ENNEAPHYLLINE, SARCOPHILLINE AND NORSARCOCAPNIDINE, NEW PHENOLIC CULARINES FROM SARCOCAPNOS PLANTS

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<u>Abstract</u> - Three new phenolic cularines have been isolated from <u>Sarcocapnos</u> species. Their structures have been elucidated by spectroscopic studies, chemical correlations and total synthesis.

Our current research on the alkaloids of the Fumariaceae <u>Sarcocapnos crassifolia</u>¹ (Desf.)DC and <u>Sarcocapnos enneaphylla</u>² (L.)DC has led us to the isolation of three new phenolic cularine alkaloids, enneaphylline <u>1</u>, sarcophylline <u>2</u> and norsarcocapnidine <u>3</u> this last being the first example of an N-norisocularine alkaloid.

Enneaphylline <u>1</u> (from <u>S.crassifolia and S.enneaphylla</u>) was obtained as colorless prims of mp: 205-207°C (EtOH), $|\alpha|_D^{25}$:+256° (c:0.8, EtOH). Its UV spectrum exhibited bands at λ_{max} (EtOH)(log ε): 226(4.04) and 284(3.85) nm. Its phenolic nature was deduced from a strong bathochromic shift to λ_{max} (EtOH, NaOH)(log ε): 223(4.33), 284(3.66) and 303(3.71) nm observed on addition of base, and from the IR spectrum (KBr), which displayed one broad band at 3440 (OH) cm⁻¹. Its molecular formula, C₁₉H₂₁NO₄, was established by high resolution MS, which showed the molecular ion at m/z (%)=327.1465 (100) (calculated 327.1470) together with fragments at m/z (%)=312 (64) and 174 (39).

The pmr spectrum, with NOEDS³, suggested the cularine type structure <u>1</u>. This was finally confirmed by direct comparison (tlc, MS, UV, pmr) with synthetic enneaphy lline, which was obtained from the mixture of sarcocapnidine <u>5</u> and enneaphylline <u>1</u> produced by phenolic oxidative coupling⁴ of crassifoline <u>4</u>⁵ (also present in the same plant). This constitutes the first evidence of the co-occurrence in the





NMe

3



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same natural source, of the isomeric cularines <u>1</u> and <u>5</u> and their putative biogenetic precursor <u>4</u>.

Sarcophylline <u>2</u> was obtained from <u>S.crassifolia</u> and <u>S.enneaphylla</u> as an amorphous substance, $|\alpha|_D^{25}$:+200° (c: 1.65, CHCl₃). The bathochromic shift of its UV spectrum in basic media λ_{max} (EtOH)(log ϵ): 219(4.01), 282(3.36); λ_{max} (EtOH/OH⁻)(log ϵ): 222(4.10), 285(3.36) and 293(3.40) nm, together with a broad signal in its IR spectrum (CHCl₃) at 3400 cm⁻¹, revealed its phenolic nature. The molecular formula C₁₉H₂₁NO₄ was established by high resolution MS, which showed the molecular ion at m/z(%): 327.1470 (100) (calculated: 327.1470) and the most important fragments at 312(56), 294(53) and 162(66), and thus permitted the phenolic group to be assigned to the C₇ position. The isocularine skeleton was deduced from its pmr spectrum³, which exhibited two methoxyl singlets and two aromatic AB quartets, and was confirmed by 0-methylation with diazomethane, which gave a product identical to authentic sarcocapnine <u>6</u>⁶. All the pmr assignments were confirmed by NOEDS experiments³.

Norsarcocapnidine <u>3</u> was obtained from <u>Sarcocapnos crassifolia</u> as an amorphous solid, $|_{\alpha}|_{D}^{25}$:+347.5°(c: 0.8, EtOH). Its phenolic nature was deduced from its UV spectrum $\lambda_{max}(\text{EtOH})(\log \epsilon)$: 225(4.16), 278(3.52) nm; $\lambda_{max}(\text{EtOH}/\text{OH}^{-})(\log \epsilon)$: 227(4.30), 284(3.73) and 293 sh (3.69) nm and IR spectrum (CHCl₃) (3450 cm⁻¹). The molecular formula $C_{18}H_{19}NO_4$ was established by high resolution MS, which showed the molecular ion at m/z(%): 313.1314 (100) (calculated: 313.1318) and fragments at m/z(%): 312(35.5), 298(21) and 160(16). Its pmr data suggested the norisocularine type structure <u>3</u>. The assignments of the chemical shifts for the -OMe groups and aromatic protons were established by NOEDS and COSY experiments. The substitution pattern was confirmed by partial synthesis from the parent sarco-capnidine <u>5</u> by means of Fremy's salt oxidation⁷ of its 0-methoxymethyl protected derivative followed by Zn-HCl reduction.

The structure of norsarcocapnidine <u>3</u> was finally confirmed by total synthesis based on our recently developed approach to cularines⁸, which in this case starts from the benzylisoquinoline intermediate <u>7</u> and constructs the cularine diaryl ether linkage by Ullmann condensation. As the Ullmann reaction gives a very low yield with N-nortetrahydrobenzylisoquinolines⁹, we decided to protect the nitrogen with the easily removable benzyl group. N-benzylation (BrCH₂Ph, KI, acetone) of <u>7</u> followed by NaBH₄ reduction of the crude product afforded the N-benzyltetrahydro-

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benzylisoquinoline <u>8</u> (72% yield), whose reaction with cupric oxide and potassium carbonate in dry pyridine gave the N-benzylated cularine <u>9</u> (94% yield). N-debenzylation of <u>9</u> (H₂, Pd/C-AcOH)¹⁰ led to the previously described N-norsarcocapnine <u>10⁶</u> (76% yield). Selective O-demethylation¹¹ of <u>10</u> (HBr, AcOH) then afforded N-norsarcocapnidine <u>3</u> (70% yield), which was identical (tlc, UV, MS, pmr) with the natural product.

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