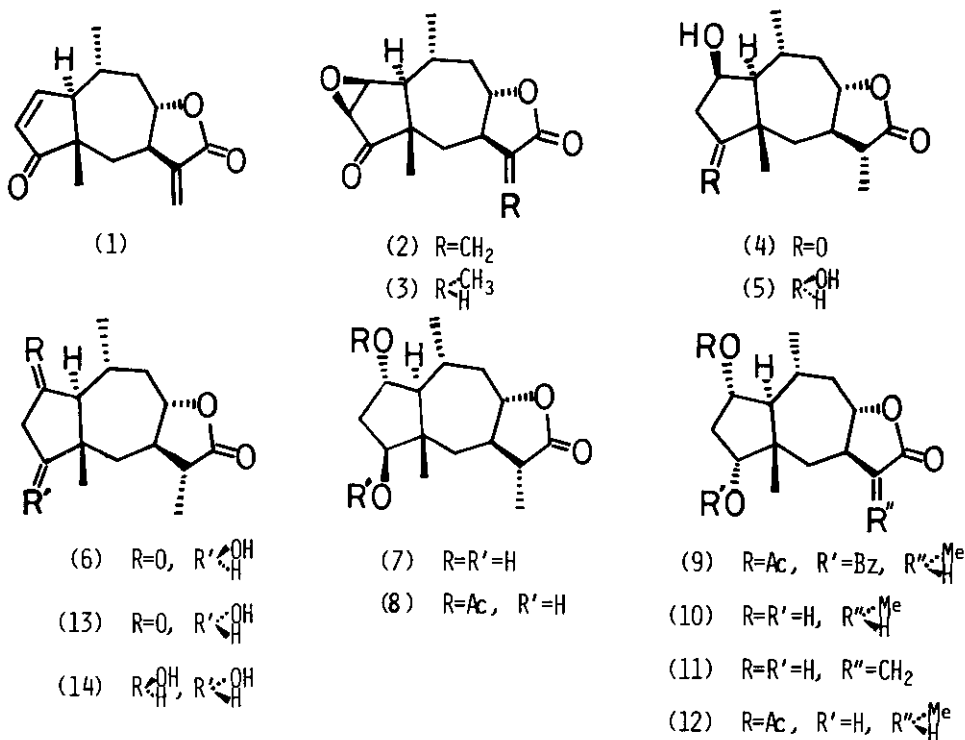


CHEMICAL TRANSFORMATION OF AROMATICIN INTO PULCHELLIN—A FORMAL
SYNTHESIS OF PULCHELLIDINE

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Abstract—A chemical transformation of aromaticin into
pulchellin implying the formal synthesis of pulchellidine was
achieved. The stereochemistry of the diol isomers of dihydro-
pulchellin was also revealed.

In the course of our studies on plant-derived bioactive sesquiterpenoids, besides eudesmanolides (pulchellins B, C, D, E and F¹) and guaianolides (gaillardin² and neogaillardin³), pulchellin,^{4,5} neopulchellin⁶ and pulchelloid A,⁷ B⁷ and C⁸ as well as pulchellidine,^{4,5,9} neopulchellidine,⁶ and pulchellon,¹⁰ were isolated as the pseudoguaianolide principles from *Gaillardia pulchella* (Compositae). Concerning synthesis of pseudoguaianolides other than ambrosanolides,¹¹ synthesis of only a few helenanolides such as helenalin,¹² aromatin¹³ and aromaticin¹⁴ has been reported, because of their interesting structural features and cytotoxic properties.¹⁵ These molecules involve a cyclopentenone moiety in the pseudoguaiane skeleton with only five or six chiral centres. From the point of view of synthesis in the preliminary study on bioactive and highly oxygenated helenanolides having many asymmetries, we now wish to report the transformation of a helenanolide, aromaticin, into pulchellin with additional two chiral centres, which implies the formal synthesis of pulchellin and pulchellidine. The starting aromaticin was provided by the facile conversion of pulchellin (1) isolated from the cultivar of *G. pulchella*.¹⁶ 2-Acetylpulchellin¹⁷ separated from 2,4-diacetylpulchellin⁴ by silica gel chromatography was submitted to Jones



oxidation and the intermediate 4-keto-2 α -acetate was heated in pyridine, giving a good yield of aromaticin (1).¹⁸ Epoxidation of 1 with alkaline hydrogen peroxide in tetrahydrofuran (THF) and methanol at -25~-15°C regio- and stereospecifically afforded 2 β ,3 β -epoxyketone (2) [mp 184-187°C, ir ν (KBr) 1760, 1649 (α -methylene- γ -lactone), 1738 cm⁻¹ (cyclopentanone)] in almost quantitative yield.

