A New and Efficient Approach to a 4H-1,3-Benzothiazine Ring that Utilises the Photo-cyclisation of N-o-Iodobenzoyl-thioamides: the Ring Transformation of Isothiazoles

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<u>Abstract</u> The photo-cyclisation of N-o-iodobenzoylthioamides has been found to provide a novel and efficient synthesis of 4H-1,3-benzothiazin-4-ones.

Approaches to the synthesis of 4H-1,3-benzothiazines so far reported utilise appropriately substituted o-mercapto-benzoic acids and benzamides, 1 N-(phenylthio-methyl)benzamides, 2 and 1,2-benzoisothiazoles. 3 We now report a new and efficient synthetic method of this ring system which involves the photo-cyclisation 4 of N-o-iodobenzoylthioamides (6)-(10), prepared from 4-aryl-3-o-iodobenzoylthio-3-isothiazoline-5-thiones (2)-(4) as described earlier. 5

A solution of the N-o-iodobenzoylthioamide (6) (0.10 g) in dry THF (580 ml) was irradiated with a high pressure mercury lamp (100 W) for 10 h; the solution was deaerated by a stream of N₂ before and during photolysis. Evaporation to dryness and silica gel chromatography of the residue with benzene followed by chloroform gave the benzothiazin-4-one (11) in 93% yield [m.p.>300°C, yellow needles (from MeCN). Found: C, 56.10; H, 2.98; N, 2.96; M⁺, 469 (FD Ms). $C_{22}H_{15}NO_{5}S_{3}$ requires C, 56.27; H, 3.22; N, 2.98%; M⁺, 469}. Its structure was assigned from spectral data [v_{max} . (nujol) 1645, 1725, and 1740 cm⁻¹ (C=0); λ_{max} . (CHCl₃) 332 (log ε 3.74) and 433 nm (4.59); δ_{H} (CF₃CO₂D) 4.10 (s, 6H), 7.33-7.53 (m, 3H), 7.70-8.03 (m, 5H), and 8.60 (d, \underline{J} 8 Hz, 1H)]. This cyclisation proceeds more inefficiently when the corresponding chloro-derivative is used; thus the compound (11) was obtained in 26% yield after 50 h of irradiation of (5) [prepared from (1)].

Similarly, the N-o-iodobenzoylthioamides (7)-(10) afforded the following benzothiazin-4-ones (12) [m.p. $>300^{\circ}$ C, 80% yield], (13) [m.p. $>300^{\circ}$ C, 83% yield], (14) [m.p. $222-223^{\circ}$ C (decomp.), 86% yield], and (15) [m.p. $239-240^{\circ}$ C (decomp.), 84%

$$\begin{array}{c}
S \\
S
\end{array}$$

$$\begin{array}{c}
C(Ar) = S \\
S
\end{array}$$

- (1) Ar=Ph, X=C1
- (5) Ar=Ph, X=C1, E=CO₂Me
- (11) Ar=Ph, E=CO₂Me

- (2) Ar=Ph, X=I
- (6) Ar=Ph, X=I, E=CO₂Me (12) Ar=p-MeC₆H₄, E=CO₂Me
- (3) $Ar=p-MeC_6H_4$, X=I
- (7) $Ar=p-MeC_6H_4$, X=I, $E=CO_2Me$ (13) $Ar=p-C1C_6H_4$, $E=CO_2Me$
- (8) $Ar=p-C1C_6H_4$, X=I, $E=CO_2Me$ (14) Ar=Ph, $E=CO_2Et$
- (4) Ar=p-ClC₆H₄, X=I
- (9) Ar=Ph, X=I, E=CO₂Et
- (15) $Ar=p-MeC_6H_4$, $E=CO_9Et$
- (10) Ar=p-MeC₆H₄, X=I, E=CO₂Et

yield], for which satisfactory microanalytical and spectral data have been obtained. Synthetical value of the above-mentioned photochemical heterocyclisation of the N-o-iodobenzoylthioamides is further obvious from the fact that their thermal cyclisation, unlike that of N-chloroacetylthioamides, 5b is very slow and unprofitable. For instance, heating of 9 in THF for 24 h gave the benzothiazinone (14) in only 7% yield, together with other products.

Although the present study was carried out from our continued interest in the ring transformation reactions of isothiazoles, 5b the reactions given herein appear promising as a synthetic method of 4H-1,3-benzothiazin-4-ones in the light of ready availability 6 of simple N-benzoylthioamides, and studies of relevance are now in progress.

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