A CONVENIENT SYNTHESIS OF 6-METHYLELLIPTICINE AND 6-METHYLOLIVACINE

Martin J. Wanner, Gerrit-Jan Koomen and Upendra K. Pandit*
Laboratory of Organic Chemistry, University of Amsterdam,
Nieuwe Achtergracht 129, 1018 WS Amsterdam, The Netherlands.

<u>Abstract</u> — Readily accessible 11-ketopyrido[4,3-b]carbazole derivatives <u>3a,b</u> have been used as central intermediates for the synthesis of 6-methylellipticine and 6-methylolivacine.

Considerable interest centres around the pyridocarbazole alkaloids ellipticine and olivacine, in view of their reported antitumour activity $^{1a-d}$. Although a number of syntheses for these alkaloids have been reported to date $^{2a-e}$, a convenient approach to the parent compounds and their derivatives, starting from readily available materials, has been lacking. In this communication we present the synthesis of both 6-methylellipticine (1) and 6-methylolivacine (2) via a general synthesis of the pyridocarbazole skeleton which has been reported by us earlier 3 .

As a part of a broader study of the application of the reaction of ester α -anions with N-alkylated nicotinic acid derivatives to the synthesis of polynuclear heterocycles, we have recently reported the syntheses of d,1-sesbanine⁴ and the pyridocarbazole derivative $\underline{3a}^3$. The conversion of $\underline{3a}$ to the corresponding ellipticine derivative $\underline{1}$ and the preparation of the analogous precursor $(\underline{3b})$, and its transformation to the related olivacine system $(\underline{2})$, constituted worthwhile synthetic targets.

The conversion of $\underline{3a}$ to $\underline{1}$ could be achieved via two routes (Scheme A). Reaction of $\underline{3a}$ with CH₃MgI (excess, THF, reflux) led directly to the formation of 6-methylellipticine (40%) in one practical step. The reaction presumably proceeds via intermediate $\underline{4}$, which undergoes a fragmentation, involving loss of ${}^{\Theta}$ OMgI, mediated by attack of the Grignard reagent (excess) on the ester carbonyl. An alternate mechanism could involve an analogous fragmentation of a lactone, formed by intramolecular reaction between the incipient alkoxide anion - generated by initial Grignard attack - and the ester group.

- (a) $Ph_3P=CH_2$, THF, 20° ; (b) KOH, EtOH/ H_2O , Δ ; (c) MeMgI, (excess), THF, Δ ;
- (d) REDAL, THF, r.t.

Scheme A

(e) Sulfolane, $160-170^{\circ}$, 20 min ; (f) PhCH₂Br , 110° , 30 min ; (g) Et₃N, r.t. 1h ; (h) N-Benzylacridinium bromide ; CH₃CN ; (i) H₂ / Pd

Scheme

The second route involved the treatment of 3a with $Ph_3P=CH_2$ (2 eq.), whereupon the exo-methylene derivative 5 was obtained in good yield (65%). Hydrolysis of 5 (KOH/EtOH/H₂O, reflux) cleanly yielded 1 as a crystalline compound, m.p. 211-212° (60%). Relevant spectral data on $\underline{5}^5$ and $\underline{1}^6$ attested to their structures. It should be emphasized that both routes are capable of variation and that 3a can serve as a central intermediate for the synthesis of diverse ellipticine analogues. The 6-methylolivacine precursor (3b) was prepared via the sequence of reactions described in Scheme B. This sequence starts with the known indolylpropionic ester 6 (Scheme B) and follows the steps $6 \rightarrow 7 \rightarrow 3 \rightarrow 3b$, in a manner analogous to that described previously for the synthesis of 3a. The only difference is represented by the use of 2-methylnicotinyl chloride hydrochloride, in place of the nicotinyl chloride hydrochloride salt. The structures of intermediates 7 and 8, and compound 3b (m.p. 165-167°), were assigned on the basis of their spectral data 7 . The keto ester 3b was converted to 2 (57%), in one practical step, by reaction with excess of RedAl. It is assumed that a hydroxy compound (9) is initially formed, which is further reduced and decomposed (perhaps via a lactone) to $\underline{2}$ under the reaction conditions. The product 2 is a crystalline compound, m.p. 228-229°, which exhibited $^{
m l}$ H NMR spectral data $^{
m 8}$ consistent with the assigned structure. The scope of the conversion of intermediates of type $\frac{3}{2}$ to ellipticine and olivacine derivatives is being actively investigated.

