TOTAL SYNTHESES OF (±)-CHANOCLAVINE I AND (±)-DIHYDROSETOCLAVINE

Mitsutaka Natsume\* and Hideaki Muratake Research Foundation Itsuu Laboratory Tamagawa 2-28-10, Setagaya-ku, Tokyo 158, Japan

Abstract: Syntheses of the title ergot alkaloids & and  $\frac{1}{2}$ *& were achieved* **from** *the* **common** *intermediate 2, obtained by a series* **of**  *reactions including our synthetic method* **of** *4-atkytindotes.* 

In the previous three papers,  $1, 2, 3$  we reported (i) a synthetic method of functionalized 4-alkylindoles such as  $\downarrow$ , (ii) its transformation into a tricyclic indole derivative *2,* which is expected to be a common intermediate for the synthesis of ergot alkaloids, and (iii) the first synthesis of  $6.7$ -secoagroclavine (2) from *2* by way of a N-protected ketone derivative *2. 2* is an important compound for the synthesis of 6,7-secoergoline type of alkaloids and this time, a synthesis of ( $\pm$ )-chanoclavine I<sup>4</sup> ( $\overline{2}$ ) was carried out as shown in Chart 1.<sup>5,6</sup>



Owing to the insoluble character of chanoclavine I in most organic solvents, identification [TLC,  ${}^{1}$ H NMR (CDCl<sub>3</sub>), and IR (CHCl<sub>3</sub>)] was performed at the stage of the compound  $l, \emptyset$ . Natural  $\overline{\lambda}$  was treated with ClCOOCH<sub>2</sub>Ph in the presence of Et<sub>3</sub>N and the resulting diacyl derivative was partially hydrolyzed<sup>7</sup> to an N-benzyloxycarbonyl alcohol, which was acetylated to afford  $Q$  of the natural origin. Both natural and synthetic  $10$ 's were treated with warm diluted alkali,  $^7$  followed by cleavage of the N-protecting group, and the recovered natural *2* was identical with 8 chanoclavine I [mixed mp, IR 1KBr)l. Synthetic *2* exhibited the same MS patern as natural  $\xi$ , thus completing a total synthesis of  $(±) - \xi$ .

 $-375-$ 



 $a$  Ph<sub>3</sub>P=CH<sub>2</sub>, THF, 0°, 46% (78%<sup>†</sup>). b OsO<sub>4</sub>, Et<sub>2</sub>O-Py, 0°+rt, 70% (80%<sup>†</sup>). **c** Ac20, Py, quant. d p-TsOH, PhH, reflux. **2:** 22%. **&Q:** 25%. **e** (il 2% KOH in t-BuOH-H<sub>2</sub>O (3:1), 55-60°, (ii) Na, liq. NH<sub>3</sub>-THF. *ca*. 80% yield for both synthetic and natural compounds.  $f$  (i) ClCOOCH<sub>2</sub>Ph, CH<sub>2</sub>Cl<sub>2</sub>-Py, Et<sub>3</sub>N, (ii) 3.5% KOH in t-BuOH-H<sub>2</sub>O (3:1), 55-60°, (iii) Ac<sub>2</sub>O, Py.

t Yield calculated on the basis of converted starting material.

## Chart 1

Construction of the tetracyclic ergoline skeleton was next attempted by assuming an intramolecular cyclization from **&J** to **&z,** if one could achieve the introduction of an aldehyde equivalent into the ketone group of **A&.** When R equaled to the benzyl group, removal of the N-protecting group from  $\frac{1}{2}$  by the catalytic hydrogenation would produce  $\frac{1}{4}$  at first and then end up in the formation of a stable D ring as  $\frac{1}{k}$ , whereas, in the case of R=Me, any reaction on  $\frac{1}{k}$  might involve the participation of an equilibrium form 12 to afford ring-opened derivatives as by-products. Based on this consideration,  $\lambda \lambda e^{5}$  and  $\lambda \lambda b^{5}$  were synthesized from  $\lambda$  (Chart 2) and submitted to the one carbon elongation reaction producing an aldehyde function. **<sup>9</sup>**

