

ISOLATION OF PHOTO-DIELS-ALDER MONO-ADDUCTS OF 4,6-DIMETHYL-
2-PYRONE AND FORMATION OF CROSS-ADDUCTS

Tetsuro Shimo, Hiroyuki Yoshimura, Hisako Uemura,
and Kenichi Somekawa*

Department of Applied Chemistry, Faculty of Engineering,
Kagoshima University, Kagoshima 890, Japan

Otohiko Tsuge

Research Institute of Industrial Science, Kyushu University,
Kasuga 816, Japan

Abstract - The Diels-Alder mono-adducts between 4,6-dimethyl-
2-pyrone and olefinic dienophiles could be isolated from the
low temperature photochemical reactions. The adducts reacted
with second olefins to afford cross-adducts with the concurrent
decarboxylation. Irradiation of the cross-adducts to
p-benzoquinone gave cage compounds.

The Diels-Alder (DA) reactions of 2-pyrones have been used as the key steps in
syntheses of colchicine,¹ barrelene² and so on, by taking advantage of easy
decarboxylation of the mono-adducts. The utilization is limited, however,
because the DA reactions take place with difficulty owing to the aromaticity
of 2-pyrones, and the resulting mono-adducts are thermally susceptible to
decarboxylation and dehydrogenation to afford bis-adducts³ and benzene
derivatives.⁴ We have previously reported that, in photochemical reactions
with olefins at room temperature, 2-pyrones having electron-withdrawing
substituents gave stable [4+2]cycloadducts,⁵ whereas ones bearing
electron-donating substituents afforded benzene derivatives with the concurrent
decarboxylation.⁶ Thus, the latter reactions at low temperature might be
expected to give [4+2]cycloadducts.

Although the mono-adducts derived from 2-pyrones are diene-equivalents, the
investigation of cross-bis[4+2]cycloadditions has been very limited.⁷ It is

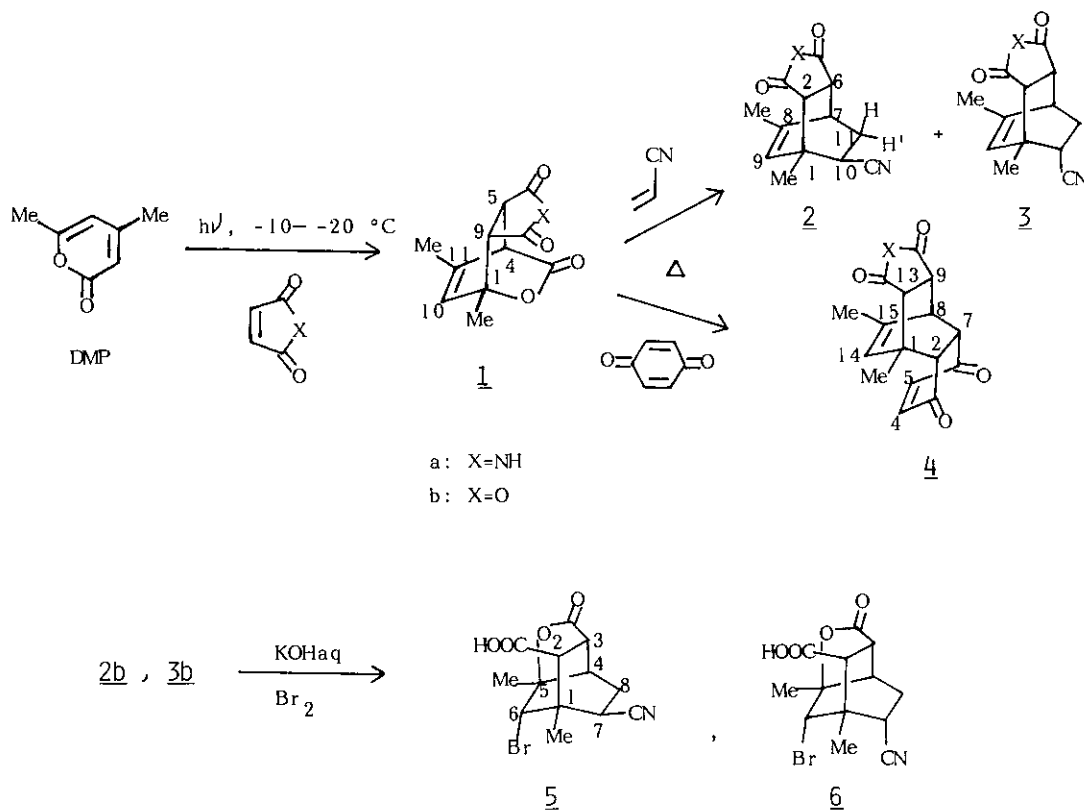
required that the mono-adducts have moderate lability in order to use them for the cross-[4+2]cycloadditions. The mono-adducts derived from 2-pyrones bearing electron-donating groups seemed to be suitable for this purpose.

