RING TRANSFORMATION OF 1,4-DITHIIN TO 1,3-DITHIOLE

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2,5-Diphenyl-1,4-dithiin-1-oxide (<u>1</u>) when heated in acetonitrile around 70°C afforded 2-formyl-2,4-diphenyl-1,3-dithiole (<u>2</u>) along with 2,4-diphenylthiophene. The yield of <u>2</u> was increased when the thermolysis was carried out in carbon tetrachloride, whereas in methanol or dimethyl sulfoxide <u>1</u> gave only the thiophene due to the extrusion of sulfur oxide. The thermal rearrangement was assumed to proceed through 1,2,5-oxadithiepin (<u>3</u>) as an intermediate in connection with the mass spectral behaviour of 1.

In contrast to the thermal reaction, photolysis of $\underline{1}$ in methanol under a nitrogen atmosphere gave rise to another rearranged product, 2-benzoyl-4-phenyl-1,3-dithicle (4) as well as 2.

Treatment of 1 with hydrochloric acid-dioxane (1:2 V/V) caused again a skeletal rearrangement to give 4. However, when gaseous HCl was introduced to a methanol solution of 1, the additive Pummerer product 5 was obtained, which was further converted into 4 by heating in 95% ethanol. The transformation of 1 to 4 initiated by the additive Pummerer reaction was explained by means of the carbonium ion rearrangement.

Ph S CI