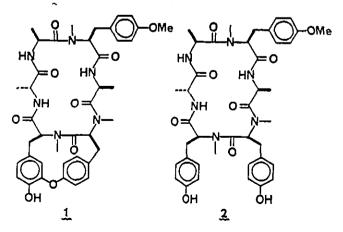
SYNTHESIS OF CYCLO-N-METHYL-L-TYR-N-METHYL-L-TYR-D-ALA-L-ALA-O,N-DIMETHYL-L-TYR-L-ALA, A CYCLIC HEXAPEPTIDE RELATED TO THE ANTITUMOR AGENT DEOXYBOUVARDIN

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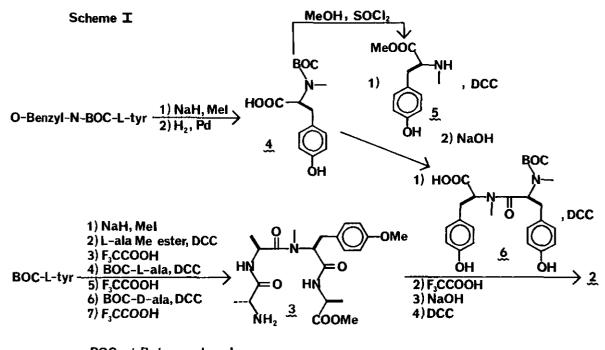
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Abstract—A synthesis of the title cyclic hexapeptide is described. Its lack of antitumor activity shows that the 14-membered ring of deoxybouvardin is needed for activity. Efforts to oxidatively couple its phenolic groups failed to give deoxybouvardin.

Bouvardin and deoxybouvardin (1), minor constituents of <u>Bouvardia ternifolia</u>, are potent antitumor agents.^{1,2} We wish to report that as part of a program of synthesizing these substances and analogs, we have synthesized cyclic hexapeptide 2, and that it lacks antitumor activity though it differs from deoxybouvardin (1) by only two hydrogen atoms.



Scheme I shows the synthetic route to 2. NMR analysis of the peptides containing N-methyl groups was complicated by the observance of approximately equal amounts of <u>cis</u> and <u>trans</u> forms about the N-methyl peptide bonds and about each BOC amide bond; in several such cases, the spectrum was simplified by remeasuring it at a temperature high enough to make rotation about such bonds faster than the NMR measurement.² Tetrapeptide 3 was prepared in 23% yield from BOC-L-tyrosine; the tripeptide obtained in the fifth step was shown by X-ray analysis³ to crystallize in the <u>cis</u> form about the peptide bond bearing the N-methyl group, whereas NMR shows <u>cis</u> and <u>trans</u> forms to be present in equal amounts in solution². Dipeptide 6 was prepared in 74% yield from 0-benzyl-N-BOC-L-tyrosine. Tetrapeptide 3 and dipeptide 6 were combined as shown to give cyclic hexapeptide 2 in 7% yield.



BOC = t-Butoxycarbonyl DCC = N,N'- dicyclohexylcarbodiimide

Cyclic hexapeptide 2, unlike 1, was found to be inactive in a P-388 test involving leukemia in mice. It is not surprising that this difference should occur, as the 14-membered ring in 1 no doubt greatly changes the conformation from those favored in 2.2 Many attempts to oxidatively couple the adjacent tyrosines of 2 to give 1 (presumably mimicking the biogenesis of 1) failed.⁴

EXPERIMENTAL SECTION

General. Melting points were determined on a Kofler hot stage and are uncorrected. NMR spectra were recorded on Varian T-60 and Bruker WM-250 spectrometers. Mass spectra were run on a Varian MAT 311A spectrometer. Elemental analyses were carried out by Huffman Laboratories, Inc., Wheat Ridge, Co.

0.N-Dimethyl-N-BOC-L-tyr. BOC-tyr (30.7 g, 0.10 mol, Vega Biochemicals) in THF (500 mL, distilled from Na) was cooled to 0° C and methyl iodide (30 mL) and NaH (26 g of 50% dispersion in mineral oil) were added. After stirring 1 h at 0°C and 25°C for 16 h, 8 mL ethyl acetate was added followed by water dropwise until gas evolution ceased. Most of the solvent was evaporated, the residue was washed 3 X 200 mL pentane, diluted with 100 mL water, acidified with citric acid, and extracted 3 X 100 mL ethyl acetate. The combined ethyl acetate solutions were washed 2 X 150 mL saturated sodium thiosulfate, once with 100 mL saturated sodium chloride, dried over magnesium sulfate, and evaporated, leaving 29 g (88%) of product as a very viscous liquid; ¹H NMR (CDCl₃, δ) 10.1 (s, COO<u>H</u>), 7.1 (~d, 2H, J=9 Hz), 6.8 (~d, 2H, J=9 Hz), 3.7 (s, OC<u>H_3</u>), 3.1 (m, C<u>H_2</u>), 2.7 (s, NC<u>H_3</u>), 1.3 (s, C(CH_3)_3).