The present paper describes the isolation of mono-adducts of 4,6-dimethyl-2-pyrone (DMP) with electrophilic olefins from the low temperature photoreaction, and the synthesis of the cross-adducts, some of which were converted into cage compounds, by using mono-adducts.

A mixture of DMP (24 mmol) and maleimide (24 mmol) in the presence of benzophenone (8.8 mmol) as a sensitizer in acetone (200 ml) was irradiated with a 400W high-pressure mercury lamp, under nitrogen, at -10 - -20 °C for 2 h to afford a DA adduct 1a, mp 101-104 °C (dec), in 60% yield. The reaction of DMP with maleic anhydride under the similar conditions gave also a DA adduct 1b, mp 90-93 °C (dec), in 52% yield. Adducts, 1a and 1b, were isolated and purified as follows: After irradiation, the solvent was removed in vacuo below 10 °C. To the residue was added chloroform (80 ml) and the solution was cooled in an ice-bath. The formed precipitate (1) was filtered and recrystallized from chloroform. The adduct 1b was heated at 40 °C for 2 h in acetonitrile to give a cyclohexadiene derivative quantitatively. The adduct 1a is rather stable, but it was also decarboxylated by heating at 70 °C to give a cyclohexadiene. The adducts 1a, 1b and the cyclohexadienes were not formed in the thermal reactions.

The exo configuration of both the adducts 1 was confirmed on the basis of spectral data.^{8,9}

We next investigated reactions of 1 with second olefinic dienophiles leading to the formation of cross-adducts (Scheme 1). The adduct 1a (1.6 mmol) was allowed to react with acrylonitrile (16 mmol) in acetonitrile (30 ml) under reflux for 20 h to give cross-adducts 2a (mp 216-219 °C, 26% yield) and 3a (mp 212-213 °C, 20% yield).¹⁰ The similar reaction of 1b with acrylonitrile for 30 h gave the corresponding cross-adducts 2b (mp 143-144 °C, 55% yield) and 3b (mp 158-161 °C, 38% yield),¹¹ which were bromolactonized to give cyclic compounds,¹² 5 (mp 256-259 °C, 69% yield) and 6 (mp 117-120 °C, 72% yield) respectively, indicating the endo configuration.¹³ Thus, both 2a and 3a were assumed to be the endo configurations of the imido group. The positions of the cyano groups in 2 and 3 were confirmed by the ¹H-NMR spectra; the signals of 10-H showed doublet-doublet pattern (J=5-7 and 9-10 Hz) and their stereochemistry was confirmed by the chemical shift

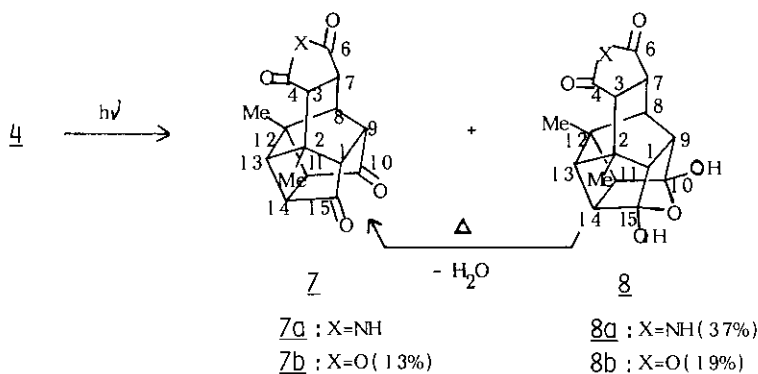


Scheme 1.

