

THE NOVEL RING OPENING OF AN OXAZOLO[5,4-d]PYRIMIDINE AND SUBSEQUENT REARRANGEMENT TO FORM AN IMIDAZO[4,5-d]PYRIMIDINE

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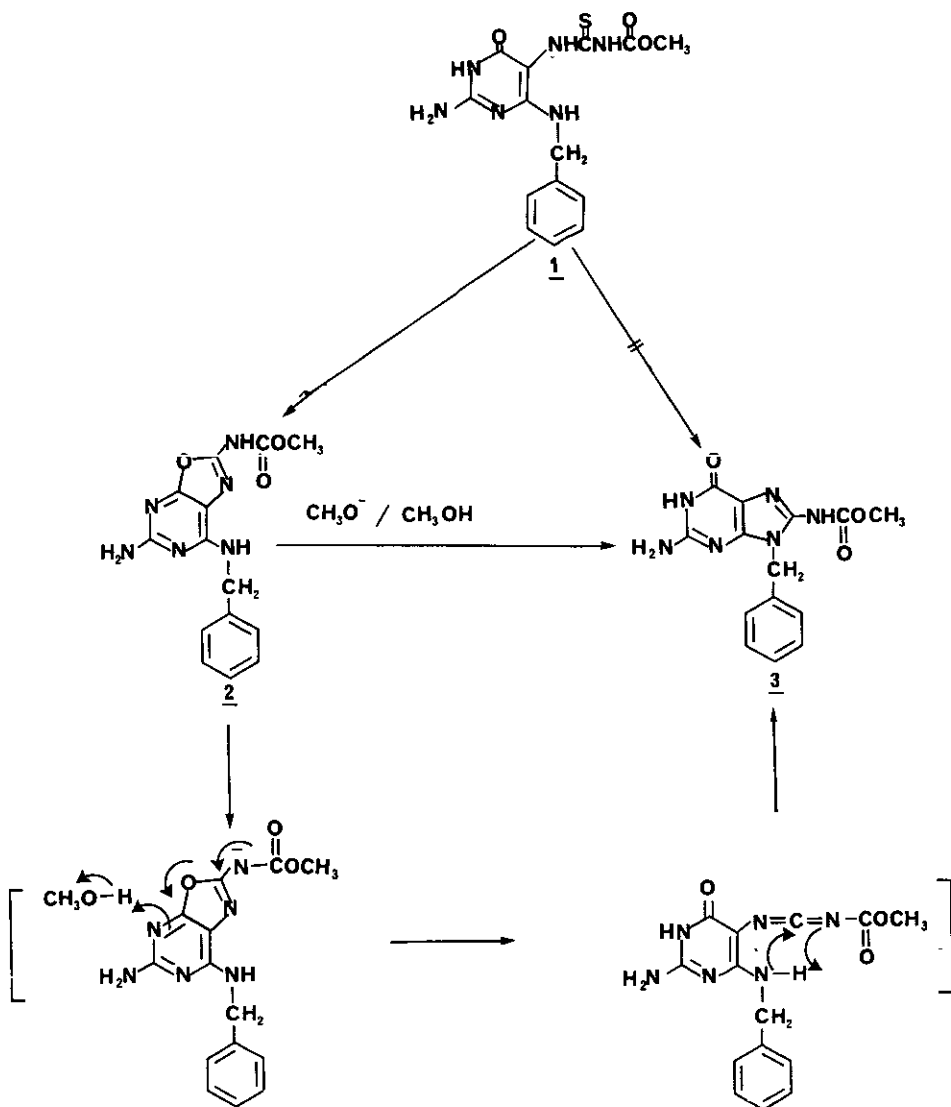
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**Abstract** - A novel rearrangement from methyl 6-amino-4-benzylaminooxazolo[5,4-d]pyrimidine-2-carbamate to methyl 9-benzylguanine-8-carbamate (3) is described.

