

STRUCTURES OF SCULPONEATIN A, B, AND C, THREE NEW DITERPENOIDS HAVING UNIQUE  
ACETAL STRUCTURES FROM *RABDOSIA SCULPONEATA*

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**Abstract**— Three novel diterpenoids, sculponeatin A, B, and C have been isolated from  
the leaves of *Rabdosia sculponeata* (Vaniot) Hara and their structures were deduced  
from spectroscopic and chemical evidence.

In the continuing search for biologically active principles of *Rabdosia* plants (Labiatae), we have  
isolated three new novel diterpenoids, sculponeatin A (1), B (2), and C (7) together with known  
enmein<sup>1</sup> from the leaves of *Rabdosia sculponeata* (Vaniot) Hara<sup>2</sup> collected in Yunnan, China.  
Sculponeatin A (1) and C (7) showed an antineoplastic activity on some transplantable animal tumors  
such as P-388, S-180 and ECS.<sup>3</sup> This paper describes the structure elucidation of new compounds  
having a unique acetalic structure.

Sculponeatin A (1), C<sub>20</sub>H<sub>24</sub>O<sub>6</sub>, mp > 300°C, [ $\alpha$ ]<sub>D</sub><sup>25</sup> -139° (c 0.21, C<sub>5</sub>H<sub>5</sub>N), contains a five membered  
ketone conjugated with an exo-methylene group [ $\lambda_{\max}$  (EtOH) 230 nm ( $\epsilon$  7030);  $\nu_{\max}$ <sup>4</sup> 1738 and 1637 cm<sup>-1</sup>;  
<sup>1</sup>H NMR<sup>5</sup>  $\delta_H$  5.38 and 6.04 (each 1H, br.s); <sup>13</sup>C NMR (d<sub>6</sub>-DMSO)<sup>5</sup>  $\delta_C$  118.1 (t), 149.9 (s)(exo-methylene)  
and 200.0 (ketone)], a  $\delta$ -lactone [ $\nu_{\max}$  1700 cm<sup>-1</sup>;  $\delta_H$  5.62 (1H, dd,  $J$  6 and 10 Hz);  $\delta_C$  170.4 (s,  
-OCO-), 78.1 (d, >CH-OCO-)], an acetal [ $\delta_H$  6.13 (1H, d,  $J$  5 Hz);  $\delta_C$  110.3 (d)], a secondary  
hydroxyl group [ $\delta_H$  6.95 (1H, br.d, OH) and 4.58 (1H, dd,  $J$  4 and 5 Hz, on addition of D<sub>2</sub>O);  
 $\delta_C$  64.1 (d)], two oxygenated methylene groups [ $\delta_H$  3.47 and 4.08 (each 1H, ABq,  $J$  8 Hz), and  
4.18 and 4.34 (each 1H, ABq,  $J$  10 Hz);  $\delta_C$  71.3 (t) and 76.1 (t)], and a tertiary methyl group  
[ $\delta_H$  1.06 (3H, s);  $\delta_C$  30.0 (q)]. The <sup>13</sup>C NMR spectrum of 1 (Table I) also showed the presence of  
additional groups: four methylenes, three methines, and three quaternary carbons. Precise <sup>1</sup>H-  
spin-spin decoupling experiments (The assignments based on the results were summarized in Table II)  
revealed the presence of the partial structures (A) and (B). From the above mentioned results,

