SYNTHESIS OF 2-AZABICYCLO [3.3.1] NONANES.

Joan Bosch<sup>\*</sup> and Josep Bonjoch. Department of Organic Chemistry, Faculty of Pharmacy, University of Barcelona, Barcelona, Spain.

All methods for the synthesis of 2-azabicyclo [3.3.1] nonanes paying special attention to those which lead to functionalized systems are reviewed. This system, present in several alka loids, is synthetically and theoretically interesting. The synthe tic routes are classified according to the method used for the f<u>1</u> nal cyclization to the morphan system.

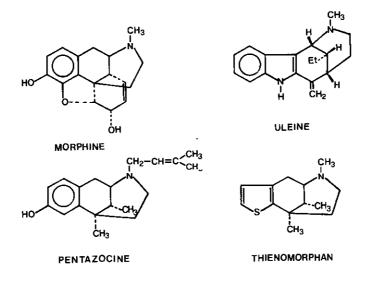
#### CONTENTS

- 1. Introduction.
- 2. First syntheses: cyclization by lactamization.
- 3. Pheny1morphans.
- 4. N-Chloramines route.
- 5. Intramolecular Michael-type cyclization.
- Other syntheses with formation of piperidine ring in the last stage.
- Formation of carbocyclic ring in the last step: Dieckmann cyclization.
- 1. Introduction.

2-AZABICYCLO 3.3.1 NONANE

The 2-azabicyclo [3.3.1] nonane system<sup>1</sup> is present in many compounds, both natural and synthetic ones. This grouping occurs in morphine<sup>2</sup>, the main opium alka-

loid, as well as in a great number of heterogeneous indole alkaloids<sup>3</sup>, such as the strychnos alkaloids (strychnine, tubifoline, condyfoline), the picralime alkaloids (akuammine, akuammicine), the aspidospermum alkaloids (uleine, dasycarpidone), and the geissospermum alkaloids (geissoschizoline); it is also present in some calabash curare alkaloids as fluorocurarine.



Among the synthetic molecules containing the 2-azabicyclo [3.3.1] nonane nucleus, morphinans<sup>4</sup> (e.g. levorphanol) and 6,7-benzomorphans<sup>4,5</sup> (e.g. pentazocine) are specially interesting since they are strong synthetic analgesics. Heteromorphans<sup>6</sup>, bioisosters of 6,7-benzomorphans, in which the benzene ring is substituted by an heteroaromatic ring are also included. In the last years, some thieno-<sup>6-9</sup>, benzo[b]thieno-<sup>10</sup>, pyrido-<sup>11-13</sup>, indolo-<sup>14,15</sup>, furo-<sup>16</sup>, and pyrrolomorphans<sup>17</sup> have been described.

Other compounds including the 2-azabicyclo [3,3,1] nonane system are the structurally interesting 2-azaadamantane<sup>18</sup>, some cyclohex[j](and[d]) indolo[2,3-f] morphans<sup>19-21</sup>, of interest in biogenetic type syntheses of indole analogues of morphine alkaloids, and some methanobenzofuro[3,2-d]azocines prepared as transformation products of thebaine<sup>22</sup>.

The systems of 2-azabicyclo [3.3.1] nonane, especially the functionalized ones, are of great interest for the study of some structure-reactivity relationships and for being intermediates in the synthesis of more complex structures.

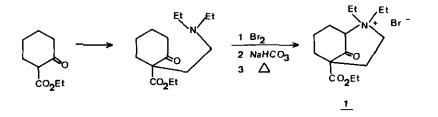
In the field of structural chemistry, 2-azabicyclo [3.3.1] nonanes allow to extend the studies, now in rapid developing, which are carried out with their carbocyclic analogues, bicyclo [3.3.1] nonanes  $^{23,24}$ . Thus, through these type of bridged systems, the reach of the Bredt's rule<sup>25</sup> was studied, the bicyclo [3.3.1] non-1-ene<sup>26</sup> and its corresponding 2-aza analogue<sup>27</sup> were synthesized and their unusual reactivity was examined  $^{26,27}$ . The conformational analysis  $^{28,29}$  and the stability of noneno lizable 1,3-dicarbonyl derivatives of these systems  $^{30}$  are further aspects of their structural interest.

In the field of organic synthesis, functionalized 2-azabicyclo[3.3.1] nona nes have been used as intermediates in the preparation of model structures related to natural products, especially of indole alkaloids<sup>31</sup>, and as precursors of complex polycyclic systems<sup>32</sup>. They have also been used as intermediates in the preparation of potentially active systems from the pharmacological standpoint<sup>12,33</sup>. In this a<u>s</u> pect some 2-azabicyclo[3.3.1] nonanes are themselves interesting, just as it happens with 5-phenylmorphans<sup>34</sup>, strong analgesics. Moreover, the study of the structureactivity relationships in synthetic analgesics has been enlarged through them<sup>35</sup>.

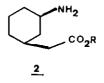
In this review on the synthesis of 2-azabicyclo[3.3.1] nonanes we are not going to take into account those approaches which lead to systems in which this grouping is part of a more complex polycyclic structure.

# 2. First syntheses: cyclization by lactamization.

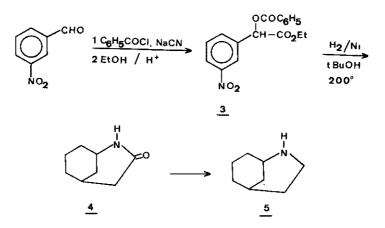
In 1947 Barltrop<sup>36</sup> reported the first reference about 2-azabicyclo[3.3.1] nonanes, in the context of the synthesis of cyclic ring systems ocurring in the mor phine molecule. It describes an approach to this system by alkylation of ethyl 2oxocyclohexanecarboxylate with 2-chloroethyldiethylamine and further cyclization through the 6-bromo derivative, to give the ammonium salt <u>1</u> in a yield lower than 1%.



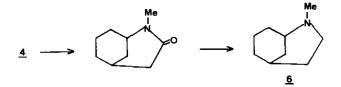
The preparation of 2-azabicyclo[2.2.2]octane by copper chromite<sup>37</sup> reduction of the lactam of <u>cis</u>-4-aminocyclohexanecarboxylic acid suggested to Cronyn<sup>38</sup> that the unsubstituted 2-azabicyclo[3.3.1] nonane (<u>5</u>) could be similarly synthesized by lactamization of <u>cis</u>-3-aminocyclohexaneacetic acid (<u>2</u>, R=H) or of that of its <u>co</u> rresponding ester (<u>2</u>, R=Et) and subsequent reduction of the resulting lactam <u>4</u>.