A satisfactory result was obtained by the condensation of tosylmethylisocyanide (TOSMIC) with  $\frac{11}{24}$  using TlOEt as a base<sup>10</sup> and subsequent treatment with p-TsOH in





a (i) LiAlH<sub>4</sub>, THF, reflux, (ii) ClCOOCH<sub>2</sub>Ph, CH<sub>2</sub>Cl<sub>2</sub>, Et<sub>3</sub>N; L<sub>2</sub>ba: 28%, L<sub>2</sub>be: 33%. b Me<sub>2</sub>CO, p-TsOH, rt; 11g: 85%, 11p: 85%. c p-TsCH<sub>2</sub>NC, T1OEt, EtOH-DME (4:1), rt. d p-TSOH, DME-H<sub>2</sub>O (6:1), rt. e H<sub>2</sub>, 10% Pd-C, CH<sub>2</sub>O-H<sub>2</sub>O, MeOH.  $e'$  (i)  $H_2$ , 10% Pd-C, MeOH, (ii) 5% KOH in MeOH-H<sub>2</sub>O (14:1), reflux, (iii)  $H_2$ , 10% Pd-C,  $CH_2O-H_2O$ , MeOH.

Chart 2



DME-H<sub>2</sub>O afforded  $\frac{1}{k}$  [MS  $m/e$ : 419 (M<sup>+</sup>), 401 (M<sup>+</sup>-H<sub>2</sub>O), 356 (M<sup>+</sup>-H<sub>2</sub>O-NH<sub>2</sub>CHO); <sup>1</sup>H NMR  $(CDCL<sub>3</sub>, 60°)^{11}$  6: 1.45 ( $\geq$ C-Me), 7.93 ( $>$ N-CHO)] and  $\downarrow$  R<sub>p</sub> [MS  $m/e$ : 419 (M<sup>+</sup>), 401 (M<sup>+</sup>- $H_2O$ ); <sup>1</sup>H NMR (CDC1<sub>3</sub>, 60°) 6: 1.20 ( $\frac{3}{2}$ C-Me), 8.35 ( $>N-CHO$ )], which were hydrogenated over 10% Pd-C in the presence of CH<sub>2</sub>O. Formation of  $(±)$ - $\frac{19b}{100}$  [mp 88-91°, MS  $m/e$ : 256 ( $M^+$ ),  ${}^{1}H$  NMR (CDCl<sub>3</sub>) 6: 1.34 (s, Me), 2.19 (br. s, OH), 2.39 (s, N-Me), 6.88 (br. s, H-2), 7.96 (br., indole NH)] and  $(\pm)$ - $2\&0$ <sup>8</sup> Imp 217-219°, MS  $m/e$ : 256  $(M^+)$ , <sup>1</sup>H NMR (CDC1<sub>3</sub>-CD<sub>3</sub>OD)  $\delta$ : 1.19 (s, Me), 2.57 (s, N-Me)] was observed in 33% and 20% yields, respectively, from LLD. The structure of 19b was confirmed by comparison with a hydrogenation product of setoclavine (vide infra).

The same series of reaction were applied to  $\frac{1}{k}$ . A mixture of the tetracyclic derivatives  $\frac{1}{6}$  and  $\frac{1}{6}$  was formed analogously, but the catalytic hydrogenation required the prolonged reaction time, and yet  $2\lambda$  [<sup>1</sup>H NMR (DMSO-d<sub>6</sub>) 6: 1.11 ( $\frac{1}{6}$ C-Me), 8.12 and 8.15 (-NHCHO), 10.58 (indole NH)] was isolated in addition to  $(t)$ dihydroisosetoclavine<sup>8,12</sup> (20<sub>2</sub>) [mp 232-236°, MS  $m/e$ : 256 (M<sup>+</sup>), <sup>1</sup>H NMR (CDCl<sub>3</sub>-CD<sub>3</sub>OD)  $\delta$ : 1.50 (s, Me), 2.43 (s, N-Me)] in 16% yield from  $\frac{1}{6}$ .  $\frac{2}{6}$  was once treated with 5% KOH in MeOH-H<sub>2</sub>O (formation of  $22$ ), followed by the catalytic hydrogenation in the presence of CH<sub>2</sub>O. ( $\pm$ )-Dihydrosetoclavine<sup>8</sup> ( $\frac{1}{2}$ ) [mp 252-256°, MS  $m/e$ : 256 (M<sup>+</sup>), <sup>1</sup>H NMR (DMSO-d<sub>c</sub>)  $\delta$ : 1.18 (s, Me), 2.37 (s, N-Me), 10.63  $(br., indole NH)$ ] was obtained in 12% yield from  $l.h.$ 

