NEW CONDENSED TRIHETEROCYCLIC SYSTEMS; PYRROLOTHIENOPYRAZINES

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<u>Abstract</u> - Cyclization of the azide derivatives of 2 (3)-(l-pyrrolyl)-3-(2)-thienylcarboxylic acids resulted in the formation of the 4,5dihydropyrrolo[l,2-a]thieno[3,2-e]pyrazin-5-one and the 4,5-dihydropyrrolo[l,2-a]thieno[2,3-e]pyrazin-5-one.

As part of our project for the synthesis and evaluation of the biological properties of tricyclic compounds containing both thiophene and pyrrole rings 1-4) we wish to report herein a convenient synthesis of hitherto unknown compounds namely 4,5-dihydropyrrolo[1,2-a]thieno[3,2-e]pyrazin-5-one and 4,5-dihydro-pyrrolo[1,2-a]thieno[2,3-e]pyrazin-5-one.

Starting with 2-amino-3-cyanothiophene ⁵⁾, the 2-(1-pyrrolyl)-3-thienylcarboxylic acid (Ib)^x; mp 178° (ethanol); nmr spectrum (DMSO) § ppm 7.41(1H,d,H5, J H5/H4 = 5.80 Hz); 7.30(1H,d,H4); 7.05(2H,m,H2',H5'); 6.23(2H,m,H3',H4'); ir spectrum (KBr) \land C=0:1680 cm⁻¹; could be obtained in two steps, first by condensation with 2,5-dimethoxytetrahydrofuran in hot acetic acid ⁶⁾ to give (Ia), followed by hydrolysis of the cyano group in alkaline medium. The latter acid (Ib) could be converted to the corresponding acid azide (Ic); ir spectrum (KBr) \land N₃:2160 cm⁻¹; using the method of Weinstock ⁷⁾. Heating the azide (Ic) whether in the solid state or in high boiling solvent affected the ring closure giving the 4,5-dihydropyrrolo[1,2-a]thieno[3,2-e]pyrazin-5-one (II).

A typical experiment is as follows : 2 g of (Ic) in 50 ml of o-dichlorobenzene are refluxed for 15 min. After cooling to room temperature a white crystalline product separated out, this was filtered, washed with ether and dried (1.6g; 92%). The structure of compound (II) was supported by ir and nmr spectra; mp 272° (sublimation in vacuo); nmr spectrum (DMSO) § ppm 7.42 (lH,dd,H8, J H8/H7 = 2.4 Hz, J H8/H6 = 1.3 Hz); 7.30(lH,d,H2, J H2/H3 = 5.7 Hz); 6.98(lH,dd,H6, J H6/H7 = 4.2 Hz); 6.92(lH,d,H3); 6.58(lH,dd,H7); ll.2(lH,s,NH); ir spectrum (KBr) C=0:1650 cm⁻¹.

On the other hand, gentle heating of the azide (Ic) in dichloromethane afforded a mixture of the starting material and the isocyanate intermediate; ir spectrum $N=C=0:2260 \text{ cm}^{-1}$. Trials to isolate this latter intermediate were unsuccessfull due to the reactivity of the pyrrole ring which favours the closure.



In a similar manner, the reaction of methyl 3-amino-2-thienylcarboxylate ⁸⁾ with 2,5-dimethoxytetrahydrofuran gave the methyl 3-(1-pyrrolyl)-2-thienylcarboxylate (IIIa); bp_{4mm}: 130°; nmr spectrum (DMSO) § ppm 7.83(1H,d,H5 J H5/H4 = 5.45 Hz; 7.17(1H,d,H4); 7.08(2H,m,H2',H5'); 6.17(2H,m,H3',H4*); 3.74(3H,s,CH₃); ir spectrum (KBr) \uparrow C=0:1700 cm⁻¹; which was converted to the corresponding acid (IIIb) by alkaline hydrolysis of the ester grouping; mp 185° (ethanol); nmr spectrum (DMSO) § ppm 7.80(1H,d,H5, J H5/H4 = 5.40 Hz); 7.17(1H,d,H4); 7.08 (2H, m,H2',H5'); 6.15(2H,m,H3',H4*); ir spectrum (KBr) \uparrow C=0:1675 cm⁻¹. Under the same conditions described above the azide (IIIc) ir spectrum (KBr) i N₃:2150 cm⁻¹, derivative of the acid (IIIb) afforded the other isomer 4,5-dihydropyrrolo[1,2-a] thieno[2,3-e]pyrazin-5-one (1.4 g 81%); mp 276° (sublimation in vacuo); nmr spectrum § ppm 7.83 (1H,dd,H8, J H8/H7 = 2.6 Hz, J H8/H6 = 1.3 Hz); 7.55(1H,d, H2, J H2/H1 = 5.7 Hz); 7.18(1H,d,H1); 6.97(1H,dd,H6, J H6/H7 = 4.2 Hz); 6.55 (1H,dd,H7); 11.5(1H,s,NH); ir spectrum (KBr) i C=0:1635 cm⁻¹. Further studies concerning these compounds and biological screening are under investigations.

REFERENCES AND A FOOTNOTE

* all new compounds gave analytical results in agreement with the proposed structures.

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