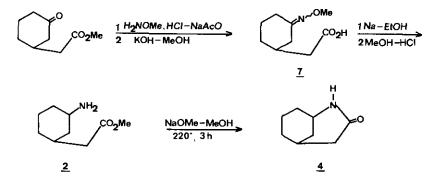
The synthesis was achieved from ethyl <u>m</u>-nitro-O-benzoylmandelate (<u>3</u>) using different procedures. The best one was the direct hydrogenation over Raney nickel in <u>tert</u>-butanol at 200°C, since in these conditions cyclization occurs and the lactam <u>4</u> was obtained in 35% of overall yield. Finally, the lactam <u>4</u> was hydrogenated to 2-azabicyclo [3.3.1] nonane (5) over copper chromite in tert-butanol (46% yield).



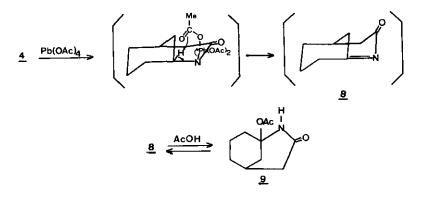
A parallel synthesis of Ginsburg<sup>39</sup> introduced some variations in the experimental aspect. Thus, reduction of ethyl <u>m</u>-nitrophenylacetate in glacial acetic acid using Adam's catalyst gave the aminoester  $\underline{2}$  (R=Et) in good yield, and subsequent lactamization by heating at 150°C afforded  $\underline{4}$  in 84% yield. Final reduction to the morphan system  $\underline{5}$  was carried out with lithium aluminum hydride. Later on, an analogous synthesis of  $\underline{4}$  from <u>m</u>-nitrophenylacetic acid has been described  $\underline{35,40}$ . N-Methylation with sodium hydride and methyl iodide gave 2-methyl-2-azabicyclo[3.3.1] nonan-3-one, and this lactam was reduced to the N-methylmorphan  $\underline{6}$  with lithium aluminum hydride



In a study of 2-azabicyclo [3.3.1] non-1-ene systems as reaction intermedia tes with a bridgehead double bond a new synthesis of the lactam <u>4</u> and of some C<sub>1</sub>substituted derivatives were reported<sup>27</sup>. The aminoester <u>2</u> was obtained from methyl 3-oxocyclohexaneacetate through 0-methyloxime <u>7</u> and subsequent reduction and esterification. Heating of <u>2</u> in the presence of sodium methoxide afforded lactam <u>4</u> in 50-60% yield.



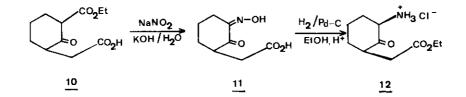
Treatment of <u>4</u> with lead tetraacetate gave 1-acetoxy-3-oxo-2-azabicyclo [3.3.1] nonane (<u>9</u>), being 2-azabicyclo[3.3.1] non-1-ene <u>8</u> the probable intermediate. From acetoxylactam <u>9</u>, some substitution reactions at C<sub>1</sub>-position were carried out, and 1-methoxy- and 1-cyanoderivatives were obtained, their formation involving again the presence of an imino intermediate of type 8.

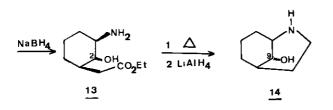


Similarly, when the alkaloid methyl homosecodaphniphyllate, which contains the 2-azabicyclo [3.3.1] nonane system, was treated with lead tetraacetate the formation of an unusually anti-Bredt's -rule imme containing the 2-azabicyclo [3.3.1] non-1-ene system was described<sup>41</sup>.

The preparation of 9-hydroxy-2-azabicyclo[3.3.1] nonane<sup>42</sup> is one of the few approaches to C<sub>9</sub>-functionalized morphans, and together with the precedent one are the only ways to morphan systems which involve the formation of an amide bond at the last step without using aromatic compounds as starting materials.

From 3-carbethoxy-2-oxocyclohexaneacetic acid (<u>10</u>) piperidine ring was for med. Thus, the replacement of the carbethoxy group by the oximino group gave <u>11</u>, which was catalytically hydrogenated affording the hydrochloride of the aminoester <u>12</u>. Reduction of this hydrochloride with sodium borohydride yielded (36%) the isolable crystalline isomer ethyl <u>c</u>-3-amino-<u>t</u>-2-hydroxycyclohexaneacetate (<u>13</u>). On he<u>a</u> ting at 150-160°C and further reduction with lithium aluminum hydride of the resulting lactam, the 9-hydroxymorphan <u>14</u> was obtained. When reduction of <u>12</u> was carried out by catalytic hydrogenation in the presence of platinum oxide, the C<sub>2</sub>-epimer alcohol of <u>13</u> was obtained. However, its cyclization afforded the same lactam than that of <u>13</u>, thus indicating that an epimerization is produced during the heating process.

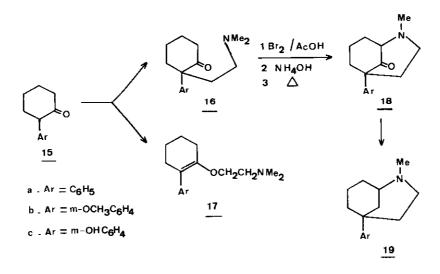




## 3. Phenylmorphans.

Once the nucleus of 2-azabicyclo [3.3.1] nonane was synthesized, the interest was focused on the preparation of 2-methyl-5-phenyl-2-azabicyclo [3.3.1] nonanes, sin ce this type of substances possess the three structural features necessary for a compound to have an analgesic activity similar to morphine<sup>4</sup>. They are (a) a phenyl nucleus, (b) a quaternary carbon attached to this nucleus, and (c) a tertiary nitrogen two carbon atoms removed from the quaternary carbon.

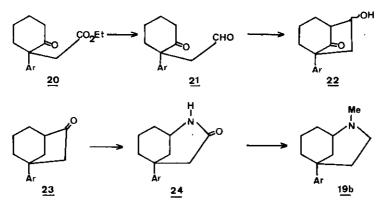
The main problem of the first syntheses of morphans <u>18</u>  $(Ar=C_6H_5^{43} \text{ and } Ar= \underline{m}-OHC_6H_4^{44})$  was the low yield (20%) in the formation of 2-dimethylaminoethyl-2-aryl cyclohexanones <u>16</u> owing to the competition of the O-alkylation process (60%)<sup>44,45</sup>. The predominance of this process is probably due to steric hindrance to C-alkylation and to stabilization of the enolate of <u>15</u> by the aromatic nucleus. However, after House's studies on the factors as favoring C-alkylation<sup>46</sup>, the synthesis was improved and <u>16b</u> was formed from <u>15b</u> in 40% yield using dimethylformamide, sodium hydride and 2-dimethylaminoethylchloride hydrochloride at room temperature. Since <u>15b</u> could be regenerated from enol ether <u>17b</u>, the process appears then to be more interesting<sup>34</sup>.



