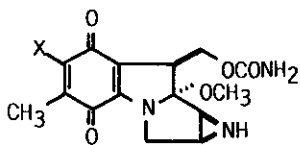


EFFICIENT SYNTHESIS OF INDOLOQUINONE DERIVATIVE BY SEVERAL OXIDATIVE DERIVATIONS OF 6-METHYLINDOLE

Shin-ichi Nakatsuka,* Kazuo Ueda, Osamu Asano, and Toshio Goto
 Laboratory of Organic Chemistry, Faculty of Agriculture,
 Nagoya University, Chikusa, Nagoya 464, Japan

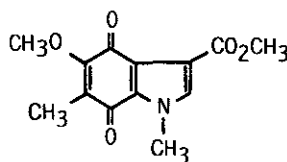
Abstract— Indoloquinone **3** was efficiently synthesized from simple 6-methylindole by several oxidation steps.

Although many synthetic studies³⁻⁵ on mitomycin^{1,2} derivatives have been reported over twenty years, no method for the synthesis of indoloquinone system by direct oxidative functionalization of benzene part of simple indole was appeared except from 5-hydroxy⁴⁻¹ or 5-methoxyindole derivative.^{6, 4-2} Recently, we have found several methods to introduce alkyl,^{7,8} acyl,⁹ and heteroatoms¹⁰ directly onto the benzene part of stabilized indole derivatives. Now, we report a new method for the synthesis of indoloquinone **3** which contains the same quinone system with mitomycin A (**1**).



mitomycin A : X=OCH₃ (**1**)

mitomycin C : X=NH₂ (**2**)



3

6-Methylindole **4** was derived to methyl 1,6-dimethylindole-3-carboxylate **5** by (1) NaH/CH₃I, (2) (COCl)₂, (3) 120 °C, (4) MeOH in 60% overall yield [MS m/z 203(M⁺); ¹H-NMR δ(CDCl₃) ppm 2.50(3H, br.s), 3.78(3H, s), 3.89(3H, s), 7.09(1H, br.d, J=8.5 Hz), 7.12(1H, br.s), 7.70(1H, s), 8.02(1H, d, J=8.5 Hz)].

Friedel-Crafts acylation¹¹ of **5** with chloroacetylchloride-AlCl₃ in CH₂Cl₂ at 25 °C

for 1.3 h afforded single chloroacetyl derivative [MS m/z 280(M⁺); ¹H-NMR δ(CDCl₃) ppm 2.66(3H, br.s), 3.81(3H, s), 3.90(3H, s), 4.81(2H, s), 7.18(1H, br.s), 7.75(1H, s), 8.49(1H, s)] in quantitative yield. The acylated position was determined by its ¹H-NMR spectra.

