SELECTIVE METHYLATION OF INTRAMOLECULAR N AND S GROUPS

--- POSSIBLE FORMATION OF \(\sigma\)-AMMONIOSULFURANES ---

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In the course of investigation on the selective methylation of intramolecular S and N groups, attention was focused on the interaction of the amino and the sulfonio groups which would yield

$$\sigma$$
-ammoniosulfuranes (2) in its ultimate (strong) stage as shown in eq. 1.

$$Me_2 N: \int_1^S Me_2 \longrightarrow Me_2 \int_2^S Me_2$$
(1)

Several compounds were prepared as its model, but no meaningful interaction was observed based on their chemical shifts of ¹H-NMR.

Me N:

Me N:

These negative results were ascribed to frontal steric hindrance among four methyl groups of 3 - 6, therefore S-substituted N-methyl-6, 7-dihydro-5H-dibenzo (b, g) (1, 5) thiazocinium salts ($\frac{7}{2}$) were prepared which would avoid such hindrance by taking trigonal bipyramidal structure ($\frac{7}{2}$).

The sulfoxide (8) was converted to 7d (X=C1) with thionyl chloride and 7d was derived to 7c (X=OMe) and 7b (X=OEt) with sodium alkoxides and also to 7a (X=Me) with lithium dimethylcuprate.

The structure of 7d was concluded to be 7d' based on 1H -NMR spectra. All 7 showed similar 1H -NMR spectra assignable to TBP and plots of chemical shifts of the N-methyl group against Hammett substituent constants of X ($o_{\widetilde{I}}$ and $o_{\widetilde{I}}$) gave very good straight lines, supporting dominance of TBP (7) for 7:

$$\delta = 1.30 \ \sigma_{m} + 2.68 \ (r = 0.989, \ n = 5) \ ; \ \delta = 1.15 \ \sigma_{T} + 2.60 \ (r = 1.000, \ n = 4)$$

As far as we know, 7a is the first example of G-sulfurane where the methyl group is attached at the apical position of TBP.