REACTIONS OF 4-ETHOXYALKYLIDENE-2-PHENYL-2-OXAZOLIN-5-ONES WITH 1,3-BINUCLEOPHILIC HETEROAROMATIC SYSTEMS

Otohiko Tsuge\* and Michihiko Noguchi
Research Institute of Industrial Science, Kyushu University 86,
Hakozaki, Higashi-ku, Fukuoka 812, Japan

<u>Abstract</u> — One-pot syntheses of pyrido[1,2-b]pyrimidine and thiazolo[2,3-b]pyrimidine derivatives were achieved by the reactions of 4-ethoxyalkylidene-2-phenyl-2-oxazolin-5-ones with 2-aminopyridines and 2-aminothiazole in refluxing ethanol, respectively. In the reaction of the oxazolinone with 2-aminopyrimidine under similar conditions, however, the expected pyrimido-pyrimidine derivative was not formed, but instead (2-pyrimidylaminomethylene)-oxazolinone and its ethanolysis product were obtained.

Since 4-methylene-2-oxazolin-5-ones have two reactive sites towards nucleophiles, their ring opening reactions are complicated  $^1$ . Among 4-methylene-2-oxazolin-5-ones, however, 4-ethoxymethylene-2-phenyl-2-oxazolin-5-one (1) $^1$  is not only a useful intermediate for the synthesis of amino acids $^{2-4}$ , but also an effective reagent reacting with binucleophiles such as diethyl 3-oxoglutarate $^4$ , acetyl-acetone $^5$ , o-phenylenediamine $^6$ , and o-aminophenol $^7$ , because the ethoxymethylene group of 1 is very reactive towards nucleophiles.

$$R^2$$
 :Nu  $R^2$  :Nu  $R=H$   $R=H$ 

In this paper we wish to report the reactions of 1 and 4-ethoxyethylidene-2-phenyl-2-oxazolin-5-one (2)<sup>8</sup> with 1,3-binucleophiles such as 2-aminopyridines, 2-aminothiazole, and 2-aminopyrimidine.

Reaction with 2-Aminopyridines. Oxazolinone 1 reacted with an equimolar amount of 2-aminopyridine (3a) in refluxing benzene for 2 h to give 2-phenyl-4-(2-pyridylaminomethylene)-2-oxazolin-5-one (4a) in 98% yield [4a: yellow needles; mp 168-169°C; IR (KBr) 3250, 1760, 1730, 1660 cm<sup>-1</sup>; H NMR (CDCl<sub>3</sub>)

 $\delta$  8.47 (1H, broad s, =CH), 8.55 (1H, broad, NH, exchanged with D<sub>2</sub>O); MS m/e 265 (M<sup>+</sup>)]<sup>9</sup>.

Next, cyclization of 4a by nucleophilic attack of the nitrogen of pyridine ring at the oxazolinone moiety has been investigated under various conditions; it has been found that upon only heating in ethanol or acetic acid 4a was transformed into the expected 4-oxopyrido[1,2-b]pyrimidine 5a in a good yield (Scheme 1). This fact suggested the possibility of one-pot synthesis of 5a from the reaction of 1 with 3a in refluxing ethanol. In fact, 1 reacted with 3a in refluxing ethanol to afford 5a in a good yield.

Similarly, the reaction of 1 with four methyl-substituted 2-aminopyridines, 3b, 3c, 3d, and 3e, in refluxing ethanol afforded the corresponding pyridopyrimidines, 5b, 5c, 5d, and 5e, in good yields

Scheme 1

except for 5e. The results are shown in Table 1.

Also, oxazolinone 2 reacted with 3a under similar conditions to give the pyridopyrimidine 6.

Structural elucidation of pyridopyrimidines 5 and 6 was accomplished on the basis of spectral data (Table 2).

Table 1.	Reactions of Oxazolinone $\underline{1}$ with 2-Aminopyridines $\underline{3}$
	in Refluxing Ethanol

2-Aminopyridine	Reaction time h	Pyridopyrimidine Yield, %	
3a (R=H)	30	5a (R=H)	88
<u>3b</u> (R=3-Me)	72	<u>5b</u> (R=9-Me)	89
3c (R=4~Me)	30	<u>5c</u> (R=8-Me)	93
<u>3d</u> (R=5-Me)	30	<u>5d</u> (R=7-Me)	91
<u>3e</u> (R=6-Me)	120	<u>5e</u> (R=6-Me)	38 <sup>a</sup>

 $<sup>^{</sup>a}4$ -(6-Methyl-2-pyridylaminomethylene)-2-phenyl-2-oxazolin-5-one  $(4e)^{11}$  was isolated in 51% yield.

