## THE ISOLATION AND STRUCTURE OF NAHAGENIN

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ABSTRACT: — The isolation and structure of a new sapogenin, nahagenin (1), from the medicinal plant Fagonia indica Linn is presented.

Fagonia indica Linn<sup>1</sup> (zygophyllaceae) is a small spiny undershrub which is widely distributed in Pakistan. An aqueous decoction of the leaves and young twigs is a popular remedy for the treatment of various skin lesions. The plant is claimed to be a remedy for cancer in its early stages and preliminary pharmacological tests of aqueous extracts on mice have shown some anti-cancer activity. An Ames mutagenicity test has also indicated marginal activity. Prior investigations have revealed triterpenoids, to carbohydrates, saponins, alkaloids, sapogenins, fatty acids and amino acids. We now wish to report the isolation of a new sapogenin, nahagenin (1).

The crude saponins, obtained from the ethanolic extract of the aerial parts of the fresh plants, were hydrolysed with 20% ethanolic HCl to afford the corresponding sapogenins. Silica gel chromatography with elution by pet. ether, pet. ether-CHCl<sub>3</sub>, CHCl<sub>3</sub>, and finally CHCl<sub>3</sub>-MeOH afforded two crystalline compounds which were further purified by crystallization. The first of these nahagenin (1), m.p. 290°C, analyzed for  $C_{30}H_{48}O_4$  and this was confirmed by high resolution mass spectrometry (m/z = 472.3540 meas, 472.3552 for  $C_{30}H_{48}O_4$ ). Major peaks in the MS occurred at m/z of 454, 436, 424, 409, 395 and 261. The IR spectrum (CHCl<sub>3</sub>) showed peaks at 1740 cm<sup>-1</sup> and 3460 cm<sup>-1</sup> suggesting a  $\delta$ -lactone and hydroxy groups. The substance readily afforded a diacetate (m/z = 556) but was found to be remarkably inert to attempted hydrolysis of the lactone.

The <sup>1</sup>H NMR showed no olefinic protons. The <sup>13</sup>C NMR (100 MHz) confirmed the presence of thirty carbons and chemical shifts and multiplicities are given

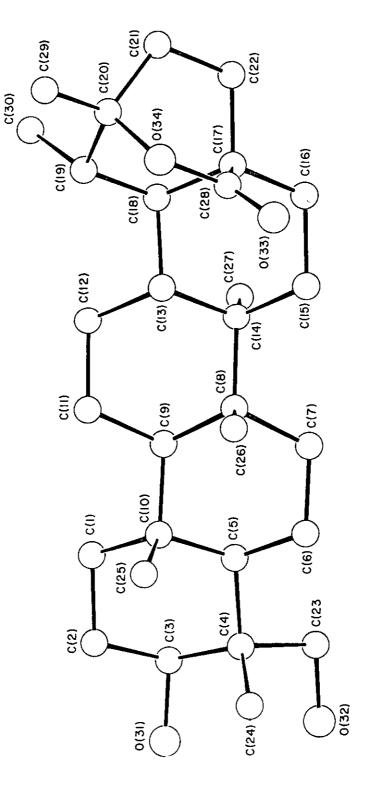


Figure 1. A computer generated perspective drawing of nahagenin. Hydrogens are omitted for clarity.

in Table 1. The carbon atoms in the A and B rings were readily recognized by comparison with corresponding signals of known pentacyclic triterpenoids. <sup>16</sup> Six quaternary centers and six methyl groups were also identified. The <sup>13</sup>C NMR displayed a resonance at  $\delta$  177.29 for the carbonyl carbon and three resonances at  $\delta$  84.27(s), 76.54(d) and 71.92(dd) for the oxygen bearing carbons C(20), C(3) and C(23) respectively.

An unambiguous structure determination was carried out by a single crystal x-ray diffraction analysis. Crystals of nahagenin formed in the monoclinic system with a = 12.370(2), b = 6.7688(8), c = 15.832(2) Å and  $\beta$  = 100.28(1)  $^{\rm O}$ . All unique diffraction maxima with 20  $\leq$  114 were collected on an automated four-circle diffractometer with variable speed, 1  $^{\rm O}$   $\omega$ -scans. Of the 1972 reflections surveyed, 1509 (77%) were judged observed (|Fo|  $\leq$  30(Fo)) after correction for Lorentz, polarization and background effects. The structure was solved by a multisolution weighted tangent formula approach with five special and one general reflection forming the variable starting set. The structure was solved by a multisolution weighted tangent formula approach with five special and one general reflection forming the variable starting set. The fullmatrix least-squares refinements with anisotropic nonhydrogens and fixed isotropic hydrogens have converged to a standard crystallographic residual of 0.0585. Figure 1 is a computer generated perspective drawing of the final x-ray model.

As expected nahagenin is a pentacyclic triterpene with a  $\delta$ -lactone added to the E ring. All of the carbocyclic rings are joined in a <u>trans-anti</u> fashion. In general bond distances and angles agree well with general accepted values. <sup>18</sup> Preliminary studies on the saponins present indicate that nahagenin is a genuine aglycone of the parent saponin, the structure of which is currently under investigation.

Carbon	Chemical Shift(s)	Multiplicity	Carbon	Chemical Shift(s)	Multiplicity
C(1)	38.51	d.d	C(16)	25.15	t
c(2)	27.28	t	C(17)	42,06	5
c(3)	76.54	d	c(18)	50.51	d
c(4)	41.86	s	C(19)	48.37	đ
c(s)	49.99	d	C(20)	84.27	s
c(6)	18.35	t	C(21)	41.98	t(?)
c(7)	32,22	t	C(22)	33.68	t
c(8)	40.51	s	c(23)	71.92	d.d
c (9)	42.83	d	C(24)	11.30	q
c(10)	37.05	5	C(25)	15.69	q
C(11)	20.96	t	C(26)	16.63	q
C(12)	27.05	t	C(27)	23.98	q
C(13)	27.59	d	C(28)	177.29	s
c(14)	41.08	s	C(29)	14.26	٩
C(15)	26.87	t	c(30)	23.98	q

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