SYNTHESES OF ELLIPTICINE AND RELATED PYRIDOCARBAZOLE ALKALOIDS - A REVIEW

Gordon W. Gribble* and Mark G. Saulnier

Department of Chemistry, Dartmouth College, Hanover, New Hampshire 03755, USA

<u>Abstract</u> - Synthetic approaches to the anticancer alkaloid ellipticine and related pyridocarbazoles covering the period 1977 through November 1984 are reviewed.

1. Introduction

The <u>Ochrosia</u>, <u>Aspidosperma</u>, <u>Tabernaemontana</u>, and <u>Strychnos</u> plant alkaloids¹ ellipticine (<u>la</u>) (5,11-dimethyl-6H-pyrido[4,3-<u>b</u>]carbazole), 9-methoxyellipticine (<u>lb</u>), and olivacine (<u>2</u>) exhibit pronounced anticancer activity in several animal and human tumor systems.² Recently, a derivative of 9-hydroxyellipticine (<u>lc</u>), 2methyl-9-hydroxyellipticinium acetate (<u>3</u>) ("elliptinium"), was commercialized³ for clinical use and is effective in the treatment of myleoblastic leukemia, advanced breast cancer, and other solid tumors.^{2a,4}



Since the original isolation⁵ of these alkaloids and the initial discovery⁶ of their anticancer activity, many synthetic approaches to the pyrido $[4,3-\underline{b}]$ carbazole ring system have been described. In 1977, Sainsbury⁷ published an extensive review on the syntheses of this ring system, detailing about 15 different synthetic routes to pyrido $[4,3-\underline{b}]$ carbazoles. A more recent review⁸, although considerably less detailed, updates the synthetic approaches to ellipticine to the middle of 1980 and

focuses on computer-generated strategies for the construction of this ring system. Following the completion of a draft of the present paper, an article appeared reviewing this field "from 1977 to December 1982."⁹ Since this latter review is incomplete within the stated time-frame and because several important papers have appeared subsequently, we have completed the present review.

The primary purpose of this review is to update those significant synthetic improvements on existing strategies and to summarize the several new strategies for the synthesis of ellipticine and related pyridocarbazoles that have appeared in the literature since 1977. We believe that coverage is complete through November 1984.

2. Classification of Strategies

All known synthetic routes to the pyrido[4,3-<u>b</u>]carbazole ring system may be assembled into eight main strategies, I-VIII, for which the key bond formations are indicated.^{8b} We feel that this classification is more useful than the original one,⁷ which focused on which rings were generated rather than which bonds were formed.



3. Strategy I

The bond forming strategy represented as I was originally employed by Stillwell¹⁰ in 1964 (Scheme 1), as reviewed by Sainsbury.⁷ This synthesis of ellipticine (<u>1a</u>) involved a Fischer-indole cyclization (82%) of decahydro-isoquinol-6-one <u>4</u>, but, since the final dehydrogenation step proceeded in only 0.3% yield, this approach was thought to be of no practical value.

Scheme 1



However, a French team¹¹ has recently used the Stillwell approach to synthesize 9-methoxy-ll-demethylellipticine (9) and 9-hydroxy-ll-demethylellipticine (10) (Scheme 2), compounds of biological interest. Thus, enone 5, prepared by Robinson annulation of ethyl vinyl ketone and the pyrrolidine enamine of 1-methyl-4-piperidone (76%), was hydrogenated to isoquinolone 6 (93%). Fisher indolization of 6 with 4-methoxyphenylhydrazine (7) afforded indole 8 (54%) with only very little of the undesired isomeric indolenine <u>11</u> being produced. Dehydrogenation of <u>8</u> gave 9-methoxy-ll-demethylellipticine (9) (26%), which could be demethylated with hot pyridine hydrochloride to afford the hydroxy derivative <u>10</u> (71%). A similar sequence starting with 1-benzyl-4-piperidone led to <u>9</u> in the same overall yield (10%).



Another recent ellipticine synthesis which encompasses Strategy I was reported by Miller.¹² In his original work (Scheme 3), Miller^{12a} utilized a Goldbergmodified Ullmann coupling reaction between acetanilide and bromoisoquinoline <u>12</u> (prepared in six steps (69%) from 2-bromo-1,4-dimethylbenzene) to give diarylamide <u>13a</u> (70%). Acid hydrolysis of <u>13a</u> gave diarylamine <u>13b</u> (90%). Unfortunately, the anticipated photochemical cyclization of <u>13b</u> failed to yield any ellipticine (<u>1a</u>), and, furthermore, a control experiment revealed that <u>1a</u> decomposed under the photolysis conditions. However, the cyclization was eventually accomplished with palladium acetate under acidic conditions to give <u>la</u> in 15-25% yield (46% based on unrecovered <u>13b</u>).





More recently, Miller^{12b} has described an improvement on his original procedure^{12a} (Scheme 4). Thus, Goldberg coupling of 2-nitroanilines <u>14a</u> and <u>14b</u> with <u>12</u> gave the corresponding diarylamines <u>15a</u> (54%) and <u>15b</u> (53%). Reduction of the nitro group and diazotization of each amine afforded the desired benzotriazoles <u>16a</u> (97%) and <u>16b</u> (94%). Pyrolytic decomposition of <u>16a</u> and <u>16b</u> in a flow system gave <u>1a</u> (69%) and 9-methoxyellipticine (<u>1b</u>) (62%), respectively. Lower yields of <u>1a</u> were realized when <u>16a</u> was heated (220°C) in polyphosphoric acid (16%) or photolyzed in methanol (33%).

Scheme 4

4. Strategy II

This approach is obviously similar to Strategy I in that both involve prior construction of an isoquinoline unit and ring closure as indicated.

Bisagni and co-workers¹³⁻¹⁷ have used Strategy II extensively in their syntheses of the potent anticancer azaellipticine derivatives such as 6,11dimethyl-5H-pyrido[3',4':4,5]pyrrolo[2,3-<u>a</u>]isoquinoline (<u>21</u>) (Scheme 5). Treatment^{13a} of 4-chloro-3-nitropyridine (<u>17</u>) with 6-amino-5,8-dimethylisoquinoline (<u>18</u>) and HCl gave the expected product <u>19</u> (29%). Catalytic hydrogenation of <u>19</u> (94%) followed by diazotization of the resulting amine furnished triazole <u>20</u> (84%). When <u>20</u> was heated in paraffin at 320 - 340°C the desired azaellipticine 21

was obtained in 41% yield. Several other approaches to <u>21</u> were unsuccessful.^{13a} Scheme 5

The same group^{13b} synthesized the 10-substituted azaellipticine derivatives <u>26</u> in analogous fashion (Scheme 6). Thus, <u>17</u> was joined to 6-amino-5methylisoguinolin-1(2H)-one (<u>22</u>) (prepared in six steps (13%) from 3-amino-2methylbenzonitrile) to afford <u>23</u> (72%). The usual sequence of catalytic hydrogenation (84%), diazotization (84%), and thermolysis (58%) gave lactam <u>24</u>. Treatment of <u>24</u> with phosphorus oxychloride led to the key chloro intermediate <u>25</u> (54%), which, when heated with various [(dialkylamino)alkyl]amines, gave the target compounds <u>26</u> (38-75%).

Screening results indicate that the presence of a [(dialkylamino)alkyl]amino side chain at the 1-position of ellipticine (10-position of <u>21</u>) greatly increases the anticancer activity. For example, <u>26c</u> is much more active than <u>21</u> in the L-1210 mouse leukemia system.^{14,15}

Using a slightly modified synthetic protocol (Scheme 7), the Bisagni group¹⁶ prepared the 6,11-dimethyl analogues of <u>26</u>. Thus, triazolopyridine <u>27</u> was converted into the chloro derivative <u>28</u> upon treatment with phosphorus oxychloride (85%). Although thermal cyclization of <u>28</u> to <u>29</u> failed, this ring closure was

achieved upon photolysis (35%). The key chloroazaellipticine <u>29</u> was converted into the aminoalkyl derivatives <u>30</u> in the usual manner.

It was also found that this route could be applied to the synthesis of the monomethyl chloro intermediate $\underline{25}$, thereby circumventing the thermal cyclization (cf., Scheme 6).

Finally, the Bisagni group¹⁷ has utilized this same Type II strategy to prepare the isomeric 7-azaellipticines (<u>33</u>; lH-pyrido[3',2';4,5]pyrrolo[2,3g]isoquinolines) (Scheme 8). Coupling of 2-chloro-3-nitropyridine (<u>31</u>) with the isoguinolinones <u>22</u> or <u>32</u>, and following the reaction sequence outlined in Scheme 6 produced azaellipticines <u>33</u>. These compounds proved to be much less active than their 9-azaellipticine analogues (<u>26</u>, <u>30</u>).

Scheme 8

5. Strategy III

This strategy has been the most prevalent in the design of ellipticine syntheses and was first employed by Cranwell and Saxton¹⁸ in one of the most efficient constructions of 6H-pyrido[4,3-b]carbazoles. This approach often involves the condensation of an indole and 2,5-hexanedione to give a 1,4dimethylcarbazole (<u>34</u>) (Scheme 9). Vilsmeier formylation generally gives the 3carboxaldehyde as the major product. Condensation with aminoacetaldehyde acetal gives an imine acetal (<u>35</u>), which, using various methods, can be cyclized to the ellipticine ring system.

