

PREPARATION OF 5-ETHOXYCARBONYL- AND 5-CYANO-5,6-DIHYDRO-3,7-DIPHENYL-4H-DIAZEPINES AND THEIR HALOGENATIONS

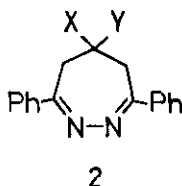
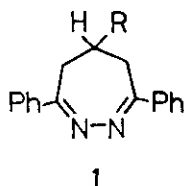
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**Abstract** — 5-Ethoxycarbonyl- and 5-cyano-5,6-dihydro-3,7-diphenyl-4H-1,2-diazepines were prepared and their halogenations were investigated under various conditions. The halogenation products were found to be strongly dependent upon the nature of 5-substituents as well as the reaction conditions.

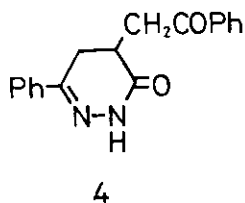
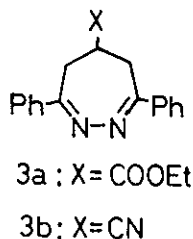
It has been found that the products in halogenations of 5,6-dihydro-3,7-diphenyl-4H-1,2-diazepines were greatly affected by the nature of substituents at the 5-position as well as by the reaction conditions. Namely, in contrast with easy contraction to pyridazines via 3,4-diazanorcaradienes in halogenations of the dihydrodiazepines (1: R=H, Me, Ph) having no electron-withdrawing groups<sup>1-3</sup>, the dihydrodiazepines (2a)<sup>4</sup> and (2b)<sup>5</sup> bearing two electron-withdrawing groups at the 5-position afforded the corresponding 3,4-diazanorcaradienes or halogenated dihydrodiazepines in the halogenations under controlled conditions. In the halogenations of the dihydrodiazepine (2c) having two strong electron-withdrawing cyano groups, a diazanorcaradine was not obtained, but instead ring-contracted pyridazines or halogenated dihydrodiazepine were formed<sup>5</sup>.



a : X=Y= COOEt  
b : X=CN, Y= COOEt  
c : X=Y=CN

However, the halogenations of dihydrodiazepines having one electron-withdrawing group at the 5-position have not been reported up to date. Thus, our attention was directed to the halogenation of such a dihydrodiazepine system in order to obtain further information concerning the effect of nature of the 5-substituent on the mode of halogenation. In this paper we wish to report the preparation of 5-ethoxycarbonyl- (3a) and 5-cyano-5,6-dihydro-3,7-diphenyl-4H-1,2-diazepine (3b), and their halogenations.

**Preparation.** After hydrolysis of 2a and 2b in a 1.5% ethanolic potassium hydroxide solution at room temperature, the resultant mixture was refluxed to give 3a and 3b in 75 and 94% yields, respectively. The structural elucidation of 3a,b was accomplished on the basis of spectral data<sup>6</sup>. Treatment of 3a with hydrochloric acid in

