A CONVENIENT SYNTHESIS OF ACETYLCHROMANS

Vinod Kumar Ahluwalia*, Krishan Kumar Arora and Ravinder Singh Jolly

Department of Chemistry, University of Delhi, Delhi-110007, India

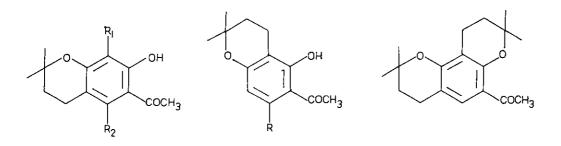
<u>Abstract</u> - A convenient one-step synthesis of acetylsubstituted 2,2-dimethylchromans involving the condensation of polyhydroxyacetophenones with 2methylbut-3-ene-2-ol in presence of orthophosphoric acid is described.

The utility of 2,2-dimethylchromans bearing an acetyl function has been shown for the synthesis of a number of naturally occurring acetylchromenes^{1,2}. Such chromans and chromenes are also useful starting materials for the synthesis of many other aromatic hemiterpenes^{3,4}. Earlier methods for synthesis of chromans involve the use of Grignard reagent⁵ on coumarins or Clemmensen reduction⁶ of chromanones. These methods are not suitable for the synthesis of acetylchromans. One attempt to synthesize acetylchromans by the condensation⁷ of 2-methylbut-3-ene-2-ol with polyhydroxyacetophenones in presence of 5% citric acid (aq.) resulted in a mixture of products from which acetylchromans could be isolated only in very poor yield (3-6%). In the past such acetylchromans have been obtained by Friedel-Crafts type reaction on the hydroxychromans^{8,9} or by acid-catalysed cyclisation of the corresponding o-hydroxyprenyl compounds¹⁰ which themselves are difficult to prepare. Recently a novel method for synthesis of such chromans have been reported from our laboratory, which involves the direct condensation of polyhydroxyacetophenones with isoprene¹.

During the course of our work on the synthesis of precocenes, acetylchromenes and related compounds, we have been able to develop another convenient and one-step synthesis of acetylchromans (yield, 60-80%). The reaction involves the condensation of 2-methylbut-3-ene-2-ol with polyhydroxyacetophenones in presence of orthophosphoric acid.

The typical experimental procedure consists in the addition of 2-methylbut-3-ene-2-ol (0.8 ml) in petroleum ether (b.p. $60-80^{\circ}$; 5 ml) to a well stirred suspension of 2,3,4-trihydroxyacetophenone (1 g), orthophosphoric acid (85%, 1 ml) and petroleum ether (5 ml) during 8 hr at 40° C and stirring continued for further 12 hr. The mixture was neutralized with sodium bicarbonate and extracted with ether. The ethereal extract was dried (Na₂SO₄) and distilled. The crude product, thus obtained, was crystallised from benzene - petroleum ether to give a colourless crystalline

compound (1.1 g), m.p.123-124^{\circ}, which was assigned the structure 6-acety1-7,8-dihydroxy-2,2dimethy1-3,4-dihydro-2<u>H</u>-1-benzopyran (1) on the basis of its NMR and direct comparison (m.p., mixed m.p. and superimposed IR) with an authentic sample¹.



$\begin{array}{c}1\\ \end{array}, \begin{array}{c}R_1 = 0H, \begin{array}{c}R_2 = H\end{array}$	2, R = H	4 ~
$^{3}_{\sim}$, $^{R}_{1} = ^{R}_{2} = ^{H}_{2}$	$\stackrel{5}{\sim}$, R = OCH ₃	
6, $R_1 = H$; $R_2 = OCH_2$		

2,4-Dihydroxyacetophenone on similar condensation gave 6-acety1-5-hydroxy-2,2-dimethy1-2<u>H</u>-1benzopyran (2), m.p.69-70°, 6-acety1-7-hydroxy-2,2-dimethy1-2<u>H</u>-1-benzopyran (3) m.p.118-119° and 6-acety1-2,2,8,8-tetramethy1-3,4,9,10-tetrahydro-2<u>H</u>, 8<u>H</u>-benzo $[1,2-\underline{b}:3,4-\underline{b}']$ dipyran (4), m.p.78-79°, in the ratio of 2:2:1 (overall yield, 70%) respectively. Similarly 2,4-dihydroxy-6methoxyacetophenone gave 6-acety1-5-hydroxy-7-methoxy-2,2-dimethy1-3,4-dihydro-2<u>H</u>-1-benzopyran (5), m.p.92-93° and 6-acety1-7-hydroxy-5-methoxy-2,2-dimethy1-3,4-dihydro-2<u>H</u>-1-benzopyran (6), m.p.52-53°, in the ratio of 1:1 (overall yield 60%) respectively. All the above products were separated by column chromatography over silica gel and structures given on the basis of ¹H NMR spectral data and direct comparison (m.p., mixed m.p. and superimposed IR) with the authentic samples¹. The yields obtained are comparable to those obtained in the reaction of isoprene with polyhydroxyacetophenones¹.

We acknowledge the financial assistance provided by CSIR, New Delhi, India.

REFERENCES

- 1. V.K.Ahluwalia and K.K.Arora, Tetrahedron, 1981, 37, 1437.
- 2. G.Cardillo, R.Cricchio and L.Merlini, Tetrahedron, 1968, 24, 4825.
- 3. R.J.Molyneux and L.Jurd, Tetrahedron, 1970, 26, 4743
- 4. S.Kumar, V.V.S.Murty and T.R.Seshadri, Indian J. Chem., 1971, 9B, 1319.
- 5. R.Adams, C.K.Cain and S.Loewe, J. Am. Chem. Soc., 1941, 63, 1977.
- 6. H.B.Bhat and K.Venkataraman, Tetrahedron, 1963, 19, 77.
- 7. B.S.Bajwa, P.Lal and T.R.Seshadri, Indian J. Chem., 1973, 11B, 100.
- 8. M.Nakayama, M.Ohno and K.Fukui, Bull. Chem. Soc. Japan, 1970, 43, 3311.
- 9. T.Backhouse and A.Robertson, J. Chem. Soc., 1939, 1251.
- 10. R.D.Allan, R.J.Wells and J.K.McLeod, Tet. Lett., 1970, 3945.

Received, 24th July, 1981