It is this approach that is employed in the commercial manufacture of $\underline{3}$ (Institut Pasteur brochure).

Recent synthetic advances that use a Type III strategy are numerous. Bisagni and his colleagues¹⁹ have patterned the preparation of various 1-substituted ellipticines (Scheme 10) after the pioneering olivacine (2) synthesis of Wenkert and Dave.²⁰ Thus, treatment of aryldiazonium salts 36 with morpholine enamines 37 qave the corresponding phenylhydrazones 38 (76-85%). Fischer indolization of 38 afforded indole ketones 39 (30-60%) which were acylated with ethyl formate and sodium hydride to give the hydroxymethylene derivatives 40 (78-95%). Etherification of 40 with isopropyl iodide followed by treatment of the isopropyl ether with excess methyllithium gave, after hydrolysis, the dihydrocarbazole aldehydes 41 (46-81% from 40). Aromatization of 41 was accomplished with manganese dioxide, but proceeded poorly with palladium, to give the 1-methyl-2-formylcarbazoles 42 (50-88%). A three-step sequence was then used to fashion the pyridine ring. Knoevenagel condensation of 42 with malonic acid gave the E-acrylic acids (66-90%) which were converted to the respective acyl azides via a mixed anhydride. Thermal cyclization of the acyl azides furnished the pyrido[4,3-b] carbazol-1(2H)-ones 43 (30-72%). The usual formation of the chloro derivatives 44 (25-88%) and aminolysis with 3-diethylaminopropylamine gave the target ellipticines 45a-d (23-78%). Catalytic hydrogenation of the two benzyloxy derivatives gave the hydroxy analogues 45e (85%) and 45f (62%).

Scheme 10

The Bisagni team has reported^{19,21} that C-1 substitution in ellipticine by the 3-(diethylamino)propylamino side chain significantly increases the anticancer activity in the L-1210 leukemia system. Furthermore, these recent²¹ testing results indicate that <u>45c</u> and <u>26c</u> exhibit both high <u>in vitro</u> cytotoxicity and <u>in vivo</u> anticancer properties. Further side chain modifications show no significant increase in biological activity over those expressed by <u>45c</u> and <u>26c</u>. The role of the [(dialkylamino)alkyl]amino side chains on the enhanced anticancer properties of these ellipticines is unknown but could involve a favorable interaction between the amino side chain and the phosphate backbone of the DNA, thus enhancing intercalation.

Synthetic investigations by Viel²² also employ a Type III approach to ellipticine derivatives. In their earlier work^{22a} (Scheme 11), a Borsche²³ indolization reaction between 4-methoxyphenylhydrazine (7) and cyclohexanone <u>46</u> or <u>47</u> gave the expected tetrahydrocarbazoles <u>48</u> (57, 67%). A Campbell²⁴ aromatization of <u>48</u> gave carbazoles <u>49</u> (73, 78%) which were formylated to give aldehydes <u>50</u> (62, 68%). Condensation with aminoacetaldehyde diethyl acetal gave the azomethine in essentially quantitative yield. Reduction of the imine double bond followed by tosylation and then treatment of <u>51</u> with acid, using conditions discovered by Jackson²⁵ for the synthesis of isoquinolines, to induce cyclization and aromatization gave the desired pyridocarbazoles. Thus, from <u>51</u> (R=CH₃) there was obtained 9-methoxyellipticine (<u>1b</u>) (29% overall yield), and from <u>51</u> (R=H) there was obtained a mixture of 9-methoxy-11-demethylellipticine (<u>9</u>) and the isomeric pyrido[3,4-c]carbazole <u>52</u> (20% overall yield). This synthesis is modeled after that originated by Cranwell and Saxton 18 and as later modified by Dalton $^{6\mathtt{a},26}$ and Guthrie. 27

In a subsequent paper, Viel^{22b} condensed 4-methoxycyclohexanone ($\underline{53}$) with 2methylphenylhydrazine ($\underline{54}$) to give tetrahydrocarbazole $\underline{55}$, which was oxidized with chloranil to $\underline{49}$ (R=H) (37%) (Scheme 12). The completion of this synthesis followed their earlier work^{22a} except that the imine $\underline{56}$, in the dimethyl series, was converted directly to 9-methoxyellipticine ($\underline{1b}$) by heating in orthophosphoric acid ($\underline{558}$)--an improvement noted by Falton.^{6a}

This approach (Scheme 12) of course precludes the formation of the isomeric indolenine Fischer cyclization products (not shown) that do form using the earlier^{22a} approach (Scheme 11).

These general methods were used by the same scientists to synthesize $8-^{28a}$ and 9-nitroellipticine^{28b} and 9-aminoellipticine.

A very similar route to ellipticine was published by Jackson^{29a} (Scheme 13) at about the same time that the Viel²² studies were reported. Thus, 3-formyl-1,4-

dimethylcarbazole (57)--originally prepared by Cranwell and Saxton¹⁸ by formylating 1,4-dimethylcarbazole (42%)--was condensed with aminoacetaldehyde diethyl acetal, and the resulting imine was hydrogenated to the amine acetal, apparently using the original¹⁸ procedure (61% overall). Whereas Cranwell and Saxton¹⁸ obtained ellipticine (<u>la</u>) from the amine acetal in low yield, Jackson and coworkers^{29a} described a greatly improved cyclization protocol. Thus, tosylation of the amine acetal (93%) followed by treatment of the derived sulfonamide <u>58</u> with HCl in dioxane gave ellipticine (<u>la</u>) in 87% yield. The presumed intermediate <u>59</u> could be isolated from the reaction mixture (13%) and converted to <u>la</u> under the same conditions.

More recently, another synthesis of <u>58</u> was reported^{29d} which used the N-benzyl derivative of <u>57</u> and employed a Birch reduction to reduce the imine double bond and effect debenzylation. The overall yield of <u>58</u> in this sequence was 66%. This same group^{29d} synthesized olivacine (<u>2</u>) in a similar manner from the appropriate monomethyl formylcarbazole but the overall yield was very low (3.5%).

The Jackson group^{29b,c} synthesized the novel but somewhat labile 8,9,10trimethoxyellipticine ($\underline{65}$) using the same general approach (Scheme 14). The synthesis required 4,5,6-trimethoxyindole ($\underline{62}$) and this compound was prepared in 62% yield from 3,4,5-trimethoxyaniline ($\underline{60}$) using the Sugasawa indole synthesis.³⁰ Condensation of $\underline{62}$ with 2,5-hexanedione afforded the desired carbazole $\underline{63}$ (35%). Vilsmeier formylation gave a separable mixture of the 3-formyl (46%), 8-formyl (37%), and 3,8-diformyl (12%) derivatives. The requisite 3formylcarbazole $\underline{64}$ was then transformed into the target ellipticine derivative $\underline{65}$ using the standard four-step methodology (78% yield from 64).

Scheme 14

Lallemand and coworkers³¹ have synthesized 7-hydroxyellipticine (<u>69</u>) by a very similar Type III strategy (Scheme 15). This material is a minor rat metabolite³¹ of ellipticine, the major metabolite being 9-hydroxyellipticine.³² Condensation of 7-methoxyindole with 2,5-hexanedione gave carbazole <u>66</u> (48%). Formylation (47%) and imine formation produced imine acetal <u>67</u> (81%). Cyclization and aromatization was accomplished by heating <u>67</u> in orthophosphoric acid, but this procedure gave 7-methoxyellipticine (<u>68</u>) in only 18% yield. Demethylation to <u>69</u> was effected with pyridinium hydrochloride (45%).

Taylor³³ has reported the synthesis of 7-fluoroellipticine (<u>71</u>) using an analogous route starting with 7-fluoroindole (Scheme 16). The usual condensation with 2,5-hexanedione, Vilsmeier formylation, and imine formation gave imine <u>70</u> (yields unreported). Transformation of <u>70</u> into 7-fluoroellipticine (<u>71</u>) was achieved using Guthrie's conditions²⁷ although the yield of the final cyclization-aromatization step was disappointing (14%), even under forcing conditions. This was ascribed to an unusually stubborn detosylation step.

Scheme 16

It is interesting to note that an $attempt^{33}$ to use the Viel²² tetrahydrocarbazole approach to <u>71</u> failed. Thus, a Borsche reaction between 2fluorophenylhydrazine and 2,5-dimethylcyclohexanone gave only the unwanted indolenine 72 as a mixture of diastereomers.

