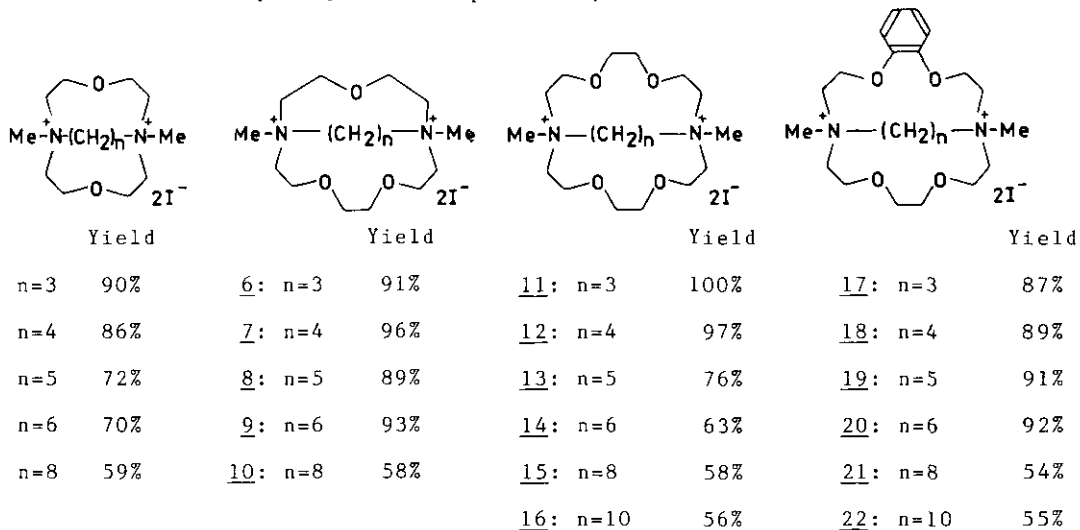


HIGH-PRESSURE SYNTHESIS OF CRYPTANDS WITH ALIPHATIC BRIDGING UNITS

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Abstract — *N,N*-Dimethyl diazacoronands react with α,ω -diiodoalkanes under high pressure (10 kbar) to give the bis-quaternary salts which are transformed into cryptands via demethylation with triphenylphosphine.

We have already published syntheses of cryptands either simple¹ and more elaborated^{2,3} by high-pressure double quaternization of *N,N*-dimethyl diazacoronands followed by demethylation. Now we report on the influence of the length of the bridging unit on high-pressure double quaternization. We selected four diazacoronands: (12)- N_2O_2 ,⁴ (15)- N_2O_3 ,⁵ (18)- N_2O_4 ,⁵ and B(18)- N_2O_4 ,⁵ and a range of α,ω -diiodoalkanes.⁶ Equimolar amounts of substrates (ca. 0.08M solution in acetone) were exposed to the pressure of 10 kbar for 20h at 25°C, in the apparatus described earlier.⁷ In all cases the white precipitate of quaternary salt was observed.



The bis-quaternary salts were demethylated with triphenylphosphine in boiling DMF⁸ to give the cryptands with aliphatic bridging units.⁹ The very high yield and selectivity of the reactions presented here makes this method very attractive for the synthesis of modified cryptands. However, the yield of double-quaternization reactions depends significantly on the bridging unit length, therefore the method presented herein possesses a practical value for the aliphatic bridges having up to six methylene groups.

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4. (12)-N₂O₂ was prepared from N-p-toluenesulphonyl-N-benzyl-1,7-diaza-4,10-dioxacyclododecane by detosylation with LiAlH₄, removal of benzyl group by hydrogenation on Pd/C, and methylation with HCO₂H/HCHO.
5. Other diazacoronands were prepared by routine methods, for example: L.C.Hodgkinson, M.R.Johnson, S.J.Leigh, N.Spencer, I.O.Sutherland, and R.F.Newton, J.Chem.Soc., Perkin I, 2193 (1979).
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9. Typically, compound **8** was demethylated with triphenylphosphine in boiling DMF for 5h to give (2.1.C₅)cryptand (80% yield). MS (17 eV) M⁺ 286(2); ¹H NMR, CDCl₃, δ (ppm): 3.9-3.4 (m, 12H, OCH₂), 2.9-2.4 (m, 12H, NCH₂), 1.8-1.4 (m, 6H, C-CH₂-C).

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