In order to confirm the structures of synthetic  $\lambda$ 9g and  $\lambda$ 9g, preparation of the authentic samples was carried out from agroclavine  $(2,3)$ .  $2,3$  was converted to setoclavine<sup>13</sup> (24) according to the procedure described in the literature<sup>4a,14</sup> and the catalytic hydrogenation<sup>15</sup> of  $24$  yielded dihydrosetoclavine<sup>13,16</sup> ( $19a$ ) (438) and  $\frac{1}{2}$  (7%). Identification of (1)- $\frac{1}{2}$ g with dihydrosetoclavine was performed by TLC (10% MeOH-CHCl<sub>3</sub>, Rf=0.2), MS, <sup>1</sup>H NMR (CDCl<sub>3</sub>-CD<sub>3</sub>OD, DMSO-d<sub>6</sub>), and <sup>13</sup>C NMR (DMSO- $d_{\epsilon}$ ) spectra.

Acknowledgement - Authors' thanks are due to Dr. S. Ohmomo of the University of Tsukuba for his exceptional cooperation in providing us with precious samples of chanoclavine I and agroclavine. A part of this work was supported by Grant-in-Aid for special Project Research from the Ministry of Education, Science and Culture, which is gratefully acknowledged.

## REFERENCES AND NOTES

1. M. Natsume and H. Muratake, *Tetrahedron Lett.*, 3477 (1979).

 $-378-$ 

- 2. M. Natsume and H. Muratake, *Heterocyctes,* 14, 445 (1980).
- 3. M. Natsume and H. Muratake, *Heterocyctes,* 14, 1101 (1980).
- 4. (a) A. Brack, A. Hofmann, R. Brunner, and H. Kobel, *HeZv. Chim. Acta.* 40, 1358 (1957). (b) Three total syntheses of chanoclavine I were reported: H. Plieninger and D. Schmalz, *Chem. Ber.*, 109, 2140 (1976); A.P. Kozikowski and H.<br>ninger and D. Schmalz, *Chem. Ber.*, 109, 2140 (1976); A.P. Kozikowski and H. Ishida, J. *Am. Chem.* Soc., 102, 4265 (1980); W. Oppolzer and J.I. Grayson, *Hetv. Chim. Acta,* **63,** 1706 (1980).
- 5. Spectral data fully supported the structures.
- 6. Preliminary experiments using the corresponding 5.10-cis compound: Application of a modified Shapiro's olefin synthesis of tosylhydrazones [P.C. Traas, H. Boelens, and H.J. Takken, *Tetrahedron Lett.,* 2287 (1976)l was found to be fruitless; and thermolysis of a-epoxysulfoxides [V. Reutrakul and W. Kanghae, *Tetrahedron Lett.,* 1377 (1977)l afforded a low yield of an inseparable mixture of  $\alpha$ ,  $\beta$ -unsaturated aldehydes.
- 7. Usage of MeOH instead of t-BuOH has a possibility of changing in part N-COO- $CH_2Ph$  to N-COOMe.
- 8. Satisfactory results of high resolution mass spectra were obtained for these compounds.
- 9. Addition of Ph<sub>3</sub>P=CHOMe or MeNO<sub>2</sub> as well as the Darzen reaction gave unsuccessful results, see Ref. 3.
- 10. O.H. Oldenziel and A.W. van Leusen, *Tetrahedron Lett.,* 163. 167 (1974).
- 11. All  $H$  NMR spectra were taken at 90 MHz.
- 12. **This** material is not isolated from the nature but corresponds to a dihydro derivative of isosetoclavine.
- 13. The material exhibited the same IR spectrum as reported in the literature.<sup>4a</sup>
- 14. S. Yamatadani and M. Abe, *Butt. Agr. Chem.* **soc.** *Japan,* 19, 94 (19551.
- 15. H. Tscherter and H. Hauth, *Helv. Chim. Acta*, 57, 113 (1974).
- 16. This alkaloid was first isolated from *Ctaviceps paspati* Stevens *et* Hall. 15

Received, 8th December, 1980