Using a somewhat different Type III strategy, Yonemitsu³⁴ has described the synthesis of several 5-substituted 5-demethylated ellipticines using his earlier published methodology 35 (Scheme 17). Sodium hydride induced condensation of formaldehyde dimethyldithioacetal S-oxide (FAMSO) with indole ester 73 (prepared 36 in two steps (78%) from indole) gave β -keto sulfoxide 74 (88%). Acid-catalyzed cyclization of 74 produced tetrahydrocarbazole 75 (81%), which on treatment with tert-butyl α -lithioacetate gave adduct 76 (97%) as a mixture of diastereomers. This material represented a key intermediate in the synthesis of the 5-methylthio $(\underline{83})$ and 5-methoxy $(\underline{91})$ derivatives. Aromatization of 76 was readily accomplished with acetic acid in hot xylene to give carbazole ester 77 (82%). This was converted in excellent overall yield to formamide 81 by the following reaction sequence: transesterification (methanol, p-toluenesulfonic acid; 98%) to 78, ammonolysis (ammonia, methanol, methoxide; 97%) to 79, dehydration (ptoluenesulfonyl chloride, pyridine; 94%) to nitrile 80, reduction (lithium aluminum hydride, aluminum chloride) to the primary amine and formylation (neat ethyl formate; 92%) to give 81. Cyclization of 81 with phosphorus oxychloride gave dihydroellipticine derivative 82 (92%). Dehydrogenation with palladium/carbon in refluxing decalin afforded 83 (63%), which could be desulfurized with Raney nickel to the known³⁷ 5-demethylellipticine (84) (61%). The overall yield of 83 from ester 73 is 27% for ten steps.

Scheme 17

Yonemitsu's synthesis of 5-methoxy-5-demethylellipticine (91) followed a different route³⁴ (Scheme 18). The hydroxy ester <u>76</u>, prepared earlier (Scheme 17),

was converted to ketone ester <u>85</u> by heating with <u>p</u>-toluenesulfonic acid in methanol (87%). The highest yield obtained for the direct conversion of $\frac{85}{10}$ to lactone $\frac{86}{10}$ was 50% (p-toluenesulfonic acid, refluxing xylene), although a three-step sequence gave 86 in 58% yield. Thus, 85 was acetylated (91%) and the resulting tertiary acetate was exposed to sodium hydride to effect elimination affording keto esters 87a/87b (14:1 ratio) (75%). Heating the 87 mixture with p-toluenesulfonic acid in toluene gave lactone carbazole 86 (81%). With 86 in hand, Yonemitsu prepared hydroxyamide 88 by treating 86 with ammonia and methoxide in methanol (95%). Methylation and dehydration afforded methoxy nitrile 89 (88%). The synthesis of 91 was completed in a manner similar to that for $\underline{83}$ (Scheme 17). Thus, reduction of 89 to the primary amine, formylation, and cyclization gave the dihydroellipticine derivative 90 (74% from 89). Oxidation to the target 5-methoxy-5-demethylellipticine (91) was achieved with activated MnO2 (52%), or better by reducing 90 to <u>92</u> and then oxidizing the latter to <u>91</u> (66% from <u>90</u>). Finally, <u>91</u> was demethylated with hydrogen bromide to give 5-hydroxy-5-demethylellipticine (93) (85%).

In 1977, Bergman and Carlsson³⁸ announced the most efficient synthesis of ellipticine (<u>la</u>) reported up to that time (Scheme 19). It is noteworthy that this synthesis is conceptually similar to Woodward's³⁹ original synthesis of

6.

ellipticine, which, although proceeding in very low yield, paved the way for Bergman's accomplishment. Acid-promoted condensation of 2-ethylindole (<u>94</u>), which was prepared via a Madelung-indole synthesis,⁴⁰ with 3-acetylpyridine gave alkene <u>95</u> (90%). Alkylation of the pyridine nitrogen with <u>n</u>-butyl bromide (100%) and rapid pyrolysis (>350°C, 5 min) of the resulting salt <u>96</u> gave ellipticine (<u>1a</u>) in 72% yield, along with about 10% of the pyrido[2,3-<u>b</u>]carbazole regioisomer <u>98</u>. Slower heating of <u>96</u> gave <u>98</u> as the major product (220°C, 30 min). The reaction is presumed to involve a 1,5-hydrogen shift leading to indole-2,3-quinodimethane <u>97</u>. Subsequent disrotatory electrocyclization, dehydrogenation, and loss of <u>n</u>-butyl bromide would afford 1a.

Scheme 19

The structure of the novel "isoellipticine" <u>98</u> was confirmed by independent synthesis^{41,42} (Scheme 20). Thus, 2-ethylindole (<u>94</u>) condensed with amine nitrile <u>99</u> to give <u>100</u> (70%), which smoothly cyclized to carbazole <u>101</u> (44%) under the influence of sodium hydride. A Skraup reaction (70%) completed the preparation of <u>98</u>.

Scheme 20

These same workers^{42,43} have synthesized olivacine (<u>2</u>) using similar methodology (Scheme 21). In this instance, like that observed by Woodward³⁹ earlier, the condensation of <u>94</u> with pyridine aldehyde <u>102</u> stopped at the 2:1 adduct <u>103</u> (65%). Thermolysis of <u>103</u> over an open flame under vacuum (<0.1 torr) for a few minutes gave olivacine (<u>2</u>) directly (40%). Scheme 21

In 1979, Takano⁴⁴ reported the synthesis of the parent 6H-pyrido[4,3b]carbazole (5,11-didemethylellipticine) (<u>110</u>) via a novel Type IV strategy (Scheme 22). Condensation of 1-benzylpiperidine-2,4-dione (<u>104</u>) with triethyl orthoformate and 2-nitroaniline (<u>14a</u>) gave vinylogous urea <u>105</u> (83%). An amine exchange reaction with 2-methyltryptamine transformed <u>105</u> into <u>106</u> (90%), which, upon heating with acetic anhydride-acetic acid (5:3), underwent a remarkable series of transformations to give, after basic hydrolysis of the mixture of <u>107</u> and <u>108</u> (6:11), carbazole <u>108</u> (46% from <u>106</u>). Reduction of <u>108</u> with lithium aluminum hydride gave amine <u>109</u> (100%), which was dehydrogenated and debenzylated with palladium in boiling decalin to give didemethylellipticine <u>110</u> (69%).

The authors⁴⁴ propose the following pathway for the conversion of $\underline{106}$ to $\underline{107}$ (Scheme 23).

Pandit⁴⁵ has recently reported the syntheses of ll-hydroxy-5,6-dimethyl-6Hpyrido[4,3-b]carbazole (120), 6-methylellipticine (121), and 6-methylolivacine (126), utilizing a mild base-catalyzed version of the Woodward-Bergman approach (Scheme 24). Indole ester <u>111</u> was readily assembled from 2-lithio-l-methylindole and ethyl pyruvate (63%). Deoxygenation to <u>112</u> was accomplished via the unsaturated ester followed by conjugate reduction with sodium borohydride (58%). Acylation of <u>112</u> with nicotinoyl chloride hydrochloride (<u>113</u>) in hot sulfolane gave keto ester <u>114</u> (55%), which was alkylated with benzyl bromide to yield the pyridinium salt <u>115</u>. This smoothly cyclized under remarkably mild conditions to dihydropyridine <u>116</u> (81% from <u>114</u>), and the 1:1 mixture of diastereomers was oxidized very efficiently with N-benzylacridinium bromide (<u>117</u>) to afford salt <u>118</u> (90%). Reductive debenzylation to <u>119</u> and base-induced decarboethyoxylation gave the desired hydroxyellipticine derivative <u>120</u> (45%).^{45a}

Scheme 24

In a subsequent paper, Pandit^{45b} used the same strategy to construct the 6methyl derivatives of ellipticine and olivacine (<u>121</u> and <u>126</u>). Thus, keto ester <u>119</u> was converted to <u>121</u> in a single step with excess methylmagnesium iodide (40%) (Scheme 25). A second route to <u>121</u> involved a Wittig reaction on <u>119</u> with methylenetriphenylphosphorane to give <u>122</u> (65%), followed by base hydrolysis and decarboxylation to yield <u>121</u> (60%).

Scheme 25

In similar fashion, Pandit^{45b} acylated indole ester <u>112</u> with 2-methylnicotinoyl chloride hydrochloride (<u>123</u>) to give <u>124</u> (30%) (Scheme 26). An identical sequence to that presented in Scheme 24 transformed <u>124</u> into <u>125</u> (49% from <u>124</u>), which, upon treatment with Red-Al, led directly to 6-methylolivacine (126) (57%).

Scheme 26

Independently, Weller 46 has recently extended this same approach to a synthesis of ellipticine (1a) (Scheme 27). Thus, keto acid 127 was prepared from indole (58%) using unspecified indole C-2 lithiation methodology, and converted to ester 128 by a Wolff-Kishner reduction and esterification (81%). Condensation of 128 with 3-acetylpyridine using Bergman's conditions (cf., Scheme 19) gave 1:1 adduct 129 (82%). Methylation and immediate exposure of the resulting pyridinium salt to methoxide gave the labile dihydropyridine 130 (62%). When the crude reaction mixture containing 130 was treated directly with the oxidizing agent, ethyl nicotinate methiodide (131), the desired pyridinium salt 132 was isolated in 78% yield from 129. Alternatively, 129 was hydrogenated to 133 which could be converted in the same fashion, via dihydropyridine 134 (84%), to pyridinium salt 132 (57% from 133). It is important to note that in these enolate additions to pyridinium salts (Schemes 24, 26, 27) 1,4-addition greatly predominates over 1,2addition. Completion of the synthesis of la was achieved by reducing 132 to the presumed dihydroellipticine 135 with Vitride (=Red-Al). Without being isolated, 135 was oxidized with 131 to the 2-methylellipticine salt 136 (85% from 132). Nucleophilic demethylation of 136 with thiophenoxide gave ellipticine (1a) (91%).

