

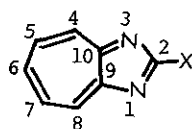
THE SUBSTITUENT EFFECT AND ASSIGNMENT OF THE  $^{13}\text{C}$ -NMR SPECTRA OF  
SOME 2-SUBSTITUTED 1,3-DIAZAZULENES

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**Abstract** — The  $^{13}\text{C}$ -NMR spectral signals of 1,3-diazazulene derivatives were unequivocally assigned by aids of  $^2\text{H}$  and  $^{15}\text{N}$  labellings. Variation of the C-2 substituents caused no appreciable effect on the chemical shifts of C-5 and C-9.

The  $^{13}\text{C}$ -NMR spectrum is a powerful tool to study the fine structures of organic compounds, and a number of data on various compounds have been accumulated. But, as far as nonbenzenoid-aromatic compounds are concerned, still relatively small number of studies have been published to date. In connection with the study on the synthesis of zoathoxanthins<sup>1</sup> and paragraccines,<sup>2</sup> we needed the NMR spectra of the 1,3-diazazulene derivatives, and we have carried out the measurements on a



series of derivatives. The 1,3-diazazulenes carry nitrogens at the most electron-rich positions of the azulene skeleton, and in fact a dipolar structure is contributed in substantially as judged on the dipole moment measurement,  $\mu=4.03$  D.<sup>3</sup> After a report by S. Ito *et al.*<sup>4</sup> in 1971, no other  $^{13}\text{C}$ -NMR of 1,3-diazazulenes has been recorded despite they belong to a fundamental non-alternant conjugate system until a recent paper by Mathias and Overberger<sup>5</sup> which described an assignment of the  $^{13}\text{C}$ -NMR spectra of two 1,3-diazazulenes. We will herein report our works to revise a part of their assignment.

The NMR spectra of 1,3-diazazulene ( $\text{X}=\text{H}$ ) and some 2-substituted derivatives ( $\text{X}=\text{a-g}$ ) is compiled in Table 1.<sup>6</sup> Most of the samples were measured in  $\text{CDCl}_3$ , but some were measured in  $\text{CD}_3\text{SOCD}_3$ . The assignment of chemical shifts is based on the mutual comparisons, the measurements on the 4,6,8-trideuterio derivatives of  $\text{X}=\text{a}$ ,  $\text{X}=\text{c}$  ( $\text{X}=\text{OMe}$ ),  $\text{X}=\text{e}$  ( $\text{X}=\text{NMe}_2$ ),  $\text{X}=\text{g}$  ( $\text{X}=\text{SMe}$ ),  $\text{X}=\text{d}$  ( $\text{X}=\text{OH}$ ), and  $\text{X}=\text{f}$  ( $\text{X}=\text{SH}$ ), and on

the 1,3-d<sup>15</sup>N-derivatives of  $\lambda$ ,  $\lambda_a$ ,  $\lambda_c$ , and  $\lambda_e$ . First of all, the signals of  $\lambda$  appeared, from high to low field, C-5, C-4, C-6, C-9 and C-2, in the same order to the assignment by Overberger,<sup>5</sup> with the figures showing a good coincidence to the papers.<sup>4,5</sup> However, the chemical shifts of  $\lambda_g$  appeared in an order of C-4, C-6, C-5, C-9 and C-2, showing a contradiction with Overberger's results. They have made the assignment by use of 2-<sup>13</sup>C-enriched derivative of  $\lambda$ , but the assignment of  $\lambda_g$  seems to be deduced by an analogy. Our revised assignment is based on the following experiments.

Thus, Table 1 shows that throughout the series of compounds, the chemical shifts of C-5 and C-9 are nearly constant, while the others of the signals suffered a considerable effect due to the C-2 substituents. This must be reasonable since C-4, C-6 and C-8 are opposite terminals to C-2 in a formal conjugate system.

Table 1. The <sup>13</sup>C-NMR Chemical Shifts of Some 1,3-Diazazulene Derivatives ( in CDCl<sub>3</sub> or CD<sub>3</sub>SOCD<sub>3</sub> ).

