POLYCONDENSED NITROGEN HETEROCYCLES. PART 15. REACTIVITY OF 3-DIAZOPYRROLES. 3¹. 1H-PYRROLO [3,2-c] CINNOLINE AND PYRROLO-[3,4-c] PYRIDAZINE

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<u>Abstract</u>- 3-Diazopyrroles of type <u>5</u> were prepared and refluxed in acetic acid. In the case of <u>5a</u>, an intramolecular coupling reaction between the diazonium group and the methoxyphenyl substituent in the 2-position leads to the 1H-pyrrolo $[\overline{3}, 2-\overline{2}]$ cinnoline derivative <u>6</u>, whilst in the case of <u>5b</u> an intramolecular coupling with the acetyl group in the 4-position affords the pyrrolo $[\overline{3}, 4-\overline{2}]$ pyridazine derivative <u>7</u>.

In connection with our investigations on pyrrolo-cinnoline derivatives, which demonstrated antigerminative properties², and on possible structure-activity relationships, we became interested in the synthesis and reactivity of 3-diazo-2-phenylpyrroles as intermediates to 1H-pyrrolo [3,2-c] cinnoline derivatives. In preceding papers in this series^{1,3}, we reported that the 3-diazopyrroles behave like the aromatic diazonium salts to give coupling reactions and decomposition, but the diazo group did not couple, under acid conditions, with the phenyl in the 2-position.

A possible explanation could be that the benzene ring was not sufficiently activated for the electrophilic reaction to take place.

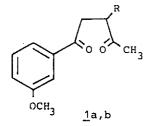
In this paper the synthesis and the reactivity of 3-diazopyrroles of type 5, in

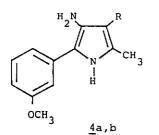
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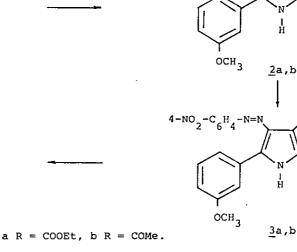
which the phenyl in the 2-position was made more reactive by the introduction of a methoxy group in the opportune position, are reported.

It is to be expected that compound 5a, which bears in the 4-position an ethoxycarbonyl group, unreactive towards the diazo group, would cyclize to give the 1H-pyrrolo [3,2-c] cinnoline ring system. In compound 5b, on the other hand, it would be interesting to observe the competition between the two possible intramolecular coupling reactions.

To this purpose the pyrroles 2a, b were prepared from the corresponding diketones 1a, b. Copulation of compounds 2a, b with the diazonium salt of 4-nitroaniline afforded the azo derivatives 3a, b, which upon reduction with stannous chloride in acetic acid, gave the aminopyrroles 4a, b (overall yield 56-71 %).

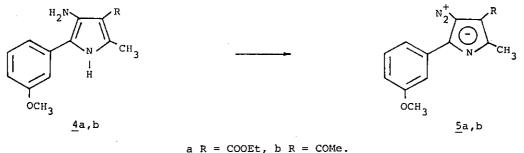






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The diazotization of compounds 4a, b gave the 3-diazopyrroles 5a, b which were isolated from the neutralized reaction mixture.

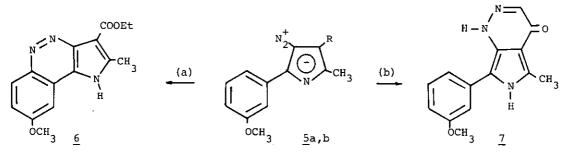


R = CODEt, D R = COME

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The diazopyrroles were characterized by analytical and spectral data. The ir spectra showed a strong absorption at 2120 cm¹ due to the diazo group stretching, while in the nmr spectra there were no signals attributable to NH group. The 3-diazopyrroles 5a, b were refluxed in acetic acid until no starting compounds were detected by t.l.c.

With compound 5a the expected 1H-pyrrolo [3,2-c] cinnoline derivative 6 was isolated in 70% yield, whilst in the case of 5b the pyrrolo [3,4-c] pyridazin-4-one 7 was obtained in 60% yield.



a R = COOEt, b R = COMe.