deviation between 2 and 3 at 2-H ($\Delta\delta=0.13-0.45$ ppm) owing to magnetic anisotropy of cyano groups.

Similarly, the adducts 1a (0.9 mmol) and 1b (0.9 mmol) reacted with *p*-benzoquinone (1.8 mmol) in acetonitrile (30 ml) under reflux for 18 h to give the cross-adducts 4a (mp>300°C, 96% yield) and 4b (mp 238-240 °C, 70% yield), respectively, whose structural elucidation was accomplished on the basis of spectral data.¹⁴ The endo-endo configuration of 4 was confirmed by the result of photoreaction mentioned below as well as by the inspection of ¹H-NMR data.

The photochemical intramolecular cyclization of the cross-adducts 4 might be expected to give the corresponding tricyclic cage γ -diketones. In fact, irradiation of a solution of 4a (2.1 mmol) in dichloromethane (100 ml) with a 400W high-pressure mercury lamp for 1 h afforded the cage γ -diketone hydrate 8a, mp>300 °C, which was converted to the diketone 7a, mp>300 °C, at 50 °C in vacuo. The similar photoreaction of 4b gave again the corresponding cage



Scheme 2.

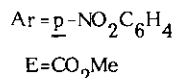
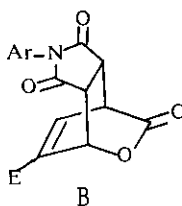
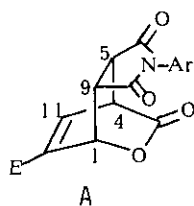
γ -diketone 7b, mp 230 °C (sub), and its hydrate 8b, mp 210 °C (sub) (Scheme 2). Structural elucidation of the cage compounds, 7 and 8, was accomplished on the basis of the elemental analyses and spectral data¹⁵; ¹H- and ¹³C-NMR spectra of 7 and 8 have no olefinic signals.

ACKNOWLEDGEMENT

This work was supported in part by a grant from Saneyoshi Scholarship Foundation.

REFERENCES

1. J. Schreiber, W. Leimgruber, M. Pesaro, P. Schudel, T. Threlfall, and A. Eschenmoser, Helv. Chim. Acta, 1961, 44, 540.
2. H. E. Zimmerman, G. L. Grunewald, R. M. Paufler, and M. A. Sherwin, J. Am. Chem. Soc., 1969, 91, 2330.
3. N. P. Shusharina, T. L. Nesterova, O. V. Polyakova, Zh. Org. Khim., 1980, 16, 1285.
4. T. Shimo, K. Somekawa, and S. Kumamoto, Nippon Kagaku Kaishi 1982, 1927.
5. T. Shimo, K. Somekawa, M. Sato, and the late S. Kumamoto, Nippon Kagaku Kaishi, 1984, 1927.
6. T. Shimo, K. Somekawa, S. Kumamoto, Nippon Kagaku Kaishi, 1983, 394.
7. N. P. Shusharina, Russ. Chem. Rev., 1974, 43, 851.
8. The formation of an exo-[4+2]adduct like 1 was also observed in the sensitized photoreaction of methyl 2-pyrone-5-carboxylate (MP). The sensitized photoreaction of MP with N-(p-nitrophenyl)maleimide under similar conditions gave the exo-adduct A, mp 220-222 °C, in 50% yield, whereas the endo-adduct B, mp 185-



188 °C, was obtained in 10% yield from the thermal reaction at 80 °C for 32 h. The exo and endo configurations were assigned on the basis of the chemical shifts of 5- and 9-H in A and B, respectively.