together with the consideration of the structures of diterpenoids isolated so far from the genus *Rabdosa*,<sup>6</sup> we presumed the partial structure (C) having B-*seco-ent*-kaurene skeleton. Consequently, only one hydroxyl group in 1 should be located at C-11. The configuration was assigned to be  $\beta$  and the C-ring was presumed to take a boat conformation from the following spectral data: The signal due to 11-H appeared as dd ( $J_{9,11}$  4 Hz,  $J_{12\alpha,11}$  5 Hz,  $J_{12\beta,11}$  0 Hz) and the signals due to 14 $\beta$ -H ( $\delta_H$  3.62) and 1 $\beta$ -H ( $\delta_H$  5.62) suffered abnormal downfield shift. Acetylation [ $Ac_2O-C_5H_5N$ ] of 1 gave the monoacetate (3),  $C_{22}H_{26}O_7$ , mp > 300° C [ $\delta_H$  2.97 (d,  $J$  11 Hz, 14 $\beta$ -H)]. Oxidation of 1 with Beckmann mixture ( $K_2Cr_2O_7-H_2SO_4$ ) afforded the dione (4),  $C_{20}H_{22}O_6$ , mp 280-283° C. In the  $^1H$  NMR spectrum of 4, the signals due to 9 $\alpha$ -H, 5 $\beta$ -H, 12 $\alpha$ -H, and 14 $\beta$ -H shifted to  $\delta_H$  3.88 (s), 3.14 (d,  $J$  5 Hz), 2.92 (d,  $J$  10 Hz) and 2.31 (d,  $J$  11 Hz), respectively. On the other hand, irradiation at the frequency of 11-H, an NOE (18%) was observed for 5-H in the  $^1H$  NMR spectrum of 1. Accordingly, sculponeatin A has an almost same structure as that of nodosin (6)<sup>7</sup> except for hemiacetal portion. The acetal structure (D) was deduced for sculponeatin A (1) from the following discussion: One of the gem-dimethyl groups at C-4 was presumed to be oxygenated from the fact that sculponeatin A has only one tertiary methyl group and two oxygenated methylene groups in the structure; In the  $^1H$  NMR spectrum, 1 showed a coupling constant,  $J_{5,6}$  5 Hz, in contrast to that ( $J_{5,6} = 0$  Hz) of nodosin (6); One more ring is needed to fulfill the degree of unsaturation; An NOE (11%) was observed for 5-H and no NOE was observed for 20-H<sub>2</sub>, on irradiation of the frequency of the tertiary methyl group. Additionally, the dihydro-compound (5) showed a negative Cotton effect in CD (MeOH) [ $\Delta\epsilon_{306}$  -0.49].<sup>8</sup> On the basis of these data, the structure of sculponeatin A was elucidated as 1.

Sculponeatin B (2),  $C_{20}H_{26}O_6$ , mp 244-246° C,  $[\alpha]_D^{25}$  -109° (c 0.14,  $C_5H_5N$ ) showed no absorption maximum above 220 nm in the UV and absorptions at 3420, 1730, and 1640  $cm^{-1}$  in the IR spectra. The  $^1H$ - (Table II) and  $^{13}C$  NMR ( $d_6$ -DMSO) (Table I) spectra suggested that sculponeatin B is an allylic alcohol in which the ketone at C-15 in 1 is reduced to an alcohol. In fact, new NMR signals [ $\delta_H$  5.71 (br.s,  $>CH-OH$ ) and  $\delta_C$  77.6 (d)] were observed in 2 instead of the signal due to a ketone group in 1 and oxidation of 2 with Beckmann mixture gave the diketone (4). The configuration of the hydroxyl group at C-15 was presumed as  $\alpha$  from the fact that the signal due to 9-H resonated ca. 0.55 ppm downfield compared with that of 1 and was confirmed by obtaining dihydrosculponeatin A (5) when 2 was treated under the conditions of Garryfoline-Çuauchichicine rearrangement.<sup>9</sup> On the basis of these findings, the structure of sculponeatin B was deduced as 2.

