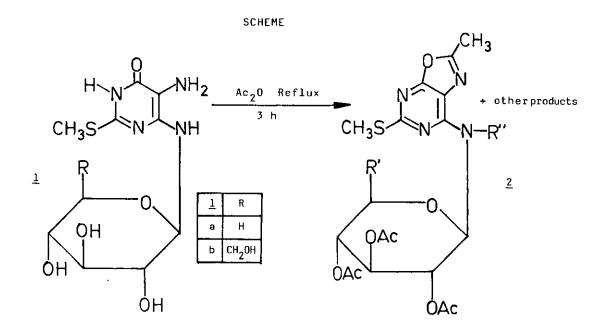
AMINOPYRIMIDINES AND DERIVATIVES. XVI¹. SYNTHESIS OF 7-GLYCOSYL-AMINO-OXAZOLO(5,4-d)PYRIMIDINES²

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<u>Abstracts</u> - Treatment of 5-amino-4-glycosylamino-6-oxo-pyrimidines $\underline{1}$ with acetic anhydride under reflux has led to 2-methyl-7-glycosylamino-oxazolo (5,4-d)pyrimidines 2.

5-Amino-4-glycosylamino-pyrimidines $\underline{1}^3$ are versatile starting materials for the synthesis of glycosides of various condensed heterocycles. Thus, glycosyl purines 4 , glycosyl triazolopyrimidines 5 and glycosyl pteridines 6 have been synthesized from them. We report herein its utility for the synthesis of 7-glycosylamino-oxazolo(5, 4-d)pyrimidines $\underline{2}$, as shown in the Scheme



<u>2</u>	R'	R"	Yield %	M *		
				Calculated	Found	
а	Н	Н		454.11578 C ₁₈ H ₂₂ N ₄ D ₈ S	454.1159	
ь	н	Ac	20	496.1263 C ₂₀ H ₂₄ N ₄ O ₉ S	496.1262	
С	CH ₂ OAc	Н	10	526.1369 C ₂₁ H ₂₆ N ₄ O ₁₀ S	526.1364	
d	CH ₂ OAc Ac 30		568.14746 C ₂₃ H ₂₈ N ₄ D ₁₁ S 568.14			

Oxazolo(5,4-d)pyrimidines have usually been synthesized by two general methods: by cyclization of 4,5-disubstituted oxazoles 7 , or, more often from either 5-amino-6-hydroxypyrimidines 8 , or 5-acylamino-6-hydroxypyrimidines $^{8a-c}$, or 5-imino-6-hydroxypyrimidines 10 by cyclization with either POCl₃ $^{8a-b}$, or PhCOCl $^{8c-d}$, or SOCl $_2^{10a,10c}$ or NBS 10b . The presence on C-4 of the pyrimidine ring of an amino group seems to orientate preferentially the cyclization xith POCl $_3$ towards the purine, yielding neverthless a small amount of oxazolopyrimidine 9b . On the other hand, the action of acid anhydrides on 5-amino-6-hydroxypyrimidines has been reported to produce the N-acylamino derivative $^{8a-b}$. No oxazolopyrimidines have been observed in these reactions, except by Patil et al. 8e . These authors have obtained an oxazolopyrimidine by mixing (EtCO) $_2^{0}$ 0 with 4,6-dihydroxy-5-aminopyrimidine (HCl salt).

In our case, the action of Ac_2O under reflux on diaminopyrimidines $\underline{1}$ has led to the formation of a significant amount of oxazolopyrimidines $\underline{2}$ and several acetylated products. The reactions have been carried out suspending $\underline{1}$ in Ac_2O and heating and keeping under reflux for 3 h. Ac_2O was completely removed by adding methanol and evaporating several times. The resulting solid was fractionated by column cromatography on silicagel, eluting with mixture of EtOAc/EtOH/hexane (2:1:7). The first fraction which was eluted from the reaction mixture obtained from 5-amino-2-methylthio-4-\$\frac{1}{2}-\frac{1}{2}-xylopyranosylamino-6-oxo-pyrimidine (\frac{1}{2}a)^3 corresponds to 2-methyl-5-methylthio-7-(2;3;4-tri-0-acetyl-\beta-\frac{1}{2}-\frac{1}{2}-xylopyranosylamino)-oxazolo(5,4-d) pyrimidine (\frac{2}{2}a), the second one corresponding to 7-acetamido-2-methyl-5-methyl-thio-7-N-(2;3;4-tri-0-acetyl-\beta-\frac{1}{2}-\frac{1}{2}-xylopyranosyl)-oxazolo(5,4-d) pyrimidine (\frac{2}{2}b).