Table 2. Physical and Spectral Data of Pyridopyrimidines 5 and  $6^a$ 

<u>5a</u>: mp 196.5-198°C<sup>10</sup>; colorless spears; IR 3300, 1665, 1640 cm<sup>-1</sup>; <sup>1</sup>H NMR & 7.17 (1H, ddd, 7- $\underline{H}$ , J= 6.5, 5.5, 2.5 Hz), 7.40-7.85 (5H, m), 7.90-8.15 (2H, m), 8.85 (1H, broad, N $\underline{H}$ ), 8.98 (1H, pseudo double t, 6- $\underline{H}$ , J=6.5, 1.0, 1.0 Hz), 9.77 (1H, s, 2- $\underline{H}$ ); MS m/e 265 (M $^+$ ).

<u>5b</u>: mp 171.5-173°C; colorless needles; IR 3410, 1665 cm<sup>-1</sup>; <sup>1</sup>H NMR  $\delta$  2.61 (3H, s), 7.04 (1H, pseudo triple d, 7-<u>H</u>), 7.40-7.70 (4H, m), 7.90-8.10 (2H, m), 8.86 (2H, dd and broad, 6-<u>H</u> and N<u>H</u>), 9.72 (1H, s, 2-H); MS m/e 279 (M<sup>+</sup>).

5c: mp 217.5-218.5°C; pale yellow prisms; IR 3400, 1650 cm<sup>-1</sup>;  $^{1}$ H NMR  $\delta$  2.49 (3H, s), 7.00 (1H, dd, 7- $\underline{\text{H}}$ , J=8.0, 1.0 Hz), 7.40-8.20 (6H, m), 8.78 (1H, broad, N $\underline{\text{H}}$ ), 8.88 (1H, dd, 6-H, J=8.0, < 1 Hz), 9.69 (1H, s, 2-H); MS m/e 279 (M $^{+}$ ).

5d: mp 206-207°C; colorless spears; IR 3300, 1650, 1635 cm<sup>-1</sup>;  $^{1}$ H NMR & 2.44 (3H, s), 7.30-7.80 (5H, m), 7.90-8.20 (2H, m), 8.86 (1H, broad, NH), 9.71 (1H, s, 2-H); MS m/e 279 (M<sup>+</sup>).

<u>5e</u>: mp 159~160°C; pale yellow needles; IR 3400, 1680, 1650 cm<sup>-1</sup>;  $^{1}$ H NMR  $_{\delta}$  3.08 (3H, s), 6.40 (1H, dd, 7-H), 7.20-8.0 (7H, m), 8.70 (1H, broad, NH), 9.43 (1H, s, 2-H); MS m/e 279 (M<sup>+</sup>).

6: mp 201-202°C; colorless spears; IR 3310, 1675 cm<sup>-1</sup>;  $^{1}$ H NMR & 2.54 (3H, s), 7.12 (1H, m, 7- $^{H}$ ), 7.40-7.60 (3H, m), 7.60-7.80 (2H, m), 7.80-8.0 (2H, m), 8.55 (1H, broad, N $^{H}$ ), 8.94 (1H, m, 6- $^{H}$ ); MS m/e 279 (M $^{+}$ ).

 $<sup>^{</sup>m a}$ IR and  $^{
m l}$ H NMR spectra were taken in KBr disks and CDCl $_{
m 3}$  solutions, respectively.

<sup>&</sup>lt;sup>b</sup>13<sub>C</sub> NMR (CDC1<sub>3</sub>) & 115.7, 119.4, 126.5, 126.8, 127.2, 128.8, 132.1, 133.1, 133.9, 141.4, 145.9, 153.5, 165.6.

Reaction with 2-Aminothiazole. Next, the reaction of oxazolinones 1 and 2 with 2-aminothiazole ( $\chi$ ) has been investigated. Even when oxazolinone 1 was heated with  $\chi$  in ethanol for 5 days, the major product was 2-phenyl-4-(2-thiazolylaminomethylene)-2-oxazolin-5-one (8), and the expected thiazolo[2,3-b]pyrimidine 9 was formed in a low yield. When 8 which was easily formed from the reaction of 1 with  $\chi$  in refluxing benzene was heated in acetic acid, however, 9 was obtained in a good yield. On the other hand, oxazolinone 2 reacted with  $\chi$  in refluxing ethanol to afford the corresponding thiazolopyrimidine 10 (Scheme 2).