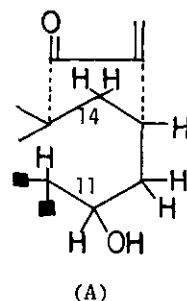
Sculponeatin C (7),  $C_{20}H_{24}O_6$ , mp 292-294° C,  $[\alpha]_D^{24}$  -163° (c 0.21,  $C_5H_5N$ ), showed an absorption maximum at 231 nm (c 7410) in the UV and absorptions at 3450, 1745, 1702, and 1640  $cm^{-1}$  in the IR spectra. The  $^1H$  and  $^{13}C$  NMR data were summarized in the Table II and I, respectively. The above mentioned spectral data are very similar to those of sculponeatin A (1). Treatment of dihydro-compound

Table I.  $^{13}\text{C}$  NMR Data of Sculponeatin A(1), B(2), and C(7)\*

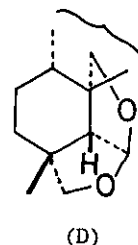
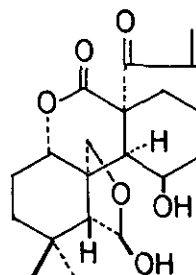
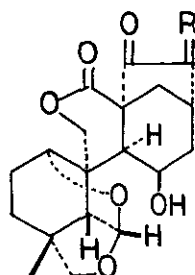
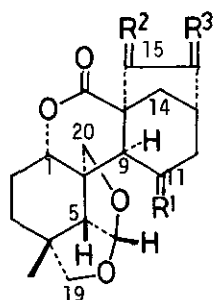
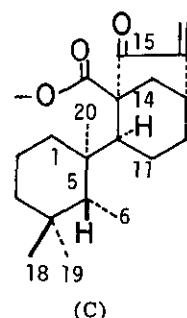
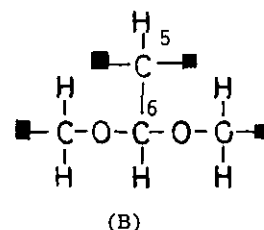
	A(1)	B(2)	C(7)		A(1)	B(2)	C(7)
C-1	78.1(d)	78.4(d)	77.5(d)	C-11	64.1(d)	63.7(d)	66.3(d)
C-2	22.2(t)	22.4(t)	29.8(t)**	C-12	39.9(t)	44.0(t)	41.0(t)
C-3	28.7(t)	28.9(t)	30.5(t)**	C-13	34.0(d)	36.2(d)	35.2(d)
C-4	40.8(s)	40.9(s)	41.0(s)	C-14	33.1(t)	32.9(t)	33.4(t)
C-5	52.8(d)	52.9(d)	52.4(d)	C-15	200.0(s)	77.6(d)	200.4(s)
C-6	110.3(d)	110.2(d)	107.9(d)	C-16	149.9(s)	158.2(s)	151.1(s)
C-7	170.4(s)	174.3(s)	171.5(s)	C-17	118.1(t)	107.8(t)	117.4(t)
C-8	54.8(s)	50.6(s)	54.6(s)	C-18	30.0(q)	30.2(q)	27.8(q)
C-9	44.7(d)	38.6(d)	40.5(d)	C-19	71.3(t)	72.5(t)	72.2(t)
C-10	49.2(s)	48.9(s)	51.0(s)	C-20	76.1(t)	77.0(t)	83.9(t)

\*  $\delta$  (multiplicity), A(1) and B(2) in  $\text{d}_6\text{-DMSO}$ , C(7) in  $\text{C}_5\text{D}_5\text{N}$ 

\*\* may be reversed.