The synthesis goes on  $^{43,44}$  converting <u>16a</u> and <u>16b</u> into their corresponding 6-bromoderivatives, later cyclization in basic media at room temperature, and dry distillation of the resulting ammonium salts. Finally, Wolff-Kishner reduction of

ketones <u>18a</u> and <u>18b</u> gave the phenylmorphan nucleus <u>19</u>. Through the <u>m</u>-methoxyphenylderivative <u>19b</u>, the corresponding phenol <u>19c</u> and its acetylderivative were obtained<sup>44</sup>.

In an alternative synthesis of  $\underline{19b}^{34}$ , alkylation of  $\underline{15b}$  with ethyl bromoacetate (sodium amide-ether) gave the C-alkyl product  $\underline{20}$  in a 80% yield. Reduction of  $\underline{20}$  with lithium aluminum hydride afforded (98%) a diol that, by oxidation under controlled conditions (chromic oxide, pyridine) gave (77%) the keto aldehyde  $\underline{21}$ . In tramolecular aldol condensation of  $\underline{21}$  yielded (45%) a mixture of epimeric keto alcohols  $\underline{22}$ , Wolff-Kishner reduction of which (93%) followed by oxidation of the hydroxy function gave (73%) 1-m-methoxyphenylbicyclo [3.2.1] octan-6-one ( $\underline{23}$ ).

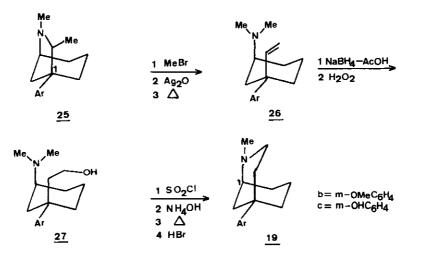


Ar = m-OCH<sub>3</sub>C<sub>6</sub>H<sub>4</sub>

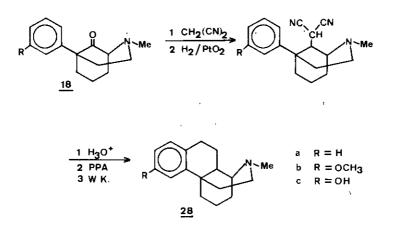
Formation of its oxime (64%) and subsequent Beckmann rearrangement, critical step of this synthesis (8%), afforded the morphan nucleus  $\underline{24}$ . Methylation of  $\underline{24}$  followed by reduction with diborane gave quantitatively phenylmorphan  $\underline{19b}$ .

Due to the strong pharmacologic action of  $({}^{+})$ -5-(<u>m</u>-hydroxypheny1)-2-methy<u>1</u> morphan <u>19c</u><sup>44</sup>, the activity of both enantiomers, previously resolved with <u>d</u>-mandelic acid<sup>34,47</sup>, was evaluated. They possess significant enantiomeric stereoselectivity in their biological actions. Thus, the <u>levo</u> isomer, an analgesic with morphine-like <u>po</u> tency, exhibits a weak narcotic antagonist activity and only a very slight physical dependence capacity<sup>47</sup>. The <u>dextro</u> isomer, four times more potent analgesic than mo<u>r</u> phine, has no antagonist activity and has a high physical dependence capacity<sup>47,48</sup>. The N-propyl, allyl, and cyclopropylmethyl derivatives of (+)-<u>19c</u> and its racemate were also prepared and their analgesic and antagonist activities studied<sup>48</sup>. Later on, the absolute configuration of the analgesic agonist-antagonist (-)-19c was established<sup>29</sup> to be 1<u>R</u>, 5<u>S</u> by single-crystal X-ray analysis of its hydrobromide salt. Both rings of the 2-azabicyclo[3.3.1] nonane system exist in chair conformations with the phenyl and methyl substituents equatorial.

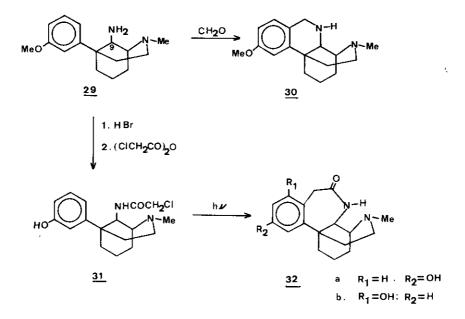
On the other hand, the enantiomer  $(+)-\underline{19c}$  was stereochemically correlated<sup>49</sup> with the normorphan  $(+)-(\underline{1R},\underline{5S},\underline{7R})-1-(\underline{m}-hydroxyphenyl)-6,7-dimethyl-6-azabicyclo <math>[\underline{3.2.1}]$  octane  $(\underline{25c})$ , compound of interest for its analgesic and antagonist properties. Thus, the methoxy derivative  $(-)-\underline{25b}$ , a precursor of  $(+)-\underline{25c}$ , gave the olefin  $(+)-\underline{26b}$  as the sole product on Hofmann degradation. Hydroboration and subsequent oxidation of  $(+)-\underline{26b}$  afforded the carbinol  $(+)-\underline{27b}$  which through the respective chloride was turned into methochloride of morphan  $\underline{19b}$ . Dry distillation followed by 0-demethylation gave a product identical in all respects to an authentic sample of the <u>dextro</u> isomer of phenylmorphan  $\underline{19c}$ . As in the process the configuration of the carbon bonded to the aromatic ring was not modified, this series of reactions provides an additional proof for the absolute configuration of normorphan  $(+)-\underline{25c}$ .



5-Phenylmorphans, with pharmacological interest on their own, have been used when functionalized as intermediates in the synthesis of more elaborated products. Thus, a Knoevenagel reaction on ketone <u>18</u> allowed obtention of tetracyclic compound <u>28</u><sup>43,50</sup>, a position isomer with respect to nitrogen attachment of N-methy<u>1</u> morphinan.

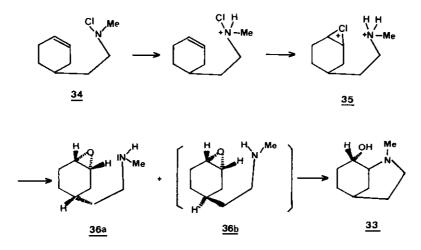


Likewise,  $9-\underline{cis}$ -amino-5-(<u>m</u>-methoxyphenyl)-2-methyl-2-azabicyclo[3.3.1]nonane (<u>29</u>), prepared <sup>32</sup> from ketone <u>18b</u>, leads by Pictet-Spengler cyclization to the tetracyclic system <u>30</u><sup>32</sup> and by photolysis of its chloroacetamido derivative <u>31</u> to 4,11b-propanopyrido[4,3-a][3]benzazepine systems of type <u>32</u><sup>51</sup>.

