A NEW SYNTHESIS OF PYRIMIDO [5,4-e]-as-TRIAZINE DERIVATIVES

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The reaction of 5-arylazo-6-arylidenehydrazino-1,3-dimethyluracils (II), prepared by the diazotization of 6-arylidenehydrazino-1,3-dimethyluracils (I), with dimethylformamide dimethylacetal afforded the corresponding 3-arylfervenulins (3-aryl-6,8-dimethylpyrimido[5,4-<u>e</u>]-<u>as</u>-triazine-5,7(6H, 8H)-diones)(V).

The discovery of the triad of antibiotics fervenulin, 2-methylfervenulone (MSD-92), and toxoflavin, has stimulated recent considerable interest in the chemistry of pyrimido[5,4-<u>e</u>]-<u>as</u>-triazines.¹ In connection with our studies on the pyrimido-[5,4-<u>e</u>]-<u>as</u>-triazines as potential medicinal agents,² we now report a new synthetic approach to 3-arylfervenulins (3-aryl-6,8-dimethylpyrimido[5,4-<u>e</u>]-<u>as</u>-triazine-5,7(6H,8H)-diones)(Va-e) by the reaction of readily accessible 5-arylazo-6-arylidenehydrazino-1,3-dimethyluracils (IIa-h) with dimethylformamide dimethylacetal (DMFDMA).

Treatment of the appropriate 6-arylidenehydrazino-1,3-dimethyluracils (Ia-e)³ with diazotized arylamines by the conventional method⁴ gave the desired starting materials (IIa-h) in 30-56% yields. Refluxing of the appropriate uracils (IIa-e) (0.0005 mol) with DMFDMA (2 ml) at 150°C for 5 h, followed by concentration of the reaction mixture <u>in vacuo</u> and addition of ethanol, caused the separation of the corresponding products (Va-e)⁵ in 20-60% yields. The yields of (Va-e) were depending upon the nature of arylidenehydrazino group of (IIa-e), <u>i.e.</u>, the uracils with an electron-withdrawing arylidenehydrazino group gave better results than those with an electron-releasing group. Similar substituent effect was also observed on the arylazo group, <u>i.e.</u>, treatment of the uracils (IIf-h), which possess a strong electron-withdrawing p-nitrophenylazo group, with DMFDMA caused pronounced improvement in the yield of (Va)(82-97%) (Table).⁶

It should be noted that the reaction of (IIa) with dimethylformamide in stead of DMFDMA under the same conditions resulted in the recovery of (IIa). Therefore, the reaction of (IIa-h) with DMFDMA leading to (Va-e) can be best explained by assuming the initial formation of the intermediate (III) through the enol form of (II), followed by 1,5-migration of the 1-methoxytrimethylamino group to give (IV), which possesses a triazahexatriene-type structure. This could undergo intramolecular cycloaddition and aromatization by loss of 1-arylamino-1-methoxytrimethylamine to give (V) as a final product. The C-O bond formation as exemplified by (III) has been speculated in the reaction of certain enols with DMFDMA⁷ and the intramolecular cycloaddition of aza analogs of hexatriene has ample precedents.⁸ To our knowledge, this is the first example in which 5-arylazopyrimidine was directly used as a nitrogen source for N-4 of the pyrimido[5,4-<u>e</u>]-<u>as</u>-triazine ring system (Scheme).

Scheme

NHN=CH | Me Ne (II)(I) DMFDMA Me MeOH Мe (III).5-migration MeO H Cycloaddition Me Me-N \cap Ме Мe MeO (V) (IV)

5-Arylazo-6-arylidenehydrazino- 1,3-dimethyluracils					3-Arylfervenulins			
Compd. ^{a)}	Rl	R ²	Mp(°C)	Yield(%)	Compd. ^{b)}	R ²	Mp(°C)	Yield(%)
Ila ,	н	н	187-188	56	Va	н	273-275	50
IIb	н	Br	213-215	47	Vb	Br	>300	60
IIc	H	C1	200-202	56	Vc	C1	280-283	56
IId	Н	OMe	144-145	31	Vđ	OMe	263-264	26
IIe	н	NMe2	176-178	30	Ve	NMe2	>300	20
IIf	NO2	Н	256-257	46	Va	н		88
IIg	NO2	Br	235-236	42	Va	н		97
IIh	NO2	Cl	223-225	34	Va	н		82

Table 5-Arylazo-6-arylidenehydrazino-1,3-dimethyluracils and 3-Arylfervenulins

a) All compounds were recrystallized from DMF.

b) All compounds were recrystallized from EtOH.

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