Table II.  $^1\text{H}$  NMR Data of Sculponeatin A(1), B(2), and C(7)\*

	A(1)	B(2)	C(7)
1-H	5.62(dd, 6, 10)	5.88(dd, 6, 10)	4.64(ddd, 1, 3, 3)
5-H	2.90(d, 5)	3.00(d, 5)	2.75(dd, 1, 4)
6-H	6.13(d, 5)	6.14(d, 5)	5.87(d, 4)
9-H	2.24(d, 4)	2.79(d, 4)	2.60(m)
11-H	4.58(dd, 4, 5)**	4.56(dd, 4, 4)**	4.44(dd, 5, 5)**
12 $\alpha$ -H			1.81(dd, 5, 15)
12 $\beta$ -H	2.46(dd, 9, 14)	2.45(dd, 8, 14)	2.47(dd, 9, 15)
13-H	3.18(dd, 5, 9)	2.94(dd, 5, 8)	3.16(dd, 4, 9)
14 $\alpha$ -H			2.10(dd, 4, 11)
14 $\beta$ -H	3.62(d, 11)	3.16(d, 11)	3.61(d, 11)
15-H		5.71(br. s)	
17-Ha	5.38(br. s)	5.23(br. s)	5.42(br. s)
17-Hb	6.04(br. s)	5.52(br. s)	6.10(br. s)
19-Ha	3.47(d, 8)	3.46(d, 8)	3.70(d, 8)
19-Hb	4.08(d, 8)	4.05(d, 8)	3.89(d, 8)
20-Ha	4.18(d, 10)	4.26(d, 9)	4.36(dd, 2, 12)
20-Hb	4.34(d, 10)	4.53(d, 9)	5.33(d, 12)
Me(18)	1.06(s)	1.09(s)	1.11(s)
11-OH	6.95(br. d)	6.48(br. d)	7.14(br. d)

\*  $\delta$  (multiplicity, coupling const. in Hz), in  $\text{C}_5\text{D}_5\text{N}$ \*\* after  $\text{D}_2\text{O}$  treatment1:  $\text{R}^1=\alpha\text{-H}$ ,  $\beta\text{-OH}$ ;  $\text{R}^2=\text{O}$ ;  $\text{R}^3=\text{CH}_2$ 2:  $\text{R}^1=\alpha\text{-H}$ ,  $\beta\text{-OH}$ ;  $\text{R}^2=\alpha\text{-OH}$ ,  $\beta\text{-H}$ ;  $\text{R}^3=\text{CH}_2$ 3:  $\text{R}^1=\alpha\text{-H}$ ,  $\beta\text{-OAc}$ ;  $\text{R}^2=\text{O}$ ;  $\text{R}^3=\text{CH}_2$ 4:  $\text{R}^1=\text{R}^2=\text{O}$ ;  $\text{R}^3=\text{CH}_2$ 5:  $\text{R}^1=\alpha\text{-H}$ ,  $\beta\text{-OH}$ ;  $\text{R}^2=\text{O}$ ;  $\text{R}^3=\alpha\text{-Me}$ ,  $\beta\text{-H}$ 7:  $\text{R}=\text{CH}_2$ 8:  $\text{R}=\alpha\text{-Me}$ ,  $\beta\text{-H}$ 

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(8), obtained by catalytic hydrogenation, with 15% HCl-MeOH gave dihydrosculponeatin A (5), showing that sculponeatin C has the same oxygenation pattern with that of sculponeatin A (1). In the  $^1\text{H}$  NMR spectrum of 7, a signal ( $\delta_{\text{H}}$  4.64, 1H, ddd,  $J$  1, 1, and 3 Hz) due to a proton attached to a carbon having an oxygen function was observed instead of the signal [ $\delta_{\text{H}}$  5.62, 1H, dd,  $J$  6 and 10 Hz, 1-H] found in that of 1. A long-range coupling was also observed between 9-H and 20-H<sub>1</sub> via W letter theory. On irradiation of 4 $\beta$ -methyl group, an NOE (14%) was observed for 9-H. From the above mentioned data, a  $\delta$ -lactone was presumed to be formed through C-7 and C-20. Five membered ether linkage was deduced to be formed between C-1 and C-6 from the fact that the signal at  $\delta$  4.64 was assigned to an equatorial hydrogen from its coupling constants and was shown to couple with 5-H ( $J$  1 Hz) via W letter theory. On the basis of these findings, the structure of sculponeatin C was elucidated as 7.

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- 4) IR spectra were recorded for KBr discs.
- 5) Unless otherwise noted, NMR spectra were recorded for  $\text{C}_6\text{D}_5\text{N}$  solution using tetramethylsilane as internal standard and the chemical shifts were expressed in  $\delta$  (ppm). The  $^{13}\text{C}$  NMR assignments were based on a combination of PND, off-resonance decoupling, single frequency selective decoupling, and comparisons with each other.
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