

SYNTHESIS OF ( $\pm$ )-1,2-DIACETOXYAPORPHINE VIA AN *o*-QUINOL ACETATE

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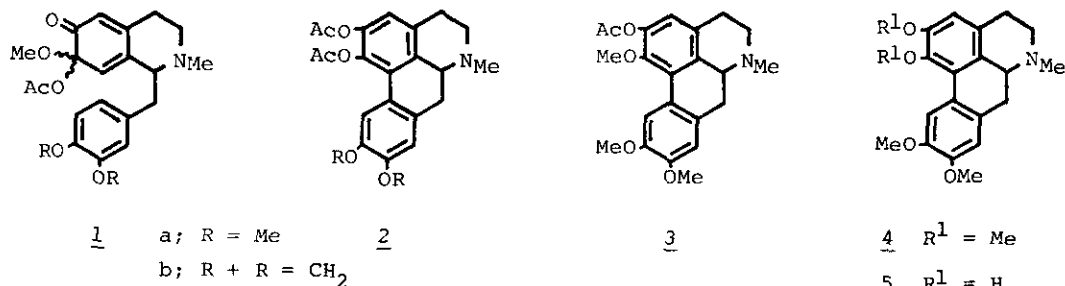
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**Abstract**——Treatment with conc.  $\text{H}_2\text{SO}_4\text{-Ac}_2\text{O}$  of a solution of *o*-quinol acetates (1) in  $\text{CH}_3\text{CN}$  gave ( $\pm$ )-1,2-diacetoxyaporphines (2) in good yields.

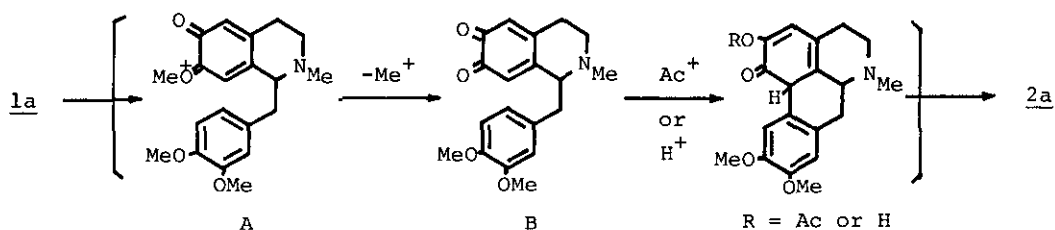
In continuation of our work on aporphine synthesis via an *o*-quinol acetate (*o*-QA)<sup>1</sup>, we found that *o*-QA (1a) in  $\text{CH}_2\text{Cl}_2$  was treated with conc.  $\text{H}_2\text{SO}_4\text{-Ac}_2\text{O}$  to give unexpectedly ( $\pm$ )-1,2-diacetoxy-9,10-dimethoxyaporphine (2a)<sup>2</sup>, m.p. 147-148°, in 13.2% yield, accompanied with ( $\pm$ )-*o*-acetylpredicentrine (3)<sup>1</sup> (oil, 18.5%). The present communication deals with the structure of 2a and an improved method for its synthesis.

The spectral data [IR  $\nu(\text{CHCl}_3)$ : 1780 (sh), 1770  $\text{cm}^{-1}$ ; NMR  $\delta(\text{CDCl}_3)$ : 2.28, 2.31 (each 3H, s,  $2 \times \text{CH}_3\text{COO}$ ), 2.56 (3H, s,  $\text{NCH}_3$ ), 3.90, 3.93 (each 3H, s,  $2 \times \text{OCH}_3$ ), 6.80, 6.88, 7.49 (each 1H, s,  $3 \times \text{arom. H}$ ); MS  $m/z$ : 411 ( $\text{M}^+$ )] and the chemical transformations by methylation<sup>3</sup> and hydrolysis to ( $\pm$ )-glaucine (4), m.p. 134-135.5° (lit.<sup>4</sup>), 136-138°, and the known ( $\pm$ )-1,2-dihydroxyaporphine (5) [HCl salt, m.p. 193-195° (lit.<sup>5</sup>), 197-198°] confirmed the structure of 2a.



A probable explanation for the formation of 2a was as follows. Namely, deacetoxylation by acetylum cation or proton occurred to give an *o*-quinonoid intermediate (A), the methyl cation of which was captured by the surrounding solvent molecule, leav-

ing an *o*-quinone (**B**)<sup>6</sup>). The intramolecular Michael reaction and the concomitant enolization of **B** produced **2a**.



Accordingly,  $\text{CH}_3\text{CN}$  was used as the more polar solvent to ensure an effective capture of the methyl cation. Thus, when **1a** was treated with the reagent in 20 ml of  $\text{CH}_3\text{CN}$ , the yield of **2a** was raised to 52%, together with 15% yield of **3**. On the other hand, **2a** was produced as a sole product in 63.3% yield, when the same reaction was conducted in 50 ml of the solvent.

Similarly, **1b** was converted to **2b**<sup>7</sup>, m.p. 195–197°, in 70.9% yield.

A typical procedure: The *o*-QA (**1a**) prepared from the 6-phenolic tetrahydroisoquinoline (100 mg) as described previously<sup>1</sup> was dissolved in  $\text{CH}_3\text{CN}$  (50 ml). To the ice-cold, stirred solution,  $\text{Ac}_2\text{O}$  (1 ml) and conc.  $\text{H}_2\text{SO}_4$  (0.1 ml, drop by drop) were added successively and stirring was continued at room temperature for 2 hr. Usual work-up of the reaction mixture gave an amorphous mass (113 mg), whose purification by preparative TLC gave **2a** (76 mg, 63.3%), m.p. 147–148° (benzene-*n*-hexane).

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2. Satisfactory analytical data were obtained for all new compounds described.
3. Cf. H. Bredreck, R. Sieber, and L. Kamphenkel, *Chem. Ber.*, 1956, **89**, 1169.
4. B. Gregso-Allcott and J. M. Osbond, *Tetrahedron Letters*, 1969, 1771.
5. S. M. Kupchan and C.-K. Kim, *J. Am. Chem. Soc.*, 1975, **97**, 5623.
6. The formation of **B** was implicitly indicated by the appearance of a red coloration during the reaction.
7. Spectral data; IR  $\nu(\text{CHCl}_3)$ : 1770  $\text{cm}^{-1}$ ; NMR  $\delta(\text{CDCl}_3)$ : 2.29, 2.30 (each 3H, s, 2 x  $\text{CH}_3\text{COO}$ ), 2.54 (3H, s,  $\text{NCH}_3$ ), 5.96 (2H, s,  $\text{OCH}_2\text{O}$ ), 6.76, 6.88, 7.41 (each 1H, s, 3 x arom. H).

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