Scheme 27

7. Strategy V

Kano's approach⁴⁷ to the pyrido $[4,3-\underline{b}]$ carbazole ring system parallels the Bergman method³⁸ in that both involve the thermal generation and cyclization of an indole-2,3-quinodimethane intermediate, but Kano's synthesis involves the bondforming strategy classified as Type V (Scheme 28). Thus, treatment of 1-(benzenesulfony1)-3-ethylindole (<u>137</u>) with lithium diisopropylamide (LDA) followed by quenching the resulting 2-lithio species with isonicotinic anhydride gave ketone <u>138</u> (75%). A Wittig reaction and base-cleavage of the protecting group gave alkene <u>139</u> (67%). Rapid thermolysis of <u>139</u> at 500°C for 7 min led directly to ellipticine (<u>1a</u>) (50%).

Scheme 28

The same group has applied this methodology to syntheses of 11demethylellipticine $(\underline{140})$, 4^7 6-thioellipticine $(\underline{141})$, 4^8 and several related heterocycles⁴⁸ that are outside the scope of this review.

An exceptionally short synthesis of the ellipticine ring system, which utilizes "tandem metalation," has been described by Snieckus⁴⁹ (Scheme 29). Thus, lithiation of N,N-diethylisonicotinamide (<u>142</u>) with <u>sec</u>-butyllithium/N,N,N',N'tetramethylethylenediamine (TMEDA) followed by sequential addition of an Nprotected indole-3-carboxaldehyde <u>143</u> and <u>sec</u>-butyllithium presumably generates <u>144</u>. Upon warming to room temperature <u>144</u> undergoes cyclization and spontaneous oxidation to give the quinones <u>145a-c</u> in fair to good yield (<u>145a</u>, 26%; <u>145b</u>, 76%; <u>145c</u>, 40%). These quinones were converted to the corresponding ellipticines (<u>1a</u>, 40%; <u>121</u>, 62%; <u>146</u>, 40%) by sequential treatment with methyllithium, 47% hydrogen iodide, and stannous chloride/hydrochloric acid. These conditions removed the MOM group in 145a giving ellipticine (1a) directly.

Scheme 29

Kutney⁵⁰ has employed tricarbonylchromium complexes in a Type V strategy to synthesize olivacine (2) and (\pm)-guatambuine (152) (Scheme 30). Thus, 1-

(benzenesulfonyl)indole was metalated at C-2 and then allowed to react with 4acetylpyridine to give alcohol <u>147</u> (65%). Hydrolysis of the protecting group (72%) and iodomethylation gave pyridinium salt <u>148</u> (89%). Reduction of this material with sodium borohydride in a two-phase system (ether-aqueous sodium hydroxide) followed by the addition of the dried ether solution to a solution of trisacetonitriletricarbonyl chromium gave the red dihydropyridine complex <u>149</u> (56%), as two diastereomers (2:1 ratio). A Vilsmeier reaction on <u>149</u> gave two products: <u>150</u> (major) and <u>151</u> (minor) (7:3 ratio) (91%) after treatment of the reaction mixture with pyridine. This mixture of <u>150</u> and <u>151</u>, which could only be separated with difficulty, was dehydrogenated to a mixture of <u>150</u> and 11-demethylellipticine (<u>140</u>) (9.5% from <u>149</u>). Demethylation of <u>150</u> with triphenylphosphine gave <u>140</u> (54%). Treatment of <u>140</u> with methyllithium and then oxidation of the dihydropyridine intermediate with iodine afforded olivacine (<u>2</u>) (54%). Finally, iodomethylation of <u>2</u> followed by sodium borohydride reduction gave (\pm)-guatambuine (<u>152</u>) (40%).

Scheme 30

The French group whose work was discussed earlier^{11,22} has also synthesized 9methoxy-11-demethylellipticine $(\underline{9})^{11}$ using the same Type V strategy (Scheme 31) that they used to prepare ellipticine (<u>1a</u>) earlier.⁵¹ Thus, 5-methoxygramine (<u>153</u>), obtained via a Mannich reaction of 5-methoxyindole (78%), and the pyrrolidine enamine of N-benzyl-4-piperidone (<u>154</u>) condensed to form ketone <u>155</u> (92%). Reaction of <u>155</u> with lithium acetylenide gave the diastereomeric alcohols <u>156</u> (83%). A Rupe rearrangement was effected by heating <u>156</u> in formic acid to afford the cyclized carbazole <u>157</u> (32%). Catalytic debenzylation proved to be difficult but <u>157</u> was eventually converted to 9-methoxy-ll-demethylellipticine (<u>9</u>) by boiling it with palladium/carbon in decalin (40%; 8% overall). As described earlier (Scheme 2) <u>9</u> could be converted to the 9-hydroxy derivative <u>10</u>.

Scheme 31

Joule has developed⁵²⁻⁵³ three quite versatile synthetic schemes for the preparation of pyridocarbazoles using the Type VI strategy. The reaction between 2-lithio-l-(benzenesulfonyl)indole (<u>158</u>) and a 3-(hydroxymethyl)isonicotinic acid lactone <u>159</u> is common to all three approaches.

8.

In the original strategy,^{52a} Joule coupled <u>158</u> with lactone <u>159a</u>, prepared⁵⁴ in three steps (16%) from dimethyl pyridine-3,4-dicarboxylate, to give keto alcohol <u>160</u> (58%) (Scheme 32). Oxidation of <u>160</u> with manganese dioxide in refluxing chloroform gave keto aldehyde <u>161</u> (44%), which was selectively protected as acetal <u>162</u> (100%) (ethylene glycol, <u>p</u>-TSA, refluxing chloroform). Base cleavage of the Nprotecting group (50% aqueous sodium hydroxide in methanol, reflux) gave <u>163</u> (96%). Simply treating <u>163</u> with 1N hydrochloric acid at 95°C in the presence of air formed the desired ellipticine-guinone <u>165</u> in 90% yield, presumably via intermediate semiguinone <u>164</u>. Reaction of <u>165</u> with an alkyl- or aryllithium followed by reduction of the derived diols with sodium borohydride gave the corresponding ellipticines (<u>1a</u>, <u>166</u>, <u>167</u>) in high yield. Moreover, treatment of <u>165</u> with only two equivalents of methyllithium followed by reduction gave 11demethylelliptine (<u>140</u>). Indeed, reaction of <u>165</u> with two equivalents of <u>n</u>butyllithium and then with one equivalent of methyllithium followed by reduction gave, with high regioselectivity, <u>5-n</u>-butyl-11-methyl-6H-pyrido[4,3-<u>b</u>]carbazole (168) in excellent yield.

In a related methodology (Scheme 33), $Joule^{52b,56}$ coupled 2-lithioindole <u>158</u> with lactone <u>159b</u>, prepared⁵⁵ in five steps (19%) from 3-ethyl-4-methylpyridine, to give alcohol <u>169</u> (58%). Oxidation with manganese dioxide (refluxing chloroform) furnished dione <u>170</u> (63%) in which the methyl ketone functionality could be selectively ketalized (ethylene glycol, <u>p</u>-TSA, refluxing chloroform) to produce <u>171</u> (80%). Alkaline hydrolysis of the sulfonamide and a Wittig reaction gave alkene <u>172</u>. Hydrogenation and acid treatment converted <u>172</u> to ellipticine (<u>1a</u>) (32% from <u>171</u>). Joule has also used this route to synthesize ll-demethylellipticine (<u>140</u>) from lactone <u>159a</u> (22% overall yield from <u>160</u>).

Scheme 33

In a more recent publication, $Joule^{53}$ has reported an improved synthesis of ellipticine (<u>la</u>) (ca. 20% overall yield from indole), although the preparation of the requisite lactone <u>159a</u> proceeds in low yield⁵⁴ (<u>vide supra</u>). Thus, keto alcohol <u>160</u> was efficiently transformed into the pyrido-oxepino-indolone <u>173</u> (83%)⁵⁷ by a novel intramolecular nucleophilic addition reaction at the indole C-3 position (Scheme 34). Refluxing <u>173</u> in basic media in the presence of air gave ellipticinequinone <u>165</u> (46%) by the presumed pathway shown.

Similarly, Joule⁵³ has synthesized the other three isomeric quinones $\frac{175a-c}{175a-c}$ from the respective pyrido-oxepino-indolones (71-89%), after much shorter reflux periods than for $\frac{173}{15} + \frac{165}{15}$ (93 h).

Sainsbury⁵⁹ has described a synthesis of benzo[k]ellipticine (<u>178</u>) using a similar scheme to that previously developed⁶⁰ for the preparation of 9-phenyl-ellipticine⁶¹ (Scheme 35). The yield of the key condensation reaction between diacetylindoxyl <u>176</u> and the 4-acetylpyridine ether <u>177</u> was less than anticipated and insufficient 178 was secured for biological evaluation.