Our attempt to synthesize methyl 9-benzylguanine-8-carbamate (3) by a direct cyclodesulfurization of 2-amino-4-benzylamino-5-[1-(3-methoxycarbonyl)thioureido]pyrimidin-6-one (1) with dicyclohexylcarbodiimide was unsuccessful. Using the reaction conditions which were specifically designed to effect a direct conversion of 1 to 3, we have obtained only 6-amino-4-benzylaminooxazolo[5,4-d]pyrimidine-2-carbamate (2). However, we have now developed reaction conditions which effect a novel ring opening of compound 2 followed by a rearrangement and recyclization to afford compound 3. In essence, this represents a facile conversion of compound 1 into compound 3. It has been well documented that various 1,2,4-oxadiazoles undergo a mononuclear heterocyclic rearrangement to various heterocyclic compounds under experimental conditions such as heating or treatment with a base, *i.e.*, potassium hydroxide or sodium methoxide in methanol<sup>2,3</sup>. Also numerous 7-aminooxazolo[5,4-d]pyrimidines are known to undergo an intramolecular rearrangement to imidazo[4,5-d]pyrimidin-6-ones by heating in formamide or dilute sodium hydroxide<sup>4,5</sup>. The mechanism postulated for the rearrangement of these oxazolo[5,4-d]pyrimidines to an imidazo[4,5-d]pyrimidine assumes that an initial nucleophilic substitution occurs at the carbon atom of the oxazole ring effecting a ring opening to an intermediate 5-acylamino-4-amino-6-oxopyrimidine anion. Subsequent annulation through nucleophilic attack of the C-4 amino group on the acyl carbonyl carbon then affords the imidazo[4,5-d]pyrimidine ring system. To the best of our knowledge, an intramolecular ring-opening-recyclization of a 7-aminooxazolo[5,4-d]pyrimidine which proceeds through a carbodiimide intermediate to an imidazo- [4,5-d]pyrimidine has not been described. We now wish to report a facile synthetic procedure for the synthesis of compound 3 from compound 2 through a presumed carbodiimide intermediate.

A mixture of compound 2<sup>1</sup> (2.45 g, 7.8 mmoles), anhydrous potassium carbonate (2.2 g, 15.6 mmoles) in anhydrous methanol (50 mL) was heated under reflux for 5 hours. The solvent was

removed in vacuo and the resulting solid was dissolved in water (20 mL). Upon the addition of an aqueous ammonium chloride solution [1.68 g, (31.2 mmoles) in water (20 mL)] the solid which had precipitated was collected by filtration. The solid was washed with cold water (10 mL) and then



methanol (5 mL) to furnish 2.13 g (87%) of crude compound **3**. The solid was recrystallized from a DMF and methanol mixture (1:1); mp 321-322° dec.; ir (KBr): 3450, 3280, 2920, 1740  $\text{cm}^{-1}$ ;

$^1\text{H}$  NMR (DMSO- $d_6$ ):  $\delta$  3.4 (s, 3 H,  $\text{CH}_3$ ), 5.1 (s, 2 H,  $\text{CH}_2$ ), 6.58 (s, 2 H,  $\text{NH}_2$ ,  $\text{D}_2\text{O}$  exchangeable), 7.3 (m, 5 H, Ar-H), 9.8 (br, 1 H, NH,  $\text{D}_2\text{O}$  exchangeable), 10.7 (s, 1 H, NH,  $\text{D}_2\text{O}$  exchangeable); uv:  $\lambda_{\text{max}}^{\text{pH } 7}$  266 nm ( $\epsilon$   $1.7 \times 10^4$ ),  $\lambda_{\text{max}}^{\text{pH } 1}$  259 nm ( $\epsilon$   $1.8 \times 10^4$ ),  $\lambda_{\text{max}}^{\text{pH } 11}$  264 nm ( $\epsilon$   $1.4 \times 10^4$ ), 273 nm ( $\epsilon$   $1.3 \times 10^4$ ), 289 nm ( $\epsilon$   $1.4 \times 10^4$ ). A reasonable mechanism for this reaction involves the initial abstraction of a proton from the 2-carbamoyl moiety of compound 2, followed by an opening of the oxazole ring to give a carbodiimide intermediate. Subsequent addition of the C-4-amino nucleophile to the carbon atom of the carbodiimide, then furnishes the imidazo[5,4-d]pyrimidin-6-one derivative 3.

The mechanism of this facile rearrangement and its application to the synthesis of other heterocyclic systems is under further investigation in our laboratory.

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