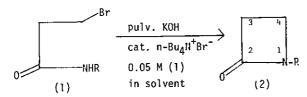
## A CONVENIENT SYNTHESIS OF MONOCYCLIC B-LACTAMS

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Abstract: The intramolecular N-alkylation of  $\beta$ -bromopropionamide (1) under phase transfer conditions gave monocyclic  $\beta$ -lactams (2) in high yields.

Synthesis of 2-azetidinones, the basic structure unit of the  $\beta$ -lactam antibiotic, is attractive to organic chemists as a synthetic target. Formation of these four-membered rings has been approached from nearly every conceivable way. I In continuation of our work on N-alkylation of lactams under phase transfer conditions, 2 we now report a facile synthesis of \beta-lactams by the formation of N-C $_4$  bond, which mimics the proposed biosynthesis,  $^3$  by cyclization of  $\beta$ -bromopropionamides (1), readily available from coupling of  $\beta$ -bromopropionylchrolide with amines, <sup>4</sup> in solid-liquid system.5



R= A.  $C_6H_5$  B. p-MeOC<sub>6</sub>H<sub>4</sub> C. p-C1C<sub>6</sub>H<sub>4</sub> J. cyclohexyl K. n-propyl

A typical procedure for the formation of  $\beta$ -lactams is as follows. To a suspension of pulverlized KOH (5.5 mmol) and  $n-Bu_AN^+Br^-$  (1 mmol) in 50 ml of dry  $CH_2Cl_2$  at room temperature was added a solution containing N-3-bromopropionyl-2-phenylglycine methyl ester (1H) (5 mmol) in 50 ml of dry CH2Cl2 over 6 hr with stirring. After completion of the addition, the reaction mixture was stirred for 30 min. The precipitate was filtered off and then washed with CH2Cl2. After removal of the combined solvent, the residue was purified by column chromatography on silica gel (CH $_2$ Cl $_2$ -MeOH) (50:1) to give the desired  $\beta$ -lactam (2H) in 83 % yield. The spectral data [  $\nu$  c=0 (neat) 1735 and 1725 cm<sup>-1</sup>;  $\delta$ (CDC1<sub>2</sub>) 2.83-3.27 (3H, m), 3.67 (1H, t, J=3.5 Hz), 3.83 (3H, s, COOMe), 5.67 (1H, s, CHCOOMe), 7.45 (5H, s, ArH); m/e 219 ( $M^{\dagger}$ )] and elemental analysis (molecular formura  $C_{12}H_{13}NO_3$ ) supported this assignment.

Table

Compd. <sup>a</sup>	mp (°C) <sup>b</sup>	Yield (%)	IR v c=o (cm <sup>-1</sup> ) <sup>c</sup>
(2A)	79-81	94	1730
(2B)	104-5	92	1725
(2C) <sup>d</sup>	137-9	94	1730
(2D) <sup>d</sup>	162-4	81	1745
(2E) <sup>d</sup>	52-3	91	1740
(2F)	oil	86	1740
(2G) <sup>d</sup>	oil	85	1745
(2H)	oi1	83	1735
(2I) <sup>e</sup>	oll	83 (50) <sup>f</sup>	1740
(2J) <sup>e</sup>	011	63 (74) <sup>f</sup>	1730
(2K)	oil	67 (94) <sup>f</sup>	1745

a) All new compounds gave satisfactory elemental analyses.

The results are summarized in Table. The cyclization of amides (1) to the  $\beta$ -lactams (2) was dependent on the concentration of the solution, the addition rate of amides to the base and the solvent. Both the high concentration over 0.05 M and the rapid addition of the amides resulted in low yields of  $\beta$ -lactams, along with the formation of substantial amounts of N-alkyl acrylz amides. With regard to the solvent employed, the  $\beta$ -lactams (2A-I) could be prepared in good yields by using  $\mathrm{CH_2Cl_2}$ , however, the use of THF for their cyclization caused complication. On the other hand, the use of THF was favoured over that of  $\mathrm{CH_2Cl_2}$  for cyclization of (1J) and (1K). In paticular, this reaction proceed at room temperature, the procedure is simple, straightforward, and easy to work up, the formation of by-products was scarcely caused, and the desired products were obtained in high yields. In conclusion, this procedure under phase transfer conditions is more convenient for N-alkyl monocyclic  $\beta$ -lactams syntheses.  $^{1,6,7,8}$  In addition, the  $\beta$ -lactams thus readily obtained have high reactivities (Fries rearrangement,  $^9$ 

b) The oil compounds were purified by chromatography, because the distillation resulted in partial decomposition of  $\beta$ -lactams.

c) (2A-E) (nujo1), (2F-K) (neat).

d) The solvent  $(CH_2Cl_2: CH_3CN)$  (19:1) was used.

e) The acrylamides of by-products were obtained in 14% and 6% yields.

f) The THF was used as a solvent.

C-C bond formation at  $C_3$ -position,  $^{10}$  azidation at  $C_3$ -position,  $^{11}$  oxidation at  $C_4$ -position,  $^{12}$  and so on  $^{13}$ ). Therefore, they would be served as potential synthetic intermediates as well as those of natural products containing the  $\beta$ -lactam ring.  $^{11}$ 

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Received, 26th January, 1980