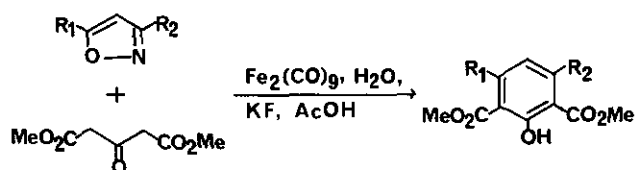


A NEW AROMATIC ANNELEMENTATION REACTION UTILIZING REACTION OF
ISOXAZOLE WITH DIIRONNONACARBONYL

Naoki Takeuchi, Kousuke Okazaki, Junji Ohki, and Seisho Tobinaga
Showa College of Pharmaceutical Sciences, Setagaya-ku,
Tokyo 154, Japan

Isioxazoles can be considered as an equivalent synthon of β -enaminones in protected form and the reductive cleavage of isioxazoles into β -enaminones by catalytic hydrogenation is widely applicable reaction for organic synthesis. Interest has centred, recently, reactions of isioxazoles with transition-metal carbonyls such as $\text{Mo}(\text{CO})_6$, $\text{Fe}(\text{CO})_5$ with $h\nu$, and $\text{Fe}_2(\text{CO})_9$ to undergo β -enaminones by reductive cleavage through aziridine intermediates. This instance is of considerable interest, for it generation of enaminones in situ in connection with a aromatic annelation reaction of the enaminones with 3-oxoglutarate.¹⁾ We report a new simple method for aromatic annelation reaction utilizing reaction of isioxazoles with diironnonacarbonyl and water followed by reaction with 3-oxoglutarate in presence of KF-AcOH .

A typical annelation reaction was carried out as follows. Diironnonacarbonyl 7.5 g (20.6 mmol) was added to a solution of 2,5-dimethylisioxazole 1 g (10.3 mmol) in benzene 50 ml with water 180 mg (10 mmol) under a nitrogen atmosphere and stirred at 45°C for 30 min. And then, a solution of dimethyl 3-oxoglutarate 3.59 g (10.6 mmol), KF 1.19 g (20.6 mmol), and AcOH 2.47 g (41.2 mmol) in dry benzene 50 ml was added and refluxed for 12 h removing water with water separator. Resulted mixture gave dimethyl 2-hydroxy-4,6-dimethyl-1,3-benzene dicarboxylate 1, mp 105-107°C, 1.25 g (50.9% yield). Some experimental results are shown as follows.



1	$\text{R}_1 = \text{R}_2 = \text{Me}$	50.9 %
2	$\text{R}_1 = \text{Me}, \text{R}_2 = \text{H}$	42.6 %
3	$\text{R}_1 = \text{Ph}, \text{R}_2 = \text{H}$	21.5 %
4	$\text{R}_1 = \text{CH}_2\text{CH}=\text{CHPh}, \text{R}_2 = \text{H}$	31.1 %
5	$\text{R}_1 = \text{CH}_2\text{CH}=\text{CH} \begin{array}{c} \text{OMe} \\ \\ \text{C}_6\text{H}_3 \\ \\ \text{OBz} \end{array}, \text{R}_2 = \text{H}$	28.0 %

(1) N. Takeuchi, K. Ochi, M. Murase, and S. Tobinaga, *J. Chem. Soc., Chem. Commun.*, 1980, 593.