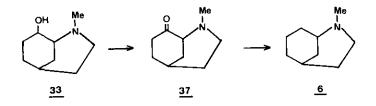


### 4. N-Chloramines route.

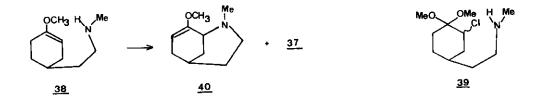
The first synthesis of a C-8 functionalized 2-azabicyclo[3.3.1] nonane was described in 1970. It concerns the obtention of 2-methyl-8-hydroxy-2-azabicyclo[3.3.1] nonane ( $\underline{33}$ )<sup>52</sup>, based on the easiness of N-chloramines for intramolecular addition to olefinic double bonds<sup>53,54</sup>. To explain such reactions, the formation of ni trenium ions<sup>55</sup> via an heterolytic process in neutral protic media (methanol reflux or tetrahydrofuran-water in the presence of silver ion catalysis) was postulated. Functionalized 6-azabicyclo[3.2.1] octanes<sup>54</sup> have been prepared in these conditions. The procedure was not applicable to the synthesis of morphan  $\underline{33}^{52}$  and it was necessary to resort to a solvolysis in 1 M sulfuric acid. The mechanism differs in the sense that chloronium ions formed by ionization of chloramine  $\underline{34}$  once protonated are now involved. Their addition to the double bond implies the formation of the cyclic chloronium cation  $\underline{35}$ , which when neutralizing the acidic aqueous reaction media would suffer a solvolysis with the formation of the diastereomeric epoxides  $\underline{36}$ .



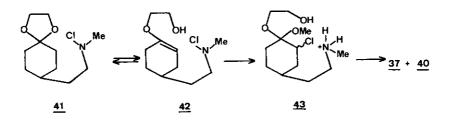
Intramolecular opening of the epoxide ring of the <u>trans</u> isomer <u>36b</u> yielded (37%) 2-methyl-8-hydroxy-2-azabicyclo [3.3.1] nonane (<u>33</u>), which was converted into 2-methyl-2-azabicyclo [3.3.1] nonane (<u>6</u>) upon oxidation with Jones reagent, thioacet<u>a</u> lization of the resulting ketone 37, and subsequent Raney nickel desulfurization.



Alternatively, ketone  $\underline{37}$  has been synthesized  $^{56,57}$  by intramolecular reaction under acidic conditions of an N-chloramine with the activated double bond of an enol ether, this reaction being already used in similar cyclizations  $^{58}$ . The required enol ether  $\underline{38}$  was prepared in six steps from p-hydroxyphenylacetic acid in 27% overall yield. The corresponding N-chloramine was obtained by reaction of  $\underline{38}$  with 1 M sodium hypochlorite solution, and by trifluoroacetic acid treatment it yielded a mixture (96%) of  $\underline{37}$  and  $\underline{40}$  in 4:1 ratio. The process involved again a chloronium ion addition to an activated double bond. When cyclization was carried out in methanol instead of pure trifluoroacetic acid, the formation in high yields of epimeric chloro ketals  $\underline{39}$  was described. These can be cyclized (75%) by acidic hydrolysis followed by base treatment to a 4:1 mixture of the elimination-cyclization product  $\underline{40}$  and the ketone  $\underline{37}$ . Acid hydrolysis of the enol ether  $\underline{40}$  afforded almost quantitatively the same ketone  $\underline{37}$ . Finally, morphan <u>6</u> was obtained by hydrogenolysis for  $\underline{37}$  in acidic conditions with platinum oxide, the procedure having been described for some azabicyclic ketones  $\underline{59}$ .

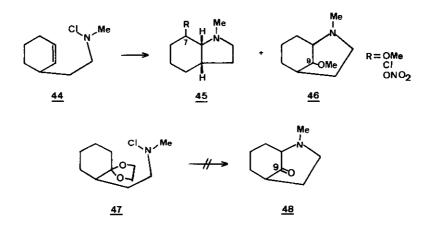


The synthesis of ketone  $\underline{37}$  has also been reported by cyclization of the Nchloramine corresponding to  $\underline{38}$  in the presence of Lewis acids. The best yields (40-50%)<sup>60</sup> were obtained with  $2nX_2$  type agents in methylene chloride at 0°. The same ketone  $\underline{37}$  was also obtained by cyclization of dioxolane  $\underline{41}$  under acid solvolysis conditions (anhydrous trifluoroacetic acid-methanol solution) through an electrophilic chlorination of the enol ether <u>42</u> formed from <u>41</u> and subsequent intramolecular displacement of the halide <u>43</u> by the amino group. Thus, ketone <u>37</u> was obtained in 50% yield, together with the enol ether <u>40</u>  $(10-30\%)^{57}$ .



As it can be deduced from the methods reviewed in this section, the N-chloramines route permits to obtain only C-8 functionalized 2-azabicyclo[3.3.1] nonanes. Contrary to what it occurs in the 2-azabicyclo[3.2.1] octane series<sup>53</sup> where the

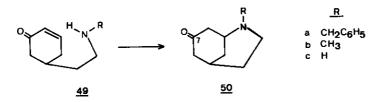
bridghed carbon can be functionalized by this procedure, the application of this type of reactions as potential way to C-9 functionalized systems has been unsuccessful<sup>61</sup>.



Thus, the silver nitrate catalyzed methanolysis of the N-chloramine  $\underline{44}$  took place<sup>61</sup> with formation of 7-substituted 1-methyl-<u>cis</u>-octahydroindole systems  $\underline{45}$  and less than 1% of the desired 2-methyl-9-methoxy-2-azabicyclo[3.3.1] nonane ( $\underline{46}$ ). On the other hand, N-chloramine  $\underline{47}$  failed to cyclize to the expected azabicyclic ke tone  $\underline{48}$ , giving instead complex reaction mixtures<sup>57</sup>.

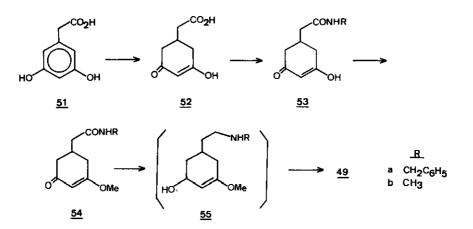
### 5. Intramolecular Michael-type cyclization.