Scheme 35

In another Type VI approach, Sainsbury⁵⁹ has demonstrated the versatility of his previously reported^{7,62,63} route to ellipticines (Scheme 36). Thus, when nitrile <u>179a</u> (prepared by the same general route⁷ depicted in Scheme 37) was treated with <u>n</u>-butyllithium and the resulting ketone exposed to mild acidic conditions the ellipticine derivative 180 was obtained (87%). In similar fashion,

nitrile <u>179b</u> with <u>n</u>-butyllithium gave <u>168</u> (78%) and <u>179a</u> with phenyllithium gave 181 (20%).

Furthermore, Sainsbury⁶⁴ has extended his "standard" ellipticine methodology^{7,62,63} to a synthesis of 8-methoxyellipticine (<u>187</u>) and 8hydroxyellipticine (<u>188</u>) (Scheme 37). This work comprises the first synthesis of <u>188</u>, which is a metabolite of ellipticine (<u>1a</u>) in <u>Aspergillus alliaceus</u>.⁶⁵ Thus, a coupling reaction between 6-methoxy-1-indoly1magnesium bromide and 3-(1chloroethy1)pyridine gave <u>182</u> (40%). Amination of <u>182</u> with <u>183</u>, followed by acetylation, and iodomethylation afforded salt <u>184</u>. Cyanide addition to <u>184</u> gave cyanodihydropyridine <u>185</u> which upon photolysis yielded cyanopyridine <u>186</u>. The usual addition of methyllithium to <u>186</u> and acid-induced cyclization gave the target 8-methoxyellipticine (<u>187</u>) (42% from <u>182</u>). Conversion to <u>188</u> was achieved under the usual conditions of hot pyridine hydrochloride (30%).

Scheme 37

The main disadvantage of the Sainsbury pyrido[4,3-b]carbazole synthesis is that the reaction between indolylmagnesium halides and pyridylethyl chloride typically proceeds in 10-30% yield. A solution to this difficulty has recently been developed in the Sainsbury laboratory. 66,67 Thus, an Emmons-Wadsworth condensation between 3-acetylpyridine and triethylphosphonoacetate gave unsaturated ester <u>189</u> (94%) (Scheme 38). Hydrogenation to <u>190</u> (98%) and DIBAL reduction to aldehyde <u>191</u> (93%) also proceeded very well. Fischer indolization of <u>191</u> with substituted phenylhydrazine hydrochlorides gave the expected pyridylethylindoles <u>192</u> (R=alkyl, alkoxy, halogen) in 65-80% yield. When 3-substituted phenylhydrazines were employed, both 4- and 6-substituted indoles were formed, with the latter predominating. The completion of the pyridocarbazole synthesis involved the same methodology described above (Scheme 37). In this fashion, the following new ellipticine derivatives were prepared: 7-chloro (<u>193a</u>), 7-fluoro (<u>193b</u>), and 7methyl (<u>193c</u>). In addition, 9-methoxyellipticine (<u>1b</u>) was prepared in 40% overall yield from 4-methoxyphenylhydrazine and aldehyde 191.

It is appropriate here to mention that $Kubo^{68}$ has reported an improved synthesis of the key Sainsbury pyridylindole <u>192</u>, R=H (Scheme 39). Thus, oxindole <u>194</u>, prepared by Sainsbury earlier,⁶¹ was converted to the 2-chloroindole <u>195</u> (78%) with phosphorus oxychloride and then by hydrogenation to <u>192</u> (R=H) (71%). In fact, the yield of <u>192</u> is higher (73% from <u>194</u>) if <u>195</u> is not isolated and purified, but simply hydrogenated directly.

Scheme 39

Kozikowski⁶⁹ has described a novel but less efficient synthesis of the Sainsbury nitrile <u>179b</u> (vide supra) that utilizes oxazole Diels-Alder methodology (Scheme 40). Indole-3-acetonitrile (<u>196</u>), which is readily prepared from gramine methiodide and potassium cyanide, was converted to methyl 2-(3-indolyl)propionate (<u>197</u>) in four steps (74%). Reaction of <u>197</u> with excess α -lithio-methylisocyanide followed by acetic acid workup gave oxazole <u>198</u> (80%). This substance underwent a [4+2] cycloaddition reaction with acrylonitrile to afford nitrile <u>179b</u> (16%). The usual^{61,63} manipulation gave ellipticine (la) in 80% yield.

Kano⁷⁰ has also utilized a Type VI thermolysis strategy to prepare ellipticines (Scheme 41). Condensation of 2-lithio-1-(phenylsulfonyl)indole (<u>158</u>) with 3-methyl-4-acetylpyridine (<u>200</u>), 2,3-dimethyl-4-acetylpyridine (<u>201</u>), and 3ethyl-4-acetylpyridine (<u>202</u>), afforded the respective 2-vinylindoles <u>203</u> (35%), <u>204</u> (14%), and <u>205</u> (24%). Thermolysis of each of the latter compounds produced 11demethylellipticine (<u>140</u>) (57%), olivacine (<u>2</u>) (57%), and ellipticine (<u>1a</u>) (30%), respectively. The last thermolysis reaction also gave <u>140</u> (43%), presumably as a result of methane elimination by a radical process during aromatization.

Scheme 41

Kano⁴⁸ has also used this pyridine-3,4-quinodimethane strategy to prepare several ellipticine analogues, such as 207 and 208, compounds which may be of biological interest but whose syntheses are outside the scope of the present review.

Quéguiner^{71a} has used a Type VI approach to synthesize several novel azaellipticines of the 6H-indolo[3,2-b]naphthyridine type (Scheme 42). Thus, a Friedländer guinoline synthesis using 1-acetylindoxyl (209) and an appropriate aminoformylpyridine <u>210</u>, prepared^{71b} from the corresponding carboxylic acid, gave directly in unspecified yield the 6H-indolo[3,2-<u>b</u>]naphthyridines <u>211a-c</u>. Moreover, iodomethylation of <u>211c</u> gave naphthyridinium salt <u>212</u>.

Kononova and Semenov⁷² have used a very similar approach to prepare 5azaellipticine.

We^{73a} have used a Type VI strategy to synthesize ellipticine (<u>1a</u>) in an approach that relies on the regioselective ring opening of cinchomeronic anhydride (Scheme 43). Thus, 1-(phenylsulfonyl)indole, prepared from indole with <u>n</u>-butyllithium and benzenesulfonyl chloride (91%), was lithiated at C-2 with LDA and the resulting anion (<u>158</u>) was treated at -100°C with cinchomeronic anhydride (<u>213</u>) to give a 92:8 mixture of keto acids <u>214</u> and <u>215</u> (78%). The major isomer <u>214</u> was hydrolyzed to <u>216</u> (~100%) which could be cyclized to keto lactam <u>217</u> upon treatment with hot acetic anhydride (~100%). Reaction of <u>217</u> with methyllithium (2 equiv) at -100°C gave diol <u>218</u> (mixture of diastereomers). This rather unstable material was directly treated with sodium borohydride to afford ellipticine (<u>1a</u>) (82% yield from <u>217</u>). The overall yield of <u>1a</u> from indole is 54%, representing one of the most efficient syntheses of ellipticine. The same sequence, when applied to 5-methoxyindole, gave 9-methoxyellipticine (<u>1b</u>) in 47% overall yield.^{73b}

Scheme 43

In identical fashion we⁷⁴ have converted the minor keto acid <u>215</u> to "isoellipticine" <u>221</u> (5,11-dimethy1-10<u>H</u>-pyrido[3,4-<u>b</u>]carbazole) (Scheme 44). Thus, base hydrolysis of <u>215</u> (95%) followed by cyclization of keto acid <u>219</u> with acetic anhydride gave keto lactam $\underline{220}$ (98%). Treatment of $\underline{220}$ with methyllithium followed by sodium borohydride gave isoellipticine (221) (91%).

Scheme 44

It has been found⁷⁵ that keto lactam <u>217</u> can be used to construct unsymmetrically substituted ellipticine derivatives (Scheme 45). Thus, sequential treatment of <u>217</u> with <u>n</u>-butyllithium, methyllithium, and then sodium borohydride gave 5-<u>n</u>-butyl-ll-methyl-6H-pyrido[4,3-<u>b</u>]carbazole (<u>168</u>) (70%) along with 18% of the 5,11-di-<u>n</u>-butyl derivative. Moreover, this methodology was also used⁷⁵ to synthesize the <u>Strychnos dinklagei</u> alkaloid 17-oxoellipticine (<u>224</u>) by using the formyl anion synthetic equivalent <u>222</u> as shown in Scheme 45. Thus, sequential treatment of <u>217</u> with 2-lithio-2-trimethylsilyl-1,3-dithiane, methyllithium, and then sodium borohydride gave <u>223</u> (25% yield from <u>217</u>). Hydrolysis of <u>223</u> with aqueous silver nitrate gave <u>224</u> (~100%).

