

NEW ANALYSIS OF REACTIVITY AND REACTIONS OF
NITROGEN-CONTAINING HETEROCYCLES

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I. New Analysis of Reactivity. In order to interpret ^{13}C -chemical shifts in both aliphatic and aromatic series, novel inductive and mesomeric substituent parameters, l and μ , were proposed. Various spectral data and σ_j constants were correlated with $[a(l + b\mu) + c]$ ($b \geq 0$). Furthermore, novel effects such as 1,3-through space interaction, α - and β -lone pair effects and analogous behavior of C-H bond to C=C bond were proposed and various phenomena and reactivities were explained clearly.¹⁾ The above equation and novel effects were shown to be useful as new method for analysis of reactivity in various kinds of amines.

II. N-S Interaction and Bond-Switch in 5-Amidino-1,2,4-thiadiazoles and Related Systems. Several examples of ring transformation have been reported, in which π -sulfurane intermediate seems to be invoked. It was found that formation of 1:1 adducts (1) from 5-imino-4,5-dihydro-1,2,4-thiadiazoles with nitriles or imidates accompany bond switch on the sulfur atom, and charge transfer interaction from the imino nitrogen atom to the sulfur atom was demonstrated by ^{13}C -NMR and X-ray crystallography.²⁾ Alkylation of 1 gave mono- and di-alkylated compounds successively and bond switch occurred in each step.³⁾ The 1:1 adducts (2) between 5-amino-3-methyl-1,2,4-thiadiazole and nitriles were shown to be equilibrated mixture of α - and β -isomers by NMR.⁴⁾ Even by protonation of 1, bond switch to protonated product (3) was observed and rapid equilibration between 1 and 3 was found by NMR. Thus, it was demonstrated that bond switch occurs easily.

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