The synthesis of C-7 functionalized 2-azabicyclo [3.3.1] nonanes can be achieved by means of an intramolecular Michael-type cyclization of an amine group upon an  $\alpha,\beta$ -unsaturated ketone.

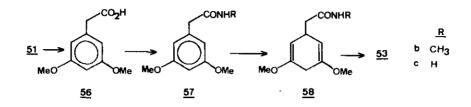


The synthesis<sup>62,63</sup> of 6-azabicyclo[3.2.1] octan-3-ones through an 1,4 intra molecular addition suggested two independent but analogous syntheses of 2-methyl-2-azabicyclo[3.3.1] nonan-7-one  $(50, R=CH_2)^{12,64}$ .

From diethyl 3-oxoglutarate, 3,5-dihydroxyphenylacetic acid 51 was obtained<sup>64</sup>, and hydrogenated (77% yield) to the dione 52 with rhodium on alumina in basic media at high temperature. The formation of amide 53 requires a laborious process. Treatment of 52 with two equivalents of isobutyl chloroformate did not only activa te the carboxyl group by forming a mixed anhydride but also protect the enolic hydroxyl as the carbonate ester. Subsequent treatment with benzylamine or methylamine, followed by hydrolysis of the intermediate carbonate and neutralization afforded the amides 53a and 53b, respectively, which were converted into enol ethers 54 with

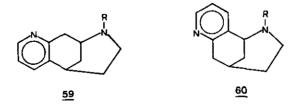


methanol and <u>p</u>-toluenesulfonic acid. Reduction of <u>54</u> with lithium aluminum hydride in tetrahydrofuran gave the unisolated amino alcohols <u>55</u> which through acid hydrolysis followed by basic treatment yielded the desired 2-alkyl-2-azabicyclo [3.3.1] nonan-7-ones (<u>50</u>) through the conjugated ketones <u>49</u>.

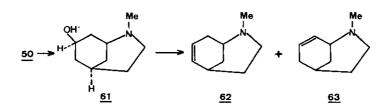


In the synthesis of Mitsuhashi and co-workers<sup>12</sup> from the same acid <u>51</u> the key intermediate <u>53</u> was obtained by an alternative way which avoids the formation of the amide group in the presence of the conjugated enolic system of <u>52</u>. Thus, 3,5dimethoxyphenylacetamides <u>57b</u> and <u>57c</u> were prepared from <u>51</u> by methylation with dimethyl sulfate, subsequent conversion of the resulting carboxylic acid (<u>56</u>) into the corresponding acid chloride and treatment with ammonia or methylamine. Reduction of <u>57</u> with sodium in liquid ammonia gave the dihydro derivatives <u>58</u>. Hydrolysis of <u>58</u> with diluted hydrochloric acid afforded quantitatively the intermediate <u>53</u>, from which the C-7 functionalized morphans <u>50b</u> and <u>50c</u> were prepared.

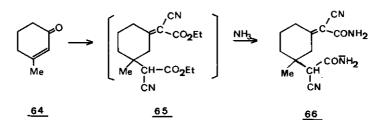
Condensation<sup>12</sup> of N-acetyl and N-formyl derivatives of 50 with 3-aminoacro lein gave the pyrido [3,2-f]morphans 59 (R=COCH<sub>3</sub>, CHO), (systematic name: 8-acyl-5,6, 7,8,9,10-hexahydro-5,9-methanopyrido [2,3-d] azocine). This synthesis is the first approach to an heteromorphan system through a strategy which employed functionalized morphans. However, the carbonyl group at C-7 position confers ambiguity in the formation of the pyridine ring since the synthesis of <u>59</u> is accompanied by that of its structural isomer 60.

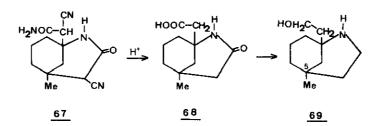


On the other hand, 2-methyl-2-azabicyclo [3.3.1] nonan-7-one (50b) was converted  $^{35}$  into a mixture of 2-methyl-2-azabicyclo [3.3.1] non-6-ene (62) and its position isomer 63 through the hydroxyderivative 61. Thus, reduction of 50b with sodium borohydride gave stereoselectively the endo amino alcohol 61 whose dehydration (methanesulfonyl chloride in pyridine) afforded the olefins 62 and 63, both of which upon catalytic hydrogenation were converted into 2-methyl-2-azabicyclo [3.3.1] nonane (6). The 6,7 double bond in 62 is postulated to provide the  $\Pi$ -electron density that is usually imparted by the aromatic ring in 6,7-benzomorphans. However this replacement of an aromatic ring by a single unsaturation does not result in compounds with significant analgesic activity.



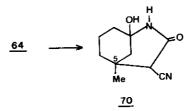
6. Other syntheses with formation of piperidine ring in the last stage.



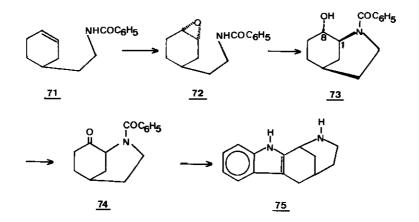


The unexpected formation of the azabicyclic system <u>67</u> when 3-methyl-2-cy clohexen -1-one (<u>64</u>) is treated with ethyl cyanoacetate and ammonia in alcohol, su ggested<sup>65</sup> the synthesis of 4-cyano-1-hydroxy-5-methyl-3-oxo-2-azabicyclo[3.3.1] nona ne (<u>70</u>) which is the first approach to 2-azabicyclo[3.3.1] nonane systems with an alkyl substituent at C-5 position.

The lactam <u>67</u> probably results from an initial Knoevenagel condensation followed by Michael addition, ammonolysis of the resulting diester <u>65</u> and final cy clization by conjugated addition. Its hydrolysis and decarboxylation afforded <u>68</u> which later on was reduced with lithium aluminum hydride to give 5-methyl-2-azabicyclo [3.3.1] nonane-1-ethanol (<u>69</u>). Likewise, when  $\alpha,\beta$ -unsaturated ketone <u>64</u> was treated with cyanoacetamide in the presence of piperidine-water, the azabicyclo <u>70</u> was obtained in 77% yield.

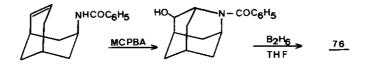


Another approach to C-8 functionalized 2-azabicyclo [3.3.1] nonanes involves a series of reactions different from the previously mentioned even through based on the same disconnections. This method consists in the formation of C<sub>1</sub>-N bond by intramolecular nucleophilic attack of an amide upon an epoxide in the presence of potassium <u>tert</u>-butoxide.