The stereospecific β -site epoxidation of the cyclopentenone ring in tenulin and its related sesquiterpenes was proposed by Toromanoff on the basis of the dynamic stereochemistry of the cyclopentenone, using a torsion angle notation.¹⁹ The configuration of epoxide (2) has now been confirmed by the X-ray analysis of dihydroepoxide (3) described below, as shown in Fig. 1. Crystal data: C₁₅H₂₀O₄, MW 264.3, monoclinic, space group, P2₁, a = 20.923(1), B = 6.912(3), c = 9.889(1), β = 98.95(5), U = 1415 Å³, z = 4, D_{calc} = 1.243 g cm⁻³, final R value = 0.054 for 2362 reflections.

Since aluminium amalgam (Al-Hg) reduction of 2 furnished a complex mixture of unseparable products, dihydroepoxyketone (3) [mp 171-172°C, ir ν (KBr) 1756 (γ -lactone), 1741 cm⁻¹ (epoxyketone)] was then prepared by sodium borohydride (NBH) reduction and subsequent Jones oxidation. Careful treatment of 3 with Al-Hg

(0°C, 3.5 h, aqueous THF) gave unstable 4-keto-2 β -ol (4) in a reasonable yield. In the 400 MHz ^1H -nmr spectrum (CDCl_3/TMS), coupling constants for H-2 α /H-1 α ($J = 4.0$ Hz), H-2 α /H-3 α ($J = 4.0$ Hz) and H-2 α /H-3 β ($J = 1.4$ Hz) indicated that the hydroxy group at C-2 must be in the β -configuration. The stereospecific NBH reduction of 4 afforded quantitatively 2 β ,4 β -diol (5) [mp 171-172°C, ir ν (KBr) 3450, 3400 (hydroxyl groups), 1747 cm^{-1} (γ -lactone)]. The regiospecific oxidation with pyridinium dichromate (PDC) and subsequent NBH reduction of diol (5) afforded 2 α ,4 β -diol (7) together with the starting 2 β ,4 β -diol (5) in the approximate ratio of 3:1. After the separation of the required diol (7) from 5 by silica gel chromatography, 7 was acetylated as usual producing 2 α -acetoxy-4 β -ol (8) [mp 160-162°C] together with 2 α -hydroxy-4 β -acetate and 2 α ,4 β -diacetate in the approximate ratio of 2:2:1. Monoacetate (8) was submitted to epimerization of the hydroxy group at C-4 using diethyl azodicarboxylate, benzoic acid and triphenyl phosphine (THF, r.t., 4.5 h) yielding 2 α -acetyl-4 α -benzoate (9) [mp 174-175°C, ir ν (KBr) 1709 (benzoate), 1736 (acetate), 1769 cm^{-1} (γ -lactone)] in a reasonable yield. In the ^1H -nmr spectrum of 9, the proton at C-4 ($\delta = 5.0$ ppm, d, $J = 6$ Hz) showed the characteristic splitting pattern of the methine for pulchellin, neopulchellin and their various derivatives bearing the 4 α -acyl group.^{4,5,6,7,8} Alkaline hydrolysis of 9 readily provided the required 2 α ,4 α -diol, i.e. dihydropulchellin (10) [mp 143-144.5°C].⁴ Finally, sequential treatment of 10 by phenylselenylation (excess LDA, $(\text{PhSe})_2/\text{HMPA}$), followed by oxidative elimination of the 11 β -phenylselenyl intermediate with hydrogen peroxide afforded a 40% yield of pulchellin (11).⁴ Furthermore, in order to confirm the configuration of the hydroxyl groups in the synthetic intermediate 2 β ,4 β -diol (5), four possible diol stereoisomers other than 10 were prepared in the following manners. NBH reduction of 2-keto-4 α -ol (13), which is made available by the regioselective PDC oxidation of dihydropulchellin⁴ yielded 2 α ,4 α -diol (10) and 2 β ,4 α -diol (14) at a ratio of 8:5. On the other hand, 2-acetyldihydropulchellin (12) [mp 161.5-162.5°C] was prepared by the usual acetylation of 10 together with diacetyldihydropulchellin⁴ at the ratio of 7:3. Jones oxidation of 12 and subsequent NBH reduction readily yielded 2 α -acetoxy-4 β -ol (8), which was further converted by hydrolysis (20% KOH/dioxane) to 2 α ,4 β -diol (7). The regiospecific PDC oxidation of 7 (r.t., 12 h) and subsequent NBH reduction of 2-keto-4 β -ol (6) gave 2 α ,4 β - (7) and 2 β ,4 β -diol (5) in the same way as mentioned above.

