A CONVENIENT SYNTHESIS OF PYRIDAZINO[4,5-b]QUINOLINES AND PYRROLO[3,4-b]quinolines

Yoshihisa Kurasawa^{*} and Atsushi Takada School of Pharmaceutical Sciences

Kitasato University, Shirokane, Minato-ku, Tokyo 108, Japan

<u>Abstract</u> — Pyridazino $[4,5-\underline{b}]$ quinolines were conveniently synthesized from the reaction of N-(1,2-bisethoxycarbonylvinyl)-<u>o</u>-aminoacetophenone with hydrazines, and 2,3-bishydrazinocarbonyl-4-methylquinolines were easily cyclized to pyrrolo-[3,4-b]quinolines.

Spectroscopic and crystallographic studies have been reported concerning the structure of maleic hydrazides^{1,2,3,4,5} and other aromatic ring-condensed cyclic hydrazides^{6,7,8} (pyridazines), which are clarified to be an oxo-hydroxy (I-a,b), but not dioxo (I-c), form in solution and solid state, as shown in Chart 1. Pyridazino[4,5-b]quinolines were prepared by the reaction of dimethyl acridinates (II) with hydrazine hydrate,^{10,11} and its structure was proven to be the 1-oxo-4-hydroxy form by Godard.¹¹ Recently, we have also synthesized pyridazino[4,5-b]quinolines conveniently from the reaction of N-(1,2-bisethoxycarbonylvinyl)-o-aminoacetophenone (III)^{12,13} with hydrazine hydrate and methylhydrazine. We now report a facile synthesis of the above condensed quinolines.

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Refluxing of III (3.3 mmol) with hydrazine hydrate (6.6 mmol) in EtOH (15 ml) for 5 hr precipitated red needles, 4-hydroxy-10-methyl-1-oxo-1,2-dihydropyridazino[4,5-<u>b</u>]quinoline monohydrazinium salt (IV-a).¹⁴ From its filtrate, yellow needles, 2,3-bishydrazinocarbonyl-4-methylquinoline (V-a),¹⁵ were obtained. When V-a (0.88 mmol) was dissolved in H₂O (20 ml) and stirred for 30 min at room





temperature, colorless needles, 2-amino-9-methyl-1,3-dioxo-1,3-dihydropyrrolo $[3,4-\underline{b}]$ quinoline (VIa)¹⁶ (86%), were obtained. The reaction in the presence of 30-fold molar and equimolar amount of hydrazine hydrate against III mainly gave IV-a and V-a, respectively. A similar reaction of III (3.3 mmol) with methylhydrazine (33 mmol) also precipitated red needles, 4-hydroxy-2,10-dimethyl-1-oxo-1,2-dihydropyridazino[4,5-<u>b</u>]quinoline monomethylhydrazinium salt (IV-b).¹⁴ From the filtrate, a yellow powder (V-b) (M⁺=289) was obtained. In order to purify the yellow powder, it was submitted



m.p. 210°(discolor) 260-263°(dec)



(VI-a) R=NH₂ m.p. 269-270° (VI-b) R=NHMe m.p. 240-241°



(V-a) R=H m.p. 323°

[(V-b) R=Me]

Chart 2*



Table I.

Molar ratio		Yield (%)		
(111)	: NH2NH2·H20	(IV-a)	(V-a)	(VI-a)
1	1	trace	12	
1	2	13	53	(47)**
1	30	88		
(111)	: NH ₂ NHMe	(IV-b)	(1I-c)	(VI-b)
1	2	trace	94	
1	.10	30		27
1	30	94		

* Satisfactory mass spectral and elemental analytical data were obtained for all new compounds.

** overall yield from (III)

to column chromatography to afford colorless needles, 9-methyl-2-methylamino-1,3-dioxo-1,3-dihydropyrrolo[3,4-b]quinoline (VI-b)¹⁶ (M⁺=234). This indicated that VI-b was formed from V-b during the column chromatography. The reaction in the presence of 30-fold and 2-fold molar amount of methylhydrazine against III predominantly resulted in the formation of IV-b and diethyl 4-methylacridinate (II-c),^{12,13} respectively. In the molar ratio of 1:2, hydrazine hydrate produced IV-a and V-a, but not II-c. This may be due to the H₂O-content in the reaction media, which presumably varies the basicity or nucleophilicity of hydrazines. The above results are summarized in Table I, which suggests that III is converted to II-c, V, VII, and IV in sequence.

Treatment of the salts IV-a and IV-b with AcOH gave (VII-a) (93%) and (VII-b) (95%), respectively. The acetate (VIII-a) was obtained from both IV-a (56%) and VII-a (50%), and the acetate (VIII-b) was likewise obtained from both IV-b (80%) and VII-b (57%).