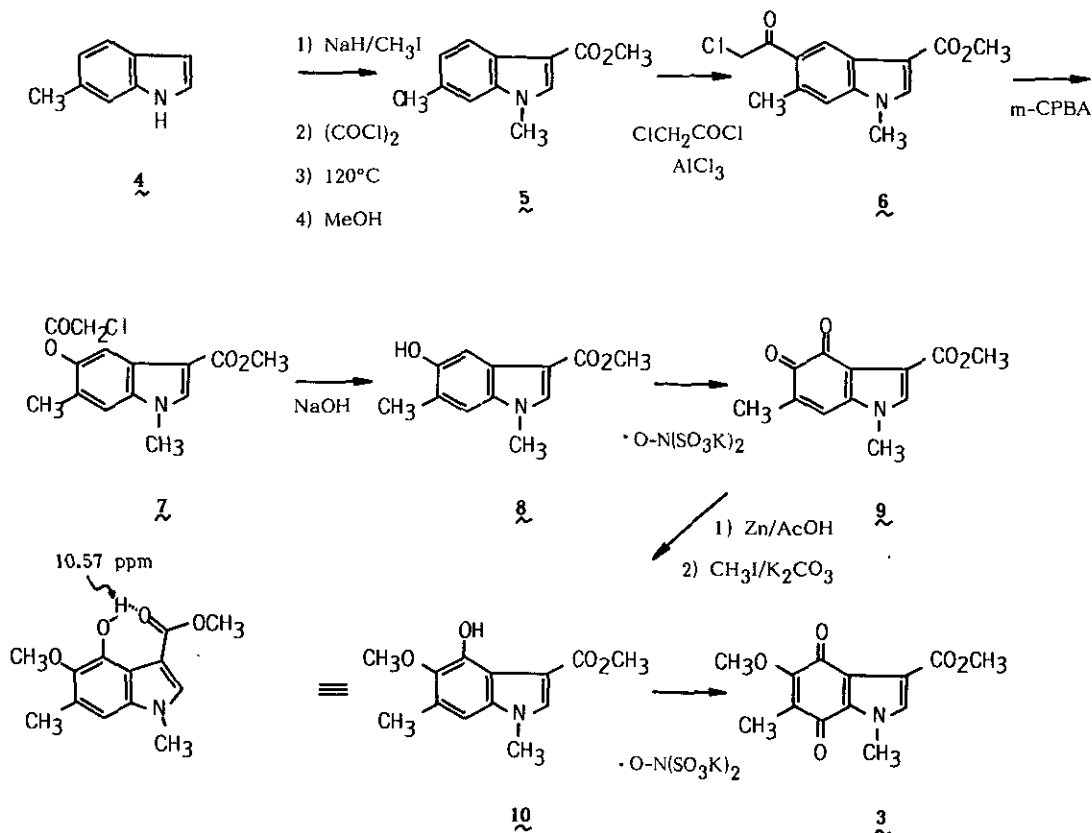
Baeyer-Villiger oxidation^{11,12} of 6 with m-chloroperbenzoic acid in the presence of powdered Na₂HPO₄ in chloroform at 25°C for 6 h afforded desired 6-chloroacetoxy derivative 7 in 59% yield [MS m/z 296(M⁺); ¹H-NMR δ(CDCl₃) ppm 2.32(3H, br.s), 3.80(3H, s), 3.88(3H, s), 4.36(2H, s), 7.18(1H, br.s), 7.73(1H, s), 7.78(1H, s)]. Hydrolysis of chloroacetyl group of 7 was achieved by treatment with 1N NaOH in MeOH at 25 °C to afford 6-hydroxy derivative 8 in quantitative yield [MS m/z 219 (M⁺); ¹H-NMR δ(CDCl₃) ppm 2.39(3H, br.s), 3.75(3H, s), 3.88(3H, s), 7.06(1H, br.s), 7.61(1H, s), 7.63(1H, s)].

Compound 8 was oxidized with Fremy's salt [·ON(SO₃K)₂] in acetone-H₂O(1:2) at 25 °C for 10 min to give orthoquinone 9 in quantitative yield [MS m/z 233(M⁺); ¹H-NMR (CDCl₃) ppm 1.97(3H, br.s), 3.66(3H, s), 3.84(3H, s), 6.94(1H, br.s), 7.20(1H, s)].

Unfortunately, treatment of 9 with Ac₂O-BF₃OEt₂⁴⁻¹ afforded no desired triacetoxo derivative. So, we planned to reoxidize reduced hydroquinone derivative such as 10. After reduction of 9 with Zn/AcOH to hydroquinone, selective methylation was achieved with CH₃I/K₂CO₃ in DMF at 25°C for 45 min to give 5-methoxy derivative 10 in 54% yield [10: MS m/z 249(M⁺); ¹H-NMR δ(CDCl₃) ppm 2.39(3H, br.s), 3.71(3H, s), 3.87(3H, s), 3.89(3H, s), 6.59(1H, br.s), 7.53(1H, s), 10.57(1H, s)]. We assume that the hydroxy group at 4-position of indole nucleus is less reactive because its stabilization by hydrogen bond with ester carbonyl group at 3-position.

In fact, a chemical shift of 4-OH in ¹H-NMR spectrum of 10 was appeared at low field (10.57 ppm). Reoxidation of 10 with Fremy's salt in acetone-phosphate buffer(pH 7) at 40°C for 2.5 h afforded desired indoloquinone 3, which contains the same quinone part with mitomycin A (1) [3: MS m/z 263(M⁺); ¹H-NMRδ(CDCl₃) ppm 1.96(3H, s), 3.87(3H, s), 3.98(3H, s), 4.07(3H, s), 7.36(1H, s)] in 65% yield.

Thus three oxygen functions at 4, 5, and 7-positions of indole nucleus could be introduced directly onto the benzene part of indole derivative and compound 3 was synthesized from simple 6-methylindole 4. These new method seems to be very usefull for the synthesis of mitomycins and related compounds. Further synthetic studies are now in progress.



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