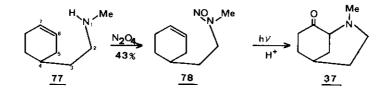


Thus, in model studies of the synthesis of the alkaloid echitamine, 2benzoyl-2-azabicyclo [3.3.1] nonan-8-one (74) was prepared and converted into the tetracyclic system 75 by Fischer indole synthesis<sup>31</sup>. Formation of alcohol 73 took place in 87% yield when the cyclization was achieved with the <u>trans</u> amido epoxide 72, and in 45% yield when the crude epoxide mixture resulting from epoxidation of the double bond of 71 was directly used. Oxidation to the ketone 74 took place in moderate yield (29%) with chromic acid in aqueous acetic acid.

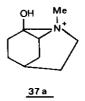
A similar strategy was used to synthesize N-benzyl-2-azaadamantan-4-ol  $(\underline{76})^{18}$  which could be formally considered as a rigid 2-azabicyclo[3.3.1] nonane due to an additional carbon bridge between the 3 and 7 positions of the morphan nucleus.



Photolysis of  $\Delta^{5,6}$ -alkenylnitrosamines<sup>66</sup> in acidified methanol afforded satisfactorily azabicyclic nucleus of 6-azabicyclo[3.2.1] octan-4-one type, either directly or through the respective oximes by a mechanism which involved aminum ra dicals<sup>67</sup>. Their addition reactions to internal olefinic bonds are highly regiospe cific, and the procedure was extended, but with moderate yield, to the synthesis of the C-8 functionalized 2-azabicyclo[3.3.1] nonane <u>37</u>. In this case cyclization was carried out with the crude  $\Delta^{6,7}$ -alkenylnitrosamine <u>78</u> prepared from the secondary amine <u>77</u> previously used in the synthesis of these systems <u>via</u> chloramines. From the crude photolysate a 6% yield of morphan <u>37</u> together with a mixture of its E (16% yield) and Z (9% yield) oximes were isolated.

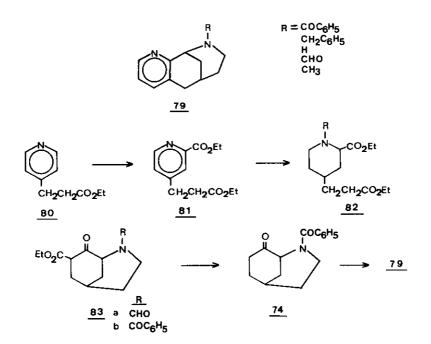


The picrate prepared from ketone 37 revealed hydroxyl absorptions at 3615 and 3345 cm<sup>-1</sup> but not carbonyl absorption in its IR spectrum. This suggested that protonation ocurred at the carbonyl oxygen but not at the amine group as in 37a.



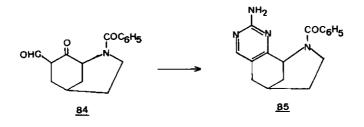
# 7. Formation of carbocyclic ring in the last step: Dieckmann cyclization.

A new strategy in the synthesis of 2-azabicyclo[3.3.1] nonan-8-ones was the formation of the carbocyclic ring from a well substituted piperidine <u>via</u> a Dieckmann cyclization followed by decarbethoxylation. This way was adopted to pre pare 5,6,7,8,9,10-hexahydro-6,10-methanopyrido[2,3-c]azocines  $(\underline{79})^{33}$ , since the presence of a  $\beta$ -keto ester moiety in the azabicyclo <u>83</u> was suitable for the subse quent elaboration of a pyridine ring.



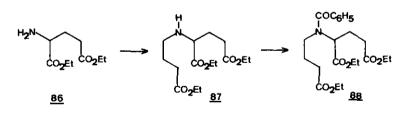
The requisite piperidine 82 was prepared from 4-pyridineacrylic acid by esterification and hydrogenation to give ethyl 4-pyridinepropionate (80). Through the corresponding N-oxide and formation of a methoxy pyridinium salt, a cyano group was introduced, and in the same reaction media was hydrolyzed and esterified to ethyl 2-carbethoxy-4-pyridinepropionate (81). Catalytic hydrogenation over platinum oxide in acetic acid afforded the piperidine derivative 82 (R=H). Acylation of the secondary amine 82 with formic acetic anhydride or benzoyl chloride and subsequent Dieckmann cyclization (sodium hydride-toluene) of the resulting acyl derivatives gave  $\beta$ -keto esters 83a and 83b respectively. Although the compound 83b reacted with hydrazine to give a pyrazolone derivative, the enolic character of the ketone group present in 83 prevented this system to react in Knoevenagel and Michael condensations or in the Reformatsky reaction. However, on hydrolysis with methanolic potassium hydroxide 83b afforded in 78% yield 2-benzoyl-2-azabicyclo[3.3.1]nonan-8-one  $(\underline{74})$ . Spectroscopic data of compound  $\underline{74}$ , specially the multiplicity of N-benzoyl signal in the nmr spectrum ( $\delta$ 7.4, singlet) differs from those reported by Dolby<sup>31</sup>  $(\delta 7.1-7.8, multiplet)$  for the same compound. The first agree with the observed magnetic equivalence of aromatic protons in dialkylbenzamides<sup>68</sup>.

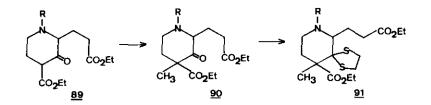
Treatment of ketone  $\underline{74}$  with ethyl formate in the presence of sodium hydride gave the  $\beta$ -keto aldehyde  $\underline{84}$  which was subjected to a condensation with guanidane to yield the amano pyrimidine derivative  $\underline{85}$ . Finally, pyridoazocane  $\underline{79}$  (R=COC<sub>6</sub>H<sub>5</sub>) was prepared (57%) by treatment of  $\underline{74}$  with 3-aminoacrolein in tracthylamine in the presence of catalytic amount of ammonium acetate at 100-110°C, and from it ,N-ben-zyl, N-formyl, N-H and N-methyl derivatives of  $\underline{79}$  were described.

