ADVANCES IN CYCLOPENTENONE SYNTHESIS FROM FURANS Giovanni Piancatelli

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Abstract  $--$  Both 2-furylcarbinols and their  $5$ -methyl derivatives are turned into the isomeric cyclopentenones through a molecular rearrangement catalyzed by acids or zinc chloride. The reaction **was** completely stereospecific and its mechanism **was** explained in terms of a thermal electrocyclic reaction of a 4n-electrons system; it showed a vide applicability and made easily available a large variety of cyclopentenones.

The oxidative ring fission of 2,5-dialkylfurans by pyridinium chlorochromate gave trans-ene dicarbonyl compounds; these were efficient building blocks for the cyclopentenone synthesis and several applications are described.

**2-(5-methyl-2-fu~yl)-1,3-dicarbbbyl** compounds, a **new** class of furan derivatives, afforded readily manipulable functionalized cyclopentenones, by a simple one-pot procedure, consisting first of the acid-catalyzed **furan** rlng opening and, then, the intramolecular cyclization of the y-diketone intermediates, directly to cyc1opentenones in the **same** medium.

The cyclopentenone molecule and its 4-hydroxy derivatives are present in several biologically active natural products as major structural features. These include prostaglandins <u>1</u>, a family of mammalian hormones  $\frac{1}{2}$ , cis-jasmone <u>2</u>, a perfume  $^2$ , and rethrolones  $\underline{3}$ , the ester components of the insecticidal pyrethrins<sup>3</sup>. These have glven rise to considerable **rnterest** and efforts to develope efficient routes to these molecules $^{\mathtt{4}}$ 



From an examination of the literature, it was possible to see that a great many syntheses of these products were available, generally based on the same approach: the prior preparation of 1,4-diketones, which are subsequently cyclized in base to cyclopentenones<sup>4</sup>. Other reported routes, for example based on modification of dioxocyclopentanes or involving **some** form of molecular rearrangement, showed some limitations (complex multi-step syntheses, or employment of expensive intermediates, or sophisticated experimental conditions), which have restricted their uti- $1$ ity<sup>4</sup>.

Our group began to work in this field about 1975; the purpose was to perform new syntheses of these compounds with the following features: inexpensive starting materials, easy to find or to prepare, simple, mild, effective experimental conditions and high yields.

We thought to utilize furans because they are easily available; then,they could he considered "depot compounds" of 1,4-diketones. In fact, in past years, furans received attentions only as precursor of the  $\gamma$ -diketones<sup>4</sup>. The well-known Büchi synthesis of 2 follows this methodology: the key intermediate  $\frac{1}{2}$  was obtained by acid-catalyzed ring opening of the dialkylfuran  $\frac{4}{3}$  (scheme 1)<sup>5</sup>

Scheme 1



Our approaches are based on the following routes:

1) Molecular rearrangement of 2-furylcarbinols

2) Cyclization of trans-ene-dicarbonyl compounds

3) One-pot cyclieation of **2-(5-methyl-2-fury1)-1,3-dicaahonyl** compounds.

1) Molecular rearrangement of 2-furylcarbinols

2-furylcarbinols 6, easily obtained by usual procedure of a Grignard reaction, could be turned into the corresponding cyclopentenones  $\overline{Z}$ , through a molecular rearrangement catalyzed by acid or zinc chloride<sup>6,7</sup>. The reaction was completely stereospecific and the mechanism was explained in terms of a thermal electrocyclic reaction of a 4  $\pi$ -electrons system; the key-step in the conversion was the formation of the carbonium ion  $\underline{8}$ , which first led to the pentadienyl cation  $\underline{9}$ , and then to the final products 7



The above result pointed out the interesting capability of 2-furylcarbinols to be precursors of the cation  $9$ , that undergoes conrotatory electrocyclic ring closure to <u>7</u> (Nazarov reaction)<sup>9</sup>; this type of cyclopentenone synthesis was often limited by the available methods to construct the precursors of the cation  $9^{10,11}$ . This reaction was completely original and showed a large applicability, also in  $\text{teroid}^{\text{12}}$  and prostaglandin field  $\text{^{13}}$ .