We⁷⁴ have also used the regioselective acylation of pyridine anhydride <u>213</u> with the previously unknown 3-lithio-1-(phenylsulfonyl)indole to synthesize isoellipticine (Scheme 46). Indole was converted⁷⁶ to 3-iodo-1-(phenylsulfonyl)indole <u>225</u> in one pot by successive treatment with <u>n</u>-butyllithium, iodine, LDA, and benzenesulfonyl chloride (88%). Halogen-metal exchange with <u>t</u>-butyllithium followed by guenching the 3-lithioindole species with cinchomeronic anhydride (<u>213</u>) gave with apparent complete regioselectively the keto acid <u>226</u> (57%). This substance was directly converted to the target isoellipticinequinone <u>227</u> upon treatment with excess LDA (41%), a transformation which presumably involves lithiation at the indole C-2 position, cyclization at the carboxylate carbonyl, and <u>in situ</u> desulfonylation. This process was improved by converting keto acid <u>226</u> to ester <u>228</u> (89%) and treating the latter with LDA to give quinone <u>227</u> (66%). In accord with the earlier work (cf., Schemes 29, 32), guinone $\underline{227}$ reacted with methyllithium and then sodium borohydride to furnish isoellipticine $\underline{221}$ (67%). Scheme 46

The same sequence when applied to 5-methoxyindole gave 7-methoxyisoellipticine (232) (Scheme 47).⁷⁴ The requisite 3-iodo derivative 229 was prepared in one pot from 5-methoxyindole (75%) and then converted to keto acid 230 in the usual fashion with complete regioselectivity (77%). Esterification (98%) and cyclization with lithium bis(trimethylsilyl)amide gave methoxyquinone 231 (60%). Treatment of 231 with methyllithium and then with sodium borohydride gave 7-methoxyisoellipticine (232) (62%).

Scheme 47

9. Strategy VII

This bond-forming protocol has been the least studied of the eight strategies discussed herein.

Moody⁷⁷ has very recently described a short synthesis of ellipticine (<u>la</u>) and isoellipticine (<u>221</u>) using a Diels-Alder Type VII strategy (Scheme 48). The

pyranoindolone <u>233</u>, which is easily prepared from indole in two steps (44%), underwent a cycloaddition reaction with 3,4-pyridyne to give, after loss of carbon dioxide, an equal amount of ellipticine (<u>la</u>) and isoellipticine (<u>lb</u>) (20% each). The pyridyne was generated in novel fashion by the thermolysis of triazene acid <u>234</u>, prepared by diazotization of 3-aminopyridine-4-carboxylic acid and treatment with dimethylamine (72%).

In a related approach, we⁷⁸ have used the Diels-Alder reaction between 3,4pyridyne and 1,3-dimethyl-4-(phenylsulfonyl)-4<u>H</u>-furo[3,4-<u>b</u>]indole (237) tosynthesize ellipticine $(\underline{1a})$ and isoellipticine $(\underline{221})$ (Scheme 49). The key furoindole 237 was prepared by two routes. In the first, 79 indole-3-carboxaldehyde was converted to the 1-phenylsulfonyl derivative 235 using LDA and benzenesulfonyl chloride (86%). Sequential treatment of 235 with methyllithium, t-butyllithium, and then acetaldehyde gave diol 236 as a mixture of diastereomers (81%). Oxidation of 236 with activated manganese dioxide followed by dehydration of the resulting lactol with a catalytic amount of trifluoroacetic acid gave furoindole 237 (30%). Alternatively, ⁷⁸ 3-ethylindole was converted to the 1-phenylsulfonyl derivative <u>137</u> using n-butyllithium and benzenesulfonyl chloride (74%). This substance was lithiated at C-2 with phenyllithium and the resulting anion was quenched with acetaldehyde to give alcohol 238 (73%). This was transformed into hydroxy ketone 241 by a sequence of oxidation, bromination, and solvolysis (86% yield from 238). Attempted recrystallization of 241 from dichloromethane gave furoindole 237 (~100%). Generation of 3,4-pyridyne in the presence of furoindole 237 was accomplished either from 3-chloro-4-iodopyridine and t-butyllithium or from 1aminotriazolo[4,5-c]pyridine and lead tetraacetate to give a mixture of the Diels-Alder adducts 242a,b (34% and 38% from the two reactions, respectively). Treatment of this mixture with sodium borohydride gave an easily separable mixture of ellipticine (1a) (23%) and isoellipticine (221) (29%).

The final bond-forming strategy to be reviewed mimics the proposed⁸⁰ biogenesis of olivacine (2), and two groups have described work along these lines.

Husson⁸¹ has used the generation and fragmentation of desethyluleine <u>248</u> in a simple biomimetic synthesis of (\pm) -guatambuine (<u>152</u>) and olivacine (<u>2</u>) (Scheme 50). The approach was patterned after Joule's⁸² synthesis of uleine and dasycarpidone alkaloids, but Husson used a different synthesis, at the outset, of desethyldasycarpidone <u>246</u>. Thus, reaction of <u>158</u> with pyridine-4-carboxaldehyde gave alcohol <u>243</u> (62%). Base hydrolysis gave <u>244</u> (57%) which was transformed into <u>246</u> using Joule's original procedure,⁸² although Joule used dimsyl sodium to isomerize <u>245</u> to the corresponding enamine isomer prior to the acetic acid-induced Mannich cyclization (to <u>246</u>). Reaction of <u>246</u> with methyllithium gave alcohol <u>247</u> (~100%) and dehydration to desethyluleine <u>248</u> occurred readily with trifluoroacetic acid (~100%). Further treatment with hot aqueous acetic acid led to carbazole <u>249</u>

(90%), which was acetylated under standard conditions to give amide $\underline{250}$ (95%). This amide was also obtained directly from $\underline{248}$ (or even $\underline{247}$) by treatment with acetyl chloride (95%). A conventional Bischler-Napieralski reaction converted $\underline{250}$ into immonium salt $\underline{251}$ which upon treatment with sodium borohydride gave (\pm)-guatambuine ($\underline{152}$) (72% from $\underline{250}$). Oxidation of $\underline{152}$ with palladium gave olivacine ($\underline{2}$) (24%). Alternatively, a Mannich reaction between amine $\underline{249}$ and acetaldehyde gave guatambuine ($\underline{152}$) in one step (33%).

A much shorter route to carbazole amine <u>249</u> was also devised by Husson⁸¹ (Scheme 51). Thus, 4-acetylpyridine was easily converted to ketal <u>252</u> by a sequence of ketalization, iodomethylation and reduction (72% overall). With elegant simplicity, piperideine ketal <u>252</u> reacted with indole in acid media to give <u>249</u> in a single operation (74%), presumably via <u>253</u> + <u>247</u> + <u>248</u> + <u>249</u>.

This latter sequence was also applied^{81a} to the synthesis of the naturally occurring⁸³ 9-methoxyolivacine (257) (Scheme 52). Thus, 5-methoxyindole and piperidine ketal 252 condensed to give carbazole-amine 254 (90%). Acetylation (63%) and Bischler-Napieralski cyclization (76%) gave (\pm)-9-methoxy-guatambuine (256), which, upon dehydrogenation and demethylation with palladium, gave 9-methoxyolivacine (257) (52%).

Husson^{81a} has also extended this methodology to a synthesis of ellipticine (<u>1a</u>) (Scheme 53). The requisite piperidine ketal <u>261</u> was prepared using Husson's⁸⁴ piperidine-functionalization methodology. Thus, oxidation of piperidine ketal <u>252</u> gave N-oxide <u>258</u> (~100%). This substance underwent a Polonovski-Potier reaction to give the conjugated immonium salt <u>259</u> which was trapped as cyano-amine <u>260</u> (45% from <u>258</u>). Reaction with methylmagnesium bromide gave piperidine ketal <u>261</u> (57%). Unfortunately, condensation between <u>261</u> and indole gave the target carbazole <u>262</u> in very low yield (5%), presumably due to steric effects in the Mannich reaction between indole and the tetrasubstituted immonium ion from <u>261</u>. Nevertheless, carbazole amine <u>262</u> gave tetrahydroellipticine <u>263</u> when treated with formaldehyde in a Pictet-Spengler reaction (53%). Dehydrogenation-demethylation under the usual conditions gave ellipticine (<u>1a</u>) (36%).

The second group to have used a biomimetic approach to the synthesis of olivacine (2) is Ninomiya and coworkers⁸⁵ (Scheme 54). Reaction of indole with 4-acetylpyridine in the presence of acetyl chloride (benzoyl chloride was also used) gave adduct 264 (35%). Hydrogenation of 264 gave 265 (55%) which, upon exposure to p-toluenesulfonic acid or boron trifluoride etherate, gave carbazole amide 266 (62%), originally prepared by Schmutz.⁸⁶ This reaction sequence leading to 266 presumably involves intermediates of the type 247 (cf., Scheme 50). Amide 266 was converted to olivacine (2) following the original procedure⁸⁶ via a Bischler-Napieralski cyclization to dihydroolivacine 267 (88%⁸⁶) and dehydrogenation to 2 (88%⁸⁶).

In conclusion, it is obvious that interest in the ellipticine family of alkaloids amongst synthetic organic chemists remains exceptionally high. As this intense search for even more potent anticancer ellipticine-derived drugs continues, we can be certain that new and efficient syntheses of pyridocarbazoles will be discovered in the future.