A: $^1\text{H-NMR}$ ($\text{DMSO-}d_6$) δ 8.37, 7.44 (each 2H, d, ArH), 7.52 (1H, dd, $J=7.5$, 2.0 Hz, 11-H), 5.84 (1H, t, $J=2.0$ Hz, 1-H), 4.16 (1H, dd, $J=7.5$, 3.0 Hz, 4-H), 3.82 (1H, dd, $J=9.5$, 2.0 Hz, 9-H), 3.80 (3H, s, CH_3), 3.64 (1H, dd, $J=9.5$, 3.0 Hz, 5-H).

B: $^1\text{H-NMR}$ ($\text{DMSO-}d_6$) δ 8.42, 7.42 (each 2H, d, ArH), 7.64 (1H, dd, $J=7.0$, 3.0 Hz, 11-H), 6.05 (1H, dd, $J=4.0$, 3.0 Hz, 1-H), 4.24 (1H, dd, $J=7.0$, 4.0 Hz, 4-H), 4.18 (1H, dd, $J=8.0$, 4.0 Hz, 9-H), 3.97 (1H, dd, $J=8.0$, 4.0 Hz, 5-H), 3.78 (3H, s, CH_3).

The exo configurations of 1 were determined on the basis of the comparison of NMR data of 1 with those of A and B, considering the shielding effect of methyl group.¹⁶

1a: IR (KBr) 1780, 1750, 1720 cm^{-1} ; $^1\text{H-NMR}$ ($\text{DMSO-}d_6$) δ 11.28 (1H, bs, NH), 6.03 (1H, bs, 10-H), 3.52 (1H, bd, $J=3.0$ Hz, 4-H), 3.35 (1H, dd, $J=3.0$, 10.0 Hz, 5-H), 3.08 (1H, d, $J=10.0$ Hz, 9-H), 1.92, 1.74 (each 3H, s, CH_3); MS m/z 177 (M^+-CO_2).

1b: IR (KBr) 1870, 1790, 1750 cm^{-1} ; $^1\text{H-NMR}$ ($\text{DMSO-}d_6$) δ 6.04 (1H, bs, 10-H), 3.92 (1H, dd, $J=3.0$, 10.0 Hz, 5-H), 3.64 (1H, dd, $J=1.8$, 3.0 Hz, 4-H), 3.52 (1H, d, $J=10.0$ Hz, 9-H), 1.92, 1.72 (each 3H, s, CH_3); MS m/z 178 (M^+-CO_2).

9. The new compounds reported herein provided satisfactory elemental analyses.

10. 2a: IR (KBr) 2240, 1770, 1710 cm^{-1} ; $^1\text{H-NMR}$ ($\text{DMSO-}d_6$) δ 11.11 (1H, s, NH), 5.48 (1H, bs, 9-H), 3.12 (1H, dd, $J=8.0$, 3.5 Hz, 6-H), 2.77 (1H, d, $J=8.0$ Hz, 2-H), 2.77 (1H, bs, 7-H), 2.56 (1H, dd, $J=10.0$, 6.0 Hz, 10-H), 1.80 (2H, m, 11-H, 11-H'), 1.68 (3H, d, $J=2.0$ Hz, 8- CH_3), 1.45 (3H, s, 1- CH_3); MS m/z 230 (M^+).

3a: IR (KBr) 2240, 1770, 1710 cm^{-1} ; $^1\text{H-NMR}$ ($\text{DMSO-}d_6$) δ 11.12 (1H, s, NH), 5.52 (1H, bs, 9-H), 2.94 (1H, dd, $J=8.0$, 3.5 Hz, 6-H), 2.86 (1H, dd, $J=10.0$, 5.0