X		C-2	C-4	C-5	C-6	C-9
$\lambda$	H	168.2	136.2	133.9	139.3	162.4
$\lambda_a$	OMe	179.3	130.4	134.9	134.6	163.6
$\lambda_b$	NHMe	175.1	125.2	134.9	130.4	165.2
$\lambda_c$	NMe <sub>2</sub>	171.8	124.8	134.7	129.6	165.5
$\lambda_d$	Ph	176.7	129.4	134.2	137.4	163.9
$\lambda_e$	SMe	182.4	130.7	134.6	135.9	163.4
$\lambda_f$	SCH <sub>2</sub> CH=CH <sub>2</sub>	181.4	130.8	134.5	135.9	163.4
$\lambda_g$	SCH <sub>2</sub> C≡CH	179.9	131.4	134.6	136.5	163.4
$\lambda_h$	SCH <sub>2</sub> Ph	181.6	130.9	134.5	135.9	163.5
$\lambda_i$	SCH <sub>2</sub> C <sub>6</sub> H <sub>4</sub> Me-p	181.7	130.8	134.5	135.9	163.5
$\lambda_j$	SCH <sub>2</sub> C <sub>6</sub> H <sub>4</sub> NO <sub>2</sub> -p	180.2	131.4	134.7	136.5	163.4
$\lambda_k$	SCH <sub>2</sub> COOMe	179.9	131.5	134.6	136.5	163.4
$\lambda_l$	SCH <sub>2</sub> COOEt	180.0	131.4	134.6	136.4	163.4
$\lambda_m$	Cl	168.6	135.2	134.7	139.3	162.7
$\lambda_n$	OH*	166.0	121.3	137.9	130.9	157.4
$\lambda_o$	NH <sub>2</sub> *	175.5	123.7	134.4	129.7	165.0
$\lambda_p$	SH*	188.3	122.9	138.2	133.0	158.3

The solvent dependencies of the chemical shifts were then examined with  $\lambda$ ,  $\lambda_a$  and  $\lambda_e$ , but except in the case of CF<sub>3</sub>COOD, no appreciable change of chemical shifts can be observed. This might be attributable to a substantial contribution of the polarized structures for  $\lambda$  and  $\lambda_e$  even in the non-polar solvents. The <sup>13</sup>C-NMR

spectra in  $\text{CF}_3\text{COOD}$  caused a considerable change of the chemical shifts for all the carbon atoms, but the substituent effect on the chemical shift,  $\Delta\delta_{\text{C}}^{\delta} = \delta_{\text{C}}^{\delta} - \delta_{\text{C}}^{\delta}$ , in this solvent, of C-5 [  $\Delta\delta_{\text{C}}^{\delta} = -2.2$ ,  $\Delta\delta_{\text{C}}^{\delta} = -1.9$  ] and C-9 [  $\Delta\delta_{\text{C}}^{\delta} = -1.7$ ,  $\Delta\delta_{\text{C}}^{\delta} = -2.1$  ] were again smaller than those of C-4 [  $\Delta\delta_{\text{C}}^{\delta} = 8.3$ ,  $\Delta\delta_{\text{C}}^{\delta} = 8.7$  ] and of C-6 [  $\Delta\delta_{\text{C}}^{\delta} = 6.0$ ,  $\Delta\delta_{\text{C}}^{\delta} = 4.9$  ] as calculated from the figures in Table 2. These also supported our assignment.

Table 2. The Solvent Dependence of the Chemical Shifts ( $\delta$ ) of Some 1,3-Diazazulene Derivatives.

		C-2	C-4	C-5	C-6	C-9
$\text{C}_5\text{H}_4\text{N}_2$ : X=H	$\text{CCl}_4$	168.4	135.2	132.4	137.6	162.3
	$\text{CDCl}_3$	168.2	136.2	133.9	139.3	162.4
	$\text{C}_6\text{D}_6$	169.2	135.7	132.8	138.0	163.0
	$\text{CD}_3\text{COCD}_3$	169.0	136.7	134.6	140.3	163.2
	$\text{C}_5\text{D}_5\text{N}$	169.0	136.3	134.0	139.6	163.0
	$\text{CD}_3\text{OD}$	167.2	137.8	136.6	142.4	162.3
	$\text{CD}_3\text{SOCD}_3$	167.8	136.1	134.3	140.1	161.8
	$\text{CF}_3\text{COOD}$	155.8	145.3	142.7	152.1	154.4
$\text{C}_5\text{H}_4\text{N}_2$ : X=SMe	$\text{CCl}_4$	182.4	129.8	133.3	134.3	163.2
	$\text{CDCl}_3$	182.4	130.7	134.6	135.9	163.4
	$\text{C}_6\text{D}_6$	183.6	130.3	133.8	134.8	164.1
	$\text{CD}_3\text{COCD}_3$	182.7	131.6	135.6	137.2	164.2
	$\text{C}_5\text{D}_5\text{N}$	182.5	131.1	134.9	136.4	163.9
	$\text{CD}_3\text{OD}$	182.3	132.3	137.0	138.7	163.8
	$\text{CD}_3\text{SOCD}_3$	180.6	130.1	135.2	137.0	162.7
	$\text{CF}_3\text{COOD}$	173.6	136.6	144.6	147.2	156.5
$\text{C}_5\text{H}_4\text{N}_2$ : X=OMe	$\text{CCl}_4$	179.4	129.2	133.5	132.9	163.4
	$\text{CDCl}_3$	179.3	130.4	134.9	134.6	163.6
	$\text{C}_6\text{D}_6$	180.6	130.0	134.1	133.5	164.3
	$\text{CD}_3\text{COCD}_3$	180.3	130.9	135.8	135.6	164.4
	$\text{C}_5\text{D}_5\text{N}$	179.9	130.6	135.2	135.0	163.9
	$\text{CD}_3\text{OD}$	179.1	131.9	137.3	137.4	163.7
	$\text{CD}_3\text{SOCD}_3$	178.6	130.3	135.3	135.3	162.8
	$\text{CF}_3\text{COOD}$	167.7	137.0	144.9	146.1	156.1

