THE ORIGIN OF THE N-FORMYL GROUP IN NATURE AND THE BIOGENESIS OF CATHARINE AND CATHARININE

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The presence of an N-formyl group in an alkaloid often points to an <u>in</u> vivo Baeyer-Villiger oxidative rearrangement of an iminium precursor. Vinblastine $(\underline{3})$ is thus shown to be the most likely progenitor for the accompanying alkaloids catharine (1) and catharinine ($\underline{2}$) in <u>Catharanthus</u> spp.

The importance of the Baeyer-Villiger type oxidative rearrangement of iminium salts in alkaloid biogenesis has only recently been recognized. The in vivo formation of the N-formyl groups in the benzylisoquinoline polycarpine¹ as well as in the benzophenanthridine derivatives iwamide, arnottianamide, and isoarnottianamide, 2 has been explained using such a process.³

The dimeric indole alkaloids catharine $(\underline{1})^4$ and catharinine $(\underline{2})^5$ have been found in a variety of Catharanthus species, ⁶ and are structurally related to the important and accompanying antitumor alkaloid vinblastine (3),





1



2



 $\underline{3}$

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Alkaloids <u>1</u> and <u>2</u> bear an N-formyl group, and the problem of their biogenesis revolves essentially around the formation of this moiety. A variety of different precursors have been assumed, all proceeding to formaldiminium salts that can undergo an ill-defined oxidation to the corresponding formamides.^{5,7}

Reconsideration of the biogenetic scheme for catharine $(\underline{1})$ and catharamine $(\underline{2})$ makes it clear that a common precursor must be the accompanying alkaloid vinblastine $(\underline{3})$. This dimeric compound , may readily lead to iminium species $\underline{4}$ and $\underline{5}$ which can suffer Baeyer-Villiger oxidative rearrangement as their key transformation in nature, to furnish eventually alkaloids $\underline{1}$ and $\underline{2}$, respectively (Scheme). ^{8,9}



Although other routes to formylation in nature are known,¹⁰ it is evident from the above that the Baeyer-Villiger type oxidative rearrangement of iminium salts must be borne in mind whenever the biogenesis of alkaloidal formamides is considered.

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References and Footnotes

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- 8. There is always a possibility that the true precursor of <u>1</u> and <u>2</u> may be a very close analog of <u>3</u>, rather than <u>3</u> itself. Additionally, it is difficult to say specifically at which stage the dehydration step required for catharine formation occurs.
- 9. We favor iminium intermediates $\underline{4}$ and $\underline{5}$ in the biogenetic scheme, over alternate Baeyer-Villiger oxidation of formal diminium salts $(\sum_{N=CH_2}^{\Theta})$ to obtain the formamides.
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