- Hz, 10-H), 2.84 (1H, bs, 7-H), 2.64 (1H, d, $J=8.0$ Hz, 2-H), 2.10 (1H, ddd, $J=12.5, 10.0, 2.0$ Hz, 11-H), 1.72 (3H, d, $J=2.0$ Hz, 8-CH₃), 1.51 (1H, ddd, $J=12.5, 5.0, 4.0$ Hz, 11-H'), 1.51 (3H, s, 1-CH₃); MS m/z 230 (M^+).
11. 2b: IR (KBr) 2250, 1840, 1780 cm^{-1} ; $^1\text{H-NMR}$ (CDCl_3) δ 5.58 (1H, bs, 9-H), 3.36 (1H, dd, $J=8.0, 2.5$ Hz, 6-H), 3.28 (1H, d, $J=8.0$ Hz, 2-H), 3.08 (1H, bs, 7-H), 2.33 (1H, dd, $J=9.0, 7.5$ Hz, 10-H), 1.94 (2H, m, 11-H, 11-H'), 1.78 (3H, d, $J=2.0$ Hz, 8-CH₃), 1.64 (3H, s, 1-CH₃); MS m/z 231 (M^+).
- 3b: IR (KBr) 2250, 1855, 1775 cm^{-1} ; $^1\text{H-NMR}$ (CDCl_3) δ 5.67 (1H, bs, 9-H), 3.21 (1H, dd, $J=8.0, 3.0$ Hz, 6-H), 3.11 (1H, bs, 7-H), 2.81 (1H, d, $J=8.0$ Hz, 2-H), 2.57 (1H, dd, $J=8.5, 5.5$ Hz, 10-H), 2.11 (1H, ddd, $J=12.0, 8.5, 2.0$ Hz, 11-H), 1.85 (3H, d, $J=2.0$ Hz, 8-CH₃), 1.75 (1H, ddd, $J=12.0, 5.5, 2.0$ Hz, 11-H'), 1.69 (3H, s, 1-CH₃); MS m/z 231 (M^+).
12. 5: IR (KBr) 2250, 1780, 1730 cm^{-1} ; $^1\text{H-NMR}$ ($\text{DMSO}-d_6$) δ 13.0 (1H, bs, COOH), 4.20 (1H, s, 6-H), 3.32 (1H, dd, $J=8.0, 3.0$ Hz, 7-H), 3.20 (1H, d, $J=6.5$ Hz, 2-H), 3.04 (1H, dd, $J=6.5, 2.0$ Hz, 3-H), 2.60 (1H, bs, 4-H), 2.28-2.12 (2H, m, 8-H, 8-H'), 1.45, 1.20 (each 3H, s, CH₃); MS m/z 327 (M^+).
- 6: IR (KBr) 2250, 1780, 1735 cm^{-1} ; $^1\text{H-NMR}$ ($\text{DMSO}-d_6$) δ 12.8 (1H, bs, COOH), 4.07 (1H, s, 6-H), 3.55 (1H, t, $J=8.0$ Hz, 7-H), 3.13 (1H, d, $J=6.0$ Hz, 2-H), 2.97 (1H, dd, $J=6.0, 2.0$ Hz, 3-H), 2.54 (1H, bs, 4-H), 2.27 (2H, dd, $J=8.0, 3.0$ Hz, 8-H, 8-H'), 1.48, 1.26 (each 3H, s, CH₃); MS m/z 327 (M^+).
13. H. Tomisawa and H. Hongo, Chem. Pharm. Bull., 1970, 18, 925.
14. 4a: IR (KBr) 1750, 1710, 1670 cm^{-1} ; $^1\text{H-NMR}$ ($\text{DMSO}-d_6$) δ 11.09 (1H, bs, NH), 6.68 (2H, s, 4-, 5-H), 5.66 (1H, s, 14-H), 3.36-3.04 (3H, m, 7-, 8-, 9-H), 2.76 (2H, d, $J_{2,7}=J_{9,13}=8.0$ Hz, 2-, 13-H), 1.55, 1.36 (each 3H, s, CH₃); MS m/z 285 (M^+).
- 4b: IR (KBr) 1850, 1780, 1675 cm^{-1} ; $^1\text{H-NMR}$ ($\text{DMSO}-d_6$) δ 6.72 (2H, s, 4-, 5-H), 5.59 (1H, s, 14-H), 3.62 (1H, dd, $J=8.0, 3.5$ Hz, 9-H), 3.36 (1H, dd, $J=8.0, 3.5$ Hz, 7-H), 3.22 (1H, m, 8-H), 3.22 (1H, d, $J=8.0$ Hz, 13-H), 2.88 (1H, d, $J=8.0$ Hz, 2-H), 1.57, 1.36 (each 3H, s, CH₃); MS m/z 286 (M^+).
15. 7a: IR (KBr) 1765, 1750, 1705 cm^{-1} ; $^1\text{H-NMR}$ ($\text{DMSO}-d_6$) δ 11.1 (1H, s, NH), 3.16 (1H, dd, $J=10.0, 3.5$ Hz, 7-H), 2.96 (1H, d, $J=10.0$ Hz, 3-H), 2.6-2.2 (6H m, 1-, 8-, 9-, 11-, 13-, 14-H), 1.18, 1.08 (each 3H, s, CH₃); MS m/z 285 (M^+).
- 8a: IR (KBr) 3400, 1765, 1700 cm^{-1} ; $^1\text{H-NMR}$ ($\text{DMSO}-d_6$) δ 11.1 (1H, s, NH), 6.94, 6.76 (each 1H, s, OH), 2.92 (1H, dd, $J=10.0, 3.5$ Hz, 7-H), 2.66 (1H, d,

