SYNTHESIS OF PERHYDRO-1,5,2-DIUXAZINES
A NOVEL TYPE OF HETEROCYCLIC COMPOUNDS

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<u>Abstract</u> — Derivatives of perhydro-1,5,2-dioxazine have been prepared by the reaction of N-substituted 2-hydroxycarbohydroxamic acids with the dimethyl acetals of aromatic aldehydes.

The six-membered ring system of 1,4,2-cloxazine ( $\underline{1}$ ) is known since 1967, when McHale  $^1$  synthesized some 5,6-dihydro-3-phenyl-1,4,2-dioxazine-5-ones ( $\underline{2}$ ). However, the isomeric heterocyclic system of 1,5,2-dioxazine ( $\underline{3}$ ) does not as yet to have been isolated.

I now wish to report the synthesis of some derivatives of this novel type of compound by a method  $^2$  used for the cyclisation of unsubstituted hydroxemic acids to 1,4,2-dioxazoles. Thus, treatment of the N-substituted 2-hydroxycarbohydroxamic acids  $\underline{4a-d}$  with the dimethyl acetals of benzaldehyde or 2-chlorobenzaldehyde in boiling benzene in the presence of p-toluenesulphonic acid gave perhydro-1,5,2-dioxazine-3-ones ( $\underline{5a-d}$ ) as crystalline products in yields of 35-60%. Attempts to prepare  $\underline{5}$  by refluxing a solution of  $\underline{4}$  and benzaldehyde in the presence of p-toluenesulphonic acid under the removal of water were unsuccessfull. The structure of  $\underline{5}$  was confirmed by elemental analysis, by ir, by  $^1$ H-nmr and by mass spectra. The infrared spectra showed a strong carbonyl band at the expected region  $^3$  (1650 - 1665 cm $^{-1}$ ). A one proton signal due to the cyclic acetalic structure was observed at  $\mathbf{d} = 6,03 - 6,50$ . Molecular peaks corresponding to the right molecular weights appeared in the mass spectra.

#### Experimental Part

### General procedure:

A mixture of 0.01 mole of hydroxamic acid  $\underline{4}^{4}$ , 0.012 mole dimethyl acetal of benzaldehyde or 2-chlorobenzaldehyde and 50 mg p-toluenesulphonic acid in 50 ml anhydrous benzene was refluxed for 4 hours. After removing the solvent the remaining oil was purified on a silica gel column (benzene/dichloromethane 1:1).

### 2-Methyl-4,4,6-triphenylperhydro-1,5,2-dioxazine-3-one (5a)

Yield 1.2g (35%); mp 99-100°C (cyclohexane), colorless prisms; ir(KBr) 1665 cm<sup>-1</sup> (C=0);  $^{1}$ H-nmr (CDCl $_{3}$ ) & 3.40 (3H, s, CH $_{3}$ ), 6.17 (1H, s, O-CH-O), 7.27 - 7.77 (15H, m, Ar $_{\rm H}$ ); MS m/e 345 (M $^{+}$ ), 317 (M $^{+}$  - CO), 239 (M $^{+}$  - PhCHO).  $^{2}$ C $_{22}$ H $_{19}$ NO $_{3}$  (345.4) calc. C 76.50 H 5.54 N 4.06

# 6-(2-Chlorophenyl)-2-methyl-4-phenylperhydro-1,5,2-dioxazine-3-one (5b)

found C 76.44 H 5.67 N 4.23

Vield 1.8g (60%); mp 88°C (cyclohexane), colorless prisms; ir(KBr) 1665 cm<sup>-1</sup>(C=0)  $^{1}$ H-nmr (CDCl<sub>3</sub>) & 3.40 (3H, s, CH<sub>3</sub>), 5.63 (1H, s,  $\stackrel{.}{>}$ CH), 6.50 (1H, s, 0-CH-0), 7.26 - 7.86 (9H, m, Ar $\stackrel{.}{H}$ ); MS m/e 303/305 (M<sup>+</sup>), 258 (M<sup>+</sup> - CH<sub>3</sub>ND).  $^{1}$ Clh<sup>4</sup>14<sup>ClNO</sup>3 (303.7) calc. C 63.28 H 4.65 Cl 11.67 N 4.61 found C 63.03 H 4.72 Cl 11.42 N 4.75

# 2-Benzyl-4,4,6-triphenylperhydro-1,5,2-dioxazine-3-one (5c)

Yield 1.7g (40%); mp 96-97°C (cyclohexane), colorless prisms; ir (KBr) 1650 cm<sup>-1</sup>. (C=0);  $^{1}$ H-nmr (CDCl<sub>3</sub>)  $^{6}$  4.96 (2H, q, N-CH<sub>2</sub>, J 7 Hz), 6.03 (1H, s, 0-CH-0), 7.26 - 7.70 (20H, m, ArH); MS m/e 421 (M<sup>+</sup>), 316 (M<sup>+</sup> - PhCH0), 301 (M<sup>+</sup> - PhCH<sub>2</sub>NO).

 $c_{28}H_{23}N0_3$  (421.5) calc. C 79.79 H 5.50 N 3.32 found C 79.94 H 5.48 N 3.32

## 2-Benzyl-6-(2-chlorophenyl)-4-phenylperhydro-1,5,2-dioxazine-3-one (5d)

Yield 1.4g (37%); mp 121°C (benzene/n-hexane), colorless prisms; ir(KGr) 1665 cm<sup>-1</sup> (C=0);  $^{1}$ H-nmr (CDCl<sub>3</sub>)  $^{6}$  4.9G (2H, q, N-CH<sub>2</sub>, J 7 Hz), 5.7C (1H, s,  $\stackrel{>}{\Rightarrow}$ CH), 6.40 (1H, s, 0-CH-0), 7.26 - 7.87 (14H, m, Ar $\stackrel{H}{=}$ ); MS m/e 379/381 (M<sup>+</sup>), 258/26C (M<sup>+</sup> - PhCH<sub>2</sub>NO).

 $C_{22}H_{18}Clno_3$  (379.9) calc. C 69.56 H 4.78 Cl 9.33 N 3.69 found C 69.72 H 4.82 Cl 9.64 N 3.71

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