The structures of compounds <u>6</u> and <u>7</u> were assigned on the basis of spectral data. The nmr spectrum of derivative <u>6</u> showed, besides substituent bands, signals due to three aromatic protons, confirming the ring closure had occurred onto the methoxyphenyl substituent and that the methoxy group occupied the 8-position of the polycondensed ring system.

The ir spectrum of the derivative <u>7</u> showed two absorption bands at 3430 and 3240 cm⁻¹ due to the NH stretchings, while in the nmr spectrum, there were all the expected signals including two exchangeable singlets for one proton each attributable to NH group, but the signal due to the acetyl methyl group was missing, confirming that intramolecular coupling to the acetyl group had taken place. Evidently, when the 3-diazopyrrole bears an acetyl group in the 4 position, intramolecular coupling with the enolic form of the acetyl group is preferred, and the phenyl group in the 2-position, even if activated towards electrophilic substitutions, is less reactive than the acetyl. But when there is no competition with the substituent in the 4 position, the diazonium group, formed in acid medium,

intramolecularly couples with the 3-methoxyphenyl group in the 2-position, affording the 1H-pyrrolo $[3, 2-\overline{3}]$ cinnoline derivative.

3-Diazopyrroles, in the intramolecular coupling reported so far^{3,5}, always showed a preferential reactivity towards the substituents on the 4-position of the nucleus. The low availability of the phenyl in the 2-position to undergo coupling reaction could be explicated according to the larger transmission of electronic effects between the 2- and 3-position (hyper-ortho) than between the 3- and 4-position (hypo-ortho) due to the high "bond fixation" which give rise to a C-2 — C-3 bond with a high π bond order as already pointed out in thiophen ring⁶.

EXPERIMENTAL

All melting points were taken on a Büchi-Tottoli capillary melting point apparatus and are uncorrected; ir spectra were determined in nujol mull with a Perkin-Elmer 299 spectrophotometer; nmr spectra were obtained with a Varian FT 80 spectrometer (TMS as internal reference); mass spectra were run on a JEOL JMS-01 SG-2 double focusing mass spectrometer operating with an electron beam energy of 75 eV and 10 KV accelerating voltage. The chromatography was performed on columns of silica gel deactivated with water (15%).

Preparation of 3-ethoxycarbonyl- (1a) and 3-acetyl-1-(3-methoxyphenyl)-1,4pentanedione (1b).

To a mixture of sodium ethoxide (40 mmoles) in absolute ethanol and ethyl acetoacetate (40 mmoles) or acetylacetone (40 mmoles), 3-methoxyphenacyl bromide (40 mmoles) was added in small portions with stirring and cooling on an ice bath. After standing at room temperature overnight, the solid was filtered off and the ethanol solution was evaporated under reduced pressure to give an oil in the case of <u>1</u>a. In the case of <u>1</u>b, the diketone was isolated by complexing it with aqueous cupric acetate. The collected solid was dissolved in dilute sulphuric acid (25%) and extracted with diethyl ether. The evaporated solution gave compound <u>1</u>b. <u>Compound 1a</u> was isolated as an oil in 60% yield, bp 202°C, ir: 1740, 1715 and 1680 (CO) cm⁻¹; nmr (DMSO-d₆) δ : 1.26 (3H,t,CH₂CH₃) 2.40 (3H,s,CH₃) 3.70 (2H,d, <u>CH₂CH) 3.86 (3H,s,OCH₃) 4.26 (2H,q,CH₂CH₃) 4.33 (1H,t,CH₂CH) 7.20 - 7.90 (4H,m,</u> $C_{6}^{H_{4}}$). Anal. Calcd. for $C_{15}^{H_{18}O_{5}}$: C, 64.73; H, 6.52. Found: C, 64.90; H, 6.45. <u>Compound 1b</u> was recrystallized from ethanol (yield 40%), mp 70 C, ir: 1710 and 1680 (CO) cm⁻¹; The nmr spectrum was in agreement with the presence of two tautomeric structures, $ArCO-CH_{2}$ -CH-COCH₃ (A) and ArCO-CH-C=C(OH)CH₃ (B), in equilibri-COCH₃ (B), in equilibri-COCH₃ (B) in ratio 4:1 respectively δ : 2.10 (3H,s,COCH₃,B) 2.36 (6H,s,COCH₃,A) 3.50 (2H,s,CH₂,B) 3.70 (2H,d,CH₂,A) 3.76 (3H,s,C(OH)CH₃,B) 3.90 (6H,s,OCH₃,A and OCH_{3} ,B) 4.25 (1H,s,OH,B) 4.56 (1H,t,CH,A) 7.20 - 7.90 (8H,m,C₆H₄,A and C₆H₄,B). Anal. Calcd. for $C_{14}H_{16}O_{4}$: C, 67.73; H, 6.50. Found: C, 67.49; H, 6.38. <u>Preparation of 4-substituted 2-(3-methoxyphenyl)-5-methylpyrroles (2a,b).</u> The preparation of these compounds was carried out according to the procedure described previously⁴.