REFERENCES AND NOTES

- (a) K.N. Kilminster, M. Sainsbury, and B. Webb, <u>Phytochemistry</u>, 1972, <u>11</u>, 389;
 (b) S. Michel, F.Tillequin, and M. Koch, <u>J. Nat. Prod.</u>, 1980, <u>43</u>, 294; (c) A. Ahond, H. Fernandez, M. J.-Moore, C. Poupat, V. Sánchez, P. Potier, S.K. Kan, and T. Sévenet, <u>J. Nat. Prod.</u>, 1981, <u>44</u>, 193.
- (a) For a review, see C. Paoletti, J.-B. Le Pecq, N. Dat-Xuong, P. Juret, H. Garnier, J.-L. Amiel, and J. Rouesse, <u>Recent Results Cancer Res.</u>, 1980, <u>74</u>, 107; (b) H. Nagasawa, M. Homma, H. Namiki, and K. Niki, <u>Eur. J. Cancer Clin.</u> <u>Oncol.</u>, 1984, <u>20</u>, 273.
- 3. Institut Pasteur.

- 4. (a) P. Juret, A. Tanguy, A. Girard, J.Y. Le Talaer, J.S. Abbatucci, N. Dat-Xuong, J.-B. Le Pecq, and C. Paoletti, <u>Eur. J. Cancer</u>, 1978, <u>14</u>, 205; (b) A. Brugarolas, M. Gracia, R. de Jager, M. Mallarme, and A. Clarysse, <u>Proc. Am.</u> <u>Assoc. Cancer Res.</u>, 1979, <u>20</u>, 310; (c) A. Clarysse, A. Brugarolas, P. Siegenthaler, R. de Jager, F. Cavalli, P. Alberto, and H. Hansen, <u>Proc. Am.</u> <u>Assoc. Cancer Res.</u>, 1980, <u>21</u>, 348; (d) N. Van-Bac, C. Moisand, A. Gouyette, G. Muzard, N. Dat-Xuong, J.-B. Le Pecq, and C. Paoletti, <u>Cancer Treat. Rep.</u>, 1980, <u>64</u>, 879; (e) P. Juret, J.F. Heron, J.E. Couette, T. Delozier, and J.Y. Le Talaer, <u>Cancer Treat. Rep.</u>, 1982, <u>66</u>, 1909; (f) P. Dodion, M. Rozencweig, C. Nicaise, M. Piccart, E. Cumps, N. Crespeigne, D. Kisner, and Y. Kenis, <u>Eur.</u> J. Cancer Clin. Oncol., 1982, <u>18</u>, 519.
- (a) S. Goodwin, A.F. Smith, and E.C. Horning, <u>J. Am. Chem. Soc.</u>, 1959, <u>81</u>, 1903; (b) R.B. Woodward, G.A. Iacobucci, and F.A. Hochstein, <u>J. Am. Chem.</u> <u>Soc.</u>, 1959, <u>81</u>, 4434.
- 6. (a) L.K. Dalton, S. Demerac, B.C. Elmes, J.W. Loder, J.M. Swan, and T. Teitei, <u>Aust. J. Chem.</u>, 1967, <u>20</u>, 2715; (h) G.H. Svoboda, G.A. Poore, and M.L. Montfort, <u>J. Pharm. Sci.</u>, 1968, <u>57</u>, 1720; (c) C.W. Mosher, O.P. Crews, E.M. Acton, and L. Goodman, <u>J. Med. Chem.</u>, 1966, <u>9</u>, 273.
- 7. M. Sainsbury, Synthesis, 1977, 437.
- 8. (a) R. Barone and M. Chanon, <u>Heterocycles</u>, 1981, <u>16</u>, 1357; (b) It is interesting to note that a computer^{8a} has suggested 253 syntheses of ellipticine involving the breaking of two bonds.
- 9. M.J.E. Hewlins, A.-M. Oliveira-Campos, and P.V.R. Shannon, <u>Synthesis</u>, 1984, 289.
- R.N. Stillwell, Ph.D. Thesis, Harvard University, 1964; <u>Diss. Abs.</u>, 1964, <u>25</u>, 2769.
- A. Gouyette, R. Reynaud, J. Sadet, M. Baillagé, C. Gansser, S. Cros, F. Le Goffic, J.-B. Le Pecq, C. Paoletti, and C. Viel, <u>Eur. J. Med. Chem.-Chim.</u> Ther., 1980, 15, 503.
- (a) R.B. Miller and T. Moock, <u>Tetrahedron Lett.</u>, 1980, <u>21</u>, 3319; (b) R.B.
 Miller and J.G. Stowell, <u>J. Org. Chem.</u>, 1983, <u>48</u>, 886.
- 13. (a) C. Rivalle, C. Ducrocq, and E. Bisagni, <u>J. Chem. Soc., Perkin Trans. 1</u>,
 1979, 138; (b) C. Ducrocq, E. Bisagni, C. Rivalle, and J.-M. Lhoste, <u>J. Chem.</u>
 <u>Soc., Perkin Trans. 1</u>, 1979, 142.

- C. Ducrocq, F. Wendling, M. Tourbez-Perrin, C. Rivalle, P. Tambourin, F. Pochon, E. Bisagni, and J.-C. Chermann, <u>J. Med. Chem</u>., 1980, 23, 1212.
- 15. J.-C. Chermann, J. Gruest, L. Montagnier, F. Wendling, P. Tambourin, M. Tourbez-Perrin, F. Pochon, C. Ducrocg, C. Rivalle, and E. Bisagni, <u>Compt.</u> <u>rend.</u>, 1977, <u>285D</u>, 945.
- C. Rivalle, C. Ducrocg, J.-M. Lhoste, and E. Bisagni, <u>J. Org. Chem.</u>, 1980, <u>45</u>, 2176.
- C. Rivalle, C. Ducrocq, J.-M. Lhoste, F. Wendling, E. Bisagni, and J.-C. Chermann, <u>Tetrahedron</u>, 1981, <u>37</u>, 2097.
- P.A. Cranwell and J.E. Saxton, <u>J. Chem. Soc.</u>, 1962, 3482. See also reference 7 for a discussion of this chemistry.
- E. Bisagni, C. Ducrocq, J.-M. Ihoste, C. Rivalle, and A. Civier, <u>J. Chem.</u> Soc., Perkin Trans. 1, 1979, 1706.
- 20. E. Wenkert and K.G. Dave, J. Am. Chem. Soc., 1962, 84, 94.
- 21. (a) C. Rivalle, F. Wendling, P. Tambourin, J-M. Lhoste, E. Bisagni, and J-C. Chermann, <u>J. Med. Chem.</u>, 1983, <u>26</u>, 181; (b) R. Lidereau, J-C. Chermann, J. Gruest, L. Montagnier, C. Ducrocq, C. Rivalle, and E. Bisagni, <u>Bull. Cancer</u>, 1980, <u>67</u>, 1.
- (a) D. Rousselle, J. Gilbert, and C. Viel, <u>Compt. rend.</u>, 1977, <u>284C</u>, 377; (b)
 J. Gilbert, D. Rousselle, C. Gansser, and C. Viel, <u>J. Het. Chem.</u>, 1979, <u>16</u>, 7.
 W. Borsche, A. Witte, and W. Bothe, <u>Ann.</u>, 1908, <u>359</u>, 52.
- 24. B.M. Barclay and N. Campbell, J. Chem. Soc., 1945, 530.
- 25. A.J. Birch, A.H. Jackson, and P.V.R. Shannon, <u>J. Chem. Soc., Perkin Trans.</u> 1, 1974, 2185.
- 26. L.K. Dalton, S. Demerac, and T. Teitei, Aust. J. Chem., 1969, 22, 185.
- R.W. Guthrie, A. Brossi, F.A. Mennona, J.G. Mullin, R.W. Kierstead, and E. Grunberg, <u>J. Med. Chem.</u>, 1975, <u>18</u>, 755.
- 28. (a) C. Gansser, X. Leveque, M. Plat, C. Viel, C. Merienne, C. Malvy, and S. Cros, <u>Farm. Ed. Sci.</u>, 1982, <u>37</u>, 283; (b) C. Gansser, C. Viel, C. Malvy, and S. Cros, <u>Farm. Ed. Sci.</u>, 1980, <u>35</u>, 887.
- 29. (a) A.H. Jackson, P.R. Jenkins, and P.V.R. Shannon, <u>J. Chem. Soc., Perkin</u> <u>Trans. 1</u>, 1977, 1698; (b) M.J.E. Hewlins, A.H. Jackson, A-M. Oliveira-Campos, and P.V.R. Shannon, <u>Chem. Ind.</u>, 1981, 338; (c) M.J.E. Hewlins, A.H. Jackson, A-M. Oliveira-Campos, and P.V.R. Shannon, <u>J. Chem. Soc., Perkin Trans. 1</u>,

1981, 2906; (d) Y. Murakami, Y. Yokoyama, and N. Okuyama, <u>Tetrahedron Lett.</u>, 1983, 24, 2189.

