## PHOTOCHEMICAL REACTIONS OF QUINOXALIN-2-THIONES

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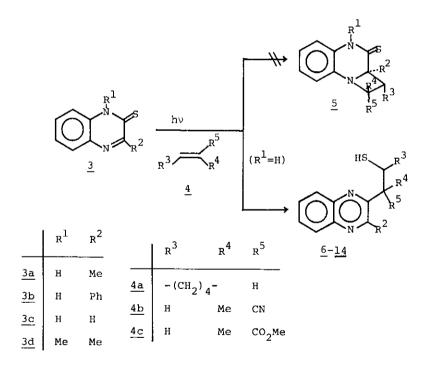
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<u>Abstract</u> — In contrast with their oxo-analogues  $(\underline{1})$ , quinoxalin-2-thiones  $(\underline{3})$  undergo photo-induced addition to cyclohexene  $(\underline{4a})$ , methacrylonitrile  $(\underline{4b})$ , and methyl methacrylate  $(\underline{4c})$  to give 2-substituted quinoxalines  $(\underline{6-14})$ .

It has been demonstrated that a variety of thiones can undergo photo-induced addition to various olefins. 1,2 However, the photochemical properties of thioamides and thioimides have been scarcely studied. Thus, we report on investigation of the photochemical reactions of quinoxalin-2-thiones (3) which illustrates an example of the dramatic difference in the photochemical behaviors of carbonyl and thiocarbonyl compounds. Recently, we reported that on irradiation quinoxalin-2-ones (1) reacted regions lectively with electron-deficient olefins to afford the 1:1-adducts of C=N double bond of 1 and olefin, azetidine derivatives (2).

Irradiation of 3-methylquinoxalin-2-thione [ $\underline{3a}$ ;  $\lambda_{\max}$ (EtOH) 218 ( $\varepsilon$ =16400), 275 (9800), and 392 nm (6200)] (200 mg) in ethylene glycol dimethyl ether (50 ml) in the presence of large excess cyclohexene ( $^{\circ}$ I ml) with a high pressure mercury lamp (Eikosha EHB-300 W) through a Pyrex filter under nitrogen for 10 h gave 2-(2-mercaptocyclohexyl)-3-methylquinoxaline ( $\underline{6}$ ) [mp 101-102°C;  $\lambda_{\max}$ (EtOH) 205 ( $\varepsilon$ =35000), 237 (30500), 312 (sh, 7000), and 318 nm (8300);  $\nu_{\max}$ (KBr) 2555, 1557, 1485, 1100, and 763 cm<sup>-1</sup>;

 $\delta_{\rm H}({\rm CDC1}_3)$  1.25-2.63 (m, 8H), 1.60 (d, 1H, J=5.9 Hz), 2.75 (s, 3H), 3.39-3.73 (m, 2H), 7.45-7.71 (m, 2H), and 7.88-8.13 (m, 2H);  $\delta_{\rm C}({\rm CDC1}_3)$  21.1(t), 22.5(q), 24.6(t), 25.2(t), 34.5(t), 40.0(d), 45.3(d), 128.0(d), 128.5(d), 128.9(d), 140.3(s),140.4 (s), 152.3(s), and 156.9(s); m/e (chemical ionozation) 259 (QM<sup>†</sup>)} in 48% isolated yield, along with the unchanged thione (3a). The structure of 6 was elucidated on the basis of spectral properties and elemental analysis. Gunchi and his co-workers described that irradiation of a mixture of 3-methylquinoxalin-2-thione (3a) and cyclohexene (4a) gave 2-cyclohexylthio-3-methylquinoxaline (16) and 2-(3-methylquinoxalyl)disulfide (17) via radical mechanism. The  $^{13}{\rm C-n.m.r.}$  spectrum of 6 showed two tertiary carbons at  $\delta$  40.0(d) and 45.3(d) and four secondary carbons at  $\delta$  21.1(t), 24.6(t), 25.2(t), and 34.5(t) and no thiocarbonyl carbon signal, 8 so we concluded that the photoproduct thus obtained is 2-(2-mercaptocyclohexyl)-3-methylquinoxaline (6). Similarly, irradiation of the quinoxalin-2-thiones (3b-c) in the presence of cyclohexene (4a) gave the corresponding 2-substituted quinoxaline (7-8).



Furthermore, 2-substituted quinoxalines (9-14) were obtained in 35-63% yields when the quinoxalin-2-thione (3a-c) was irradiated in the presence of electron-deficient olefins such as methacrylonitrile (4b) and methyl methacrylate (4c) under the same conditions as described above. On the other hand, 1,3-dimethylquinoxalin-2-thione

 $(\underline{3d})$  was inert to the photolysis in ethylene glycol dimethyl ether in the presence of olefins (4a-c) under the similar conditions

Table. Yield of the 2-Substituted Quinoxalines (6-14)

				Yield (%) a				
Compd.	R <sup>2</sup>	R <sup>3</sup>	$R^4$	R <sup>5</sup>	Product	Recovered	( <u>3</u> )	
<u>6</u>	Me	-(CH <sub>2</sub>	) <sub>4</sub> -	H	48	5		
7	Ph	-(CH <sub>2</sub>	) 4-	H	50	∿0		
8	Ħ	-(CH <sub>2</sub>	) 4-	н	43	25	$(\underline{15})^{b}$ 32	
<u>9</u>	Me	H	Ме	CN	54	23		
10	Ph	Н	Me	CN	35	trace		
11	Н	Н	Me	CN	35	trace		
12	Me	Н	Ме	CO <sub>2</sub> Me	59	trace		
<u>13</u>	Ph	Н	Me	co <sub>2</sub> Me	35	15		
14	Н	н	Me	CO <sub>2</sub> Me	63	trace		

<sup>&</sup>lt;sup>a</sup>Isolated yield.  $^{b}$ The yield of 3-phenylquinoxalin-2-one (15)

A reasonable mechanism for the formation of 2-substituted quinoxalines  $(\underline{6}-\underline{14})$  is presented in Scheme, which the thietane  $(\underline{18})$ , formed initially by the photochemical [2+2]cycloaddition of thiocarbonyl to olefin, 9 undergoes aromatization to give the final products  $(\underline{6}-\underline{14})$ . This ready mode of carbon-carbon bond formation described here would provide a convenient method for the preparation of the 2-substituted quinoxalines.

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- 8.  $^{13}$ C-n.m.r. spectrum of  $\underline{3a}$ ;  $\delta_{\rm H}$  (DMSO-D<sub>6</sub>) 24.7(q), 115.7(d), 125.6(d), 128.0(d), 130.1(d), 131.6(s), 134.9(s), 161.4(s), and 175.1(s).
- 9. The regiochemistry of photocycloaddition is in accord with the previously published works in thioamide  $^{3a}$  and thioimide  $^{4b,f}$  photochemistry.

Received, 17th September, 1984