It made easily accessible a large variety of cyclopentenone derivatives, most of those unknown.



In order to investigate the role of the key intermediate  $\underline{8}$  in the rearrangement, several furan derivatives  $\underline{14}$  (a, b, c) were prepared  $^{14}$ .



As to be expected, 14a and 14b were stable, also under drastic experimental conditions. On the contrary, 14c was a very reactive intermediate; the strong electron donor effect of the methoxy group favored the formation of the cation 15, increasing both reaction rate and yield<sup>14</sup>. 4-Ylidene butenolides 16 were obtained, but their formation was in agreement with the presence of a common intermediate, the reactivity of which was governed by the methoxyl group on furan ring<sup>14</sup>. The compounds  $\frac{7}{2}$ , isomerized into  $3$ -oxo-5-hydroxy-cyclopentene derivatives  $\frac{17}{2}$ , are interesting intermediates for prostaglandin and rethrolone syntheses<sup>15</sup>.



This isomerization **was** achieved in convenient manner, by an intramolecular migration of the alcoholic function<sup>16,17</sup>. The conversion is shown to occur through an alumina-catalyzed process of intramolecular hydration. The experimental results, reported in the scheme 2, were in agreement with a mechanism via the enolate ion  $21$ , which rearranged into the more stable one 22  $(scheme 2)<sup>17</sup>$ 





2) Cyclization of trans-ene dicarbonyl compounds

During the studies on the reactivity of pyridinium chlorochromate **(FCC),** the synthesis of trans-ene dicarbonyl compounds  $24$  was carried out<sup>18</sup>, through an oxidative ring fission of furan derivatives 23 with PCC.



These compounds were rarely considered useful for cyclopentenone synthesis<sup>19</sup>; only the cis-isomers were utilized to build up cyclopentenones by an intramolecu- $20,21$ . On the contrary their great versatility in this field

has been demonstrated (Scheme 3).

Scheme 3



In fact, base-catalyzed cyclization afforded only 5-methoxycyclopentenone 25 (thermodynamically controlled product)<sup>18</sup>; the same reaction with cyclic secondary amines gave 4-amino- and 5-aminocyclopentenones  $26$  and  $27$ , never prepared before (kinetically controlled products)<sup>22</sup>. One-pot procedure allowed to convert directly  $24$  into 4-hydroxy derivatives, first by the trans-cis photoisomerization of 24 and then by the condensation in base to cyclopentenones  $28^{23}$ . At last, enedicarbonyl compounds, treated with sodium iodide and hydrochloric acid in acetone, were rapidly and quantitatively reduced to their saturated analogues  $29^{24}$ . This reaction **was** also available in the steroid field and was completely stereospecific. The known available methods for this reduction either require expensive reagents or afford only low yields<sup>24</sup>.

This mild, new reduction failed with butenedioic acids and esters, or with  $\alpha$ ,  $\beta$ unsaturated monocarbonyl compounds. 'This different behaviour was fully in agreement with the proposed mechanism<sup>24</sup> (Scheme 4).



**Scheme 4** 



**This mechanism explalns why the procedure fails in the above-mentioned cases: the elimination of iodide ion requires the presence of a strongly electron-withdrawing**  substituent at the C-atom  $\alpha$  to the C-atom bearing the iodine to ensure the neces**sary delocalization of the resultant. negative charge.** 

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**3)** One-pot cyclization **of 2-(5-methy1-2-furyl)-1,3-dicarbonyl** compounds. Recently, the synthesis of a new class of furans, **2-(5-methyl-2-furyl)-1,3-dicar**bony1 compounds **35,** has been reported by an intermolecular condensation of 2-methyl-2,5-dihydro-2,5-dimethoxyfuran 33, with an appropriate active methylene compound  $34^{25}$ .



Compounds **35** can he considered precursors **of** 1,4-diketones of type **36** (with an adjacent active group),capable of undergoing cyclization in the same acidic medium in which the opening of furan ring occurred.



