SYNTHESIS OF NOVEL 5-ALKOXYCARBONYL-5,6-DIHYDROPYRIMIDIN-4(3<u>H</u>)-ONES FROM 3-SUBSTITUTED 2-ALKOXYCARBONYL-2-PROPENOATES AND AMIDINES

Hidetsura Cho, Masaru Ueda, Yoshiko Ohnaka and Mariko Hayashimatsu Suntory Institute for Biomedical Research, 1-1-1, Wakayamadai, Shimamotocho, Mishimagun, Osaka, 618, JAPAN

<u>Abstract</u> A cyclization of 3-substituted 2-alkoxycarbonyl-2propenoates with acetamidine, benzamidine, guanidine, or 1,1-dimethylguanidine hydrochloride in the presence of 1-2 eq of metal alkoxide afforded novel 5-alkoxycarbonyl-5,6-dihydropyrimidin-4(3<u>H</u>)ones in good yield (\mathbb{R}^1 =Me, Et, ⁱPr; \mathbb{R}^2 =phenyl, furyl, thienyl, pyridyl, alkyl; \mathbb{R}^3 =Me, C₆H₅, NH₂, NMe₂). Substitution of a functional group on the phenyl ring influenced the <u>trans-cis</u> relationship between the 5-proton and the 6-proton of the dihydropyrimidone skeleton, namely the ratio of <u>trans/cis</u> increased from <u>ortho</u> substitution (3/2-2/1) and <u>meta</u> (9/1) to <u>para</u> (only <u>trans</u>).

Various cyclizations of α,β -unsaturated carbonyl compounds with amidines were reported to afford pyrimidone or pyrimidine derivatives.¹ In such cases the compound has a good leaving group at a position β to a carbonyl group. However, a survey of the literature reveals only a few papers² on cyclizations of amidines with 3-aryl-2-propenals or 2-propenal (acrolein) and no reactions of amidines with 3-aryl-2-propenoates which have no leaving group at the β position. By the latter reactions several types of hydroxytetrahydropyrimidines and dihydropyrimidones can be obtained, respectively.

As a part of investigations on the synthesis of new heterocyclic compounds with potent biological activity, it became necessary to study the preparation of 5alkoxycarbonyl-5,6-dihydropyrimidin-4(3H)-one derivatives. In this communication, we wish to report briefly a facile synthesis of novel 5-alkoxycarbonyl-6-aryl(or alkyl)-5,6-dihydropyrimidin-4(3H)-ones.

A mixture of 1.0 eq of diethyl benzylidenemalonate and 1.1 eq of acetamidine hydrochloride in the presence of 2.0 eq of NaOEt in EtOH was refluxed for 1 h under Ar. After evaporation of alcohol, the residue was dissolved in CHCl, and washed with brine, dried and evaporated to leave the compound (1a) 3 in 87% yield. Similarly, several derivatives (1b-1j) were synthesized in good yield (see table). The position of the C=N double bond was determined as shown in the figure, because of the long range coupling (J=1-2 Hz) of the methyl group with the methine proton α to the aromatic ring. Among some bases, metal alkoxide is the most suitable base to afford free amidine from the corresponding hydrochloride. The cyclization appears to be initiated via Michael addition of acetamidine to a double bond and then to complete the cyclization with an ester group. The position of the functional group on the aromatic ring of the compound influenced the ratio of stereoisomers Namely, both the compounds (1b) $(X=o-NO_2)$ and (1d) (X=o-Br) consist of the isomers $(\underline{\text{trans}}/\underline{\text{cis}}=2/1)$ $(\underline{\text{trans}}$ $J_{5,6}=12.0$ Hz, $\underline{\text{cis}}$ $J_{5,6}=6.5$ Hz). Differentiation between trans and cis signals is clear in (1d) and (7d) at the 5-proton and the 6-proton but in other examples trans protons often overlap cis protons. However, the NMR signal at C-5 ethyl or C-2 methyl group and TLC behavior prove them to be stereoisomers. The ratio of stereoisomers increased from ortho substitution (2/1) and meta (9/1) to para. Usually, a para-substituted compound was obtained stereoselectively as the trans isomer. Presumably, this is due to the repulsion between the ester group and ortho nitro or ortho bromo group.