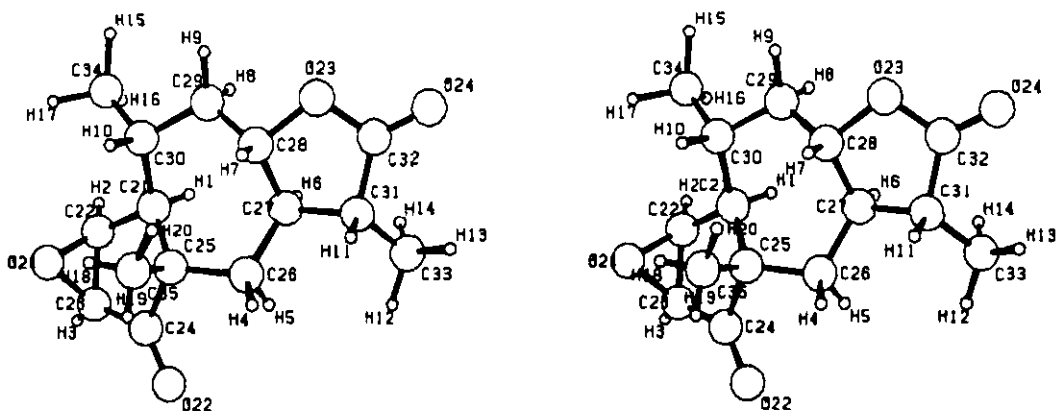


Fig. 1. Stereoview of 2 β ,3 β -epoxy-11 β H-dihydropulchellin (3) (B molecule)

Spectroscopic properties of the three diols thus obtained, i.e., 5, 7 and 10 were all in accord with those of the corresponding isomers described above. The transformation of aromaticin into pulchellin has now been accomplished. This means the formal synthesis not only of 11, but also of pulchellidine,^{4,9} which is derived stereospecifically by the Michael addition of piperidine to 11.¹⁰ The similar conversion of aromatin into neopulchellin or neopulchellidine is under investigation.

ACKNOWLEDGEMENTS

One of the authors (S.I.) acknowledges a research grant from the Miyata Prize partly used for this work. The authors are grateful to Mr. H. Yamanaka and Miss S. Takei of the Joint Laboratory, School of Medicine, Keio University, for measurement of MS and IR/UV spectra, respectively.

REFERENCES

- 1) W. Herz and S. Inayama, *Tetrahedron*, 1964, 20, 341. H. Yoshioka, T. J. Mabry, N. Dennis and W. Herz, *J. Org. Chem.*, 1970, 35, 627, and references loc. cit.
- 2) T. A. Dullforce, G. A. Sim, D. N. White, J. E. Kelsey and S. M. Kupchan, *Tetrahedron Lett.*, 1969, 973, and references loc. cit.
- 3) S. Inayama, T. Kawamata and M. Yanagita, *Phytochemistry*, 1973, 12, 1743.
- 4) M. Yanagita, S. Inayama and T. Kawamata, *Tetrahedron Lett.*, 1970, 131, and references loc. cit.

- 5) T. Sekita, S. Inayama and Y. Iitaka, Acta Crystallogr., 1971, 1327, 877 (Cf. ibid., 1970, 135).
- 6) M. Yanagita, S. Inayama and T. Kawamata, ibid., 1970, 3007. S. Inayama, K. Harimaya, H. Hori, T. Kawamata, T. Ohkura, H. Nakamura and Y. Iitaka, Heterocycles, 1982, 19, 1801.
- 7) S. Inayama, K. Harimaya, T. Ohkura and T. Kawamata, Heterocycles, 1982, 17, 219.
- 8) S. Inayama, K. Harimaya, H. Hori, T. Kawamata, T. Ohkura, I. Miura and Y. Iitaka, ibid., 1983, 20, 1501.
- 9) M. Yanagita, S. Inayama, T. Kawamata, T. Ohkura and W. Herz, Tetrahedron Lett., 1969, 2073. T. Kawamata and S. Inayama, Chem. Pharm. Bull., 1971, 19, 643.
- 10) S. Inayama, T. Kawamata and T. Ohkura, Chem. Pharm. Bull., 1975, 23, 2998. Idem, Tetrahedron Lett., 1978, 1557.
- 11) C. H. Heathcock, S. L. Graham, M. C. Pirrung, F. Plavic and C. T. White, "Total Synthesis of Natural Products", Vol. 5, J. W. Apsimon Ed., Willey, New York, 1982.
- 12) Y. Ohfuné, P. A. Grieco, C. -L, Wang, G. Majetich, J. Am. Chem. Soc., 1978, 100, 5946. M. R. Roberts and R. H. Schlessinger, ibid., 1979, 101, 7626.
- 13) P. T. Lansbury, D. G. Hangauer and J. P. Vacca, J. Am. Chem. Soc., 1980, 102, 3964.
- 14) P. T. Lansbury and D. G. Hangauer, Jr., Tetrahedron, 1981, 37, 371.
- 15) S. M. Kupchan, M. A. Eakin and A. M. Thomas, J. Med. Chem., 1971, 14, 1147.
- 16) S. Inayama, K. Harimaya, H. Hori, T. Ohkura, T. Kawamata, M. Hikichi and T. Yokokura, Chem. Pharm. Bull., 1984, 32, 1135.
- 17) K. Aota, C. N. Coughlan, M. T. Emerson, W. Herz, S. Inayama and Mazhar-ul-Haque, J. Org. Chem., 1970, 35, 1448.
- 18) J. Romo, P. Joseph-Nathan and G. Diaz, Tetrahedron, 1964, 20, 79.
- 19) E. Toromanoff, Tetrahedron, 1980, 36, 2809.

Received, 25th October, 1984