REGIOCONTROLLED CYCLISATION TO THE ISOMERIC DIPYRIDAZINO[4,5- \underline{b} : 4',5'- \underline{e}][1,4]THIAZINES — COMPETENCE AND PERFORMANCE OF BENZYL-AMINO AND THIOETHER FUNCTIONS ON THE 3(2<u>H</u>)-PYRIDAZINONES AS REGIOCONTROLLING ELEMENTS

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<u>Abstract</u> — The 5,5'-thioethers (5a-c) perform a regiocontrolled cyclisation affording either the $10\underline{H}$ -dipyridazino[1,4]thiazine-1,9diones (1'a-c) (by Ullmann mode) or the 10-benzyl derivatives of 1,6-dione isomers (2a-c) (via Smiles rearrangement), only depending on the acidity of media employed. The 4,5'-thioethers (5a-c) exhibit also the regiospecific control to give exclusively the $10\underline{H}$ dipyridazino[1,4]thiazine-1,6-diones (2'a-c) in acidic conditions or the 10-benzyl-4,6-dione isomers (3a-c) in basic conditions, respectively. Some principal features of the regiocontrolled cyclisation are discussed.

In the preceding paper¹ we have reported the synthesis of three types of dipyridazino[4,5- \underline{b} :4',5'- \underline{e}][1,4]thiazine derivatives ($\underline{l}a$, $\underline{l}a$ and $\underline{l}a$) by the condensation between the amino-mercaptopyridazinone ($\underline{l}'a$ or $\underline{l}'a$) and the dichloro-pyridazinone ($\underline{l}a$) followed by benzylation. The following paper² describing the conversion of 2,7-disubstituted 10<u>H</u>-dipyridazino[4,5- \underline{b} :4',5'- \underline{e}][1,4]thiazine derivatives ($\underline{l}'a$ -c) into the corresponding 2,6-disubstituted 9<u>H</u>-dipyridazino[4,5- \underline{b} :4',5'- \underline{d}]pyrrole-1,5(2<u>H</u>,6<u>H</u>)-diones ($\underline{l}'a$ -c) by base-induced extrusion of sulphur, prompted us to exploit a regiocontrolled synthesis of all of the individual members belonging to the dipyridazino[4,5- \underline{b} :4',5'- \underline{e}][1,4]thiazines ($\underline{l}a$ -c, $\underline{l}a$ -c and $\underline{l}a$ -c).









 χ_{a-c} (R'=PhCH₂)
 χ_{a-c} χ'_{a-c}
 χ'_{a} (R'=H)
 χ'_{a-c}
 χ'_{a} (R'=H)
 χ'_{a-c}

The present work deals with the exclusive synthesis of the individual dipyridazino-[4,5-<u>b</u>:4',5'-<u>e</u>][1,4]thiazine derivatives (χ 'a-c or χ a-c; χ 'a-c or χ a-c) by regiocontrolled cyclisations of the thioethers, 5-(2-substituted 4-benzylamino-2,3dihydro-3-oxo-pyridazinyl)-5'-(2'-substituted 4'-chloro-2',3'-dihydro-3'-oxopyridazinyl)sulphides (χ a-c) and the isomeric 4,5'-sulphides (χ a-c), depending on the acidity of medium employed. The thioethers, 4-(2-substituted 5-benzylamino-2,3dihydro-3-oxo-pyridazinyl)-5'-(2'-substituted 4'-chloro-2',3'-dihydro-3'-oxopyridazinyl)sulphides, χ a-c (a: mp 235-236°C, 63%; b: mp 244-246°C, 69%; c: mp 239-241°C, 71%) were easily prepared by the interaction of the 5-benzylamino-4mercaptopyridazinones (χ a-c) and the corresponding 4,5-dichloropyridazinones (χ a-c) similarly as the formation of the thioethers (χ a-c).²

Much attention has been long paid to the regiocontrolled synthesis of polyazaphenothiazines, including pyridazine or pyridazinone rings, $^{3-6}$ however, any paper concerned with examination of benzylamino or thioether functions as regiocontrolling elements for the cyclisation, has not yet been found. Heating the thioethers (5a-c) with fortified acetic acid (with concd. HCl) under reflux for 4 h caused an Ullmann mode of cyclisation, attended with debenzylation, to afford 2,8-disubstituted 10<u>H</u>-dipyridazino[4,5-<u>b</u>:4',5'-<u>e</u>][1,4]thiazine-1,9(2<u>H</u>,8<u>H</u>)-diones (χ 'a-c) in high yield (a: 92%, b: 90%, c: 93%). Contrary to this, the cyclisation of the thioethers (5a-c) in a sodium hydroxide solution, had proceeded <u>via</u> Smiles rearrangement, to give the corresponding 2,7-disubstituted 10-benzyl-10<u>H</u>-dipyridazino-



[4,5- \underline{b} :4',5'- \underline{e}][1,4]thiazine-1,6(2<u>H</u>,7<u>H</u>)-diones (2a-c: a, 79%; b, 63%; c, 81%).² Similarly, another thioethers (8a-c) exhibited the alternative competence for the regiocontrolled cyclisation, performing the formation of 2,7-disubstituted 10<u>H</u>pyridazino[4,5- \underline{b} :4',5'- \underline{e}][1,4]thiazine-1,6(2<u>H</u>,7<u>H</u>)-diones (2'a-c: a, 63%; b, 85%; c, 79%) in acidic conditions, and of 3,7-disubstituted 10-benzyl-10<u>H</u>-dipyridazino[4,5- \underline{b} :4',5'- \underline{e}][1,4]thiazine-4,6(3<u>H</u>,7<u>H</u>)-diones (3a-c: a, 76%; b, 64%; c, 81%) in bacic conditions. Assigned structures for the dipyridazino[1,4]thiazines, 1'a-c, 2'a-c, 2a-c and 3a-c, were supported by their spectral and microanalytical data, and also confirmed by the alkylation of 1'a-c and 2'a-c to the corresponding 10-benzyl derivatives 1a-c and 2a-c, respectively.