Scheme 2

Structural elucidation of 8, 9, and 10 was again accomplished on the basis of spectral data.

8: mp 188-189°C; yellow needles; IR (KBr) 3100-2400, 1775 cm<sup>-1</sup>;  $^{1}$ H NMR (CDCl<sub>3</sub>) & 7.20, 7.41 (each 1H, d, J=3.5 Hz), 7.40-7.70 (3H included dd at & 7,41, m), 7.90-8.10 (2H, m), 7.99 (1H, s, =CH), 11.90 (1H, broad, NH); MS m/e 271 (M<sup>+</sup>).

9: mp 187-188°C; pale yellow needles; IR (KBr) 3400, 1660 cm<sup>-1</sup>;  $^{1}$ H NMR (CDCl<sub>3</sub>) & 7.32 (1H, d, J=5.0 Hz), 7.40-7.70, 7.90-8.10 (each 3H, m), 8.67 (1H, broad, NH), 9.43 (1H, s); MS m/e 271 (M<sup>+</sup>).

10: mp 206-208°C; colorless spears; IR (KBr) 3300, 1655 cm<sup>-1</sup>;  $^{1}$ H NMR (CDCl<sub>3</sub>) & 2.42 (3H, s), 6.96, 7.86 (each 1H, d, J=5.0 Hz), 7.30-7.60 (3H, m), 6.80-7.10 (4H (1H was exchanged with D<sub>2</sub>0), m, aromatic protons + NH); MS m/e 285 (M<sup>+</sup>).

Reaction with 2-Aminopyrimidine. When oxazolinone 1 was heated with 2-aminopyrimidine (11) in ethanol for 30 h, 2-phenyl-4-(2-pyrimidinylaminomethylene)-2-oxazolin-5-one (12) and its ethanolysis product 13 were formed in 38 and 31% yields, respectively. The compound 12 was obtained in 81% yield from the same reaction in refluxing dioxan for 1 h. Even when 12 was heated in acetic acid for 24 h, however, 12 was unchanged.

$$1 + \left( \frac{N}{N} \right) \xrightarrow{NH_2} Ph \xrightarrow{N} O + \left( \frac{N}{N} \right) \xrightarrow{NHCOPh} \frac{N}{NHCH = CCO_2Et}$$

$$11 \qquad 12 \qquad 13$$

12: mp 243-244°C (dec); yellow needles; IR (KBr) 3300-2400, 1795 cm<sup>-1</sup>; MS m/e 266 (M<sup>+</sup>).

13: mp 181-182°C, colorless needles; IR (KBr) 3310, 1695, 1660 cm<sup>-1</sup>;  $^{1}$ H NMR (CDCl<sub>3</sub>) & 1.37 (3H, t, J=7.0 Hz), 4.29 (2H, q, J=7.0 Hz), 6.78 (1H, t, J=5.0 Hz), 7.30-7.60 (3H, m), 7.80-8.00 (2H, m), 8.29 (1H, d, =CH, J=11.5 Hz, changed to a singlet when treated with D<sub>2</sub>0), 8.30 (1H, broad, NH, exchanged with D<sub>2</sub>0), 8.42 (2H, d, J=5.0 Hz), 9.72 (1H, broad d, NH, J=11.5 Hz, exchanged with D<sub>2</sub>0); MS m/e 312 (M<sup>+</sup>).

## References and Notes

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- Although compound 4a is recorded in Ref. 1, its synthetic method and physical properties are not described. All new compounds in this paper gave satisfactory elemental analyses.
- 10. It has been reported that the treatment of 4a with sodium ethoxide in ethanol, followed by decomposition of the formed lemon-yellow compound with water afforded 5a, mp 168°C1. However,

5a obtained here melted at 196.5-198 $^{\circ}$ C.

11. <u>4e</u>: mp 194-195°C; IR (KBr) 3300, 1775, 1730 cm<sup>-1</sup>; <sup>1</sup>H NMR (CDC1<sub>3</sub>)  $\delta$  2.50 (3H, s), 6.72 (1H, dd, J=8.0, 1.0 Hz), 6.88 (1H, dd, J=7.5, 1.0 Hz), 7.40-7.70 (4H, m), 7.90-8.10 (2H, m), 8.45 (1H, broad, N<u>H</u>), 8.48 (1H, s, =C<u>H</u>); MS m/e 279 (M<sup>+</sup>).

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