Methyl 0,N-Dimethyl-N-BOC-L-tyr-L-alaninate. 0,N-Dimethyl-N-BOC-L-tyr (20.3 g, 66 mmol) and methyl L-alaninate (6.8 g, 66 mmol, freshly prepared from the hydrochloride by bubbling ammonia through a CHCl₃ solution of the hydrochloride, filtering, and evaporating) in CH₂Cl₂ (200 mL) was protected with a drying tube, cooled to 0°C, and DCC (15 g) was added. After stirring 1 h at 0°C and 16 h at 25°C, urea was filtered off and the solution was washed with 5% HOAc, water, and saturated KHCO₃ solution, dried over MgSO₄, and evaporated, leaving 24.6 g (95%) of product as a viscous oil containing some urea crystals; ¹H NMR (CDCl₃, δ) 7.1 (~d, 2H, J=9 Hz), 6.8 (~d, 2H, J=9 Hz), 6.6 (m, N<u>H</u>), 4.7 (m, Tyr C<u>H</u>), 4.6 (quintet, J=7 Hz, Ala C<u>H</u>), 3.8 (s, OC<u>H₃</u>), 3.8 (s, OC<u>H₃</u>), 3.2 (m, C<u>H₂</u>), 2.8 (s, NC<u>H₃</u>), 1.4 (s, C(C<u>H₃</u>)₃), 1.4 (d, J=7 Hz, Ala CC<u>H₃</u>).

Methyl 0,N-Dimethyl-L-tyr-L-alaninate. Methyl 0,N-dimethyl-N-BOC-L-tyr-L-alaninate (15.9 g, 40 mmol) was allowed to stand for 15 min at 25°C with 160 mL of trifluoroacetic acid and then most of the solvent was removed under vacuum at 40°C. CH_2Cl_2 (20 mL) was added and the amine was extracted with 500 mL and then 100 mL of 0.1 N HC1. The aqueous solutions were washed 2 X 20 mL CH_2Cl_2 , made basic with solid KHCO3, saturated with solid NaCl, and extracted 3 X 100 mL CH_2Cl_2 . Drying over MgSO4 and evaporating left 8.4 g (76%) of amine as an oil; ¹H NMR (CDCl₃, δ) 7.6 (broad d, Ala NH), 7.1 (~d, 2H, J=9 Hz), 6.8 (~d, 2H, J=9 Hz), 4.6 (quintet, J=7 Hz, Ala CH), 3.7 (s, OCH_3), 3.7 (s, OCH_3), 2.3-3.3 (m, CH_2 and Tyr CH), 2.3 (s, NCH₃), 1.5 (s, Tyr NH), 1.4 (d, J=7 Hz, CCH_3).

On standing for several months at 25°C, this substance is converted into the diketopiperazine

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<u>cyclo-0,N-dimethyl-L-tyr-L-ala</u> hydrate, mp 78-80°C; ¹H NMR (CDCl₃, ⁸) 7.0 (~d, 2H, J=9 Hz), 6.8 (~d, 2H, J=9 Hz), 6.5 (broad s, NH), 4.2 (~t, J=4 Hz, Tyr CH), 4.0 (m, Ala CH), 3.8 (s, OCH₃), 3.2 (m, CH₂), 3.1 (s, NCH₃), 2.0 (s, <u>H</u>₂O), 0.7 (s, CCH₃).⁵

Anal. Calcd for C₁₄H₁₈N₂O₃·H₂O: C, 59.99; H, 7.19; N, 9.99. Found: C, 60.22; H, 7.22; N,

Methyl N-BOC-L-ala-0,N-dimethyl-L-tyr-L-alaninate. Methyl 0,N-dimethyl-L-tyr-L-alaninate (6.18 g, 21 mmol) was coupled with excess BOC-L-ala (Vega Biochemicals) as described above except the washings with HOAc solution and water were omitted to minimize diketopiperazine formation. The crude viscous oll containing urea crystals (10.6 g, 109 %) was not further purified; ¹H NMR showed coupling to be complete (NCH3 due to <u>cis</u> and <u>trans</u> peptide bonds at δ 2.9 and δ 3.0; no NCH3 at δ 2.3).

