

THE REACTION OF O-ETHYL SUCCINIMIDE WITH ARYLLITHIUM COMPOUNDS :
AN EFFICIENT METHOD FOR THE INTRODUCTION OF 3-ETHOXYCARBONYL-
PROPIONYL GROUP TO AROMATIC COMPOUNDS

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Abstract -- O-Ethyl succinimide (I) was found to be an efficient reagent for the introduction of 3-ethoxycarbonylpropionyl group (-COCH₂CH₂COOEt) to aromatic compounds by a single operation under mild conditions.

In the previous communication¹⁾, we have demonstrated synthetically useful reactions of O-ethyl succinimide (I)²⁾ with primary and secondary amines, especially o-aminophenylcarbonyl derivatives affording quinazolones and quinazolines in good yields. In this paper we wish to report an efficient synthetic method for the introduction of 3-ethoxycarbonylpropionyl group to aromatic compounds by the use of O-ethyl succinimide (I) and aryllithium compounds.

It has been reported that the reaction of I with active methylene compounds such as ethyl cyanoacetate and ethyl acetoacetate gave 5-alkylidene-2-pyrrolidones (II)³⁾. We attempted to apply this reaction to other nucleophiles. In this process, it was found that the reaction of I with aryllithium gave directly ethyl 4-aryl-4-oxobutyrates (III) in moderate yields. Several examples are listed in Table. We also found that lithium acetylide reacted with I to give III (Run 6).

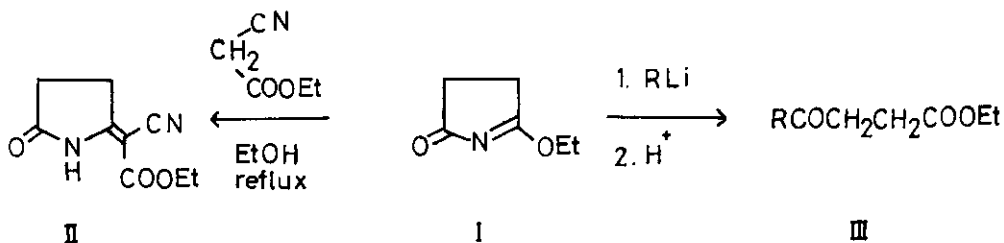


Table. Reaction of I with Aryllithium (or Lithium Acetylide)^{*1}

Run	Starting Material	Product ^{*2}	Isolated ^{*3} Yield (%)	bp ^{*4} (or mp) (°C)
1	bromobenzene		67	132 (2 mmHg)
2	p-dibromobenzene	Br-	69	(57-58)
3	p-bromochlorobenzene	Cl-	69	(56-57)
4	m-bromotoluene	CH ₃ -	54	110 (1 mmHg)
5	1-bromonaphthalene		67	157-160 (1 mmHg)
6	phenylacetylene		40	130 (1 mmHg)

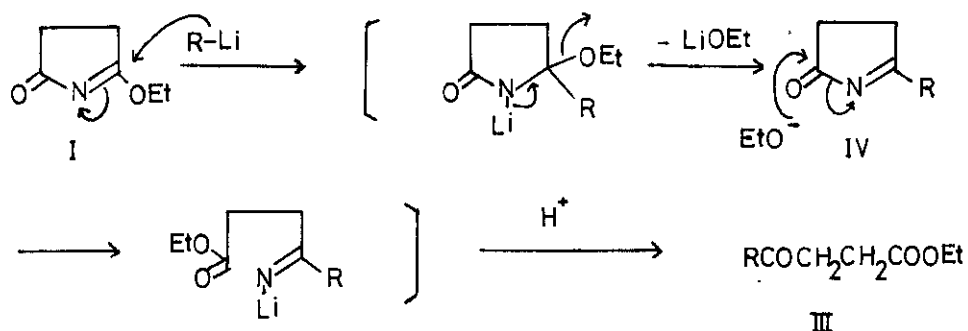
*1 Organolithium compounds were prepared by mixing of starting materials and n-BuLi in ether at 0° for 15 min.

*2 All products gave satisfactory analytical results and spectral data.

*3 Yields have not been optimized.

*4 Bp is the bath temperature of Kugelrohr.

The mechanism for this reaction is proposed as follows : Nucleophilic attack of the



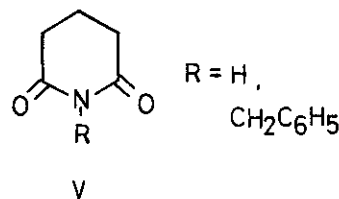
ethoxide anion generated from the addition-elimination reaction would take place at the carbonyl carbon of intermediate (IV) to give ketoester (III) after the cleavage of C-N bond and acidic work-up.

A typical procedure is described for the synthesis of ethyl 3-benzoylpropionate (Run 1) :

A solution of n-BuLi (3.3 mmol) was added at 0° to a solution of bromobenzene (3

mmol) in 5 ml of dry ether under argon. After stirring at 0° for 15 min, a solution of I (3 mmol) in 2 ml of dry ether was added at -78°. The reaction mixture was stirred at -78° for 1 h and then at room temperature for 1 h. By usual work-up, chromatography, and distillation, a pure colorless oil was obtained in 67% yield.

In conclusion, it is noted that this reagent (I) is useful and convenient for the direct introduction of 3-ethoxycarbonylpropionyl group on the lithiated position of aromatic compounds by a single operation under mild conditions. This method could overcome the disadvantage of Friedel-Crafts acylation⁴⁾ which is not always selective for aromatic substitution reactions. As it is reported⁵⁾ that glutarimides (V), homologs of succinimide, reacted with phenyllithium to give 6-hydroxy-6-phenyl-2-oxopiperidines, the reaction of I with aryllithium, different from the reaction of real imides, provides an alternative utility.



Research on the scope and limitation of this reaction is currently being investigated.

REFERENCES AND NOTE

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2. This material (bp 112°/2 mmHg), prepared from Ag salt of succinimide and EtI by the method of literature [K. Matoba and T. Yamazaki, *Chem. Pharm. Bull. (Tokyo)*, 1974, 22, 2999] is stable for at least several months by the storage under argon in the refrigerator.
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