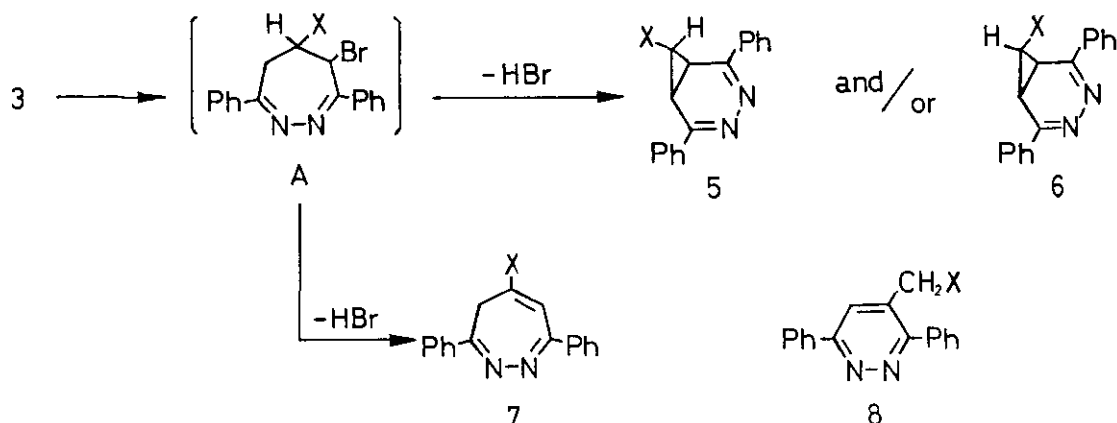


ethanol afforded 4,5-dihydro-4-phenacyl-6-phenyl-3(2H)-pyridazinone (4)<sup>7</sup> in 94% yield. The formation of 4 is closely similar to that of 4-cyanopyridazinone from 2b in the same treatment<sup>5</sup>.

**Bromination.** When 3a was treated with an equivalent of bromine in methanol at room temperature, 7-exo-ethoxycarbonyl-2,5-diphenyl-3,4-diazanorcaradiene (5a); yellow needles, mp 233-234 °C (lit.<sup>4</sup> mp 233-234 °C), was obtained in 96% yield. On the other hand, the bromination of 3b under similar conditions gave the 7-exo-cyano- (5b) and 7-endo-cyano-2,5-diphenyl-3,4-diazanorcaradiene (6b) in 49 and 34% yields, respectively. The structures of 5b and 6b were confirmed on the basis of spectral data<sup>8</sup>. In addition, 6b was identical with an authentic sample prepared from the hydrolysis of the 7-endo-cyano-7-exo-ethoxycarbonyldiazanorcaradiene<sup>5</sup>.

In the reaction with an equivalent of N-bromosuccinimide in carbon tetrachloride under reflux, however, 3a and 3b afforded the 4H-diazepines (7a; 24%) and (7b; 3%), and pyridazines (8a; 32%) and (8b; 73%), respectively<sup>9</sup>. Although it has been reported that the photolysis of 7,7-dimethyl-2,5-diphenyl-3,4-diazanorcaradiene gave the corresponding 4H-diazepine<sup>10</sup>, the formation of 7a,b is the first example for that from halogenation-dehydrohalogenation process of dihydroazepines.

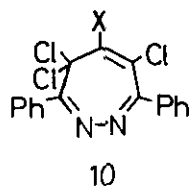
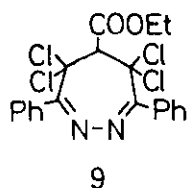
The pathways for the formation of 5 - 8 can be readily understood. Although the



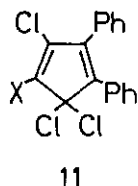
a: X=COOEt, b: X=CN

monobromide A could not be isolated, it is clear that the diazanorcaradienes, 5 and 6, are formed via A by internal nucleophilic displacement. In the reaction with N-bromosuccinimide in carbon tetrachloride under reflux, A undergoes the competitive internal nucleophilic displacement and 1,2-dehydrobromination to yield 5 and 6, and 7, respectively. However, the formed 5 and 6 are transformed, by the ring opening catalyzed by the generated hydrogen bromide under the conditions, into the stable pyridazines 8.

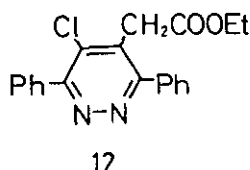
**Chlorination.** When an equivalent of sulfuryl chloride was slowly added to a solution of 3a or 3b in dichloromethane at room temperature, 5a (54%) or 5b (47%) and 6b (3%) were obtained together with tarry materials, respectively. The tetrachlorodihydrodiazepine (9) or trichloro-4H-diazepine (10b) was formed in 87 or 77% yield, respectively, however, when 3a or 3b was allowed to react rapidly with six equivalents of sulfuryl chloride in dichloromethane at room temperature. Treatment of 9 with triethylamine in benzene at room temperature afforded the trichloro-4H-diazepine (10a) in quantitative yield. Thermolysis of 10a and 10b in benzene under reflux for 5 h afforded the ring-contracted cyclopentadienes 11a and 11b in 43 and 29% yields, respectively, with the elimination of nitrogen. The dechlorination of 9 with sodium iodide in boiling acetone for 2 h gave 4-chloro-5-ethoxycarbonyl-3,6-diphenylpyridazine (12) in 82% yield, whereas 8a was obtained in 21% yield together with tarry materials in dechlorination with zinc dust in ethanol under reflux for



