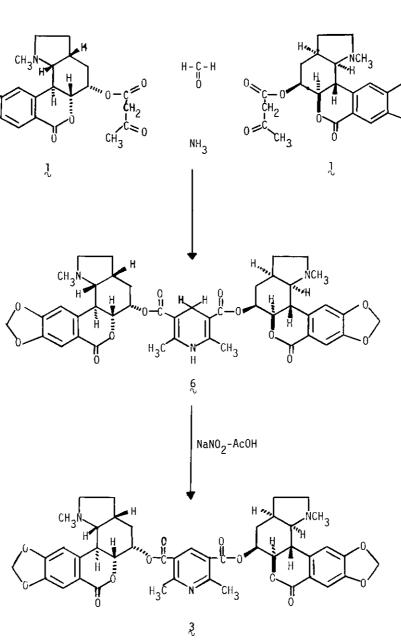
BIOGENETIC-TYPE SYNTHESIS OF CLIVIMINE FROM CLIVACETINE

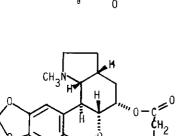
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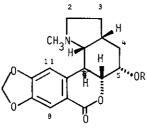
Abstract — The structures of clivacetine  $(\frac{1}{2})$  and clivatine  $(\frac{4}{2})$  were confirmed by the partial synthesis of clivacetine  $(\frac{1}{2})$  from clivonine  $(\frac{2}{2})$ . Clivimine  $(\frac{3}{2})$  was obtained by biogenetic-type synthesis from clivacetine  $(\frac{1}{2})$ , which seems to be a biosynthetic precursor of clivimine  $(\frac{3}{2})$ .

Recently we reported<sup>1</sup> the isolation of a novel alkaloid clivacetine (1,), having an acetoacetyl function, from <u>Clivia miniata</u> Regel.<sup>2</sup> (<u>Amaryllidaceae</u>), and proposed structure 1 for clivacetine on the basis of its physical and spectral data. This paper describes the partial synthesis of clivacetine (1,) from clivonine (2) and the biogenetic-type synthesis of clivimine (3) from clivacetine (1), which seems to be a biosynthetic precursor of 3.

In order to confirm the structure of clivacetine  $(\frac{1}{2})$ , the partial synthesis of  $\frac{1}{2}$  from 2 was carried out as follows: heating a mixture of diketene (0.76 ml), 2 (90 mg), and chloroform (20 ml) in the presence of triethylamine (0.08 ml)(at 80° for 30 min.) gave O-acetoacetylclivonine ( $\frac{1}{2}$ )(55 mg, 48.2%)[mp 153-155° (lit.<sup>1</sup> mp 152-155°); [ $\alpha$ ]<sup>24</sup><sub>D</sub> + 51.9°(CHCl<sub>3</sub>)(lit.<sup>1</sup> [ $\alpha$ ]<sup>24</sup><sub>D</sub> + 53.8°(CHCl<sub>3</sub>); Found: C, 62.65; H, 5.81; N, 3.73%. Calcd. for C<sub>21</sub>H<sub>23</sub>NO<sub>7</sub>: C, 62.83; H, 5.78; N, 3.49%]. The spectral data and the melting point of O-acetoacetylclivonine were fully consistent with those of clivacetine ( $\frac{1}{2}$ )<sup>1</sup> from the natural source. Moreover, the partial synthesis of  $\frac{1}{2}$  presented the final proof of the structure ( $\frac{4}{2}$ ) for clivatine,<sup>3</sup> since we have already reported<sup>1</sup> that O-acetylclivatine ( $\frac{5}{2}$ ) derived from  $\frac{4}{2}$  is the same compound as that prepared by reduction of  $\frac{1}{2}$  with sodium borohydride, followed by O-acetylation of the resultant product.