- **35,** treated in acidic medium, underwent opening of the furan ring and subsequent cyclization to give cyclopentenones (scheme  $5)^{26}$ . The substituent pattern of the final products depended upon whether the ring-opened intermediate 36 cyclized directly or underwent first an acidic-catalyzed  $\beta$ -diketone fission. This procedure provided a flexible **sequence** for the preparation of various synthetically interesting cyclopentenones, most of those never prepared before<sup>26</sup>.





## Conclusion

It can be seen from the above discussion that furans are highly versatile building blocks for organo-chemical synthesis of cyclopentenones; further studies on the reactivity of these intermediates should find out new their original applications, particularly **in the fieldofnaturallyoccurring** substances. ACKNOWLEDGEMENT

I express my deep gratitude to my colleagues.A Scettri and M. D'Auria, and other co-workers cited in the References, whose efforts made the above advances in cyclopentenone chemistry possible

## REFERENCES

- 1) S. Bergström, Science, 1967, 157, 382.
- 2) E.L. Saul, Amer. Perfume Essent. Oil Record, 1943, 45, 27.
- 3) L. Crombie and M. Elliot, Fortschr<u>. Chem. Org. Naturstoffe</u>, 1961, 19, 120.
- 4) R.A. Ellison, Synthesis, 1973, 397.
- 5) G. Büchi and H. Wüest, J. Org. Chem., 1966, 31, 977.
- 6) G. Piancatelli, A. Scettri, and S. Barbadoro, Tetrahedron Letters, 1976, 3555.
- 7) G. Piancatelli, A. Scettri, **G.** David,and M. DIAuria, Tetrahedron, 1978, 32, 2775.
- 8) R.B. Woodward and **R.** Hoffman, "The Conservation of Orbital Symmetry', Verlag Chemie: Weinheim/Bergstr., Germany, 1970, pp. 45-58.
- 9) I.N. Nazarov and I.I. Zaretskaya, Zh. Obshch. Khim., 1957, 27, 693.
- 10) S. Hacini, R. Pardo,and M. Santelli, Tetrahedron Letters, 1979, 4553.
- 11) J.P. Marino and R.J. Linderman, J. Org. Chem., 1981, 46, 3696.
- 12) **G.** Piancatelli and **A.** Scettri, Tetrahedron Letters, 1977, 1131.
- 13) G. Piancatclli and A. Scettri, Tetrahedron, 1977, *3\_3,* 69.
- 14) M. D'Auria, G. Piancatelli, and A. Scettri, Tetrahedron, 1980, 36, 3071.
- 15) G. Stork, C. Kowalski, and G. Garcia, J. Am. Chem. Soc., 1975, 97, 3258.
- 16) **G.** Piancatelli and A. Scettri, Synthesis, 1977, 116.
- 17) A. Scettri, G. Piancatelli, M. D'Auria, and G. David, Tetrahedron, 1979, 35, 135.
- 18) G. Piancatelli, A. Scettri, and M. D'Auria, Tetrahedron, 1980, 36, 661.
- 19) A.J. Birch, K.S. Keogh, and V.R. Memdapur, Aust. **J.** Chem., 1973, **25,** 2671.
- 20) M.B. Floyd, J. Org. Chem., 1978, 43, 1641.
- 21) T. Shono, Y. Matsumura, H. Hamaguchi, and K. Nakamura, Chem. Lett., 1976, 1249.
- 22) R. D'Ascoli, M. D'Auria, A. De Mico, G. Piancatelli, and A. Scettri, J. Org. Chem., 1980, 4\_5, 4500.
- 23) G. Piancatelli, A. Scettri, and M. D'Auria, Ital. Patent, nº 47769, A/81, 12.2.81.
- 24) M. D'Auria, G. Piancatelli, and A. Scettri, Synthesis, 1980, 245.
- 25) R. D'Ascoli, M. D'Auria, G. Piancatelli, and A. Scettri, Tetrahedron, 1979, 35, 2905.
- 26) R. D'Ascoli, M. D'Auria, C. Iavarone, G. Piancatelli, and A. Scettri, J. Org. Chem., 1980, 4J, 4502.

Received. 26th April, 1982