Finally, the assignment was assured by the isotope labellings: The  $^{13}\text{C}$ -NMR spectra of 4,6,8-trideuterio derivative of  $\text{C}_5\text{H}_4\text{N}_2$ , illustrated in Fig. 1, clearly retained the signals at  $\delta$  138.2, 158.3 and 188.3, to which we assigned as C-5, C-9, and C-2. In addition, the observations of the  $^{15}\text{N}$ - $^{13}\text{C}$ -couplings by the  $^{13}\text{C}$ -NMR measurements

on the di-<sup>15</sup>N-labelled compounds<sup>7</sup> of  $\mathcal{L}$ ,  $\mathcal{Z}_a$ ,  $\mathcal{Z}_c$ , and  $\mathcal{Z}_e$ . The NMR of this enriched sample of  $\mathcal{Z}_e$  is illustrated in Fig. 2.

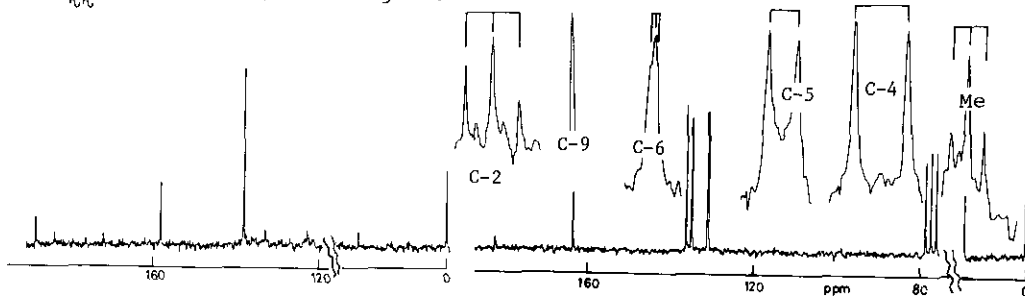


Fig. 1: The <sup>13</sup>C-NMR Spectrum of 4,6,8-Trideuterio Derivative of  $\mathcal{Z}_a$  ( in CD<sub>3</sub>SOCD<sub>3</sub> )

Fig. 2: The <sup>13</sup>C-NMR spectrum of Di-<sup>15</sup>N-Derivative of  $\mathcal{Z}_e$  ( in CDCl<sub>3</sub> )

Table 3 summarized the results of  $n_J(^{15}\text{N}-^{13}\text{C})$ .

Table 3. The  $n_J(^{15}\text{N}-^{13}\text{C})$  of some Di-<sup>15</sup>N-Azazulenes ( Hz ).

$n_J(\text{C-x})$	$^1J(\text{C-2})$	$^1J(\text{C-9})$	$^2J(\text{C-4})$	$^3J(\text{C-5})$	$^4J(\text{C-6})$	$^3J(\text{Me})$
$\mathcal{L}$	4.0	1.5	7.5	4.0	0	-
$\mathcal{Z}_a$	2.6	1.2	8.1	4.4	0	2.3
$\mathcal{Z}_c$	3.2	2.0	8.5	4.4	0	1.6
$\mathcal{Z}_e$	4.1	0	8.1	4.4	0.5	2.4

The above evidences are consistent in each other, and the assignment deduced should be true. Detailed features of the spectra will be discussed elsewhere.

#### References and Notes

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4. J. Tsunetsugu, S. Ito, H. Ohtani, T. Nishida, and K. Kushida, 'Abstract Papers of Third International Symposium of Heterocyclic Chemistry', p 523 ( Sendai, Aug., 1971 ).
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6. The <sup>13</sup>C-NMR spectra were obtained by measurements with an FX 100 Model Spectrometer ( JEOL Co. ) at 25 MHz with FT mode.
7. The <sup>15</sup>N-NMR of  $\mathcal{Z}_e$  in CDCl<sub>3</sub> exhibited a signal at 111.4 ppm up field from the external standard, NH<sub>4</sub>NO<sub>3</sub>. An <sup>15</sup>N-NMR study on the  $\mathcal{L}$  and  $\mathcal{Z}$  will be a subject of another paper.

Received, 31st January, 1979