Similarly from 5-amino-2-methylthio-4- β - $\underline{\mathbb{D}}$ -glucopyranosylamino-6-oxo-pyrimidine $(\underline{\mathbf{1b}})^3$, the first fraction corresponds to 2-methyl-5-methylthio-7-(2;3;4;6-tetra-0-acetyl- β - $\underline{\mathbb{D}}$ -glucopyranosylamino)-oxazolo(5,4-d)pyrimidine $(\underline{\mathbf{2c}})$, and the second to 7-acetamido-2-methyl-5-methylthio-7-N-(2;3;4;6-tetra-0-acetyl- β - $\underline{\mathbb{D}}$ -glucopyranosyl)-oxazolo(5,4-d)pyrimidine $(\underline{\mathbf{2d}})$. Next fractions correspond to diverse acetylated products, but no purines have been detected.

When ${\rm Ac}_2{\rm O}$ has been removed by pouring the solution over ${\rm NaHCO}_3{\rm -1ce}$ (and subsequent extraction with ${\rm CHCl}_3$ several non-glycosidic products, like 7-acetamido-2-methyl-5-methylthio-oxazolo(5,4-d)pyrimidine, have been observed in the mixture, thus indicating that an acetolysis of the glycosidic bond has taken place in some extent. The structural and configurational assignment of $\underline{2}$ has been made on the basis of high resolution mass spectrometry (exact mass measurement) and ${}^1{\rm H-nmr}$ data shown in the Scheme and the Table, respectively.

TABLE $^{1}\text{H-NMR}$ Data of Compounds 2

	Me-2	Me-5 <u>H</u>		<u>Ac</u> -N-7	Sugar protons					
			<u>H</u> N-7		<u>H</u> -1'	<u>H</u> -21	<u>H</u> -3'	<u>H</u> -4'	<u>H</u> ~51	<u>H</u> -61
<u>2a</u>	2.55 s	2.55 s	6.20 d		5.75 pt	5.25 pq	5.00 m	5.00 m	4.10 pq 5'e	
			9Hz						3.52 pt 5'a	
<u>2b</u>	2.66 s	56 2.71 s		2.14	5.90 d 9Hz	5.24 m	5.24 m	4.98 m	4.20 5'e	
									3.50 pt 5'a	
<u>2c</u>	2.60 s	2.62 s	6.30 d 9Hz		5.80 pt	5.14 pq	5.45 pt	5.14 pq	3.97 wide	4.29 pq 4.12 pd
<u>2d</u>	2.67 s	2.67 s		2.16	6.05 d 9Hz	5.05- -5.35 m	5.05- -5.35 m	5.05- -5.35 m	3.80- -4.00 m	4.15- -4.30; m

a = axial; d = doublet; e = equatorial; m = multiplet; s = singlet;pq = pseudoquadruplet; pt = pseudo-triplet. The observed coupling in the $^1\text{H-nmr}$ spectra of $\underline{2a}$ and $\underline{2c}$ between the anomeric proton (as a pseudo-triplet) and another proton at lower field (exchangeable by D, therefore corresponding to $\underline{\text{H-N-C}}_7$) excludes the possibility of the isomeric purines of the formulated oxazolopyrimidines.

As for 2b and 2d, structures 3 could also account for both mass and nmr data.

Addition of the nmr shift reagent $\operatorname{Eu(DPM)}_3$ has shifted approximately 0.3 ppm downfield one of the $-\operatorname{C(0)-CH}_3$ signals, the rest of the acetyl signals remaining almost unchanged. The doublet corresponding to the anomeric proton suffers a similar shift of about 0.5 ppm downfield, strongly suggesting its vicinity to the site of complexation of the Eu chelate. For N- and 0-acetylated polyfunctional molecules, the preferential site of complexation has been reported to be the N-Ac rather than 0-Ac¹¹. The formerly mentioned results allow therefore to reject structures 3. As an additional proof, de-acetylation with NaOMe has been performed on 2b and 2d. Protons corresponding to $H-N-C_7$ (doublets, exchangeable by D) have been observed in the respective ^1H-nmr spectra.

The β -pyranosyl configurations of the sugar moleties have been assigned according to the values of the coupling constants of the anomeric protons ($J_{1,2}$ = 9 Hz in all cases). For 2a and 2c the respective pseudo-triplets became doublets when \underline{H} -N-C $_7$ was exchanged by D.

Finally, experiments of double resonance on the 400 MHz 1 H-nmr spectra of $\underline{2b}$ and $\underline{2c}$ have allowed the assignment of the rest of the sugar protons.

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