THE STRUCTURE OF CAMMACONINE FROM ACONITUM VARIEGATUM

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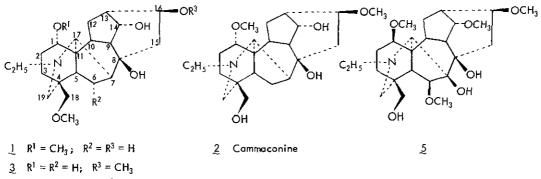
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The novel structure assigned earlier to the C₁₀-diterpenoid alkaloid, commaconine, contained a C(16) hydroxyl and a C(18) methoxyl group. On the basis of ¹³C NMR spectral analysis, these assignments have been reversed.

In 1971 Mollov and co-workers¹ reported the isolation of a new alkaloid, cammaconine, from *Aconitum* variegatum and on the basis of chemical correlation with isotalatizidine, oxidation reactions, and spectral analysis, assigned structure 1 to cammaconine. Thus cammaconine would be the first member of the C₁₉-diterpenoid alkaloids² to bear a C(16) β - hydroxy instead of a C(16) β - methoxy group. This unusual feature prompted us to examine the ¹³C NMR spectrum of cammaconine.

We report here a revision of the structure of cammaconine to <u>2</u> on the basis of ¹³C NMR spectral analysis. A recent analysis²⁻⁵ of ¹³C NMR spectra of more than 100 *Aconitum* - and *Lycoctonine*-type alkaloids and their derivatives provides a foundation for unambiguous assignments of all resonances for cammaconine. A correlation



$$_{4}$$
 R¹ = R³ = CH₃; R² = OCH₃

of the ¹³C NMR spectrum of cammaconine was made with isotalatizidine (3), chasmonine (4), lycoctonine (5) and other published ¹³C NMR data for related alkaloids (Table 1). The presence of a primary hydroxy carbon signal at 68.8 ppm and the absence of a methoxy carbon signal at ~ 59 ppm indicated that a hydroxy group instead of a methoxy group is present at C(18) in cammaconine. Furthermore, the signals at 82.3 and 56.3 ppm also revealed the presence of a methoxy group at C(16) instead of at C(18) in cammaconine. All other signals are also in agreement with the new structure (2) assigned for cammaconine.

| The above evidence indicates that cammaconine possesses a $C(18)$ hydroxyl rather than a $C(16)$ hydroxyl | |
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| group . It is therefore identical with the alkanolamine of the alkaloid aconorine . 5 | |

| | | C Chemical sinns and Assignments for Caminaconne (2) and Refered Arabids | | | | | | | | |
|-------|------|--------------------------------------------------------------------------|------|-------|----------|-------|------|---------------|------|--|
| | 2 | 3 | 4 | 5 | ···· | 2 | 3 | 4 | 5 | |
| C(1) | 86.3 | 72.3 | 86.1 | 84 .2 | C(14) | 75.6 | 75.6 | 75.5 | 84.0 | |
| C(2) | 25.8 | 29.2 | 26.0 | 26.1 | C(15) | 38.3 | 42.3 | 39.2 | 33.7 | |
| C(3) | 33.2 | 2 9 .7 | 35.2 | 31.6 | C(16) | 82.3 | 82.4 | 2, 82 | 82.7 | |
| C(4) | 39.1 | 37.3 | 39.5 | 38.6 | C(17) | 63.0 | 63,7 | 62.4 | 64.8 | |
| C(5) | 46,0 | 41.7 | 48.8 | 43.3 | C(18) | 68.8 | 79.0 | 80,8 | 67.6 | |
| C(6) | 24.6 | 25.0 | 82.5 | 90.6 | C(19) | 53.1 | 56.6 | 54.0 | 52.9 | |
| C(7) | 45.9 | 45.3 | 52.8 | 88.3 | N-CH₂ | 49.5 | 48.4 | 49.3 | 51,1 | |
| C(8) | 73.0 | 74.3 | 72.6 | 77.5 | I CH₃ | 13.7 | 13.0 | 13.6 | 14,1 | |
| C(9) | 47.0 | 46.7 | 50,3 | 49.7 | C(1') | 56.5* | - | 56.3 | 55,7 | |
| C(10) | 37.6 | 40.4 | 38.4 | 38.0 | C(6') | - | - | 57.2 | 57.7 | |
| C(11) | 48.8 | 48.7 | 50.4 | 48.9 | C(14') | - | - | - | 58.0 | |
| C(12) | 27,7 | 26.8 | 28.6 | 28.8 | C(16') | 56.3* | 56.2 | 55 <i>.</i> 9 | 56.2 | |
| C(13) | 45.6 | 44.1 | 45.7 | 46.1 | C(18') | - | 59.3 | 59.2 | - | |

Table 1. ¹³C Chemical shifts and Assignments for Cammaconine (2) and Related Alkaloids[†]

[†] In ppm downfield to TMS; Solvent, CDCl₃. ^{*}Values within any vertical column are interchangeable.

REFERENCES

- 1. M. A. Khaimova, M. D. Palamareva, N. M. Mollov, and V. P. Krestev, Tetrahedron, 27, 819 (1971).
- S. W. Pelletier and N. V. Mody in "The Alkaloids", Vol. 17, R.H.F. Manske and R. Rodrigo Eds. Academic Press, New York, 1979, Chapter 1.
- 3. S. W. Pelletier and Z. Djarmati, J. Am. Chem. Soc., 98, 2626 (1976).
- 4. S. W. Pelletier, N. V. Mody, R. S. Sawhney, and J. Bhattacharyya, Heterocycles, 7, 327 (1977).
- 5. S. W. Pelletier, N. V. Mody, and R. S. Sawhney, Can. J. Chem., 57, 1652 (1979).
- 6. V. A. Tel'nov, M. S. Yunusov, S. Yu. Yunusov, and B. Sh. Ibragimov, Khim. Prir. Soedin., 11, 814 (1975).

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