- T. Sugasawa, M. Adachi, K. Sasakura, and A. Kitagawa, <u>J. Org. Chem.</u>, 1979, <u>44</u>, 578.
- 31. J-Y. Lallemand, P. Lemaitre, L. Beeley, P. Lesca, and D. Mansuy, <u>Tetrahedron</u> Lett., 1978, 1261.
- 32. (a) V. Peinhold, L. Bittman, R. Bruni, K. Thrun, and D. Silveira, Proc. Am. Assoc. Cancer Research, 1975, <u>16</u>, 135; (b) P. Lesca, P. Lecointe, C. Paoletti, and D. Mansuy, <u>Compt. rend.</u>, 1976, <u>282D</u>, 1457; (c) V.N. Reinhold and R.J. Bruni, <u>Biomed. Mass Spectrom.</u>, 1976, <u>3</u>, 335; (d) P. Lesca, P. Lecointe, C. Paoletti, and D. Mansuy, <u>Biochem. Pharmacol.</u>, 1977, <u>26</u>, 2169; (e) C. Paoletti, P. Lecointe, P. Lesca, S. Cros, D. Mansuy, and N. Dat-Xuong, <u>Biochimie</u>, 1978, <u>60</u>, 1003.
- 33. G.N. Taylor, J. Chem. Res., S, 1981, 332.
- 34. (a) Y. Oikawa, M. Tanaka, H. Hirasawa, and O. Yonemitsu, <u>Heterocycles</u>, 1981,
 <u>15</u>, 207; (b) Y. Oikawa, M. Tanaka, H. Hirasawa, and O. Yonemitsu, <u>Chem. Pharm.</u>
 <u>Bull. Japan</u>, 1981, <u>29</u>, 1606.
- 35. Yonemitsu has previously reported the synthesis of <u>la</u> and <u>2</u> by a similar route: Y. Oikawa and O. Yonemitsu, <u>J. Chem. Soc., Perkin Trans. 1</u>, 1976, 1479.
- 36. Y. Oikawa, H. Hirasawa, and O. Yonemitsu, Tetrahedron Lett., 1978, 1759.
- 37. J. Bergman and H. Goonewardena, Acta Chem. Scand., 1980, B34, 763.
- 38. J. Rergman and R. Carlsson, Tetrahedron Lett., 1977, 4663.
- R.B. Woodward, G.A. Iacobucci, and F.A. Hochstein, <u>J. Am. Chem. Soc.</u>, 1959, <u>81</u>, 4434.
- 40. W.E. Noland, L.R. Smith, and K.R. Rush, J. Org. Chem., 1965, 30, 3457.
- 41. J. Bergman and R. Carlsson, Tetrahedron Lett., 1978, 4051.
- R. Carlsson, Ph.D. Dissertation, Royal Institute of Technology, Stockholm, 1978.
- 43. J. Bergman and R. Carlsson, Tetrahedron Lett., 1978, 4055.
- 44. S. Takano, K. Yuta, S. Hatakeyama, and K. Ogasawara, <u>Tetrahedron Lett.</u>, 1979, 369.
- 45. (a) M.J. Wanner, G-J. Koomen, and U.K. Pandit, <u>Heterocycles</u>, 1982, <u>17</u>, 59; (b)
 M.J. Wanner, G-J. Koomen, and U.K. Pandit, <u>Heterocycles</u>, 1982, <u>19</u>, 2295; (c)
 M.J. Wanner, G-J. Koomen, and U.K. Pandit, <u>Tetrahedron</u>, 1983, <u>39</u>, 3673.

- 46. D.D. Weller and D.W. Ford, Tetrahedron Lett., 1984, 25, 2105.
- 47. S. Kano, E. Sugino, S. Shibuya, and S. Hibino, J. Org. Chem., 1981, 46, 2979.
- Kano, N. Mochizuki, S. Hibino, and S. Shibuya, <u>J. Org. Chem.</u>, 1982, <u>47</u>, 3566.
- 49. M. Watanabe and V. Snieckus, J. Am. Chem. Soc., 1980, 102, 1457.
- 50. J.P. Kutney, M. Noda, N.G. Lewis, B. Monteiro, D. Mostowicz, and B.R. Worth, Can. J. Chem., 1982, <u>60</u>, 2426.
- 51. F. Le Goffic, A. Gouyette, and A. Ahond, Tetrahedron, 1973, 29, 3357.
- 52. (a) D.A. Taylor, M.M. Baradarani, S.J. Martinez, and J.A. Joule, <u>J. Chem.</u> <u>Res., Synop.</u>, 1979, 387; <u>J. Chem. Res., Miniprint</u>, 1979, 4801; (b) D.A. Taylor and J.A. Joule, <u>J. Chem. Soc., Chem. Comm.</u>, 1979, 642.
- 53. W.R. Ashcroft, M.G. Beal, and J.A. Joule, <u>J. Chem. Soc., Chem. Comm.</u>, 1981, 994.
- 54. W.R. Ashcroft, M.G. Beal, and J.A. Joule, <u>J. Chem. Soc., Perkin Trans. 1</u>, 1981, 3012.
- D.I.C. Scopes, M.S. Allen, G.J. Hignett, N.D.V. Wilson, M. Harris, and J.A. Joule, <u>J. Chem. Soc., Perkin Trans. 1</u>, 1977, 2376.
- 56. M.M. Cooper, G.J. Hignett, and J.A. Joule, <u>J. Chem. Soc., Perkin Trans. 1</u>, 1981, 3008.
- M.G. Beal, W.R. Ashcroft, M.M. Cooper, and J.A. Joule, <u>J. Chem. Soc., Perkin</u> <u>Trans. 1</u>, 1982, 435.
- 58. M.M. Cooper, G.J. Hignett, R.F. Newton, J.A. Joule, M. Harris, and J.D. Hinchley, J. Chem. Soc., Chem. Comm., 1977, 432.
- 59. M. Driver, I.T. Matthews, and M. Sainsbury, <u>J. Chem. Soc., Perkin Trans. 1</u>, 1979, 2506.
- 60. M. Sainsbury, B. Webb, and R.F. Schinazi, <u>J. Chem. Soc., Perkin Trans. 1</u>, 1975, 289.
- 61. K.N. Kilminster and M. Sainsbury, J. Chem. Soc., Perkin Trans. 1, 1972, 2264.
- 62. M. Sainsbury and R.F. Schinazi, J. Chem. Soc., Chem. Comm., 1975, 540.
- 63. M. Sainsbury and R.F. Schinazi, J. Chem. Soc., Perkin Trans. 1, 1976, 1155.
- 64. D. Dolman and M. Sainsbury, <u>Tetrahedron Lett.</u>, 1981, <u>22</u>, 2119.
- 65. M.M. Chien and J.P. Rosazza, Drug Metabol. Dispos., 1979, 7, 211.
- 66. M. Sainsbury and D.K. Weerasinghe, J. Chem. Soc., Chem. Comm., 1981, 630.
- 67. M. Sainsbury, D.K. Weerasinghe, and D. Dolman, <u>J. Chem. Soc.</u>, Perkin Trans. 1, 1982, 587.

- 68. A. Kubo and T. Nakai, Synthesis, 1980, 365.
- 69. A.P. Kozikowski and N.M. Hasan, J. Org. Chem., 1977, 42, 2039.
- 70. S. Kano, E. Sugino, and S. Hibino, <u>Heterocycles</u>, 1982, <u>19</u>, 1673.
- 71. (a) F. Nivoliers, A. Decormeille, A. Godard, and G. Quéguiner, <u>Tetrahedron</u> <u>Lett.</u>, 1980, <u>21</u>, 4485; (b) A. Decormeille and G. Quéguiner, <u>J. Het. Chem.</u>, 1976, 13, 387.
- 72. V.V. Kononova and A.A. Semenov, Khim. Geterotsikl. Soedin., 1982, 1211.
- 73. (a) M.G. Saulnier and G.W. Gribble, <u>J. Org. Chem.</u>, 1982, <u>47</u>, 2810; (b) Unpublished results from our laboratory.
- 74. M.G. Saulnier and G.W. Gribble, J. Org. Chem., 1983, 48, 2690.
- 75. M.G. Saulnier and G.W. Gribble, Tetrahedron Lett., 1983, 24, 3831.
- 76. M.G. Saulnier and G.W. Gribble, J. Org. Chem., 1982, 47, 757.
- 77. C. May and C.J. Moody, J. Chem. Soc., Chem. Comm., 1984, 926.
- 78. G.W. Gribble, M.G. Saulnier, M.P. Sibi, and J.A. Obaza-Nutaitis, <u>J. Org.</u> Chem., 1984, 49, 4518.
- 79. M.G. Saulnier and G.W. Gribble, Tetrahedron Lett., 1983, 24, 5435.
- 80. P. Potier and M.-M. Janot, <u>C.R. Acad. Sci., Paris</u>, 1973, <u>276C</u>, 1727.
- 81. (a) R. Besselievre and H.-P. Husson, <u>Tetrahedron Suppl. No. 1</u>, 1981, <u>37</u>, 241;
 (b) R. Besselievre and H.-P. Husson, Tetrahedron Lett., 1976, 1873.
- 82. A. Jackson, N.D.V. Wilson, A.J. Gaskell, and J.A. Joule, <u>J. Chem. Soc.</u>, (C), 1969, 2738.
- 83. R.H. Burnell and D.D. Casa, Can. J. Chem., 1967, 45, 89.
- B.S. Grierson, M. Harris, and H.-P. Husson, <u>J. Am. Chem. Soc.</u>, 1980, <u>102</u>, 1064.
- 85. T. Naito, N. Iida, I. Ninomiya, J. Chem. Soc., Chem. Comm., 1981, 44.
- 86. J. Schmutz and H. Wittwer, Helv. Chim. Acta, 1960, 43, 793.

Received, 17th December, 1984