1 : R=COCH<sub>2</sub>COCH<sub>3</sub> 2 : R=H 4 : R=COCH<sub>2</sub>CH(OH)CH<sub>3</sub> 5 :  $R=COCH_2CH(OAc)CH_3$  Clivacetine  $\binom{1}{2}$  has an interesting structure from a biogenetic viewpoint, since it seems to be a possible biosynthetic precursor of a unique <u>Amaryllidaceae</u> alkaloid clivimine  $\binom{3}{2}^4$  having a 2,6-dimethyl-pyridine-3,5-dicarboxyl function. The biogenetic-type synthesis of  $\frac{3}{2}$  has been realized by the Hantzsch pyridine synthesis<sup>5</sup> of  $\frac{1}{2}$ . Treatment of  $\frac{1}{2}$  (30 mg) in ethanol with 35% formalin (0.07 ml) and 25% ammonium hydroxide (0.26 ml) (at 100° for 30 min.) gave dihydroclivimine  $\binom{6}{2}$  (17 mg, 57.2%) as pale yellow needles, mp 233-237° (dec.) [IR (KBr) 1720 cm<sup>-1</sup>; Mass m/e 795 (M<sup>+</sup>)]. The NMR spectrum of  $\frac{6}{2}$  in CDCl<sub>3</sub> exhibited a characteristic pair of signals at  $\delta$  2.18 and 2.15 (total 6H) due to 2,6-dimethyl protons in the dihydropyridine ring, and other pairs of singals at  $\delta$  7.74 and 7.67 (total 2H, each s, 2 X C-11-H), 7.47 and 7.40 (total 2H, each s, 2 X C-8-H), and 2.52 and 2.48 (total 6H, each s, 2 X NCH<sub>3</sub>). In this NMR spectrum the pairs of signals<sup>6</sup> of the aromatic protons at C-11 and C-8, C-, and N-methyl protons suggest that dihydroclivimine  $\binom{6}{2}$  may have two conformational isomers with respect to its dihydropyridine ring.

Dehydrogenation<sup>5d</sup> of 6 (8 mg) with sodium nitrite (20 mg) and acetic acid (0.5 ml) gave clivimine<sup>1</sup> (3.5 mg, 43.9%) [mp 254-256°(dec.)(lit.<sup>1</sup> mp 258-259° (dec.); IR (KBr) 1720 cm<sup>-1</sup>; ORD [M]<sup>21</sup> (nm) (MeOH) 0° (342), -8821°(325)(trough), 0°(311), +17643° (296), +16761° (291), +28670° (278)(peak), +8821° (261) (trough), +105858° (241)(peak); NMR  $\delta$  (CDCl<sub>3</sub>) 8.52 (lH, s,  $\gamma$ -H in the pyridine ring), 7.48 (2H, s, 2 X C-11-H), 7.44 (2H, s, 2 X C-8-H), 6.06 and 6.03 (each 2H, d, J=2 Hz, 2 X AB type of OCH<sub>2</sub>O), 5.54 (2H, m, 2 X C-5-H), 4.18 (2H, dd, J<sub>5a-11b</sub>=12, J<sub>5a-5</sub>=3 Hz, 2 X C-5a-H), 2.87 (6H, s, 2- and 6-CH<sub>3</sub> in the pyridine ring), and 2.34 (6H, s, 2 X NCH<sub>3</sub>)]. This compound appeared identical with clivimine (3) from the natural source on direct comparison of their IR, ORD, and NMR spectra and in the mixed melting point test.

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- 3. In <u>Heterocycles</u> (1977,  $\xi$ , 551), W. Döpke proposed that the structure of clivatine was 4, but gave no details about the presence of the  $\beta$ -hydroxy-butyryl group in 4.
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- 5. a) A. Hantzsch, <u>Ann.</u>, 1882, <u>215</u>, 1; b) <u>idem</u>, <u>Chem</u>. <u>Ber</u>., 1885, <u>18</u>, 1744;
  c) <u>idem</u>, <u>ibid</u>., 1886, <u>19</u>, 289; d) B.E. Norcross, G. Clement, and
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- 6. Similar pairs of signals were observed in the NMR spectrum of an ester of lycoramine and 2,6-dimethyl-1,4-dihydropyridine-3,5-dicarboxylic acid : δ (CDCl<sub>3</sub>)
  2.16 and 2.12 (total 6H, each s, 2- and 6-CH<sub>3</sub> in the dihydropyridine ring), 3.82 and 3.80 (total 6H, each s, 2 X OCH<sub>3</sub>) (unpublished result).

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