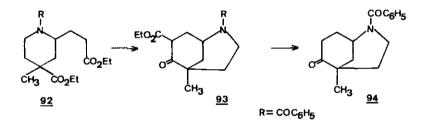


-524-

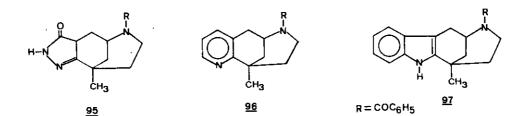
Through a Dieckmann cyclization of a suitably polysubstituted piperidine, a 2-azabicyclo [3.3.1] nonan-6-one<sup>69</sup> which possesses a quaternary carbon in C-5 was obtained for the first time. Thus, monoalkylation of diethyl glutamate (<u>86</u>) with ethyl 4-bromobutyrate followed by benzoylation gave diethyl N-benzoyl-N-(3-carbethoxypropyl)glutamate (<u>88</u>). Ring closure of this triester was accomplished by the Dieckmann reaction with sodium ethoxide-benzene as cyclizing agent. The alkylation of  $\beta$ -keto ester <u>89</u> (methyl iodide, potassium carbonate, acetone) led to a diastereomeric mixture <u>90</u> which, after conversion into the ethylenedithioketal <u>91</u> (boron trifluoride etherate, 80°C) and subsequent desulfurization (Raney nickel), was con verted into the diester <u>92</u>, also in the form of diastereoisomeric mixture (1:1, glc). On standing, the <u>trans</u> isomer was obtained, and the resulting <u>cis</u>-enriched mixture was subjected to Dieckmann cyclization (sodium hydride, toluene) to give the desired functionalized 2-azabicyclo[3.3.1]nonane system <u>93</u>, which was decarba<u>1</u> koxylated to <u>94</u> by treatment with sodium chloride in wet dimethylsulfoxide (155°C, 3 h).







These functionalized morphans are versatile intermediates for the elabora tion of condensed heterocyclic systems, and provide a new approach to the synthesis of heteromorphans. From  $\beta$ -keto ester <u>93</u>, the pyrazolo [3,4-f] morphan <u>95</u> was prepared. From ketone <u>94</u>, the pyrido [2,3-f] morphan <u>96</u> by reaction with 3-aminoacrolein and indolo [2,3-f] morphan <u>97</u> by Fischer indole synthesis were prepared<sup>70</sup>. In this hetero condensed morphans the heteroaromatic ring is fused unambiguously between the 6 and 7 position of the 2-azabicyclo [3.3.1] nonane system, and they possess a methyl substituent on the carbon atom attached to the aromatic ring.



#### REFERENCES

- In 1947 Barltrop suggested the denomination morphan for the 2-azabicyclo[3.3.1] nonane system.
- 2. M. Gates and G. Tschudi, J. Am. Chem. Soc., 78, 1380 (1956).
- 3. K. Nakanishi, T. Goto, S. Ito, S. Natori, and S. Nozoe, "Natural Products Chemis try", Vol. II, Kodansha Scientific books, Tokyo, 1975, p. 255-422.
- 4. (a) E. L. May and L. J. Sargent, "Analgetics", ed. by G. de Stevens, Academic Press, New York, 1965; (b) J. Hellerbach, O. Schnider, H. Besendorf, and B. Pell mont; N. B. Eddy and E. L. May, "Synthetic Analgesics", Part II, Pergamon Press, Oxford, 1966.
- 5. D. C. Palmer and M. J. Strauss, Chem. Rev., 77, 1 (1977).
- 6. J. Bosch, R. Granados, and F. López, <u>J. Heterocyclic Chem.</u>, <u>12</u>, 651 (1975).
- 7. T. A. Montzka and J. D. Matiskella, J. Heterocyclic Chem., 11, 853 (1974).
- M. Alvarez, J. Bosch, R. Granados, and F. López, <u>J. Heterocyclic Chem.</u>, <u>15</u>, 193 (1978).
- 9. M. Alvarez, J. Bosch, and J. Canals, An. Quim., 71, 807 (1975).

- 10. M. Alvarez, J. Bosch, and M. Feliz, J. Heterocyclic Chem., 15, 1089 (1978).
- 11. D. Kishore, P. K. Khandelwal; and B. C. Joshi, Arch. Sci., 27, 39 (1974).
- 12. J. Adachi, K. Nomura, and K. Mitsuhashi, Chem. Pharm. Bull., 24, 85 (1976).
- 13. J. Adachi, K. Nomura, S. Yamamoto, and K. Mitsuhashi, <u>Chem. Pharm. Bull.</u>, <u>24</u>, 2876 (1976).
- 14. G. C. Morrison, R. O. Waite, A. N. Caro, and J. Shavel Jr., <u>J. Org. Chem.</u>, <u>32</u>, 3691 (1967).
- 15. J. Bosch and F. Boncompte, An. Quim., 75, 357 (1979).
- 16. J. Bosch, R. Granados, and R. Llobera, An. Quim., 75, 360 (1979).
- 17. J. Bosch and D. Mauleón, unpublished results.
- 18. R. J. Schultz, W. H. Staas, and L. A. Spurlock, <u>J. Org. Chem.</u>, 38, 3091 (1973).
- G. C. Morrison, R. O. Waite, F. Serafin, and J. Shavel Jr., <u>J. Org. Chem.</u>, <u>32</u>, 2551 (1967).
- 20. G. C. Morrison, R. O. Waite, and J. Shavel Jr., J. Org. Chem., 32, 2555 (1967).
- 21. H. Zinnes, F. R. Zuleski, and J. Shavel Jr., <u>J. Org. Chem.</u>, <u>34</u>, 3165 (1969).
- 22. M. Mokotoff and L. J. Sargent, <u>J. Org. Chem.</u>, <u>33</u>, 3551 (1968).
- 23. J. A. Peters, Synthesis, 321 (1979).
- 24. (a) T. Momose and O. Muraoka, <u>Chem. Pharm. Bull.</u>, <u>26</u>, 288, 2217, 2589 (1978);
  (b) T. Momose, O. Muraoka, and S. Atarashi, <u>Heterocycles</u>, <u>12</u>, 37 (1979).
- 25. G. L. Buchanan, Chem. Soc. Rev., 3, 41 (1974).
- 26. (a) J. A. Marshall and H. Faubl, <u>J. Am. Chem. Soc.</u>, <u>92</u>, 948 (1970); (b) J. R. Wiseman and W. A. Pletcher, <u>J. Am. Chem. Soc.</u>, <u>92</u>, 956 (1970); (c) K. B. Becker, <u>Helv. Chim. Acta</u>, <u>60</u>, 81 (1977); (d) M. Kim and J. D. White, <u>J. Am. Chem. Soc.</u>, <u>99</u>, 1172 (1977); (e) H. O. House, W. A. Kleschick and E. J. Zaiko, <u>J. Org. Chem.</u>, <u>43</u>, 3653 (1978); (f) K. J. Shea and S. Wise, <u>J. Am. Chem. Soc.</u>, <u>100</u>, 6519 (1978); (g) K. Becker and R. W. Pfluger, <u>Tetrahedron Letters</u>, 3713 (1979).
- 27. M. Toda, H. Niwa, K. Ienaga, and Y. Hirata, Tetrahedron Letters, 335 (1972).
- 28. (a) J. P. Schaeffer, J. C. Lark, C. A. Flegal, and L. M. Honig, <u>J. Org. Chem.</u>, <u>32</u>, 1372 (1967); (b) J. A. Peters, J. M. Van der Toorn, and H. VanBekkun, <u>Tetrahedron</u>, <u>30</u>, 633 (1974); (c) J. A. Peters, <u>Tetrahedron</u>, <u>34</u>, 3313 (1978); (d) V. S. Mastryukov, M. V. Popik, O. V. Dorofeera, A. V. Golubinskii, L. V. Vilkov, N. A. Belikova, and N. L. Allinger, <u>Tetrahedron Letters</u>, 4339 (1979).
  29. T. G. Cochran, J. <u>Med. Chem.</u>, <u>17</u>, 987 (1974).
- 30. (a) J. R. Hargreaves and P. W. Hickmott, Tetrahedron, 23, 3151 (1967);