a : X = COOEt, b : X = CN



a : X = COOEt, b : X = CN



2 h. Structural elucidation of 9 — 12 was accomplished on the basis of spectral data<sup>11</sup>.

#### REFERENCES AND NOTES

1. R. G. Amiet, R. B. John, and K. R. Markham, *J. C. S. Chem. Commun.*, 1965, 128.
2. R. G. Amiet and R. B. John, *Aust. J. Chem.*, 1968, 21, 1279.
3. O. Tsuge and K. Kamata, *Heterocycles*, 1975, 3, 15.
4. O. Tsuge, K. Kamata, and S. Yogi, *Bull. Chem. Soc. Jpn.*, 1977, 50, 2153.
5. K. Kamata and O. Tsuge, *J. Heterocyclic Chem.*, 1986, 23, 557.
6. All the new compounds in this paper gave satisfactory elemental analyses.

3a: Mp 101-102 °C; colorless needles; IR (KBr) 1725 cm<sup>-1</sup>; <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ=1.18 (3H, t), 2.98 (4H, m), 3.58 (1H, m, CH), 4.15 (2H, q), 7.3-7.6 (6H, m), 7.8-8.1 (4H, m); <sup>13</sup>C NMR (CDCl<sub>3</sub>) δ=14.1 (q), 28.2 (t, 4- and 6-C), 49.8 (d, 5-C), 61.5 (t), 127.1, 128.8, 130.4 (each d), 136.5 (s), 157.5 (s, 3- and 7-C), 172.8 (s, C=O); MS m/z 320 (M<sup>+</sup>).

3b: Mp 187-188 °C; colorless needles; IR (KBr) 2240 cm<sup>-1</sup>; <sup>1</sup>H NMR (DMSO-d<sub>6</sub>) δ=2.94 (4H, m), 4.12 (1H, m, CH), 7.4-7.65 (6H, m), 7.85-8.15 (4H, m); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>) δ=28.3 (t, 4- and 6-C), 34.4 (d, 5-C), 121.2 (s, C=N), 126.8, 128.6, 130.3 (each d), 135.6 (s), 155.4 (s, 3- and 7-C); MS m/z 273 (M<sup>+</sup>).

7. 4: Mp 190-191 °C; colorless needles; IR (KBr) 3230 (NH), 1690, 1670 cm<sup>-1</sup>; <sup>1</sup>H NMR (DMSO-d<sub>6</sub>) δ=2.5-3.9 (5H, m), 7.3-8.2 (10H, m), 11.13 (1H, s, NH); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>) δ=125.7, 128.1, 128.7, 128.8, 129.4, 133.4 (each d), 136.1, 136.8 (each s), 150.0 (s, 6-C), 169.1 (s, 3-C), 197.9 (s, PhCO); MS m/z 292 (M<sup>+</sup>).

8. 5b: Mp 183-185 °C (dec); yellow needles; IR (KBr) 2240 cm<sup>-1</sup>; <sup>1</sup>H NMR (CF<sub>3</sub>COOD)

$\delta=2.46$  (1H, t,  $\text{CH}$ ,  $J=4.9$  Hz), 4.15 (2H, d,  $\text{CH}$ ,  $J=4.9$  Hz), 7.6-8.1 (6H, m), 8.1-8.4 (4H, m);  $^{13}\text{C}$  NMR ( $\text{CF}_3\text{COOD}$ )  $\delta=13.0$  (d, 7-C), 28.8 (d, 1- and 6-C), 116.2 (s,  $\text{C}\equiv\text{N}$ ), 130.7 (s), 131.4, 132.1, 139.2 (each d), 165.9 (s, 2- and 5-C); MS  $m/z$  271 ( $\text{M}^+$ ).