Although the <u>para</u>-substituted dihydropyrimidones (p-CN, p-SMe, p-Br) were easily prepared as described above, an exceptional result was observed in the case of the $p-NO_2$ derivative (2).⁴ The cyclization of 1.0 eq of 2-ethoxycarbonyl-3-p-nitrophenyl-2-propenoate with 1.0 eq of acetamidine hydrochloride in the presence of 2.0 eq of NaOEt gave both dihydropyrimidone (1k) and 4-pyrimidone (2) in 40% and 15 % yield, respectively. Therefore, we examined some reaction conditions in detail. It was interesting to find that the choice of 1.0 eq of NaOEt and 0.5 eq of the ester to 1.0 eq of acetamidine hydrochloride was necessary to obtain exclusively the desired dihydropyrimidone (1k) in 74% yield. This type of reaction can be





Table	

No	Compour	nd XorY	R	Base/ROH	Yield	l mp °C	(solvent)	NMR (ĉ)	5&6	protons	J=Hz
1	la	Н	Et	NaOEt/EtOH	87%	129-13	1°(Me_CO- n-C_H,)	3.46	(1H, (1H,	d, J=12 dd, J=1	$\frac{1}{2}, 2$
2	1b	0-NO2	Et	NaOEt/EtOH	77%	oil	— ——- 6 —⊥4 ⁻	3.86	(1H, (1H,	d, J=12 brd,J=1	$\frac{2}{2}, \frac{2}{1}$
3	1c	m-NO ₂	Et	NaOEt/EtOH	64%	110-11	5°(C6H6-C6H14	$(3.47)^{3.47}_{5.13}$	(1H, (1H,	d, J=12 dd, J=1	$\frac{2}{2}, \frac{2}{2}$
4	1d	o-Br	Et	NaOEt/EtOH	70%	oil	trans <u>Cis</u>	3.99 5.29	(1H, (1H, (1H, (1H,	d, J=10 brd, J= d, J=6. dd,J=6.	.5), 10.5, 0.5) 5), 5, 2)
5	le	m-Br	Et	NaOEt/EtOH	87%	oil		3.40	(1H (1H	d, J=12 brd,J=1	$\frac{1}{2}, \frac{1}{2}$
6	lf	p-Br	Et	NaOEt/EtOH	55%	oil		3.39	(1H, (1H,)	d, J=12 brd,J=12	$\frac{-7}{0.5}$
7	1g	p-CN	Me	NaOMe/MeOH	56%	178-18	0°(C6H14CHC1	3.45 3,5.07	(1H, (1H,)	d, J=12 prd.J=12),
8	1h	p-SMe	ⁱ Pr	NaOPr ^j PrOH	31%	109-11	0°(Et ₂ 0)	3.40	(1H, (1H,)	d, J=12 ord,J=12), ,0.2)
9	1i	2,6-diCl	Et	NaOEt/EtOH	53%	135-13	6.5° (EtOH- CcH.)	4.28	(1H, (1H,	d, J=12 dd,J=12	$\frac{1}{2.7}$
10	1j 3	,4,5-triOMe	Et	NaOEt/EtOH	90%	powd	er 6-14-	3.50	(1H, (1H,	d, J=12 d. J=12),),)
11	1k	p-NO2	Et	NaOEt/EtOH	74%	oil		3.46	(1H, (1H,)	d, J=12	$\frac{1}{1}, \frac{1}{1}$
12	2	p-NO2	Et	NaOEt/EtOH	40%	226-22	9" (MeCOOEt-		<u>, </u>		<u>/•••</u>
13	3a	0	Et	NaOEt/EtOH	65%	94~95.	5° (Et ₂ 0 ⁴	3.67	(1H, (1H,	d, J=10 dd.J=10),
14	3b	S	Et	NaOEt/EtOH	64%	76-79 °	$(Et_{2}^{0-C_{6}H_{14}})$	3.55	(1H, (1H,	d, J=10 dd,J=10,	$\frac{1}{1}, \frac{1}{5}$