$J=10.0$ Hz, 3-H), 2.2-1.8 (6H, m, 1-, 8-, 9-, 11-, 13-, 14-H), 1.28, 0.99 (each 3H, CH_3); ^{13}C -NMR (DMSO- d_6) δ 180.2, 179.5 (each s, C=O), 109.7, 106.1 (each s, 10-, 15-C), 59.2, 44.6 (each s, 2-, 12-C), 54.5, 51.1, 46.1, 42.8, 40.3, 39.5, 37.9 (each d, 1-, 3-, 7-, 8-, 11-, 13-, 14-C), 22.2, 19.7 (each q, CH_3); MS m/z 285 ($\text{M}^+-\text{H}_2\text{O}$).

7b: IR (KBr) 1860, 1780, 1755 cm^{-1} ; ^1H -NMR (DMSO- d_6) δ 3.60 (1H, dd, $J=11.0$, 3.5 Hz, 7-H), 3.38 (1H, d, $J=11.0$ Hz, 3-H), 2.98 (1H, ddd, $J=9.0$, 6.5, 3.0 Hz, 14-H), 2.8-2.28 (5H, m, 1-, 8-, 9-, 11-, 13-H), 1.20, 1.06 (each 3H, s, CH_3); MS m/z 286 (M^+).

8b: IR (KBr) 3400, 1860, 1775 cm^{-1} ; ^1H -NMR (DMSO- d_6) δ 7.01, 6.85 (each 1H, s, OH), 3.34 (1H, dd, $J=10.5$, 3.5 Hz, 7-H), 3.09 (1H, d, $J=10.5$ Hz, 3-H), 2.28 (1H, ddd, $J=10.0$, 4.0, 2.0 Hz, 14-H), 2.56-1.80 (5H, m, 1-, 8-, 9-, 11-, 13-H), 1.22, 0.97 (each 3H, s, CH_3); ^{13}C -NMR (DMSO- d_6) δ 174.4, 173.0 (each s, C=O), 110.0, 106.3 (each s, 10-, 15-C), 52.4, 44.9 (each s, 2-, 12-C), 54.7, 51.1, 46.3, 43.1, 40.3, 40.2, 39.6, 38.3 (each d, 1-, 3-, 7-, 8-, 11-, 13-, 14-C), 21.7, 19.5 (each q, CH_3); MS m/z 286 ($\text{M}^+-\text{H}_2\text{O}$).

16. K. Somekawa, T. Watanabe, and S. Kumamoto, Nippon Kagaku Kaishi, 1978, 412.

Received, 30th June, 1986