**6b**: Mp 260-262 °C (dec); yellow needles; IR (KBr) 2240  $\text{cm}^{-1}$ ;  $^1\text{H}$  NMR ( $\text{CF}_3\text{COOD}$ )  $\delta=4.18$  (3H, s,  $\text{CH}$ ), 7.6-8.1 (6H, m), 8.1-8.4 (4H, m);  $^{13}\text{C}$  NMR ( $\text{CF}_3\text{COOD}$ )  $\delta=8.3$  (d, 7-C), 29.4 (d, 1- and 6-C), 112.9 (s,  $\text{C}\equiv\text{N}$ ), 131.0 (s), 131.4, 132.2, 139.2 (each d), 165.3 (s, 2- and 5-C); MS  $m/z$  271 ( $\text{M}^+$ ).

9. **7a**: Mp 130-131 °C; yellow prisms; IR (KBr) 1710  $\text{cm}^{-1}$ ;  $^1\text{H}$  NMR ( $\text{CDCl}_3$ )  $\delta=1.36$  (3H, t), 2.19 (1H, dd, 4-H,  $J=12.9$ , 1.8 Hz), 4.36 (2H, q), 4.57 (1H, dd, 4-H,  $J=12.9$ , 2.2 Hz), 7.3-7.7 (7H, m), 7.8-8.4 (4H, m);  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ )  $\delta=14.2$  (q), 29.6 (t, 4-C), 61.9 (t), 126.6, 127.5, 128.2, 128.6, 130.2, 130.3 (each d), 134.6, 134.9 (each s), 137.1 (s, 5-C), 149.9, 155.5 (each s, 3- and 7-C), 164.1 (s,  $\text{C}=\text{O}$ ); MS  $m/z$  318 ( $\text{M}^+$ ).

**7b**: Mp 182-183 °C (dec); pale yellow needles; IR (KBr) 2210  $\text{cm}^{-1}$ ;  $^1\text{H}$  NMR ( $\text{CDCl}_3$ )  $\delta=2.53$  (1H, dd, 4-H,  $J=13.5$ , 1.5 Hz), 4.01 (1H, dd, 4-H,  $J=13.5$ , 1.8 Hz), 6.98 (1H, distorted t,  $=\text{CH}$ ), 7.3-7.6 (6H, m), 7.7-8.1 (4H, m);  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ )  $\delta=33.0$  (t, 4-C), 115.5, 116.5 (each s), 127.4, 128.0, 128.9, 129.0, 130.7, 130.8, 132.2 (each d), 133.8, 136.1 (each s), 148.4, 154.1 (each s, 3- and 7-C); MS  $m/z$  271 ( $\text{M}^+$ ).

**8a**: Mp 139-140 °C (lit.<sup>4</sup> mp 139-140 °C). **8b**: Mp 156-157 °C; colorless prisms; IR (KBr) 2240  $\text{cm}^{-1}$ ;  $^1\text{H}$  NMR ( $\text{CDCl}_3$ )  $\delta=3.82$  (2H, s,  $\text{CH}_2$ ), 7.4-7.6 (8H, m), 8.0 (1H, s, 5-H), 8.0-8.2 (2H, m);  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ )  $\delta=21.7$  (t), 116.0 (s,  $\text{C}\equiv\text{N}$ ), 123.7 (d, 5-C), 127.1, 129.0, 129.1, 129.5, 129.7, 130.5, 135.4 (s), 158.3, 159.1 (each s, 3- and 6-C); MS  $m/z$  271 ( $\text{M}^+$ ).