No	Compound	х	R or R'	Base/ROH	Yield	ld mp°C (solvent) NMR(δ) 5&6 protons J=F
15	4		Et	NaOEt/EtOH	86%	<pre>% 108-110^{(C}₂H₁₄-Et₂O 3.95 (1H, d, J=10), </pre>
16	5		Et	NaOEt/EtOH	98%	<pre>3.33 (1H, d, J=8), 3.80 (1H, m, W_L=12)</pre>
17	6a	p-SMe	Et	NaOEt/EtOH	70%	$180-181^{\circ} \left(\frac{n-C_{0}H_{14}}{Me_{2}C_{0}} - \frac{3.51}{5.14} \right) \left(\frac{1H}{1H}, d, J=11 \right)$
18	6b	m-NO2	Et	NaOEt/EtOH	85%	* $178.5-180^{\circ}$ ($C_{6}H_{14}^{-}$ 3.58 (1H, d, J=12), EtOAc 5.37 (1H, d, J=12)
19	7a	m-CF3	R=Et R'=H	NaOEt/EtOH	62%	% 200.5-203° () 3.58 (1H, d, J=10.8) 4.94 (1H, d, J=10.8)
20	7Ь	o-Cl	R=Et R'=H	NaOEt/EtOH	70%	<pre>% 169-172 (MeOH-CHCl 3.51 (1H, d, J=7.2 -Me_CO 5.22 (1H, d, J=7.2</pre>
21	7c	m-NO ₂	R=Et R'=Me	NaOEt/EtOH	92%	% 211-213 (MeOH) 3.52 (1H, d, J=7.2), 5.12 (1H, d, J=7.2)
22	7đ	0-N0 ₂	R=Me R'≕Me	NaOMe/MeOH	90%	$\begin{array}{r} \underline{\text{trans}} 3.48 (1\text{H}, \text{d}, \text{J=6.0}), \\ 5.42 (1\text{H}, \text{d}, \text{J=6.0}), \\ 5.42 (1\text{H}, \text{d}, \text{J=6.0}), \\ () \underline{\text{cis}} 3.16 (1\text{H}, \text{d}, \text{J=5.3}), \\ 5.42 (1\text{H}, \text{d}, \text{J=5.3}) \end{array}$

* The NMR spectra were taken in CDCl, solution except for (7a-d).

(7a), (7b), (7d)---DMSO-d₆, (7c)---CDCl₃:CD₃OD=1:1

applied to α -alkoxycarbonyl-2-propencate with another heterocyclic moiety or an alkyl moiety at the β position. Thus, furyl, thienyl, pyridyl, and cyclohexyl dihydropyrimidinones [(3a), (3b), (4), (5)] were obtained.

Next, the cyclization of dialkyl benzylidenemalonate and benzamidine hydrochloride was carried out under the same reaction condition to yield 5-alkoxycarbonyl-2,6-diphenyl-5,6-dihydropyrimidin-4($3\underline{H}$)-one (6a) or (6b) as a sole compound (<u>trans</u>, J=11 or 12 Hz), respectively.