The principal features of the regiocontrolled cyclisation are assumed as follows: (1) Any of the thioethers (ξ_a -c, ξ_a -c) bears a benzylamino function as an internal nucleophile which might attack exclusively either C⁴' atom (in an acidic medium) or c^5 ' atom (in a basic medium) on the B ring.

(2) The thioether linkage, C^5 -S- C^5' or C^4 -S- C^5' , joining together A and B rings, is also essential to the regiocontrolled cyclisation and exerts first a competence of a nucleofuge and then that of a nucleophile (Smiles rearrangement).

(3) On the transition state (A), leading to the product by Ullmann mode of cyclisation, elimination of the benzyl function from the quaternary nitrogen might be more favourable than deprotonation, due to the release from the steric crowding of the neighbourhood.

(4) On the transition state (\underline{R}) , leading to the product <u>via</u> Smiles rearrangement, the deprotonation might be more feasible than the elimination of the bulky group. (5) The oxo and chloro functions, pertinently built in the B ring, might cooperate essentially the competence and performance of the benzylamino and thioether functions as the regiocontrolling elements.



Physical Properties*)

- l'a) mp>300°C. IR v_{max}^{KBr} cm⁻¹: 1640(CO), 3310(NH). ¹H-NMR(CF₃CO₂H) δ: 3.35(6H, s, N-CH₃×2), 7.00(2H, s, 4-H and 6-H).
- <code>l'b) mp>300°C. IR v_{max}^{KBr} cm⁻¹: 1640(CO), 3340(NH). ¹H-NMR(CF₃CO₂H) &: 7.14(10H, s, C₆H₅×2), 7.28(2H, s, 4-H and 6-H).</code>
- l_{c} mp>300°C. IR v_{max}^{KBr} cm⁻¹: 1645(CO), 3360(NH). ¹H-NMR(CF₃CO₂H) δ: 5.07(4H, s, N-CH₂×2), 7.05(10H, s, C₆H₅×2), 7.23(2H, s, 4-H and 6-H).
- 1b) mp 224-225°C. IR $v_{\text{max}}^{\text{KBr}}$ cm⁻¹: 1640(CO). UV $\lambda_{\text{max}}^{\text{EtOH}}$ nm(log ϵ): 228(4.13), 267 (4.08), 312(4.03). ¹H-NMR(CDCl₃) δ : 5.62(2H, s, N-CH₂), 7.48(15H, s, C₆H₅×3), 7.58(2H, s, 4-H and 6-H).
- 3a) mp 275-276°C. IR $v_{\text{max}}^{\text{KBr}}$ cm⁻¹: 1640(CO). UV $\lambda_{\text{max}}^{\text{EtOH}}$ nm(log ϵ): 234(4.42), 276 (4.47), 316(4.16). ¹H-NMR(CDCl₃) δ : 3.62(6H, s, N-CH₃×2), 4.68(2H, s, N-CH₂), 6.69(2H, s, 1-H and 9-H), 7.30-7.50(5H, m, C₆H₅).
- 3b) mp>300°C. IR v_{max}^{KBr} cm⁻¹: 1650(CO). UV λ_{max}^{EtOH} nm(log ε): 232(4.43), 271(4.49), 312(4.20). ¹H-NMR(DMSO-<u>d_6</u>) δ : 4.83(2H, s, N-CH₂), 7.26-7.60(17H, m, C₆H₅×3, 1-H and 9-H).
- 3c) mp 272-274°C. IR $v_{\text{max}}^{\text{KBr}}$ cm⁻¹: 1640(CO). UV $\lambda_{\text{max}}^{\text{EtOH}}$ nm(log ε): 233(4.48), 273 (4.49), 315(4.21). ¹H-NMR(DMSO-<u>d</u>₆) δ : 5.20(2H, s, N-CH₂), 5.29(4H, s, N-CH₂×2), 7.15-7.51(17H, m, C₆H₅×3, 1-H and 9-H).
- fa) mp 236°C. IRv_{max}^{KBr} cm⁻¹: 1620, 1650(CO), 3290(NH). ¹H-NMR(CDCl₃) δ : 3.61, 3.72 (each 3H, s, N-CH₃), 4,59(2H, d, <u>J</u>=7Hz, NH-CH₂), 7.30(5H, s, C₆H₅), 7.18, 7.68(each 1H, s, 6-H, 6'-H).
- f(b) mp 244-246°C. IR v_{max}^{KBr} cm⁻¹: 1620, 1660(CO), 3310(NH). ¹H-NMR(DMSO-<u>d</u>₆) δ: 4.63 (2H, d, <u>J</u>=7Hz, NH-C<u>H</u>₂), 7.21-7.50(10H, m, C₆H₅×2), 7.27(5H, s, C₆H₅), 7.35, 7.80(each 1H, s, 6-H, 6'-H).
- *) 2'a-c, 2a-c and 2a-c: see ref. 2.
 All compounds gave satisfactory elemental analyses (C, H and N).

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