10. H. E. Zimmermann and W. Eberbach, *J. Am. Chem. Soc.*, 1973, **95**, 3970.

11. **9**: Mp 100-101 °C; colorless prisms; IR (KBr) 1750  $\text{cm}^{-1}$ ;  $^1\text{H}$  NMR ( $\text{CDCl}_3$ )  $\delta=1.30$  (t), 4.33 (2H, q), 5.10 (1H, s,  $\text{CH}$ ), 7.4-7.6 (6H, m), 7.6-7.9 (4H, m);  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ )  $\delta=14.0$  (q), 62.6 (t), 78.6 (s, 4- and 6-C), 85.6 (d, 5-C), 127.8, 129.2, 129.9 (each d), 135.2 (s), 151.8 (s, 3- and 7-C), 163.3 (s,  $\text{C}=\text{O}$ ); MS  $m/z$  456, 458, 460, 462, 464 ( $\text{M}^+$ ).

**10a**: Mp 114-115 °C (dec); pale yellow prisms; IR (KBr) 1750  $\text{cm}^{-1}$ ;  $^1\text{H}$  NMR ( $\text{CDCl}_3$ )  $\delta=1.38$  (3H, t), 4.45 (2H, q), 7.3-8.0 (10H, m);  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ )  $\delta=14.0$  (q), 61.8 (s, 4-C), 63.2 (t), 127.8, 128.5, 128.6, 129.0, 129.8, 131.0 (each d), 133.8, 134.9 (each s), 136.8, 142.5 (each s, 5- and 6-C), 151.2, 153.5 (each s, 3- and

7-C), 161.3 (s, C=O); MS m/z 420, 422, 424, 426 ( $M^+$ ).

**10b** : Mp 123-124 °C (dec); yellow prisms; IR (KBr) 2220  $\text{cm}^{-1}$ ;  $^1\text{H}$  NMR ( $\text{CDCl}_3$ )  $\delta$ =7.4-7.7 (8H, m), 7.7-7.9 (2H, m);  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ )  $\delta$ =73.6 (s, 4-C), 111.5 (s, C $\equiv$ N), 123.3 (s, 5-C), 128.2, 128.8, 129.7, 130.4, 131.5 (each d), 132.7, 134.3, 136.3 (each s), 150.0, 152.3 (each s, 3- and 7-C); MS m/z 373, 375, 377, 379 ( $M^+$ ).

**11a** : Mp 100-101 °C; colorless prisms; IR (KBr) 1740  $\text{cm}^{-1}$ ;  $^1\text{H}$  NMR ( $\text{CDCl}_3$ )  $\delta$ =1.20 (3H, t), 4.18 (2H, q), 7.2-7.6 (10H, m);  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ )  $\delta$ =14.3 (q), 61.8 (t), 104.2 (s, 5-C), 111.8, 123.1 (each s), 127.6, 128.0, 128.2, 128.7, 129.0, 129.5, 130.0, 131.2 (s), 133.6 (s), 136.7 (s), 170.4 (s, C=O); MS m/z 392, 394, 396, 398 ( $M^+$ ).

**11b** : Mp 136-137 °C; yellow prisms; IR (KBr) 2210  $\text{cm}^{-1}$ ;  $^1\text{H}$  NMR ( $\text{CDCl}_3$ )  $\delta$ =7.2-7.6 (m);  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ )  $\delta$ =111.2 (s, C $\equiv$ N), 119.5 (s), 126.2 (d), 128.3, 128.5, 128.7, 129.5, 129.6, 129.9, 131.7 (s), 133.4 (s), 147.5 (s), 149.1 (s); MS m/z 345, 347, 349, 351 ( $M^+$ ).

**12** : Mp 99-100 °C; colorless needles; IR (KBr) 1735  $\text{cm}^{-1}$ ;  $^1\text{H}$  NMR ( $\text{CDCl}_3$ )  $\delta$ =1.23 (3H, t), 3.86 (2H, s), 4.20 (2H, q), 7.4-7.7 (8H, m), 7.7-8.0 (2H, m);  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ )  $\delta$ =14.1 (q), 36.8 (t, CH<sub>2</sub>COOEt), 61.8 (t), 128.4, 128.9, 129.3, 129.7, 129.8, 130.0 (each d), 132.1, 135.5, 136.4, 138.8 (each s), 158.7, 161.7 (each s, 3- and 6-C), 169.0 (s, C=O); MS m/z 352, 354 ( $M^+$ ).

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