<u>Compound 2a (R=COOC₂H₅)</u> was recrystallized from ethanol (yield 90%), mp 137 C; ir: 3320 (NH) 1655 (CO) cm⁻¹; nmr (DMSO-d₆) &: 1.30 (3H,t,CH₂CH₃) 2.53 (3H,s,CH₃) 3.83 (3H,s,OCH₃) 4.28 (2H,q,CH₂CH₃) 6.93 (1H,d,CH,J=2.5 Hz, a singlet appeared upon exchange with deuterium oxide) 6.80 - 7.50 (4H,m,C₆H₄) 11.70 (1H,broad, exchangeable NH); ms: M⁺= 259. Anal. Calcd. for C₁₅H₁₇NO₃: C, 69.48; H, 6.61; N, 5.40. Found: C, 69.78; H, 6.49; N, 5.61.

<u>Compound 2b (R=COCH₃)</u> was recrystallized from ethanol (yield 80%), mp 159°C; ir: 3190 (NH) 1625 (CO) cm⁻¹; nmr (DMSO-d₆) &: 2.36 (3H,s,CH₃) 2.45 (3H,s,CH₃) 3.83 (3H,s,OCH₃) 7.05 (1H,d,CH,J=2.5 Hz, a singlet appeared upon exchange with deuterium oxide) 6.75 - 7.50 (4H,m,C₆H₄) 11.70 (1H,broad,exchangeable NH); ms: $M^+=$ 229. Anal. Calcd. for C₁₄H₁₅NO₂: C, 73.34; H, 6.59; N, 6.11. Found: C, 73.51; H, 6.51; N, 6.00.

Preparation of 4-substituted 2-(3-methoxyphenyl)-5-methyl-3-(4-nitrophenylazo) pyrroles (3a,b).

The preparation of these compounds was carried out according to the procedure described previously¹.

<u>Compound 3a (R=COOC_2H_5)</u> was recrystallized from ethanol (yield 85%), mp 203°C; ir: 3290 (NH) 1685 (CO) cm⁻¹; nmr (DMSO-d₆) &: 1.26 (3H,t,CH₂<u>CH₃</u>) 2.43 (3H,s,CH₃) 3.93 (3H,s,OCH₃) 4.33 (2H,q,<u>CH₂CH₃</u>) 7.06 - 8.66 (8H,m,2xC₆H₄) 11.35 (1H,broad, exchangeable NH); ms: M^+ = 408; Anal. Calcd. for $C_{21}H_{20}N_4O_5$: C, 61.76; H, 4.94; N, 13.72. Found: C, 61.44; H, 4.99; N, 13.71.