Methyl L-Ala-O,N-dimethyl-L-tyr-L-alaninate. The entire crude product from the reaction above was cleaved with trifluoroacetic acid as described above, yielding 4.6 g (60% over two reactions) of amine, mp 136-137°C; ¹H NMR (CDCl₃, 6) shows <u>cis</u> and <u>trans</u> forms about the N-methyl-bearing peptide bond, present in about equal amounts.² An X-ray study showed the crystals to be of <u>cis</u> peptide.³

Anal. Calcd for C₁₈H₂₇N₃O₅; C, 59.16; H, 7.45; N, 11.50. Found: C, 59.28; H, 7.67; N,

Merhyl N-BOC-D-ala-L-ala-0,N-dimethyl-L-tyr-L-alaninate. Methyl L-ala-0,N-dimethyl-L-tyr-L-alaninate (4.5 g, 15 mmol) was coupled with excess BOC-D-ala (Vega Biochemicals) as described above. The crude foam obtained (6.6 g, 106%) was used directly in the next reaction.

Methyl D-Ala-L-ala-O,N-dimethyl-L-tyr-L-alaninate (3). Cleavage of the BOC grouping from 4.0 g (9 mmol) of the above tetrapeptide with trifluoroacetic acid gave 1.8 g (61% over two reactions) of crystalline 3, mp 138-142°C; ¹H NMR (CDCl₃) shows <u>cis</u> and <u>trans</u> forms about the N-methyl-bearing peptide bond in about equal amounts, & 2.9 and 3.0 (s, NCH₃).

<u>O-Benzyl-N-methyl-N-BOC-L-tyr.</u> O-Benzyl-N-BOC-L-tyr (55.6 g, 0.15 mol, mp 112-113°C, 1it. 108-110°C⁶) in dry THF (700 mL) was cooled to 0°C and methyl iodide (42 mL) and NaH (25.5 g of 50% dispersion in mineral oil) were added. Reaction and workup as above for 0,N-dimethyl-N-BOC-L-tyr gave 57.4 g (99%) of viscous oil; ¹H NMR (CDCl₃, δ) 9.6 (broad s, COOH), 7.3 (broad s, C₆H₅), 7.1 (~d, 2H, J=9 Hz), 6.8 (~d, 2H, J=9 Hz), 5.0 (s, CH₂), 4.6 (m, CH), 3.1 (m, CH₂), 2.7 (s, NCH₃), 1.3 (s, C(CH₃)₃).

N-Methyl-N-BOC-L-tyr (4). O-Benzyl-N-methyl-N-BOC-L-tyr (30 g, 78 mmol) and acetic acid (3 mL) in dioxane (150 mL) were stirred with 5% Pd on charcoal (5 g) at 25°C under a hydrogen atmosphere. After 60 h, the theoretical amount of hydrogen (1.9 L) had been taken up. Filtering

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and evaporation of the solvent left a viscous oil which on trituration with hexane crystallized to give 22.5 g (98%) of product, mp 141-144°C (dec); ¹H NMR (CDCl₃, δ) 8.8 (broad s, COO<u>H</u> and ArO<u>H</u>), 7.0 (~d, 2H, J=9 Hz), 6.7 (~d, 2H, J=9 Hz), 4.1 (m, C<u>H</u>), 3.1 (m, C<u>H</u>₂), 2.7 (s, NC<u>H</u>₃), 1.3 (s, C(CH₃)₃).

Anal. Calcd for C_{15H21}NO₅: C, 61.00; H, 7.17; N, 4.74. Found: C, 60.52; H, 7.08; N, 4.84. Methyl N-Methyl-L-tyrosinate (5). 4 (4.0 g, 14 mmol) in 130 mL of methanol was stirred and cooled to 0°C and 10 mL of SOCl₂ was added dropwise. After refluxing overnight, the volatiles were blown off with nitrogen. The residual oil was washed with ether, taken up in 400 mL of chloroform, and ammonla was bubbled through this solution for 10 min. Filtration and evaporation left 2.4 g (100%) of 5, mp 109-111°C (1it.⁷ 111-112°C).