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- 5. <u>5</u>: Unstable oil (65%); IR (CHCl₃): 1725, 1615, 1590 cm⁻³. ¹H NMR (CDCl₃): δ 1.08 (t, J=7, CH₃); 1.90 (s, CH₃); 3.67 (s, N-CH₃); 4.12 (m, CH₂); 5.98 (s,=CH); 6.02 (s,=CH); 7.3-7.5 (m, C_{4,7,8} and 9-H); 8.08 (d, J=8, C₁₀-H); 8.53 (d, J=6, C₃-H); 9.32 (s, C,-H).
- 6. <u>1</u>: M.p.: 211-212° (60%); IR (CHCl₃): 1595, 1470 cm⁻¹. ¹H NMR (CDCl₃): δ 3.00 (s, 5-CH₃); 3.14 (s, 11-CH₃); 4.08 (s, N-CH₃); 7.30 (t, J=8, C₈-H/C₉-H); 7.38 (d, J=8, C₇-H); 7.58 (t, J=8, C₈-H/C₉-H); 7.86 (d, J=7, C₄-H); 8.32 (d, J=8, C₁₀-H); 8.46 (d, J=7, C₃-H); 9.64 (s, C₁-H). MS (M⁺) 260.1307; Calcd. for C₁₈N₁₆N₂: 260.1301.
- 7. (a) $\underline{7}$: Oil (30%); IR (CHCl₃): 1730, 1620, 1580 cm⁻¹. ¹H NMR (CDCl₃): δ 1.25 (t, J=7, CH₃); 1.66 (d, J=7, CH₃); 2.53 (s, CH₃); 3.76 (s, N-CH₃); 4.25 (q, J=7, CH₂); 5.05 (q, J=7, CH); 6.5-7.5 (m, arylprotons + pyridine C₅-H); 7.70 (dxd, J=7, J=1.5, pyridine C₄-H); 8.66 (dxd, J=5, J=1.5, pyridine C₆-H).
 - (b) 8: M.p.: 174-177° (60%); IR (CHCl₃): 1740, 1655, 1615 cm⁻¹. ¹H NMR (CDCl₃): δ 1.16 (t, J=7, CH₃); 2.10 (s, CH₃); 3.50 (s, CH₃); 3.86 (s, N-CH₃); 4.28 (q, J=7, CH₂); 6.32 (s, Ø-CH₂); 7.3-7.5 (m, 8H-Ar); 8.22 (d, J=7, C₄-H); 8.41 (m, C₁₀-H); 9.98 (d, J=7, C₃-H).
 - (c) <u>3b</u>: M.p.: 165-176° (81%). IR (CHCl₃): 1730, 1640, 1570 cm⁻¹. ¹H NMR (CDCl₃): δ 1.06 (t, J=7, CH₃); 1.95 (s, CH₃); 3.22 (s, CH₃); 3.78 (s, N-CH₃); 4.17 (q, J=7, CH₂); 7.35 (d, J=6, C₄-H); 7.40 (m, 3H-Ar); 8.45 (m, C₁₀-H); 8.63 (d, J=6, C₃-H).
- 8. $\underline{2}$: M.p.: 228-229° (57%). IR (CHCl₃): 1625 (sh), 1600 cm⁻¹. ¹H NMR (CDCl₃):

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