<u>Compound 3b (R=COCH₃)</u> was recrystallized from ethanol (yield 70%), mp 203°C; ir: 3210 (NH) 1645 (CO) cm⁻¹; nmr (DMSO-d₆) ^{*} δ : 2.25 (3H,s,CH₃) 2.29 (3H,s,CH₃) 3.84 (3H,s,OCH₃) 7.42 - 8.42 (8H,m,2xC₆H₄) 12.23 (1H,s,exchangeable NH); ms: M⁺= 378. Anal. Calcd. for C₂₀H₁₈N₄O₄: C, 63.48; H, 4.80; N, 14.81; Found: C, 63.61; H, 4.98; N, 14.78.

<u>Preparation of 4-substituted 3-amino-2-(3-methoxyphenyl)-5-methylpyrroles (4a,b)</u>. These compounds were prepared according to the procedure described previously (method B)¹.

<u>Compound 4a (R=COOC₂H₅)</u> was recrystallized from benzene (yield 83%), mp 146°C; ir: 3440 and 3350 (NH₂) 3280 (NH) 1640 (CO) cm⁻¹; nmr (DMSO-d₆) δ : 1.30 (3H,t, CH₂CH₃) 2.50 (3H,s,CH₃) 3.83 (3H,s,OCH₃) 4.30 (2H,q,CH₂CH₃) 4.93 (2H,s,exchangeable NH₂) 6.70 - 7.45 (4H,m,C₆H₄) 11.00 (1H,s,exchangeable NH); ms: M⁺= 274; Anal. Calcd. for C₁₅H₁₈N₂O₃: C, 65.67; H, 6.61; N, 10.21. Found: C, 65.44; H, 6.41; N, 10.21.

<u>Compound 4b (R=COCH₃)</u> was recrystallized from ethanol (yield 80%), mp 215°C (decomp.); ir: 3440 and 3340 (NH₂) 3180 (NH) 1680 (CO) cm⁻¹; nmr (DMSO-d₆) δ : 2.33 (3H,s,CH₃) 2.50 (3H,s,CH₃) 3.80 (3H,s,OCH₃) 5.36 (2H,s,exchangeable NH₂) 6.70 - 7.50 (4H,m,C₆H₄) 11.00 (1H,s,exchangeable NH); ms: M⁺= 244. Anal. Calcd. for C₁₄H₁₆N₂O₂: C, 68.83; H, 6.60; N, 11.47. Found: C, 69.04; H, 6.71; N, 11.40. <u>Preparation of 4-substituted 3-diazo-2-(3-methoxyphenyl)-5-methylpyrroles (5a,b)</u>. These compounds were prepared according to the procedure described previously³. The neutralized reaction mixture was extracted with diethyl ether (3 x 100 ml). The organic layer, dried on sodium sulphate, was evaporated under reduced pressure and purified by chromatography (eluant light petroleum (bp 50 - 70°C)- ethyl acetate 8:2 for 5a and light petroleum (bp 50 - 70°C)- ethyl acetate 1:1 for 5b). <u>Compound 5a (R=COOC₂H₅)</u> was recrystallized from light petroleum (bp 50 - 70°C) (yield 70%), mp 73°C (decomp.); ir: 2120 (NEN) 1670 (CO) cm⁻¹; nmr (DMSO-d₆) δ : 1.30 (3H,t,CH₂CH₃) 2.50 (3H,s,CH₃) 3.86 (3H,s,OCH₃) 4.35 (2H,q,CH₂CH₂) 7.10 - 7.60

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 $(4H,m,C_{6}H_{4})$; ms: $M^{+}= 285$; Anal. Calcd. for $C_{15}H_{15}N_{3}O_{3}$: C, 63.15; H, 5.30; N, 14.73. Found: C, 62.89; H, 5.19; N, 14.49.

<u>Compound 5b (R=COCH₃)</u> was recrystallized from light petroleum (bp 50 - 70°C) (yield 70%), mp 73°C (decomp.); ir: 2120 (N=N) 1625 (CO) cm⁻¹; nmr (DMSO-d₆) &: 2.46 (3H,s,CH₃) 2.56 (3H,s,CH₃) 3.85 (3H,s,OCH₃) 7.05 - 7.56 (4H,m,C₆H₄); ms: M⁺= 255. Anal. Calcd. for $C_{14}H_{13}N_{3}O_{2}$: C, 65.87; H, 5.13; N, 16.46. Found: C, 65.66; H, 4.99; N, 16.21.