Methyl N-BOC-N-methyl-L-tyr-N-methyl-L-tyrosinate. 5 (2.1 g, 10 mmol) was coupled with 4 (3.0 g, 10 mmol) as described above, yielding 4.06 g (84%) dipeptide as a foam. A sample was chromatographed on silica gel (1:10 MeOH-CH₂Cl₂) to remove urea, but the product still did not crystallize. The ¹H NMR spectrum (CDCl₃) showed broad absorptions in the expected regions which sharpened on warming to 120°C.²

N-BOC-N-methyl-L-tyr-N-methyl-L-tyr (6). Methyl N-BOC-N-methyl-L-tyr-N-methyl-L-tyrosinate (3.76 g, 7.7 mmol) was stirred under N₂ with 20 mL of 2 N NaOH for 2.5 h, evaporated slightly to remove MeOH, diluted to 150 mL with water, washed 3 X 100 mL CH₂Cl₂, acidified to pH 2 with 1N HCl, and quickly saturated with NaCl and extracted 3 X 100 mL of EtOAc. After drying over MgSO₄, evaporation left 3.32 g (91%), mp 190-198°C (dec). Recrystallization from CHCl₃ gave 6, mp 208-210°C (dec); ¹H NMR (CDCl₃, low solubility) showed the absence of OCH₃.

Methyl N-BOC-N-methyl-L-tyr-N-methyl-L-tyr-D-ala-L-ala-0, N-dimethyl-L-tyr-L-alaninate. 6 (1.28 g, 2.7 mmol) was coupled with 3 (1.18 g, 2.7 mmol) as described above, giving 2.59 g (108%) of crude hexapeptide as a foam which was reacted without further purification.

Methyl N-Methyl-L-tyr-N-methyl-L-tyr-D-ala-L-ala-O,N-dimethyl-L-tyr-L-alaninate. Removal of the BOC grouping from the above hexapeptide with trifluoroacetic acid gave 1.2 g (56% over two reactions) of amine, mp 105-112°C; ¹H NMR (CDCl₃, δ) shows the BOC grouping to be removed, 6.6-7.1 (m, aromatic), 3.8 (s, OCH₃), 2.6-3.4 (m, NCH₃ and CH₂), 0.9-1.4 (m, CCH₃).

Cyclo-N-methyl-L-tyr-N-methyl-L-tyr-D-ala-L-ala-O,N-dimethyl-L-tyr-L-ala (2). The total product from the previous reaction was stirred for 2 h under N_2 with 40 mL of 2 N NaOH, vacuum was applied for a short time to remove MeOH, the pH was adjusted to 6.5 with 1N HCl, and the solvent was evaporated. DMF (300 mL), N-hydroxybenzotriazole (511 mg), and DCC (344 mg) were added. After stirring 48 h under N_2 , the mixture was concentrated to 70 mL, filtered to remove NaCl and urea, washed with CHCl₃, and evaporated to give 2.3 g of solid. Chromatography on silica gel (20:25:2 pentane-CH₂Cl₂-MeOH) gave a fraction of 200 mg, mp 250-265°C, which on washing with CHCl₃ gave 140 mg (12% over two reactions) of 2 as small white crystals, mp 280-290°C. TLC indicated about 150 mg additional 2 to be present in the adjacent chromatography fractions. Fourier transform ¹H and ¹³C NMR spectra were obtained and showed broad absorptions in the expected places, but the signals were weak due to very low solubility in CDCl₃. The formation of the last amide bond was supported by the *insolubility of this* substance in hot IN HCl or saturated KHCO₃; it was soluble in 2N NaOH as expected. MS: no molecular ion peak observed with EI and CI, but FAB using glycerol gave a very strong peak for MH⁺ at 759, and with NaCl added, a very strong peak for MNa⁺ at 781.⁸ The ¹H NMR spectrum, unlike that of $1,^2$ showed very broad absorptions at room temperature due to the presence of 8 forms differing in configurations about the 3 N-methyl peptide bonds. Amino acid analysis indicated the peptide to contain equimolar amounts of ala and N-methyl-tyr.

Anal: Calcd for C₄₀H₅₀N₆Og + 6.4% non-CHN impurity: C, 59.28; H, 6.22; N, 10.37. Found: C, 59.62; H, 6.30; N, 10.08.

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