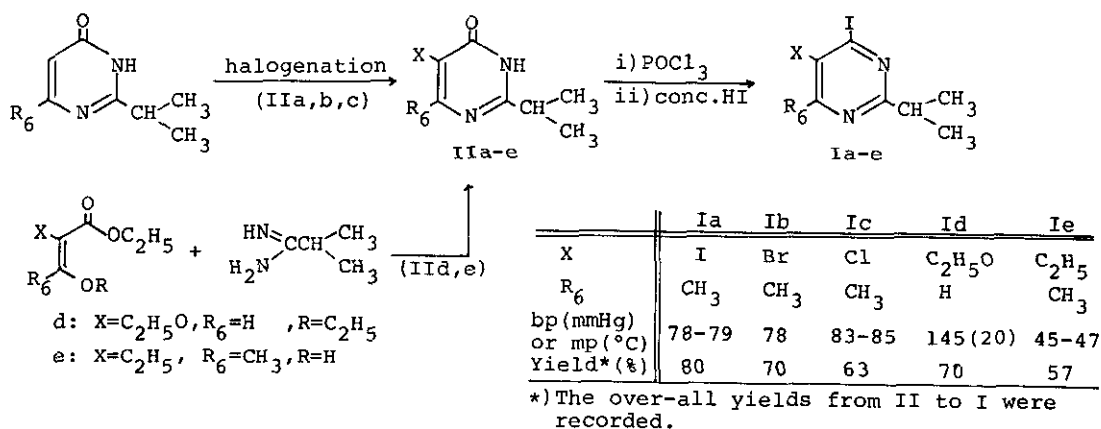
Abstract — On the olefinic coupling reaction of halopyrimidines, the presence of a 5-substituent was found to facilitate the coupling at the 4-position. Namely, 5-substituted (X=I, Br, Cl, C₂H₅O, C₂H₅) 4-iodo-2-isopropylpyrimidines (Ia-e) reacted with styrene to give the corresponding 4-styryl derivatives in 46, 46, 40, 87, and 51 % yields, respectively. The syntheses of the starting material, 4-iodopyrimidines (Ia-e) are also described.

We have recently reported the following result on the cross-coupling reaction of monoiodo(or bromo)pyrimidines with olefins.¹ In the presence of a palladium-tri-phenylphosphine complex, olefins such as acrylonitrile, ethyl acrylate, and styrene reacted with 5-iodo(or bromo)pyrimidines to give the cross-coupled products. In contrast with this observation, the same olefins did not coupled with 2- and 4-iodopyrimidines. For instance, the reaction of 4-iodo-2-isopropyl-6-methylpyrimidine resulted in the formation of 2,2'-diisopropyl-6,6'-dimethyl-4,4'-bipyrimidine independent of the presence of olefins.



Scheme 1

In this communication, we wish to report the coupling reaction of 4,5-diiodo-2-isopropyl-6-methylpyrimidine (Ia) and its related compounds (Ib-e) with styrene affording the result which seemed to conflict with the above description. The pyrimidines (Ia-e) employed in the investigation were prepared from the 5-substituted 4-pyrimidinones (IIa-e), which were obtained by the halogenation of the corresponding 4-pyrimidinones or by ring-closing reactions, through 4-chloro derivatives as illustrated in Scheme 2.



Scheme 2

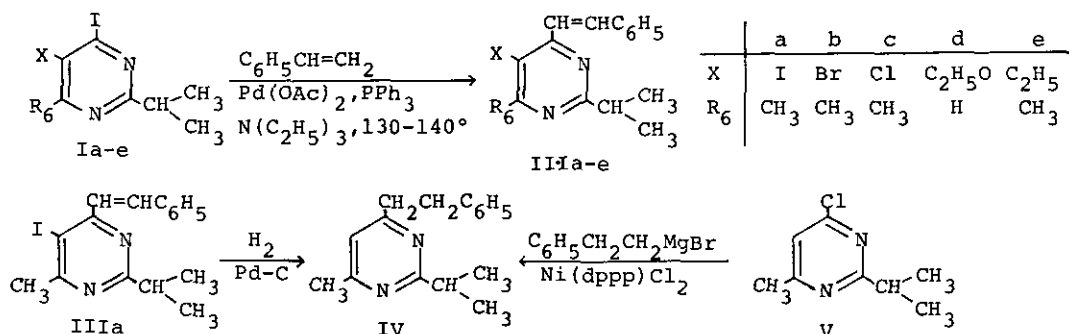
When 4,5-diiodo-2-isopropyl-6-methylpyrimidine (Ia) was heated with styrene in triethylamine in the presence of a catalytic amount of palladium-triphenylphosphine complex,² 5-iodo-2-isopropyl-6-methyl-4-styrylpyrimidine (IIIa), mp 84-85°, C₁₆H₁₇IN₂, was obtained in 46 % yield, and the corresponding 5-styryl isomer was not isolated. Although the NMR spectrum (CDCl₃) of IIIa [1.35 ppm (6H, d, J=7.0 Hz), 2.68 ppm (3H, s), 2.80-3.40 ppm (1H, m), 7.00-7.80 ppm (5H, m), 7.52 ppm (1H, d, J=15.0 Hz), and 8.08 ppm (1H, d, J=15.0 Hz)] is consistent with its trans olefin structure, the spectrum gives no information on differentiation of IIIa from the 5-isomer. The catalytic reduction of IIIa over palladium-charcoal gave 2-isopropyl-6-methyl-4-phenethylpyrimidine (IV), bp 148° (2 mm Hg), which was identical with the authentic specimen prepared from 4-chloro-2-isopropyl-6-methylpyrimidine (V) according to the reported manner.³ Thus it became apparent that Ia coupled with styrene in a fashion quite different from what might have been

anticipated by the reaction of the monoiodopyrimidines.

Further investigation was made to confirm the scope of this unusual reaction.

Namely, 5-bromo-4-iodo-2-isopropyl-6-methyl- (Ib), 5-chloro-4-iodo-2-isopropyl-6-methyl- (Ic), 4-iodo-5-ethoxy-2-isopropyl- (Id), and 5-ethyl-4-iodo-2-isopropyl-6-methyl-pyrimidine (Ie) were allowed to react with styrene under identical conditions. In all the cases, the corresponding 4-styryl compounds (IIIb-e) were obtained in the following yields [IIIb: mp 69-70.5°, 46 %; IIIc: mp 77-79°, 40 %; IIId: bp 170° (2 mmHg), 87 %; and IIIe: bp 165° (1 mmHg), 51 %].

Based on the results of the elemental analysis and the mass spectra of the products [IIIb: $C_{16}H_{17}BrN_2$, $m/e=316, 318 (M^+)$; IIIc: $C_{16}H_{17}ClN_2$, $m/e=272, 274 (M^+)$; IIId: $C_{17}H_{20}N_2O$, $m/e=268 (M^+)$; and IIIe: $C_{18}H_{22}N_2$, $m/e=226 (M^+)$], the olefinic coupling reaction was clearly demonstrated to occur at the 4-position of these pyrimidine derivatives (Ib-e).



Scheme 3

The superior activity of an iodo-(or bromo-)substituent at the β -position of N-heteroaromatics toward the olefinic coupling reaction has already reported not only on halopyrimidines but on halogenated pyridines, quinolines, and isoquinolines.^{1,4} Accordingly, our present result forms striking contrast to the previous observations mentioned above, although the role of the neighbouring substituents in this reaction is still obscure.

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References

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