Action of acetic acid on the 3-diazopyrroles 5a,b.

The 3-diazo compounds $\underline{5}a$, b (10 mmoles) were refluxed in acetic acid until no starting material was detected by t.l.c. (8 h in the case of $\underline{5}a$ and 0.5 h in the case of $\underline{5}b$). The mixture was poured onto crushed ice. The solid precipitate was collected. In the case of $\underline{5}a$ the solid was purified by refluxing in ethanol for a long time. The insoluble residue was identified as ethyl 8-methoxy-2-methylpyrro-lo $[3,2-\underline{c}]$ cinnolin-3-carboxylate (6) (yield 70%), mp 272°C; ir: 3410 (broad NH) 1700 (CO) cm⁻¹; nmr (DMSO-d₆) &: 1.39 (3H,t,CH₂CH₃) 2.85 (3H,s,CH₃) 4.03 (3H,s, OCH₃) 4.41 (2H,q,CH₂CH₃) 7.54 (1H,dd,J₀=9.38 and J_m=2.64 Hz,H-7) 7.91 (1H,d,J_m= 2.64 Hz,H-9) 8.40 (1H,d,J₀=9.38,H-6); ms: M⁺= 285. Anal. Calcd. for C₁₅H₁₅N₃O₃: C, 63.15; H, 5.30; N, 14.73. Found: C, 63.35; H, 5.21; N, 14.49. The ethanol solution was evaporated under reduced pressure. The residue was chromatographed on p.l.c. plate using light petroleum (bp 50 - 70°C)-ethyl acetate (8:2) as eluant to give compound $\underline{2}a$ (yield 3%).

In the case of <u>5</u>b the crude solid was chromatographed: elution with light petroleum (bp 50 - 70°C)-ethyl acetate (3:7) gave compound <u>2</u>b (yield 5%). Further elution with ethyl acetate gave 7-(3-methoxyphenyl)-5-methyl-1,6-dihydropyrrolo <u>[6,4-c]</u> pyridazin-4-one (<u>7</u>). This compound was recrystallized from ethanol (yield 60%), mp 273 C°(decomp.); ir: 3430 and 3220 (NH) 1600 (broad CO) cm⁻¹; nmr (DMSO-d₆) &: 2.68 (3H,s,CH₃) 3.86 (3H,s,OCH₃) 6.64 - 7.48 (5H,m,C₆H₄ and CH) 12.46 (1H,s,exchangeable NH) 12.76 (1H,s,exchangeable NH); ms: M⁺= 255. Anal. Calcd. for $C_{14}H_{13}N_3O_2$: C, 65.87; H, 5.13; N, 16.46. Found: C, 65.80; H, 5.24; N, 16.61.

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REFERENCES AND NOTES

- An account of this work was presented at "IV Convegno Nazionale Divisione Chimica Farmaceutica della Società Chimica Italiana, Palermo Oct. 1983.
- Part 2: G. Dattolo, G. Cirrincione, A.M. Almerico, G. Presti and E.Aiello, <u>Heterocycles</u>, 1983,20,829.
- G.Dattolo, G. Cirrincione, A.M. Almerico, I. D'Asdia and E.Aiello, <u>Heterocycles</u>, 1982,<u>19</u>,681 and references cited therein.
- G. Dattolo, G. Cirrincione, A.M. Almerico and E. Aiello, <u>Heterocycles</u>, 1983,20,255.
- E. Aiello, G. Dattolo, G. Cirrincione, S. Plescia and G. Daidone, <u>J.Heterocycl.Chem.</u>, 1979,16,209.
- 5) F. Angelico, <u>Gazz.Chim.Ital.</u>,1909,<u>39</u>,II,134.
- G. Consiglio, D. Spinelli, S. Gronowitz, A.B. Hornfeldt and R. Noto, <u>Chemica Scripta</u>, 1982,19,46 and references cited therein.

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