Finally, we studied the reaction of guanidine or 1,1-dimethylguanidine with α , β unsaturated esters.⁵ In this case, since the product was insoluble in water and most organic solvents except for DMSO, the crystals precipitated as the reaction proceeded. A solution of 1.0 eq of diethyl 3-nitrobenzylidenemalonate in EtOH was added to a stirred mixture of 1,1-dimethylguanidine hydrochloride and 2.0 eq of NaOEt in EtOH. After reflux for 2 h, the solvent was evaporated under reduced pressure to leave the crystals. Then, filtration and washing with water gave the desired compound (7c) in 92% yield. The NMR spectrum of (7d) showed a <u>trans</u> and <u>cis</u> mixture (3:2) but the others (7a-c) only <u>trans</u> isomer.

Consequently, the cyclization of 3-substituted 2-alkoxycarbonyl-2-propenoates with amidine hydrochloride in the presence of 1-2 eq of metal alkoxide provided suitably substituted 5,6-dihydropyrimidin-4(3 \underline{H})-ones in good yield (R¹=Me, Et, Prⁱ; R²=aryl, alkyl; R³=Me, C₆H₅, NH₂, NMe₂).

The pharmacological activity of 5,6-dihydropyrimidin-4(3H)-ones will be reported

in due course.

ACKNOWLEDGEMENT

The authors thank Professor Teruhisa Noguchi and Dr. Minoru Morita for their encouragement throughout the investigation and Dr. Fumio Satoh for his kind discussions. We also thank Mr. Keiyuu Shima and Miss Yumi Takeuchi for their helpful experiments.

REFERENCES AND NOTES

- a) H.-D. Stachel, <u>Chem. Ber</u>., 1962, <u>95</u>, 2172; b) W. D. Rudorf, A. Schierhorn, and M. Augustin, <u>Tetrahedron</u>, 1979, <u>35</u>, 551; c) K. T. Potts, M. J. Cipullo, P. Ralli, and G. Theodoridis, <u>J. Org. Chem</u>., 1983, <u>48</u>, 4841; d) S. Kohra, Y. Tominaga, Y. Matsuda, and G. Kobayashi, <u>Heterocycles</u>, 1983, <u>20</u>, 1745; e) S.-J. Lee and J. M. Cook, <u>Heterocycles</u>, 1983, <u>20</u>, 87.
- 2) a) K. K. Weinhart and M. Mark, <u>Eur. Patent Appl</u>. EP 24776, 11 Mar. 1981, <u>C.A.</u>, 95 (11): 97837a: 4-phenyl- and 5-phenyl-1,4,5,6-tetrahydropyrimidine derivatives have been reported to have anticonvulsion activity by Syntex Inc.
 b) W. Wendelin, <u>Monatsh. Chem.</u>, 1974, <u>105</u>, 382, <u>C.A.</u>, 81 (21): 136094p ---2-amino-6-phenyl-1,4,5,6-tetrahydropyrimidin-4-ol
 c) A. L. Vais and V. P. Mamaev, <u>Khim. Geterosikl. Soedin.</u>, 1977, (5), 674, <u>C.A.</u>, 87 (11): 84937t ---2,6-diphenyl-1,4,5,6-tetrahydropyrimidin-4-ol
 d) A. Weis, F. Frolow, D. Zamir, and M. Bernstein, <u>Heterocycles</u>, 1984, <u>22</u>, 657 ---2-phenyl-1,4,5,6-tetrahydropyrimidin-4-ol was prepared and oxidized with KMnO_A or Ag₂O to give 2-phenyl-5,6-dihydropyrimidin-4-one.
- All new compounds gave satisfactory IR, NMR, MS spectral data and combustion analyses.
- 4) R. M. Dodson and J. K. Seyler, <u>J. Org. Chem</u>., 1951, <u>16</u>, 461 --- The similar oxidative cyclization was reported; 2,4,6-triphenylpyrimidine was obtained in a good yield from the condensation of benzamidine and benzalacetophenone in an alcoholic solution of KOH.
- 5) Y. H. Kim and N. J. Lee, <u>Heterocycles</u>, 1983, <u>20</u>, 1769 ---Reaction of guanidine derivative with simple β -nonsubstituted α,β -unsaturated ester was reported.

Received, 10th May, 1984