CONDENSATION OF 2-HYDROXYTRYPTAMINE WITH DIMETHYL 4-ETHYL-4-FORMYLPIMELATE Esahak Ali, Pulak K. Chakraborty and Satyesh C. Pakrashi* Indian Institute of Chemical Biology, Calcutta-700 032, India

<u>Abstract</u> - Condensation of 2-hydroxytryptamine (<u>1</u>) with dimethyl 4-ethyl-4-formylpimelate (<u>2</u>) yielded the stereoisomeric oxindoles, <u>8A</u>, <u>8B</u>, <u>8C</u>, and <u>8D</u> which were separated and their stereochemistry assigned. The compounds <u>8A-D</u> or their N-methyl derivatives, <u>5A-D</u> could be selectively reduced to the oxindole-amines <u>9</u> or <u>3</u> by treatment with phosphorus oxychloride followed by sodium borohydride but the amines were isomerised under the reaction condition.

The condensation of 2-hydroxytryptamine $(\underline{1})$ with dimethyl 4-ethyl-4-formylpimelate ($\underline{2}$) is a potentially useful route for the total synthesis of several indole alkaloids. In an attempted synthesis of the alkaloid vincatine ($\underline{3}$)¹, the reaction of 1-methyl-2-hydroxytryptamine ($\underline{4}$) with $\underline{2}$ was reported² to yield two stereoisomers of the oxindole $\underline{5}$. Condensation of $\underline{1}$ with $\underline{2}$ was later used for the total synthesis of ($\underline{+}$) - vincadifformine ($\underline{6}$)³ and ($\underline{+}$) - quebrachamine ($\underline{7}$)⁴. This reaction was also reported to yield two isomers of $\underline{8}$, the more polar one being amenable to cyclisation to vincadifformine³ or aspidospermidine⁴ skeleton. A recent publication⁵ by the French group prompted us to report the work carried out in this laboratory on the condensation of $\underline{1}$ with $\underline{2}$, which showed that the reaction yields all the four possible isomers of $\underline{8}$.

Reaction of 2-hydroxytryptamine (<u>1</u>) with the ester <u>2</u> in refluxing methanol in presence of acetate buffer (pH 4.6) yielded, as reported^{2,3} earlier, two products (ratio 6:4, total yield 76%) separable by thin layer or column chromatography over silica gel. The nmr spectra, however, indicated that each of them is a mixture of two compounds which could be resolved by reversed phase HPLC (μ -Bondapak C-18, MeOH:H₂O/2:1). We have also been able to successfully separate the four isomers <u>8A-D</u> by fractional crystallisation, their relative proportion (by HPLC) being A:B:C:D = 18:22.5:32:27.5. The corresponding N-methyl derivatives <u>5A-D</u> were prepared by methylation with MeI/NaH in DMF.

As expected, the uv spectra of all the compounds <u>BA-D</u> and <u>5A-D</u> were almost superimposable⁶. The mass spectra of the isomers <u>BA-D</u> were also almost identical [for <u>BA</u>, <u>m/z</u> 370 (M⁺, 100%), 341 (M-Et, 6%), 339 (M-OMe, 11%), 283 (M-CH₂CH₂COOMe, 6%), 225 (<u>a</u>, 16%), 212 (<u>b</u>, 23%), 196 (<u>c</u>, 39%), 159 (<u>d</u>, 59%) and 138 (<u>e</u>, 67%)], with appropriate mass shifts in the spectra of the N-methyl derivatives. The ir spectra of the compounds were also very close. Since the chemical shifts of the C-21 methine proton in all the eight compounds were observed in the narrow range of δ 4.12-4.20 (Table) all of them must have the same stereochemistry at the C/D ring junction, i.e. <u>trans</u>. The C-7 configuration of the isomers could be unambiguously deduced by comparison of the ¹³C chemical shifts of C-7 and C-9 with those reported for model compounds⁷. A distinction between the isomers <u>BA</u> and <u>BB</u> and between <u>8C</u> and <u>8D</u> could be made from the chemical shifts of the C-17 and C-19 which were expected to resonate at higher field in the axial isomer than in the corresponding equatorial isomer⁸. The shielding of the CH₃-CH₂- groups in the pmr spectra of <u>8B</u> and <u>8C</u> can then be explained as due to anisotropy of the aromatic ring.

The isolation of all the four possible isomers of $\underline{8}$ thus, disproves the earlier contention² that two of the isomers are unlikely to be produced because of steric interactions. We have also observed that the lactams <u>BA-D</u> are not isomerised by acetic acid or pyridine at 100°C, the usual conditions for equilibration of oxindole alkaloids⁹.

For the synthesis of indole alkaloids, it is necessary to selectively reduce the δ -lactam functions of 5 and 8. Diborane or lithium aluminium hydride reduction of 5 and 8 have been reported to cleave the C_7-C_{21} bond^{1,4}. Potassium borohydride reduction of a derivative of 9 was also reported⁴ to cleave this bond. We could, however, achieve the desired reduction by treatment of the lactams 5 and 8 with POCl₃ followed by sodium borohydride reduction¹⁰ when the oxindole amines 3 and 9 were produced in reproducible yields of 60-65%. However, the stereochemical integrity was not preserved during the reduction. In fact, reduction of any of the isomers of 5 or 8 produced apparently the same mixture of the isomers of 3 or 9. It could be confirmed that isomerisation of 5 and 8 did not take place during the POCl₃ treatment. Evidently, the different stereoisomers of 3 and 9 are equilibrated under the conditions of borohydride reduction. Thus, though the δ -lactam group in 5 and 8 could be selectively reduced, the stereospecific synthesis of vincatine (3) could not be achieved. That goal could, however, be reached by an alternative route reported in the succeeding communication⁸.



Compound	mp (°C)	ir (nujol), cm ⁻¹	pmr (90/100 MHz), δ CDCl ₃			
			CH ₃ CH ₂	OMe	21-н	NH/NMe
8A*	122-124	3413,1705,1618	0.86	3.60	4.20	9.46
8 B *	238-240	3100,1718,1613	0.63	3.63	4.20	9.56
8C**	236-238	3236,1720,1646,1616	0.74	3.56	4.16	9.43
8D**	256-258	3170,1720,1611	0.56	3.56	4.20	9.35
5A	126-128	1730,1706,1643,1613	0.83	3.60	4.16	3.30
5B	152-154	1730,1706,1643,1613	0.58	3.70	4,20	3.30
5C	198-200	1738,1693,1640,1613	0.74	3.52	4.12	3.22
5D	212-214	1724,1693,1635,1613	0.57	3.63	4.13	3.24

Table. Physical data of the lactams <u>8A-D</u> and <u>5A-D</u>

*Isolated from the less polar product. **Isolated from the more polar product.

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- 6. For <u>8A</u>, λ_{max} (EtOH) (Log ϵ) : 209 (4.50), 256 (3.88) and 286 (3.20) nm.
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