- (b) R. Gelin, S. Gelin, and R. Dolmazon, <u>Bull. Soc. chim. France</u>, 1049 (1973);
  (c) P. W. Hickmott, G. J. Miles, G. Sheppard, R. Urbani, and C. T. Yoxall, <u>J. Chem. Soc. Perkin I</u>, 1514 (1973); (d) D. Gravel and S. Rahal, <u>Can. J. Chem.</u>, 53, 2671 (1975).
- 31. L. J. Dolby and S. J. Nelson, <u>J. Org. Chem.</u>, <u>38</u>, 2882 (1973).
- 32. H. Ong and E. L. May, J. Heterocyclic Chem., 8, 1007 (1971).
- 33. J. Adachi, K. Nomura, S. Shiraki, and K. Mitsuhashi, <u>Chem. Pharm. Bull.</u>, <u>22</u>, 658 (1974).
- 34. M. E. Rogers and E. L. May, J. Med. Chem., 17, 1328 (1974).
- 35. R. Cavestri and M. Mokotoff, J. Med. Chem., 20, 1493 (1977).
- 36. J. A. Barltrop, J. Chem. Soc., 339 (1947).
- 37. E. Ferber and H. Bruckner, Chem. Ber., 76, 1019 (1943).
  - 38. M. W. Cronyn, J. Org. Chem., 14, 1013 (1949).
  - 39. D. Ginsburg, J. Org. Chem., 15, 1003 (1950).
  - 40. J. H. Dygos, <u>Diss. Abst. Int. B</u>, <u>31</u>, 3913 (1971); C. A. <u>76</u>, 3668d.
  - 41. M. Toda, Y. Hirata, and S. Yamamura, Chem. Comm., 1597 (1970).
  - 42. F. Ramirez and J. W. Sargent, 'J. Am. Chem. Soc., 77, 6297 (1955).
  - 43. E. L. May and J. G. Murphy, <u>J. Org. Chem.</u>, <u>19</u>, 618 (1954).
  - 44. E. L. May and J. G. Murphy, <u>J. Org. Chem.</u>, <u>20</u>, 1197 (1955).
  - 45. J. G. Murphy and E. L. May, J. Org. Chem., 19, 615 (1954).
  - 46. H. O. House, "Modern Synthetic Reactions" 2nd ed, W. A. Benjamin, Menlo Park, Calif., 1972, p. 520-526.
  - 47. E. L. May and M. Takeda, J. Med. Chem., 13, 805 (1970).
  - 48. H. H. Ong, T. Oh-ishi, and E. L. May, J. Med. Chem., 17, 133 (1974).
  - 49. M. Takeda, H. Inoue, K. Noguchi, Y. Honma, M. Kawamori, G. Tsukamoto, Y. Yam<u>a</u> waki, S. Saito, K. Aoe, T. Date, S. Nurimoto, and G. Hayashi, <u>J. Med. Chem.</u>, 20, 221 (1977).
  - 50. E. L. May, J. Org. Chem., 23, 947 (1958).
  - 51. H. H. Ong and E. L. May, J. Org. Chem., 37, 712 (1972).
  - 52. P. G. Gassman and J. H. Dygos, Tetrahedron Letters, 4749 (1970).
  - 53. P. Gassman, F. Hoyda, and J. H. Dygos, J. Am. Chem. Soc., 90, 2716 (1968).
  - 54. P. G. Gassman and J. H. Dygos, Tetrahedron Letters, 4745 (1970).
  - 55. P. G. Gassman, Accounts Chem. Res., 3, 26 (1970).
  - R. Tadayoni, A. Heumann, R. Furstoss, and B. Waegell, <u>Tetrahedron Letters</u>, 2879 (1973).

- 57. R. Furstoss, R. Tadayoni, G. Esposito, J. Lacrampe, A. Heumann, and B. Waegell, Can. J. Chem., 54, 3569 (1976).
- 58. A. Heumann, R. Furstoss, and B. Waegell, <u>Tetrahedron Letters</u>, 993 (1972).
- 59. (a) L. P. Reiff and H. S. Aaran, <u>Tetrahedron Letters</u>, 2329 (1967); (b) R. Furs toss, G. Esposito, P. Teissier, and B. Waegell, <u>Bull. Soc. chim. France</u>, 2485 (1974).
- J. Lacrampe, A. Heumann, R. Furstoss, and B. Waegell, <u>J. Chem. Res.(S)</u>, 334 (1978).
- 61. M. Mokotoff and R. R. Sprecher, Tetrahedron, 30, 2623 (1974).
- 62. W. J. Gensler, C. D. Gatsonis, and Q. A. Ahmed, J. Org. Chem., 33, 2968 (1968).
- 63. R. Furstoss, P. Teissier, and B. Waegell, Chem. Comm., 384 (1970).
- 64. M. Mokotoff and R. C. Cavestri, J. Org. Chem., 39, 409 (1974)
- 65. M. W. Cronyn and G. H. Riesser, J. Am. Chem. Soc., 75, 1664 (1953).
- 66. Throughout this discussion the radical center will be designated as position 1.
- R. A. Perry, S. C. Chen, B. C. Menon, K. Hanaya, and Y. L. Chow, <u>Can. J. Chem.</u>, 54, 2386 (1976).
- 68. V. I. Stenberg, S. P. Singh, and N. K. Narain, J. Org. Chem., 42, 2244 (1977).
- 69. J. Bosch and J. Bonjoch, presented at the 1st European Symposium on Organic Chemistry, Cologne (1979).
- 70. J. Bosch and J. Bonjoch, unpublished results.

Received, 21st January, 1980