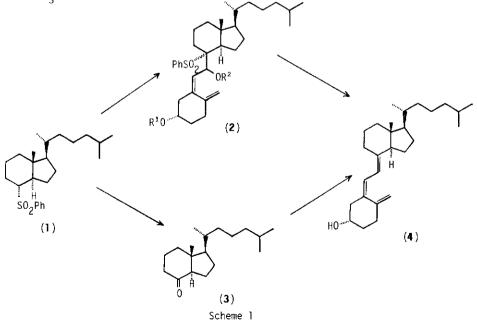
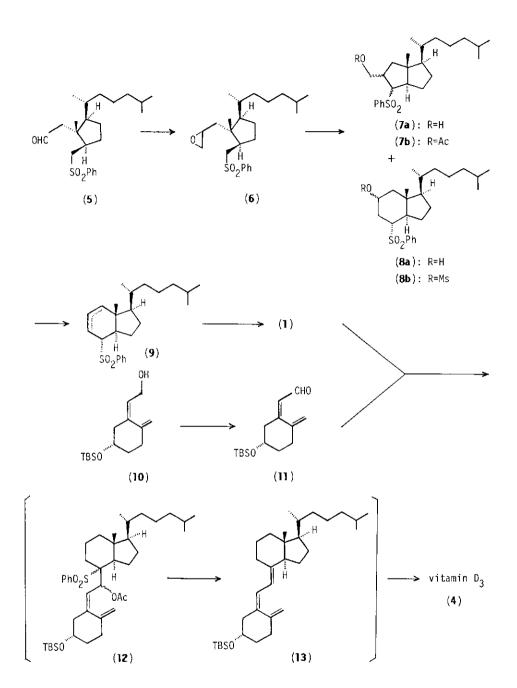
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<u>Abstract</u> — An intramolecular alkylation of the phenylsulfonyl epoxide (6), which was readily obtained from the aldehyde (5), gave a separable mixture of the alcohols (7a) and (8a). The alcohol (8a) was then dehydrated via the corresponding mesylate (8b) to afford the olefin (9) which on hydrogenation furnished (+)-8 α -phenylsulfonyl-des-AB-cholestane (1). Further this product was converted into vitamin D₂ (4).

In the preceeding paper,¹ we described a first total synthesis of (+)-8 α -phenylsulfonyl-des-AB-cholestane (1) which could be a potential intermediate for vitamin D₃ (4) either by Julia's synthesis^{2,3} via β -hydroxyphenylsulfonyl derivative (2) or other types of reaction via Grundmann's ketone (3). The studies on the synthesis of vitamin D₃ (4) via 1 through 2 have been limited in contrast with that⁴ via 3





Scheme 2

partially because of the difficulty of obtaining 1. So, we have undertaken the studies on exploring the facile synthesis of $(+)-8\alpha$ -phenylsulfonyl-des-AB-choles-tane (1) and here wish to report its alternative synthesis and its conversion into vitamin D₂.

Epoxidation (Me₃SI, n-BuLi, THF, 0°C, 1.5 h) of the aldehyde $(5)^{1}$ gave the oxirane (6) $[m/z; 265 (M^+-SO_Ph)]$ which was subjected to the intramolecular cyclization (LDA, THF, -78°C, 30 min) giving the alcohols (8a) $[m/z; 265 (M^+-SO_2Ph)]$ and (7a) $[m/z; 265 (M^+-SO_2Ph)]$ in 50 % and 30 % yields respectively. In the ¹H-NMR spectrum, the signals observed at 3.55 - 3.76 ppm as multiplet due to methylene protons of hydroxymethylene moiety in the compound (7a) was shifted to 3.80 - 4.21 ppm in its acetoxy derivative (7b). The α configuration of a phenylsulfonyl group at C-8 in (8a) was deduced from the coupling constants (3.02 ppm, d,d,d, J=12, 12, 4 Hz) of C-8 H in the NMR spectrum. This was eventually confirmed by a conversion of (8a) into (1). Then, the compound (8a) was converted into the target compound (1) in 70 % overall yield via the mesylate (8b) and olefins (9) by a succesive treatment (MsCl, pyridine, 0 ℃, 1 h; LiBr, Li₂CO₃, DMF, 150°C, 4 h; H₂, Pd-C, AcOEt, room temperature, 10 h). The compound (1) thus obtained was identical with the authentic sample prepared previously¹ in all aspects including optical rotation. The metallated sulfone (1) was condensed (LDA, THF, -78° C) with the ring A component (11), obtained by oxidation (MnO2, THF, room temperature) of corresponding allvl alcohol (10).⁵ Treatment of the reaction mixture with acetyl chloride gave a mixture of diastereoisomeric β -acetoxy-sulfones (12). This was reduced (5 % Na-Hg, MeOH-THF, -20 $^{\circ}C \sim$ room temperature, 7 h) to the triene (13), whose desilylation (n- $Bu_A NF$, THF, room temperature, 2 h) gave vitamin D₃ (4) in 51 % overall yield. The 3,5-dinitrobenzoate of the synthetic vitamin $D_3^{(mp)}$ (mp) 129 - 130°C, $[\alpha]_D^{20}$ +95.9°) was identical with authentic vitamin D_3 3,5-dinitrobenzoate (lit.⁶, mp 128 - 129 c, lit.⁷, $[\alpha]_{D}^{20}$ +97% in mp and spectral (IR and ¹H-NMR) comparisons.

Thus, we could disclose an alternative route for the synthesis of $(+)-8\alpha$ -phenyl-sulfonyl-des-AB-cholestane (1). Furthermore, such the compound (1) was converted into vitamin D₃ (4) by Juria's olefin synthesis.

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