PHOTOCYCLIZATION OF SUCCINIMIDE DERIVATIVES WITH AMIDO GROUP IN SIDE CHAINS. SYNTHESIS OF TRICYCLIC PYRROLO[1,2-a]PYRAZINE RING SYSTEMS. 1

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Abstract ----- Succinimide derivatives (2) with amido group in their side chains were irradiated in acetonitrile to give biand tricyclic pyrrolo[1,2-a]pyrazines (3).

The photoreactions of cyclic imides have been extensively studied giving rise to a variety of new heterocycles. As to synthesizing various multicyclic fused heterocycles with two nitrogen atoms, we have already reported the photolysis of N-substituted phthalimides such as $N-(\omega-\text{methylanilino}^3)$ and $\omega-\text{amino}^4$ alkylphthalimides. Furthermore, efficiencies of these reactions were highly improved by incorporating an acyl group on the nitrogen atom in their N-alkyl side chains (1). As an extension of the application of these reactions to the alicyclic imide system, we now wish to report an easy synthesis of bi- and tricyclic aliphatic systems containing two nitrogen atoms.

Substrates $\underline{2}$ and $\underline{4}$ were prepared from succinimidoacetyl chloride and the corresponding amines, and succinimide and the corresponding ω -amidoalkyl acetate, respectively. Their photolysis was performed with a 60 or 120 W low-pressure mercury lamp in acetonitrile (10-24 mM) in an atmosphere of nitrogen and the products were purified by silica gel column chromatography. The results are summarized in Table 1.

The photolysis of $\underline{2}$ which lacks γ - and δ -hydrogens relative to the imide carbonyl gave bi- and tricyclic pyrazine derivatives $\underline{3}$ in moderate yields. In all cases a new carbon-carbon bond was formed between the imide carbonyl and the carbon

adjacent to nitrogen in the side chain by initial ε -hydrogen abstraction. These structures were assigned on the basis of the elemental analysis and spectral data. In the IR spectra of <u>3a-e</u> the bands in the regions 3200-3500 and 1625-1700 cm⁻¹ indicated the presence of the cyclol and lactam moieties, respectively. In the ¹H-NMR spectra of <u>3b-d</u>, the methylene signal on the pyrazine ring showed two doublets at 3.46-3.72 and 3.97-4.95 ppm, which exhibit the large coupling constant (J=18-19 Hz) due to the geminal methylene protons. In the ¹³C-NMR spectra of <u>3b-d</u>, doublets in the off-resonance decoupled spectra appeared at 62.8-65.8 ppm, suggesting the presence of a tertiary carbon due to a new carbon-carbon bond formation. In the case of <u>2b-d</u> two diastereoisomers (<u>3-i</u> and <u>3-ii</u>) were isolated, though their stereochemistry is still unknown.

Upon the irradiation of $\underline{4a}$ which also lacks γ -hydrogen, dipyrrolo[1,2-c:2',1'-e]-imidazoline $\underline{5}$ was obtained in a moderate yield, a product of δ -hydrogen abstraction. In the IR spectrum of $\underline{5}$ the bands at 3280, 1690, and 1660 cm⁻¹ indicated the presence of the cyclol and lactam moieties, respectively. The photolysis of $\underline{4b}$ -e having γ -hydrogens relative to the imide carbonyl gave mainly keto lactam δ by a ring expansion reaction.

Interestingly, such γ -hydrogen abstraction was not observed with the phthalimide derivatives (1). ^{1b} In contrast with the photoreactions of the aromatic imides (phthalimides) 1 which are believed to proceed mainly by an electron transfer mechanism, ^{1b-4} it is worth noting that the reactions of their aliphatic counterparts (succinimides) 4 are initiated predominantly by typical γ -hydrogen abstraction, being parallel to the behavior of the common carbonyl. ⁶ Although further work is required for the determination of stereochemistry of the photoproducts, this method provides a simple route to the tricyclic nitrogen heterocycles, which will serve as a convenient source of synthetic intermediates for such alicyclic amine systems.

Table I Thomproduces from 2 and 4.	Table 1	Photoproducts	from	2	and	4.
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Substrate	e mp °C	Time(h)	Product	mp °C	Yield(%)
<u>2a</u>	95.5 - 98	1.5	<u>3a</u>	110.5-112	60
<u>2b</u>	141 - 142	1.5	<u>3b-i</u>	135 - 138	11
			<u>3b-ii</u>	81 (dec.)	43
2c*	120 - 122.5	5	<u>3c-i</u>	143 (dec.)	13
			<u>3c-ii</u>	102.5-104.5	9
<u>2d</u> *	88.5 - 91	2	<u>3d-i</u>	160 - 163	29
			<u>3d-ii</u>	142 - 144	21
2e*	185.5-186.5	4	<u>3e</u>	200 (dec.)	26
<u>4a</u> **	138.5-140	1.5	<u>5</u>	131 - 133	51
<u>4b</u>	100 - 101.5	1.5	<u>6b</u>	154 - 156	41
4c	87 - 87.5	1.5	<u>6c</u>	115.5-117.5	31
		:	succinimide		16
<u>4d</u> *	oil	3.5	<u>6đ</u>	115.5-117	23
			succinimide		18
<u>4e</u>	^{bp} 0.6 ¹⁶⁵	2	<u>6e</u>	78 - 81	25

- * A 120 W low-pressure mercury lamp was used.
- ** in methanol

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