The compounds VI-a and VI-b were converted to IV-b (50%) and IV-a (30%) in the presence of excess of methylhydrazine and hydrazine hydrate, respectively. In this reaction, the intermediates were assumed to be the compounds V-a and V-b, from the results shown in Table I.

The compounds IV-a, VII-a, and VIII-a are assigned as the 1-oxo-4-hydroxy form, from the finding of Godard¹¹ described above. The compounds IV-b, VII-b, and VIII-b are also assumed to be the 1-oxo-4-hydroxy form, since the UV spectral patterns of these compounds are similar to those of IV-a, VII-a, and VIII-a, respectively. The signal for 10-methyl protons were observed at δ 3.40(VII-a), 3.38 (VII-b), 3.39(VIII-a), and 3.37(VIII-b), which were not varied significantly. Therefore, the compounds IV-b, VII-b, and VIII-b and VIII-b and VIII-a), and 3.37(VIII-b), which were not varied significantly.

In conclusion, we found that pyridazino[4,5-b]quinolines were synthesized directly from the enamine adduct III, and 2,3-bishydrazinocarbonyl-4-methylquinolines V-a and V-b were easily converted to pyrrolo[3,4-b]quinolines.

Table II.

Compound	λ _{max} (EtOH) nm (logε)
(IV-a)	215.0(4.36) 248.0(4.72) 287.0(3.94)
(IV-b)	214.0(4.39) 249.0(4.71) 293.0(3.98)
(VII-a)	216.0(4.28) 248.5(4.71) 282.5(3.93)
(VII-b)	216.0(4.34) 249.0(4.69) 290.0(3.96)
(VIII-a)	228.5(4.37) 253.5(4.69) 296.0(3.87) 340.0(3.74) 357.5(3.66)

References and Footnotes

- 1. O. Ohashi, M. Mashima, and M. Kubo, Can. J. Chem., 1964, 42, 970.
- 2. H. P. Fritz, F. H. Koehler, and B. Lippert, Chem. Ber., 1973, 106, 2918.
- 3. D. M. Miller and R. W. White, Can. J. Chem., 1956, 34, 1510.
- 4. A. R. Katritzky and A. J. Waring, J. Chem. Soc., 1964, 1523.
- 5. T. Otterson, Acta Chem. Scand., 1973, 27, 797; Idem, ibid., 1973, 845.
- 6. G. Adembri, F. DeSio, R. Neshi, and M. Scotton, J. Chem. Soc.(C), 1968, 2857.
- 7. A. Leberre and B. Dumaitre, Bull. Soc. Chim. Fr., 1970, 4376.
- 8. E. Domagalina, I. Kurpiel, and N. Koktysz, Rocktz. Chem., 1969, 43, 775.
- 9. G. B. Barlin, Aust. J. Chem., 1979, 32, 459.
- 10. D. Kreysig, G. Kempter, and H. H. Stroh, Z. Chem., 1969, 9, 230.
- 11. A. Godard, G. Queguiner, and P. Pastour, Bull. Soc. Chim. Fr., 1972, 1588.
- 12. E. C. Taylor and N. D. Hendel, J. Org. Chem., 1967, 32, 1666.
- J. B. Henderickson, R. Rees, and J. F. Templeton, <u>J. Amer. Chem. Soc.</u>, 1964, 86, 107.
- 14. Further reflux of (IV-a) and (IV-b) in solutiond of hydrazine hydrate and methylhydrazine in EtOH gave analytically pure samples, respectively.
- Compound (V-a): δ(DMSO-d₆) 9.22(2H, br.s, CON<u>H</u>NH₂), 8.47-7.50(4H, m, aromatic),
 4.17(4H, br.s, CONHN<u>H₂</u>), 2.65(3H, s, 4-Me), v(KBr) 3220(NH), 1655 and 1630(C=0).
- 16. δ(DMSO-d₆) 8.47-7.73(4H, m, aromatic), 5.15(2H, s, 2-NH₂), 3.03(3H, s, 10-Me), (VI-a); 8.43-7.63(4H, m, aromatic), 5.85(1H, q, J=6 Hz, 3-NHMe), 3.02(3H, s, 10-Me) 2.60(3H, d, J=6 Hz, 3-NHMe) (VI-b). ν(KBr) 1773 and 1700 (C=0) (VI-a), 1770 and 1715 (C=0) (VI-b). λ_{max} (EtOH) nm(logc) 215.0(4.36), 248.0(4.72), 287.0(3.94) (VI-a), 214.0(4.39), 249.0(4.71), 293.0(3.98) (VI-b).
 G. Adembri, F. DeSio, R. Neshi, and M. Scotton, -

J. Hetero. Chem., 1976, 13, 1155.

 $\delta(DMSO-d_6)$ 5.25

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