WATER-INDUCED AZOMETHINE YLIDE GENERATION FROM N-(SILYLMETHYL)THIO-IMIDATES, SYNTHETIC EQUIVALENTS OF NONSTABILIZED NITRILE YLIDES

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Abstract — N-(Silylmethyl)thioimidates undergo a water-induced desilylation generating azomethine ylide 1,3-dipoles which cycloadd to a variety of electron-deficient olefins providing 1- or 2-pyrrolines after the elimination of thiol, indicating that the thio-imidates are synthetic equivalents of nonstabilized nitrile ylides.

N-(Silylmethyl)imines owe their importance in heterocyclic synthesis much to the applications toward the generation of azomethine ylide 1,3-dipoles under mild conditions. The general route into the ylides consists of the initial quaternarization on the nitrogen and the subsequent desilylation with a fluoride ion^1 . A variety of azomethine ylides have been prepared according to this method and found wide applications in organic synthesis². A rather exceptional generating method discovered in our laboratory involves the action of water to N-(silylmethyl)imines in polar solvents such as HMPA or DMF³. Cycloadditions of the ylide \underline{A} thus formed to olefins provide pyrrolidines as shown below.

TMS
$$\stackrel{}{\nearrow}_{N}$$
 $\stackrel{}{\nearrow}_{Ph}$ $\stackrel{}{\longrightarrow}_{H_{2}O}$ $\stackrel{}{\longrightarrow}_{N}$ $\stackrel{}{\longrightarrow}_{H_{2}O}$ $\stackrel{}{\longrightarrow}_{N}$ $\stackrel{}{\longrightarrow}_{H_{2}O}$ $\stackrel{}{\longrightarrow}_$

The present research is aiming at an extension of our method to an N-(silylmethyl)-imine \underline{B} bearing a leaving group at the imine carbon. The cycloadditions of ylide \underline{C} to olefinic dipolarophiles, followed by elimination of the leaving group, would lead to 1-pyrroline derivatives. By this sequence, the imine \underline{B} will become the

synthetic equivalent of a nonstabilized nitrile ylide \underline{D} . We have chosen some N-(silylmethyl)thioimidates as candidates for B.

Scheme 1.

The thioimidates $\underline{1-4}$ were readily available in one-pot reactions of trimethylsilylmethyl isothiocyanate⁴ with Grignard reagents and then with methyl iodide or benzoyl chloride as shown in Scheme 1 ($\underline{1}$: 83%; $\underline{2}$: 71%; $\underline{3}$: 71%; $\underline{4}$: 76%). This route might allow the introduction of a variety of substituents at the imine carbon of thioimidates⁵.

The reaction of $\underline{1}$ with N-methylmaleimide $\underline{5}$ in the presence of an equivalent amount of water in HMPA at room temperature (method A) gave an excellent yield of the 1-pyrroline $\underline{10}$ (Scheme 2 and Table 1)⁶. Although some initial cycloadduct $\underline{\underline{E}}$ was detected in the crude reaction mixture, it suffered from the ready and complete

Table 1. Cycloadditions of 1-4 to Symmetrically Substituted Olefins.

N-Silylmethyl thioimidate	Olefin	Reaction co	nditions ^{a)} Time/h	Product	Yield/% ^{c)}
<u>1</u>	<u>5</u>	A	24	10	98
		В	16	<u>10</u>	25
		С	16	<u>10</u>	44
<u>2</u>	<u>5</u>	D	36	<u>10</u>	70
<u>1</u>	<u>6</u>	A	24	<u>11</u>	80
		В	24	<u>11</u>	38
<u>1</u>	<u>7</u>	A	24	<u>11</u>	62
<u>1</u>	8	Α	24	<u>12</u>	70
<u>1</u>	<u>9</u>	Α	24	<u>13</u>	77
<u>3</u>	<u>5</u>	Α	24	<u>14</u>	15
<u>4</u>	<u>5</u>	Α	24	<u>15</u>	10
<u>3</u>	<u>6</u>	E	24	<u>16</u>	23

a) Each equivalent amounts of the thioimidates and olefins were used. b) A: H₂O (equiv) in HMPA at rt; B: TMSOTf-CsF (each equiv) in HMPA at 60 °C; C: PhCOF (equiv) in MeCN at 60 °C; D: AcOH (equiv) in HMPA at 60 °C; E: AcOH (equiv) in HMPA at rt. c) All isolated yields.

elimination of methanethiol during the chromatographic procedure over silica gel providing $\underline{10}$. The conventional generating method using TMSOTf-CsF (method B) or PhCOF (method C) resulted in poorer yield of $\underline{10}$. The thioimidate $\underline{2}$ bearing benzoylthic moiety as a leaving group showed no advantage over the methylthio-substituted thioimidate 1.

It is surprising that the same trans-1-pyrroline $\underline{11}$ was the only product either in the reaction of $\underline{1}$ with dimethyl fumarate $\underline{6}$ or with dimethyl maleate $\underline{7}$, because the cycloaddition of nonstabilized azomethine ylide to olefins is known to be highly stereospecific³. The initial cycloadducts \underline{F} and \underline{G} must have been trans and cis, respectively. The thiol elimination at the 2-3 bond leads to the same 2-pyrroline \underline{H} . The sterically hindered enamine of \underline{H} undergoes a hydrogen migration forming the sterically more favored trans-1-pyrroline $\underline{11}$.

Scheme 2.

Similar reactions of $\underline{1}$ with di(t-butyl) fumarate $\underline{8}$ and fumaronitrile $\underline{9}$ gave the 1-pyrroline $\underline{12}$ and 2-pyrroline $\underline{13}$, respectively (Scheme 2 and Table 1). The cyano moiety at the 3-position of $\underline{13}$ is small in size so that it tolerates the fixation of double bond at the 2-position.

Although the thioimidates $\underline{3}$ and $\underline{4}$ carrying an alkyl substituent at the imine carbon underwent similar cycloadditions to $\underline{5}$ and $\underline{6}$, the yields of $\underline{14}$ - $\underline{16}$ were unfortunately

quite poor. This low yield might be partly due to the lack of ylide-stabilizing ability of the alkyl substituent. In these cases, the major reaction path was the polymerization of olefins employed.

The reactions of $\underline{1}$ with a variety of unsymmetrically substituted olefins $\underline{17-21}$ produced the regionelective 1-pyrrolines 22-26 in good yields (Table 2).

Table 2. Cycloadditions of $\underline{1}$ to Unsymmetrically Substituted Olefins.^{a)}

a) Conditions: ${\rm H_2O}$ (equiv) in HMPA at rt for 24 h. b) All isolated yields.

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- 5. Other substituents such as alkyl, aryl and vinyl moieties can be successfully introduced (unpublished results).
- 6. All new compounds reported herein gave satisfactory spectral and analytical results.

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